

Supporting Information for:

Insertion/Isomerization Polymerization of 1,5-Hexadiene: Synthesis of Functional Propylene Copolymers and Block Copolymers

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General Considerations

All manipulations of air- and/or water-sensitive compounds were carried out under dry nitrogen using a Braun Labmaster drybox or standard Schlenk line techniques. Polymer ^1H spectra were recorded on a Varian Inova (^1H , 400 MHz) spectrometer and referenced versus residual non-deuterated solvent shifts. ^{13}C NMR spectra of the polymers were recorded on a Varian VXR-400 (100 MHz) spectrometer and referenced versus residual non-deuterated solvent shifts. The polymer samples were dissolved in $\text{C}_2\text{D}_2\text{Cl}_4$ in a 5 mm O.D. tube by heating to 120°C in an oil bath.

Molecular weights (M_n and M_w) and polydispersities (M_w/M_n) were determined by high temperature gel-permeation chromatography (GPC). All analyses were performed with a Waters Alliance 2000 liquid chromatograph equipped with a Waters DRI detector and a Jordi styrene-divinylbenzene linear mixed-bed column. The GPC columns were eluted with 1,2,4-trichlorobenzene (TCB) containing 0.1 wt % Irganox 1010 at 140°C at 1.0 mL/min and were calibrated using 23 monodisperse polystyrene standards. Polymer samples were typically placed in a 140°C oven for 24 h to eliminate supermolecular aggregates prior to molecular weight measurements. DSC analyses were performed on a Seiko DSC 220C instrument using EXSTAR 6000 processing software. The measurements were made in crimped aluminum pans under nitrogen with a heating rate of $10^\circ\text{C}/\text{min}$ from -50 - 200°C , and reported values originate from the second heating scan. The sample containing 96% propylene (Table 1, Entry 4) was annealed at 80°C for one hour to ensure sufficient time for crystallization.

Materials

Toluene was purified over columns of alumina and copper and degassed by freeze-pump-thaw cycles. Propylene (Matheson, polymer grade) was purified through a mixed bed column (R&D Separations, BOT-4). Methylaluminoxane (PMAO-IP, 12.9 wt % Al in toluene, Akzo Nobel) was concentrated in vacuo to dryness (10^{-3} mm Hg) at 40 °C for 18 h to remove residual trimethylaluminum, providing a solid white powder. 1,5-Hexadiene was purchased from Aldrich, dried over activated 3 Å molecular sieves, and distilled before use. **1** was synthesized according to a published procedure (Tian, J.; Hustad, P. D.; Coates, G. W. *J. Am. Chem. Soc.* **2001**, *123*, 5134-5135). All other chemicals were commercial materials and were used as received.

Polymer Synthesis

Polymerization of 1,5-Hexadiene using 1/MAO. A Schlenk-type flask equipped with a magnetic stir bar was charged with a PMAO (0.29 g, 5.0 mmol), toluene (85 mL), and 1,5-hexadiene (11.0 g). The flask was equilibrated at 0 °C and **1** (0.018 g, 0.020 mmol, [Al]/[Ti] = 250) in toluene (4 mL) was added via gas-tite syringe. After 20 min, the reaction was carefully quenched with methanol/HCl and the polymer was precipitated in copious methanol/HCl, filtered, washed with methanol, and then dried *in vacuo* to constant weight (0.67 g, $M_n = 268,000$, $M_w/M_n = 1.27$).

Hydrogenation of Poly(1,5-Hexadiene). For NMR studies, the unsaturated poly(1,5-hexadiene) was hydrogenated according to the procedure of Wu et al. (Wu, Z.; Benedicto, A. D.; Grubbs, R. H. *Macromolecules* **1993**, *26*, 4975-4977). In a slightly modified experiment, a toluene solution (10 mL) of the polymer (0.20 g, 0.90 mmol C=C) and *p*-toluenesulfonylhydrazide (*p*-TsNHNH₂) (1.36 g, 7.30 mmol) was heated at 80 °C. After 3 h, another 8 equiv. *p*-TsNHNH₂ (1.36 g, 7.30 mmol) was added and the slurry was stirred for 15 h. At this point, another 3 equiv. *p*-TsNHNH₂ (0.45 g, 2.4 mmol) was added and the mixture was again stirred at 80 °C for 10 h. The saturated polymer was then isolated by precipitation into copious methanol, filtered, and dried *in vacuo* to constant mass (0.18 g isolated yield).

General Procedure for 1,5-Hexadiene/Propylene Copolymerization. A 6 ounce Lab-Crest® pressure reaction vessel (Andrews Glass) equipped with a magnetic stir bar was first conditioned under dynamic vacuum and high temperature and then charged with PMAO (0.29 g,

5.0 mmol), toluene (85 mL), and 1,5-hexadiene (11.0 g). The reactor was the equilibrated at 0 °C, the atmosphere was exchanged with propylene three times, and the solution was saturated under propylene pressure. The catalyst **1** (0.018 g, 0.020 mmol, [Al]/[Ti] = 250) in toluene (4 mL) was then added via gas-tite syringe. After 4 h, the reactor was vented and the polymer was precipitated in methanol/HCl, filtered, washed with methanol, and then dried *in vacuo* to constant weight.

Synthesis of Poly(propylene)-*block*-poly(propylene-co-1,5-Hexadiene). A 6 ounce Lab-Crest® pressure reaction vessel (Andrews Glass) equipped with a magnetic stir bar was first conditioned under dynamic vacuum and high temperature and then charged with PMAO (0.29 g, 5.0 mmol), toluene (85 mL). The reactor was the equilibrated at 0 °C, the atmosphere was exchanged with propylene three times, and the solution was saturated under propylene pressure (10 psi). The catalyst **1** (0.018 g, 0.020 mmol, [Al]/[Ti] = 250) in toluene (4 mL) was then added via gas-tite syringe. After 4 h, a 5 mL sample of the polymerization medium was removed, quenched with methanol, and the polymer was filtered and dried *in vacuo* to constant weight ($M_n = 51,500$, $M_w/M_n = 1.11$). 1,5-Hexadiene (11.0 g) was then added to the reactor via gas-tite syringe. After another 2 h, the reactor was vented and the polymer was precipitated in methanol/HCl, filtered, washed with methanol, and then dried *in vacuo* to constant weight (0.98 g, $M_n = 93,300$, $M_w/M_n = 1.11$).

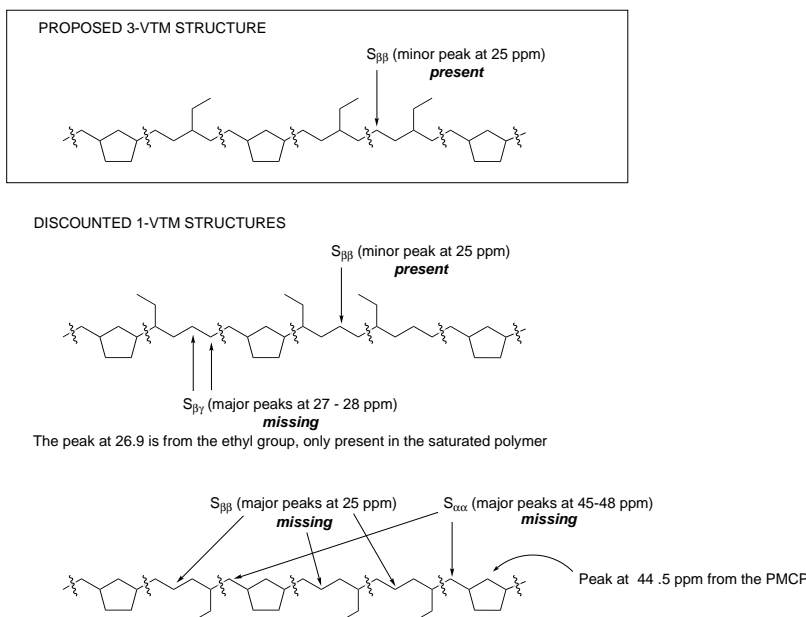


Figure S1. ^1H NMR spectrum (400 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 80°C) of the 1,5-hexadiene polymer from **1**/MAO (Table 1, Entry 1).

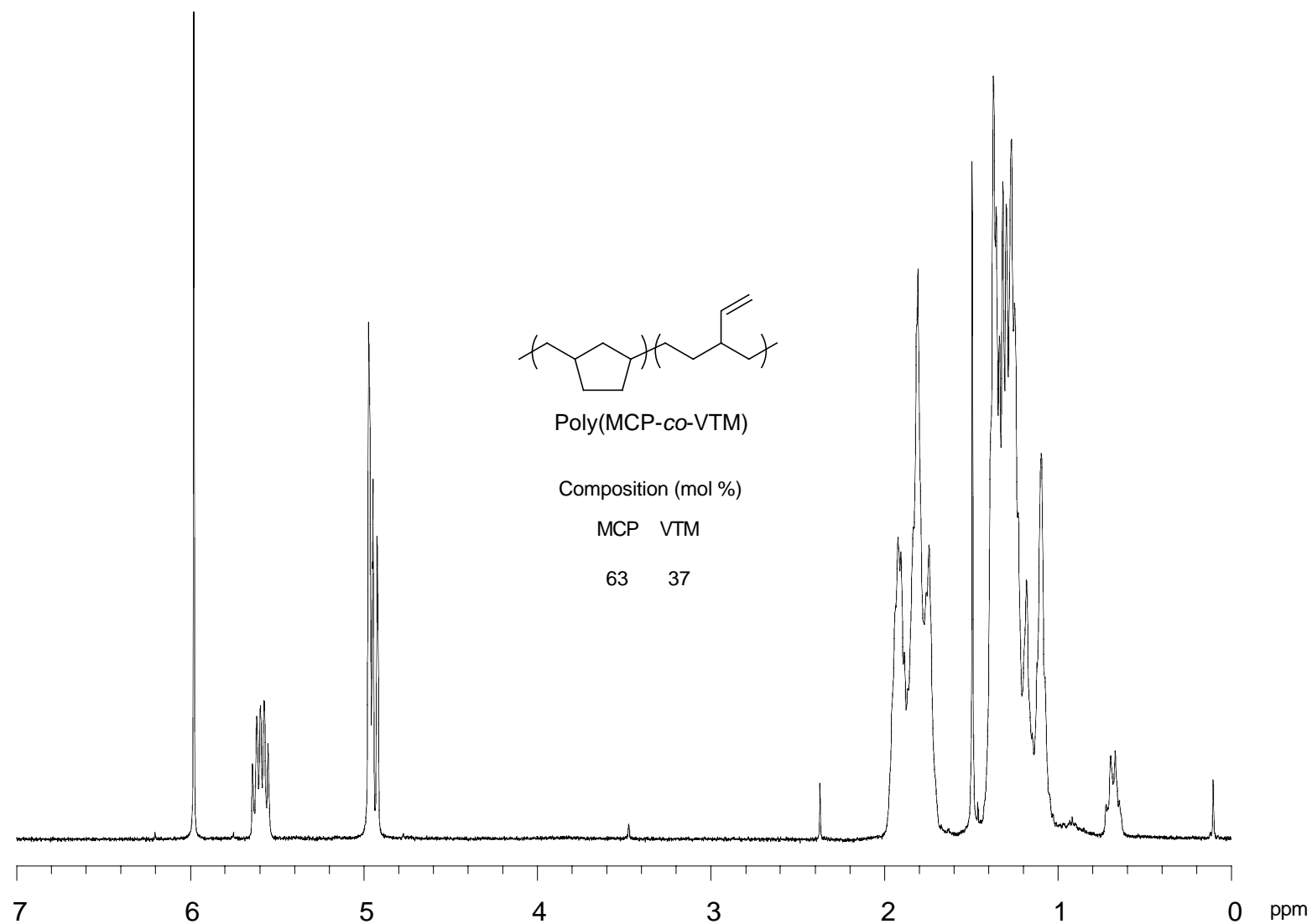


Figure S2. ^{13}C NMR spectrum (100 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 90°C) of the 1,5-hexadiene polymer from **1**/MAO (Table 1, Entry 1).

S5

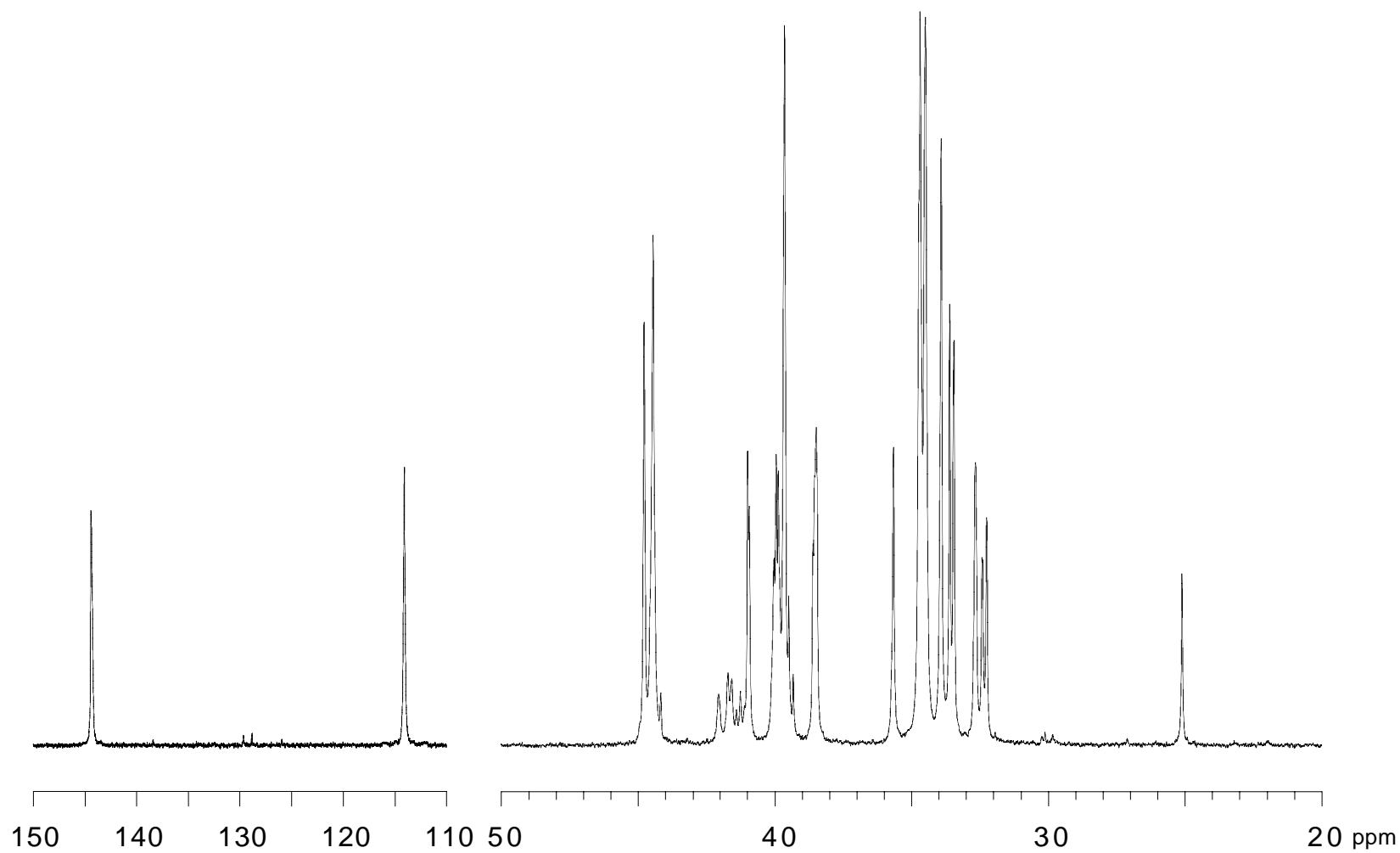


Figure S3. ^{13}C NMR spectrum (100 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 90 $^\circ\text{C}$) of the saturated hexadiene polymer from **1**/MAO and subsequent treatment with *p*-TsNHNH $_2$.

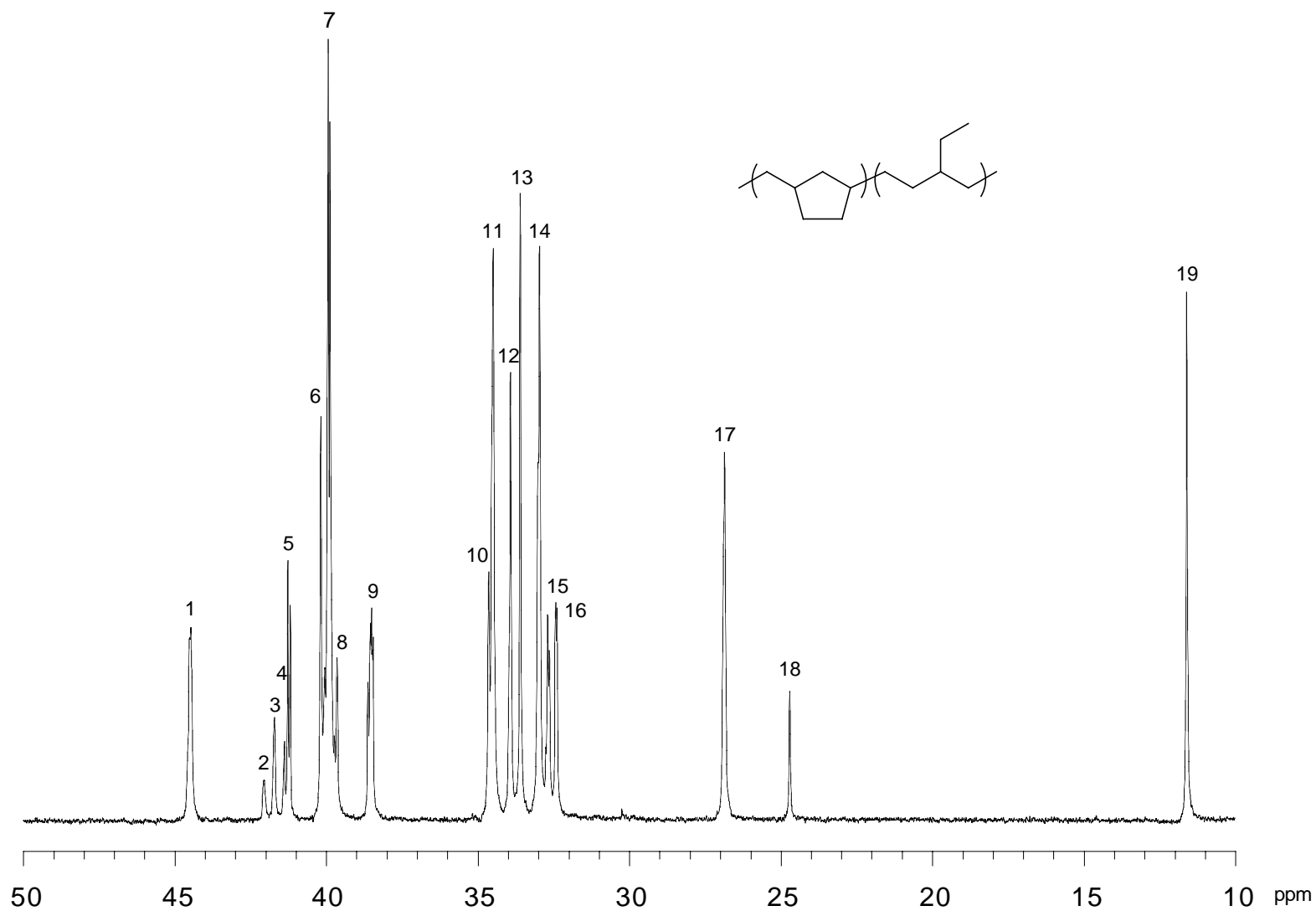


Figure S4. ^{13}C NMR spectrum (a) and DEPT spectrum (b) (100 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 90 °C) of the saturated hexadiene polymer from **1**/MAO and subsequent treatment with *p*-TsNHNH₂.

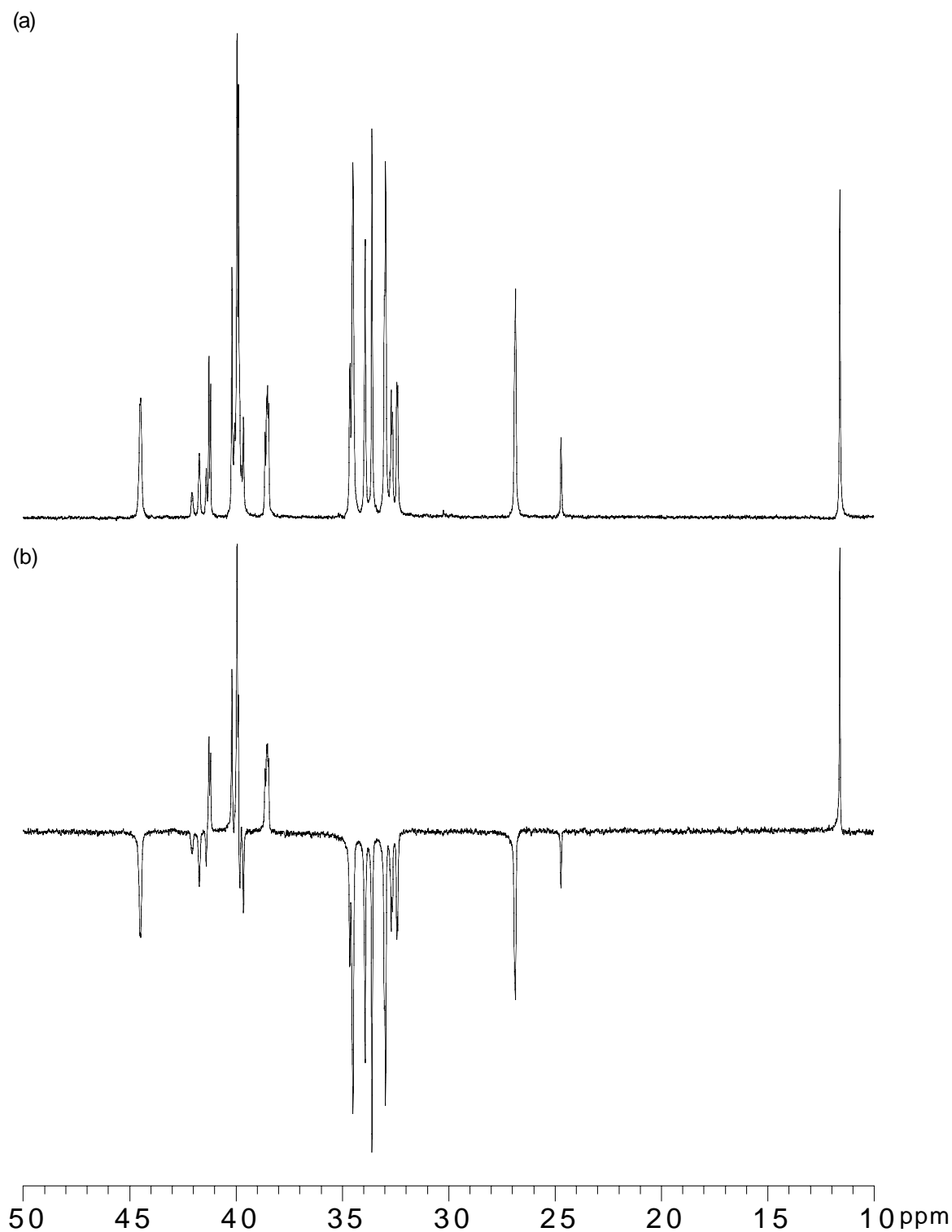
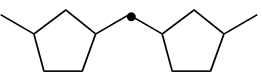
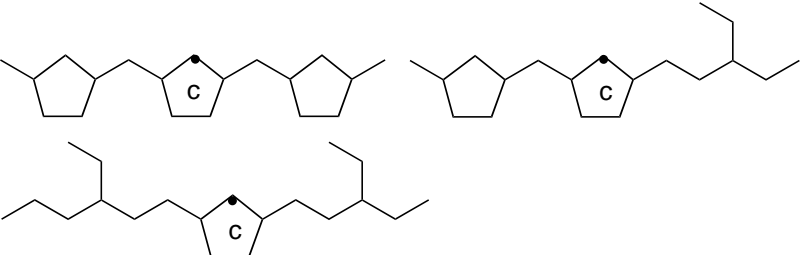
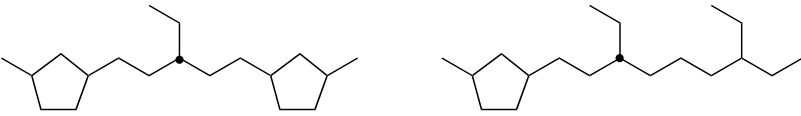
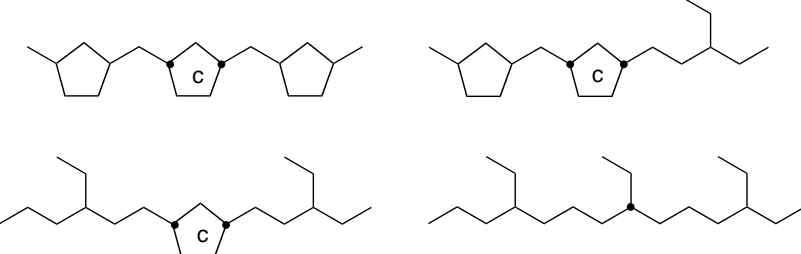
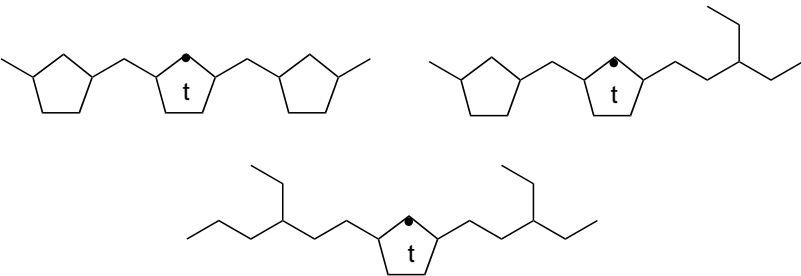
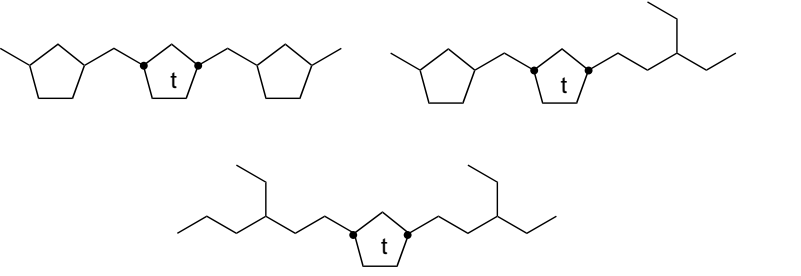
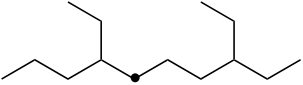
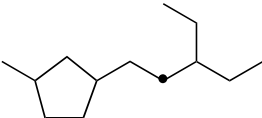
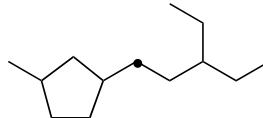

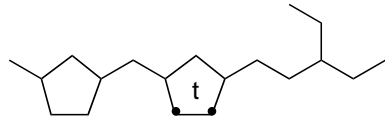
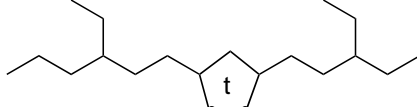
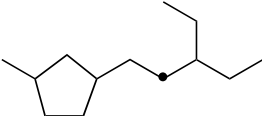
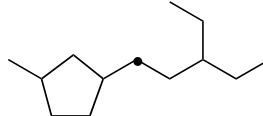
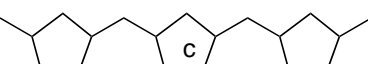
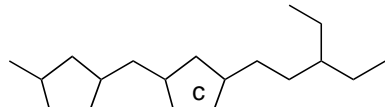
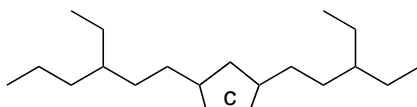
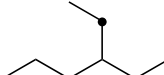
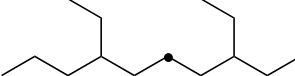
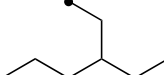


Table S1. Tentative ^{13}C NMR peak assignment for hexadiene polymers following hydrogenation of the pendant vinyl substituents.

Peak	Shift, ppm	Microstructure and Assignment*	Carbon
1	44.5		$\text{S}_{\alpha\alpha}$
2 - 4	42.0 - 41.2		C_2
5	41.2		$\text{T}_{\gamma\gamma}$ $\text{T}_{\gamma\delta}$
6,7	40.1-39.8		$\text{C}_{1,3}$ $\text{T}_{\delta\delta}$
8	39.8 - 39.6		C_2
9	38.6 - 38.5		$\text{C}_{1,3}$ $+$ $\text{T}_{\delta\delta}$
10	34.6		$\text{S}_{\alpha\gamma}$

11	34.5	 or 	$S_{\alpha\beta}$
12, 13	33.9 - 33.6	  	$C_{4,5}$
14	33.0	 or 	$S_{\alpha\beta}$
15, 16	32.7 - 32.4	  	$C_{4,5}$
17	26.9		CH_2CH_3
18	24.7		$S_{\beta\beta}$
19	11.6		CH_2CH_3

c : *cis*-ring, t : *trans*-ring

*Assignments based on ^{13}C and DEPT spectra and related assignments for (a) ethylene/hexadiene copolymers (Choo, T. N.; Waymouth, R. M. *J. Am. Chem. Soc.* **2002**, 124, 4188-4189), (b) ethylene/butene copolymers (Cheng, H. N. *Macromolecules* **1991**, 24, 4813-4819), (c) ethylene/propylene copolymers (Wang, W. J.; Zhu, S. *Macromolecules* **2000**, 33, 1157-1162 and references therein), and (d) poly(1,5-hexadiene) (Resconi, L.; Waymouth, R. M. *J. Am. Chem. Soc.* **1990**, 112, 4953-4954).

Figure S5. ^1H NMR spectrum (400 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 80 $^\circ\text{C}$) of the propylene/hexadiene copolymer from **1**/MAO (Table 1, Entry 3).

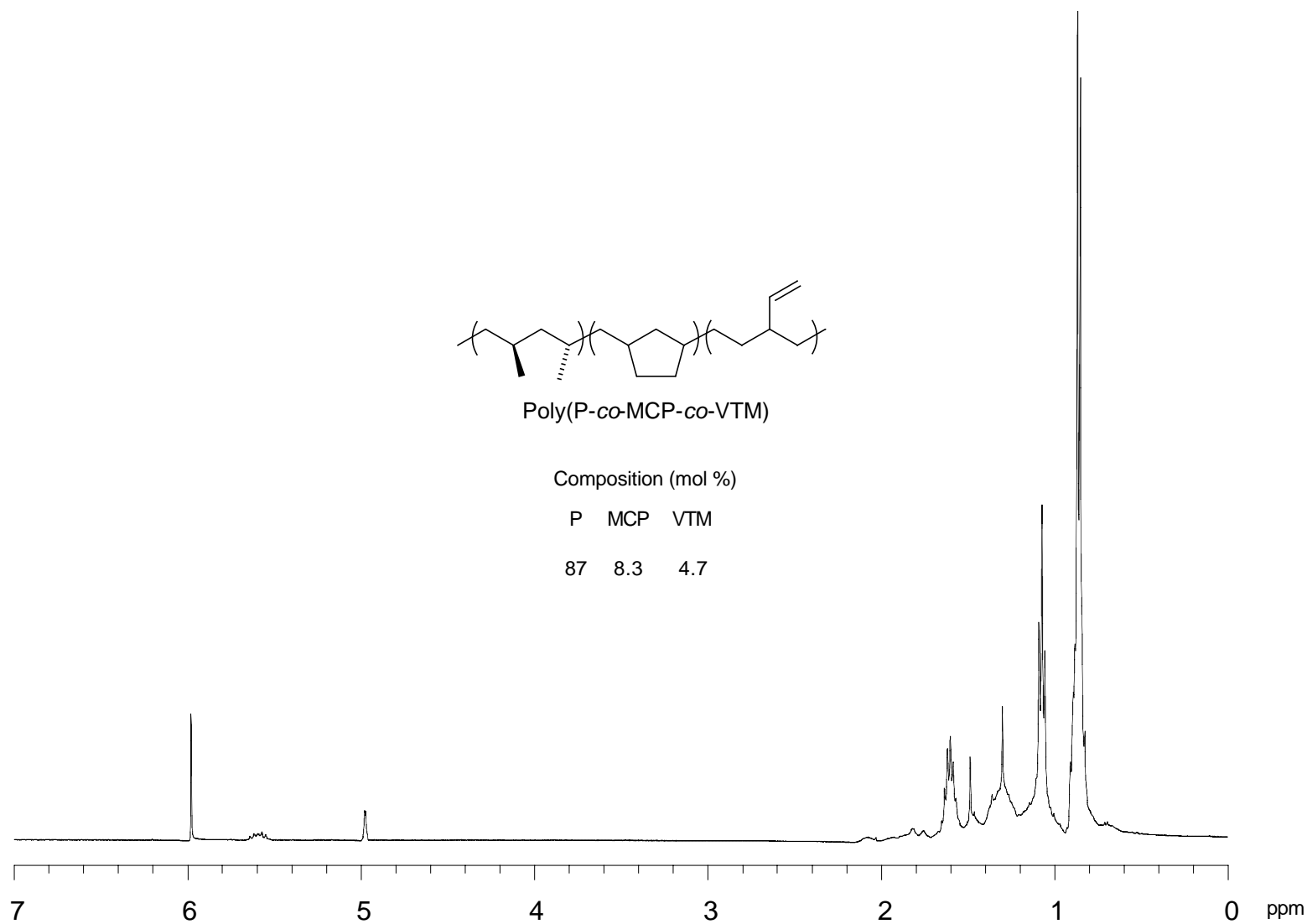


Figure S7. GPC data for the polypropylene-*block*-poly(propylene-*co*-hexadiene) copolymer.

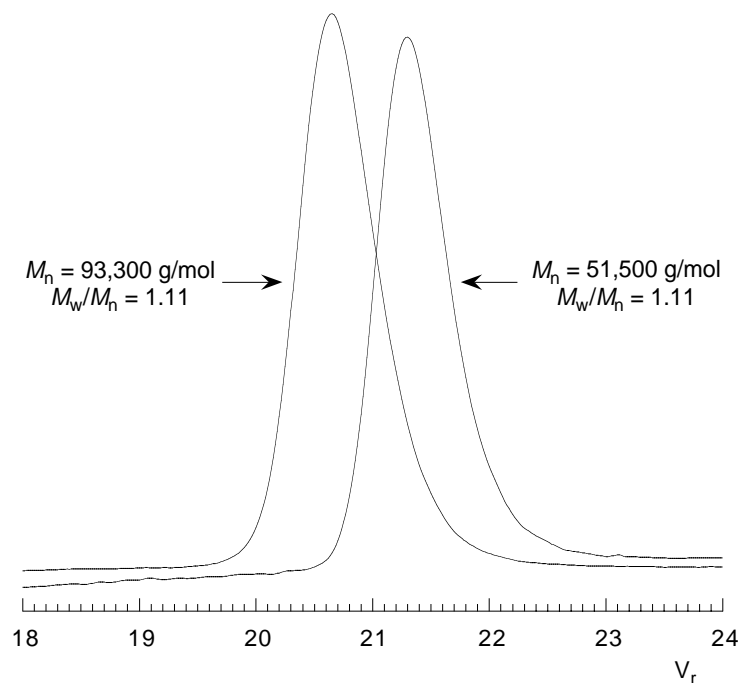


Table S2. DSC characterization of hexadiene polymers and hexadiene/propylene copolymers.

Entry	Composition (mol %) ^a			T_g^b (°C)	T_m^b (°C)
	P	MCP	VTM		
1	0	63	37	-19.3	-
2	100	0	0	-	146
3	87	8.3	4.7	2.8	-
4	96	1.5	2.5	0.6	93.2
5 ^c	100/93 ^d	0/4.3 ^d	0/2.7 ^d	2.6	137

^a Calculated from the ¹H NMR spectra. ^b Determined by DSC. ^c After reaction of **1**/MAO with propylene for 4 hours, 1,5-hexadiene was added and polymerization was conducted for another 2 hours. ^d Data for the initial polypropylene block and the final *syndio*-poly(propylene)-*block*-poly(propylene-*co*-hexadiene) diblock polymer.