Supporting Information

Synthesis of (-)-Nakamurol A and Assignment of Absolute Configuration of Diterpenoid (+)-Nakamurol A

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Table 1. ¹³C NMR Chemical shifts of compounds 1-21^{a,b}

	1	3	4	5	7 ^c	8^d	9	10	12
C 1	31.9	33.3	41.8	31.8	32.1	30.4	28.9	33.3	31.8
C2	21.4	26.5	21.9	21.4	21.2	22.9	21.7	21.6	21.4
C3	31.1	30.5	30.4	30.3	30.4	30.3	30.4	30.7	31.2
C4	30.4	43.1	31.0	30.7	31.0	31.5	32.6	30.9	30.4
C5	39.5	39.0	37.9	39.1	39.5	37.6	38.3	39.4	39.5
C6	33.9	35.5	32.2	32.2	33.0	27.4	32.2	32.2	33.9
C7	33.1	34.0	37.6	38.0	38.3	31.7	35.3	38.6	33.1
C8	149.6	199.5	213.4	216.2	213.1	146.7/147.0	203.2	213.6	149.3
С9	43.2	124.0	49.0	53.5	48.4	114.5/115.4	152.4	50.6	41.9
C10	42.4	171.2	35.9	44.6	46.5	40.8/41.0	44.8	46.4	42.1
C11	18.2			58.3	22.7	23.8/24.0	120.5	21.7	23.6
C12	41.7				50.2	46.1/46.4		33.4	33.3
C13	73.6					165.2		139.0	139.5
C14	145.3								
C15	111.5								
C16	27.6							114.8	114.1
C17	106.5								106.2
C18	16.3	15.2	16.1	16.3	16.3	16.3	15.7	16.5	16.4
C19	16.4	16.0	15.3	15.2	16.0	15.5	15.7	16.0	16.5
C20	18.3		24.6	19.7	18.6	24.1/24.9	27.2	18.9	18.4

^a Diterpene numbering is used in this table. ^b Values for compounds **1**, **3-5**, **10**, **12**, **13**, **15**, and **21** were assigned on the basis of HSQC spectra. ^c (COOMe): 170.3, 170.0, 52.4, 52.5. ^d (COOMe): -two epimers- 168.6, 169.1, 52.8, 52.9.

Table 1 (continued). ¹³C NMR Chemical shifts of compounds 1-21^{a,b}

	13	14 ^e	15 ^{f,g}	16 ^h	17 ⁱ	18 ^{<i>j</i>}	19	20	21
C 1	31.9	31.9	32.4	32.0	32.1	31.2	31.2	30.8	31.9
C2	21.3	21.4	21.3	21.2	20.4	21.4	21.4	22.6	21.4
С3	31.1	31.2	30.3	30.9	30.7	32.0	31.9	31.1	31.1
C4	30.5	30.4	30.7	30.4	30.5	30.4	30.4	30.5	30.4
C5	39.5	39.5	39.5	39.6	39.7	39.5	39.5	39.5	39.5
C6	34.0	33.9	33.0	33.5	33.2	33.9	33.9	33.0	33.9
C7	33.1	33.1	38.2	32.5	32.2	33.1	33.1	31.9	33.0
C8	149.2	149.6	211.6	148.8	146.1	149.2	149.5	149.5	149.4
С9	42.2	43.1	52.8	43.5	37.8	42.4	42.2	42.0	42.7
C10	42.4	42.4	46.7	42.6	42.6	42.3	42.4	42.2	42.3
C11	18.1	18.2	28.4	29.7	53.4	22.4	22.6	23.6	19.2
C12	43.0	41.6				40.1	38.7	33.9	37.6
C13	209.0	73.7				161.2	140.7	140.7	62.7
C14		145.0				114.7	122.9	124.4	62.5
C15		111.6				167.2	59.5	59.2	61.5
C16	30.1	28.1				19.1	16.5	21.2	16.9
C17	106.3	106.3		107.9	108.1	106.3	106.2	106.3	106.6
C18	16.4	16.4	16.3	16.3	16.2	16.3	16.3	16.4	16.3
C19	16.4	16.5	15.8	16.4	16.6	16.4	16.4	16.4	16.4
C20	18.2	18.4	18.8	18.3	18.8	18.3	18.3	18.4	18.3

 $^{^{}e}$ Values taken from an NMR spectrum of a mixture of **1** and **14**. f The COSY and HSQC spectra of the corresponding sulfone were recorded. g (SPh): 136.5, 130.0, 129.3, 128.7, 126.1. h (SPh): 128.2, 127.7, 124.8. i (SO₂Ph): 140.3, 133.4, 129.0, 128.1. j (OCH₃): 50.8.

Table 2. ¹H NMR Chemical Shifts of Vinylic Protons at C(17) for Labdane-type Diterpenoids and Compounds 1 and 14. Comparison Between C(13) Epimers

Compounds with configuration 9S,13R

Compounds with configuration 9S,13S

	=CH2	$\Delta\delta$	ref		=CH ₂	Δδ	ref
compound 14	4.45 / 4.80	0.35	*	nakamurol-A (1)	4.50 /4.82	0.32	1, *
manool	4.46 / 4.80	0.34	22	13-epimanool	4.50 / 4.80	0.30	22
3α-hydroxymanno	ol 4.49 / 4.82	0.33	23	3α-hydroxy-13-epimanool	4.53 / 4.83	0.30	20
3α-OTBS manool	4.46 / 4.79	0.33	20	3α-OTBS 13-epimanool	4.50 /4.80	0.30	20

^{*} This work

Experimental Section³⁵

General: All reactions were carried out under an argon atmosphere with dry, freshly distilled solvents under anhydrous conditions. Analytical thin-layer chromatography was performed on SiO₂ (silica gel 60 F₂₅₄, Merck), and the spots were located with iodoplatinate reagent or 1% aqueous KMnO₄. Chromatography refers to flash chromatography and was carried out on SiO₂ (silica gel 60, SDS, 230-240 mesh ASTM). Drying of organic extracts during workup of reactions was performed over anhydrous Na₂SO₄. Evaporation of solvent was accomplished with a rotatory evaporator. Chemical shifts of ¹H and ¹³C NMR spectra are reported in ppm downfield (δ) from Me₄Si. The ¹³C NMR spectra, when unambiguous assignation was not available, are reported as follows: chemical shift (multiplicity determined from DEPT spectra). Only noteworthy IR absorptions (cm⁻¹) are listed. Microanalyses and HRMS were performed by the Centro de Investigación y Desarrollo (CSIC), Barcelona.

(4aS,5R,8aS)-4a,5,8a-Trimethyl-3,4,4a,5,6,7,8,8a-octahydro-1*H*-naphthalen-2-one (4): 1 H NMR (500 MHz, COSY) 0.84 (s, 3H, Me-19), 0.88 (s, 3H, Me-20), 0.89 (d, J = 6.5 Hz, 3H, Me-18), 1.07 (dm, J = 10.5 Hz, H-1_{eq}), 1.3-1.4 (m, 1H, H-3), 1.50-1.60 (m, 4H, H-1_{ax}, H-2, H-3), 1.68 (td, J = 14.5, 4.5 Hz, 1H, H-6_{ax}), 1.69 (dd, J = 14.5, 2.1 Hz, 1H, H-9_{eq}), 1.89 (ddd, J = 14.5, 6.5, 2.1 Hz, 1H, H-6_{eq}), 2.14 (dm, J = 12 Hz, 1H, H-4_{ax}), 2.18 (dm, J = 14 Hz,1H, H-7_{eq}), 2.39 (td, J = 14.5, 5 Hz, 1H, H-7_{ax}), 3.04 (d, J = 14.5 Hz, 1H, H-9_{ax}); 13 C NMR (75 MHz, HSQC), see Table 1.

(1*S*,4a*S*,5*R*,8a*S*)-4a,5,8a-Trimethyl-1-(2,2-dimethoxycarbonylethyl)-3,4,4a,5,6,7,8,8a-octahydro-1*H*-naphthalen-2-one (7). To a solution of dimethyl malonate (0.09 mL, 0.78 mmol) in benzene (2 mL) and MeOH (0.5 mL) was added sodium methoxide (86 mg, 1.59 mmol), and the mixture was heated at reflux. Then, a solution of mesylate 6 (94 mg, 0.31 mmol) in benzene (3 mL) was slowly added and the stirring was maintained for 1 h. The reaction mixture was cooled to room temperature and, after addition of 2 N HCl (25 mL), extracted with Et₂O. The dried organic extracts were concentrated and purified by chromatography (5% EtOAc in hexane) to give, in order of elution, enol lactone 8 (32 mg, 33%), as a mixture of epimers, and malonate 7 (15 mg, 14%).

Compound 7: IR (NaCl) 1708, 1736 cm⁻¹; ¹H NMR (300 MHz) 0.68 (s, 3H), 0.82 (s, 3H), 0.88 (d, J = 6.6 Hz, 3H), 1.38–1.55 (m, 3H), 1.65 (td, J = 14.4, 4.5 Hz, 1H), 1.78–1.92 (m, 2H), 2.13–2.23 (m, 2H), 2.26–2.35 (m, 1H), 2.43 (td, J = 14, 6 Hz, 1H), 3.06 (d, J = 11 Hz, 1H), 3.50 (dd, J = 11.4, 3.9 Hz, 1H), 3.71 and 3,72 (2 s, 3H each); ¹³C NMR (75 MHz, DEPT), see Table 1.

Methyl (2*RS*,6a*S*,7*R*,10a*R*)-6a,7,10a-Trimethyl-3-oxo-2,3,5,6,6a,7,8,9,10,10a-decahydro-1*H*-benzo[*f*]chromene-2-carboxylate (8): IR (NaCl) 1743, 1771 cm⁻¹; ¹H NMR (300 MHz): 0.81 and 0.82 (2 s, 3H), 0.82 and 0.83 (2 d, J = 6.5 Hz, 3H), 0.88 and 0.94 (2 s, 3H), 2.39-2.75 (m, 2H), 3.46 (dd, J = 12.9, 6.9 Hz, 0.5H), 3.63 (dd, J = 6.3, 4.8 Hz, 0.5H), 3.75 and 3.82 (2 s, 3H); ¹³C NMR (75 MHz, DEPT), see Table 1.

(1*R*,4a*S*,5*R*,8a*S*)-1-(But-3-enyl)-4a,5,8a-trimethyl-3,4,4a,5,6,7,8,8a-octahydro-1*H*-naphthalen-2-one (11): ¹³C NMR (75 MHz, DEPT) 17.7 (q), 19.3 (q), 21.6 (t), 23.9 (q), 28.4 (t), 29.7 (d), 31.1 (t), 33.5 (t), 35.9 (t), 37.5 (s), 39.0 (t), 45.1 (s), 56.8 (d), 114.6 (t), 138.8 (d), 213.6 (s). Values taken from nNMR spectrum of a mixture of **10** and **11**.

(1S,4aS,5R,8aS)-1-Phenylsulfanylmethyl-4a,5,8a-trimethyl-3,4,4a,5,6,7,8,8a-octahydro-1*H*-naphthalen-2-one (15). To a solution of enone 9 (125 mg, 0.6 mmol) in THF (3 mL) were added

thiophenol (0.09 mL, 0.9 mmol) and triethylamine (4 μ l, 0.03 mmol). The reaction mixture was stirred at room temperature for 24 h, poured into saturated aqueous NaHCO₃ and extracted with Et₂O. The dried organic extracts were concentrated and purified by chromatography (2% EtOAc in hexane) to give **15** (128 mg, 67%): ¹H NMR (300 MHz) 0.69 (s, 3H), 0.81 (s, 3H), 0.87 (d, J = 6.6 Hz, 3H), 1.0-1.5 (m, 7H), 1.69 (dd, J = 14, 4.6 Hz, 1H), 1.90 (ddd, J = 14.4, 6, 2.6 Hz, 1H), 2.27 (m, 1H), 2.49 (td, J = 13.6, 6.2 Hz, 1H), 2.73 (d, J = 11 Hz, 1H), 3.29 (d, J = 9 Hz, 1H), 3.35 (dd, J = 11, 9 Hz, 1H), 7.28 (m, 5H); ¹³C NMR (75 MHz, DEPT), see Table 1.

(1*R*,4a*S*,5*R*,8a*S*)-2-Methylene-1-phenylsulfanylmethyl-4a,5,8a-trimethylperhydronaphthalene (16). *n*-BuLi (1.6 M in hexane, 1.98 mL, 3.16 mol) was added to a cooled (0 °C) dispersion of methyltriphenylphosphonium bromide (1.13g, 3.16 mol) in THF (10 mL). After stirring for 1 h, a solution of 15 (100 mg, 0.316 mmol) in THF (12 mL) was added dropwise, and the reaction mixture was heated at reflux for 24 h. The reaction mixture was cooled at room temperature and quenched with water, and the aqueous layer was extracted with EtOAc. The organic extracts were washed with brine, concentrated, and purified by chromatography (hexane) to give 16 (75 mg, 75%): 1 H NMR (200 MHz) 0.70 (s, 3H), 0.75 (s, 3H), 0.79 (d, J = 6.6 Hz, 3H), 1.05-1.50 (m, 8H), 1.61 (dq, J = 13.8, 4.6, 2.5 Hz, 1H), 2.11 (ddd, J = 13.2, 5.0, 2.8 Hz, 1H), 2.25 (m, 1H), 3.03 and 3.05 (2d, J = 12 Hz, 1H each), 3.04 (s, 1H), 4.63 (s, 1H), 4.98 (s, 1H), 7.25 (m, 5H); 13 C NMR (50 MHz, DEPT), see Table 1.

(1*R*,4a*S*,5*R*,8a*S*)-1-Benzenesulfonylmethyl-2-methylene-4a,5,8a-trimethylperhydronaphthalene (17). Method A. To a solution of sulfide 16 (250 mg, 0.8 mmol) in MeOH (5 mL) was added dropwise a solution of oxone® (1.47 g, 2.34 mmol) in water (5 mL). The reaction mixture was stirred at room temperature for 16 h, diluted with CH_2Cl_2 and extracted. The organic layer was dried and concentrated, and the resulting oil was purified by chromatography (10% EtOAc in hexane) to give 17 (275 mg, 98%): 1 H NMR (300 MHz, COSY) 0.56 (s, 3H, Me-20), 0.74 (s, 3H, Me-19), 0.80 (d, J = 6.6 Hz, 3H, Me-18), 1.10-1.55 (m, 7H), 1.6 (ddd, J = 13.8, 4.4, 2.8 Hz, 1H, H-6_{eq}), 2.03 (ddd, J = 13.6, 4.8, 2.6 Hz, H-7_{eq}); 2.20 (m, 1H, H-4), 2.22 (td, J = 14, 4.2 Hz, H-7_{ax}), 3.19 and 3.37 (2m, 1H each, H-11), 3.40 (d, J = 9.2 Hz, 1H, H-9), 4.47 and 4.78 (2s, 1H each, H-17), 7.58 (m, 3H), 7.89 (m, 2H); 13 C NMR (75 MHz, HSQC, DEPT), see Table 1.

Method B. To a solution of sulfide **15** (110 mg, 0.35 mmol) in MeOH (2 mL) was added dropwise a solution of oxone® (0.6g, 1.05 mmol) in water (2 mL). The reaction mixture was stirred at room

temperature for 4 h and extracted with CH₂Cl₂. The dried organic extracts were dried, concentrated, and purified by chromatography (10% EtOAc in hexane) to give (1S,4aS,5R,8aS) 1benzenesulfonylmethyl-4a,5,8a-trimethyl-3,4,4a,5,6,7,8,8a-octahydro-1*H*-naphthalen-2-one, not shown, (120 mg, 99%): ¹H NMR (600 MHz, COSY): 0.54 (s, 3H, Me-20), 0.82 (s, 3H, Me-19), 0.91 $(d, J = 6.6 \text{ Hz}, 3H, \text{Me-}18), 1.35 \text{ (m, }1H, \text{H-}1), 1-40 \text{ (m, }1H, \text{H-}3), 1.50 \text{ (m, }1H, \text{H-}1), 1.55 \text{ (m, }1H, \text{H-}1), 1.55 \text{ (m, }1H, \text{H-}1), 1.55 \text{ (m, }1H, \text{H-}1), 1.50 \text{$ 2), 1.60 (m, 1H, H-3), 1.65 (m, 1H, H-6ax), 1.76 (m, 1H, H-2), 1.91 (ddd, J = 14.5, 6.2, 2.5 Hz, 1H, H- 6_{eq}), 2.16 (ddd, J = 13.2, 4.5, 2.4 Hz, H- 7_{eq}), 2.41 (dqd, J = 12.3, 6.6, 4 Hz, 1H, H-4), 2.47 (td, J = 13.4, 6.4 Hz, H- 7_{ax}), 2.85 (d, J = 13.9 Hz, 1H, H-11), 3.75 (d, J = 8.5 Hz, 1H, H- 9_{ax}), 3.99 (dd, J = 14.2, 8.5Hz, 1H, H-11), 7.54 (td, J= 7.8, 1.2 Hz, 2H, m-ArH), 7.62 (tt, J = 7.8, 1.2 Hz, 1H, p-ArH), 7.90 (dd, J = 7.8, 1.2 Hz, 2H, o-ArH); ¹³C NMR (75 MHz, HSQC) 16.1 (C-19), 16.4 (C-18), 19.1 (C-20), 20.8 (C-2), 30.3 (C-3), 31.7 (C-4), 32.7 (C-1), 33.0 (C-6); 37.5 (C-7), 39.7 (C-5), 46.0 (C-9), 46.8 (C-10), 50.6 (C-10), 11), 128.0 (o-Ar), 129.1 (m-Ar), 133.6 (p-Ar), 139.9 (ipso-Ar), 208.4 (C-8). A solution of TiCl₄ (0.1 mL, 0.95 mmol) in CH₂Cl₂ (2 mL) was added slowly to a cooled (-40 °C) mixture of zinc dust (0.3 g, 4.5 mmol) and CH₂Cl₂ (0.2 mL) in THF (3 mL). The mixture was stirred for 30 min and a solution of the above ketone sulfone (55 mg, 0.16 mmol) in THF (2 mL) was added dropwise. The reaction was allowed to reach room temperature and stirred for 20 h. The crude was poured into 2 N HCl and extracted with Et₂O. The dried organic extracts were concentrated and purified by chromatography (10% EtOAc in hexane) to yield 17 (25 mg, 46%).

(*E*)-(1*R*,4a*S*,5*R*,8a*S*)-3-Methyl-5-(4a,5,8a-trimethyl-2-methylene-decahydronaphthalen-1-yl)pent-2-en-1-ol (20): $[\alpha]_D$ -6.4 (*c* 0.05, CH₂Cl₂); ¹H NMR (300MHz) 0.62 (s, 3H), 0.73 (s, 3H), 0.79 (d, J = 6.6 Hz, 3H) 1.10-1.60 (m, 10H), 1.75 (s, 3H), 1.80-1.95 (m, 1H), 2.05 (m, 1H), 2.10-2.25 (m, 3H), 2.35 (m, 1H), 2.58 (d, J = 9 Hz, 1H), 4.09 (br d, J = 6 Hz, 2H), 4.54 (s, 1H), 4.86 (s, 1H), 5.43 (br t, J = 6 Hz, 1H); ¹³C NMR (75 MHz, DEPT), see Table 1.





































































































