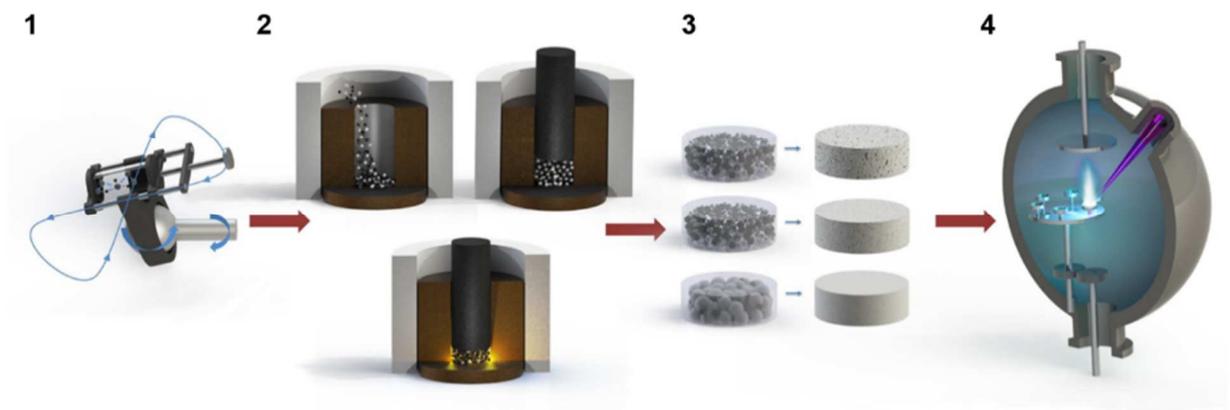


# Large Area Deposition of MoS<sub>2</sub> By Pulsed Laser

## Deposition With *In-Situ* Thickness Control

Martha I. Serna<sup>†</sup>, Seong H. Yoo<sup>‡</sup>, Salvador Moreno<sup>††1</sup>, Yang Xi<sup>†</sup>, Juan Pablo Oviedo<sup>†</sup>, Hyunjoo Choi<sup>‡</sup>, Husam N. Alshareef<sup>~</sup>, Moon J. Kim<sup>†</sup>, Majid Minary-Jolandan<sup>††1</sup>, Manuel A. Quevedo-Lopez<sup>†\*</sup>



**Figure S1. Target fabrication and process flow chart for the fabrication of MoS<sub>2</sub> Thin films.**

### Target fabrication

Powders of MoS<sub>2</sub> and S were mixed and the hot pressed to create three different targets with different densities as reported in Table 1. Target 100°C / 2 μm and 75°C / 2μm were fabricated with the same particle size and different hot press conditions, while target 75°C / 43μm was fabricated with a larger (43μm) MoS<sub>2</sub> particle size and 75°C for 3 hours with a powder

compression of 18 ton-force. Figure S1. shows the target non-scale particle size. The thin film deposition was carried in separate processes with identical conditions. The films are named with the same nomenclature as the corresponding targets.

**Table 1. MoS<sub>2</sub> target fabrication conditions. Targets 100°C / 2 μm to 75°C / 43 μm have a sulfur particle size of 20 μm with variable MoS<sub>2</sub> particle size.**

Processing Condition	100°C / 2 μm	75°C / 2 μm	75°C / 43 μm	Commercial
MoS <sub>2</sub> size (μm)	2	2	43	N/A
Milling Process	SPEX milling	SPEX milling	SPEX milling	Commercial brand
Hot pressed condition	100 °C 1h / 25 Ton	75 °C 3h / 18 Ton	75 °C 3h / 18 Ton	N/A
MoS <sub>2</sub> :S ratio	1:1	1:1	1:1	1:0
Relative density %	88.1	88.7	99.9	N/A

### Precursor costs

PLD:

For the process of 10 min of deposition time the target cost is estimated as follows:

MoS<sub>2</sub> and S powders for targets 75°C / 43μm, 75°C / 2μm, 100°C / 2μm

For each target 11.17 gr. of MoS<sub>2</sub> powder and 2.23 gr. Sulfur powder were used.

- Sulfur ( 20 μm particle size)  
US\$ 125.7 / 50 gr. (\$2.514/ gr)
  
- MoS<sub>2</sub> ( 43 μm particle size)  
US\$ 64.15 / 100 gr. (\$0.6415 / gr)
  
- MoS<sub>2</sub> ( 2 μm particle size)  
US\$ 93.6 / 100 gr (\$0.936 / gr)

**Table 2. Materials cost for MoS<sub>2</sub> target fabrication**

Target	Sulfur (US\$ / g)	MoS <sub>2</sub> (US\$ / g)	Sulfur (g)	MoS <sub>2</sub> (g)	Target cost (US\$)
100°C / 2 μm	2.5	0.936	2.23	11.17	16.06
75°C / 2 μm	2.5	0.936	2.23	11.17	16.06
75°C / 43 μm	2.5	0.641	2.23	11.17	12.77

## CVD

Prices as of April 6 2016

- Mo(CO)<sub>6</sub>

0.01 sscm per 26 hours = 15.600 cm<sup>3</sup>; MoCl<sub>5</sub> density is 2.928 g/cm<sup>3</sup>; 45.6768 MoCl<sub>5</sub> grams were used for a total of \$ 565.478 dollars

- (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>S

0.4 sscm per 26 hours = 624 cm<sup>3</sup>; \$1.22 dollars/ ml, for a total of \$761.28 dollars

- H<sub>2</sub>

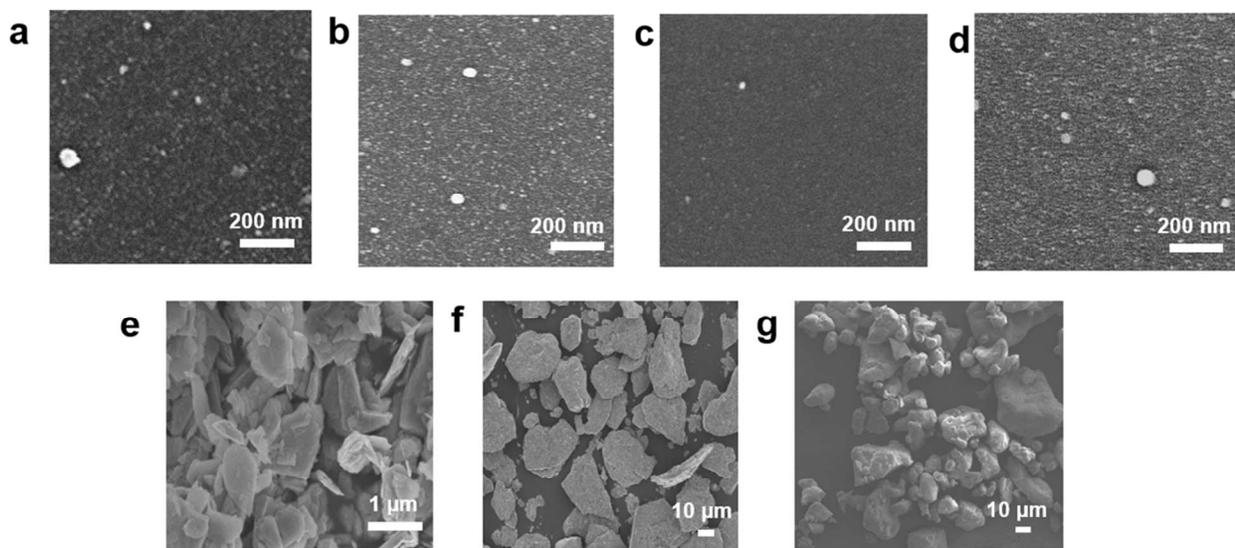
5 sscm per 26 hours = 7800 cm<sup>3</sup>; \$4.22 dollars/ lt, for a total of \$32.94 dollars

- Ar

150 sscm per 26 hours = 234000 cm<sup>3</sup>; \$4.61 dollars/ lt, for a total of \$1078.74 dollars

## Powders and thin films characterization (SEM)

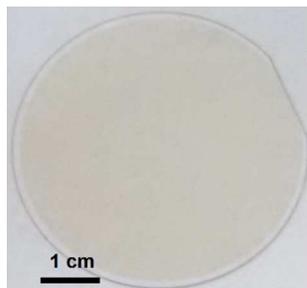
The morphology of the MoS<sub>2</sub> thin films and the powders used for the target fabrication, was imaged using Scanning Electron Microscopy (SEM) with an electron microscope Zeiss Supra-40. **Figures S2 a,b,c,d,e,f** show the top view of each thin film and the particle shape of Sulfur and molybdenum disulfide powders. The film produced by target 75°C / 43 μm had less particles on the surface and it appeared more superficially homogeneous than the other thin films (**Figure S2d**)



**Figure S2. Scanning electron microscopy of thin films (top view) and MoS<sub>2</sub> and Sulfur powders. Figures S2 a, b, c, d show the top view of thin films 100°C / 2 μm, 75°C / 2μm, 75°C / 43 μm and T, respectively. Figures S2 e-f show MoS<sub>2</sub> powder of 2 μm and 43 μm. Figure S2 g shows the Sulfur particle size.**

**Thin film 75°C / 43μm deposited 5.08 cm diameter sapphire wafer**

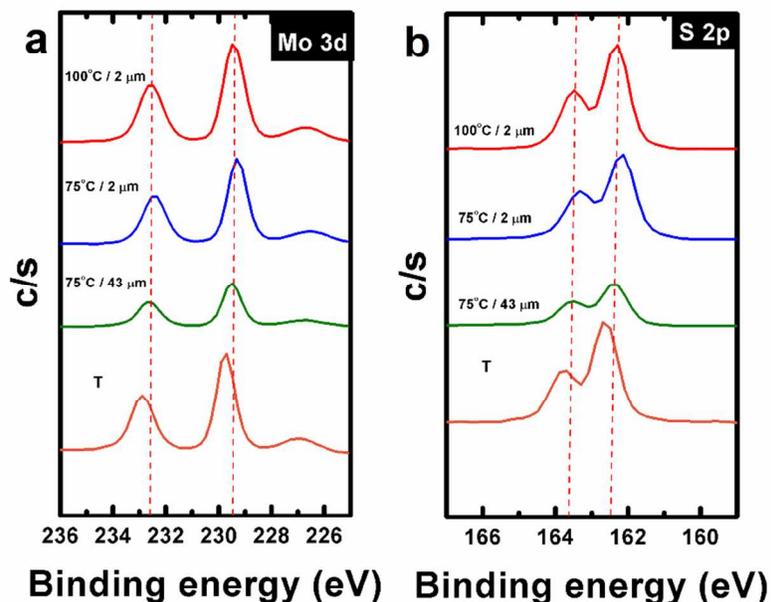
With thin film  $75^{\circ}\text{C} / 43\mu\text{m}$  being homogeneous and presenting less particle density, it was the ideal candidate to deposit it on a 50.8 mm sapphire wafer. Thin film  $75^{\circ}\text{C} / 43\mu\text{m}$  homogeneously covered the entire sapphire wafer. (**Figure S3**)



**Figure S3.  $\text{MoS}_2$  thin film  $75^{\circ}\text{C} / 43\mu\text{m}$  on a two inches diameter sapphire wafer**

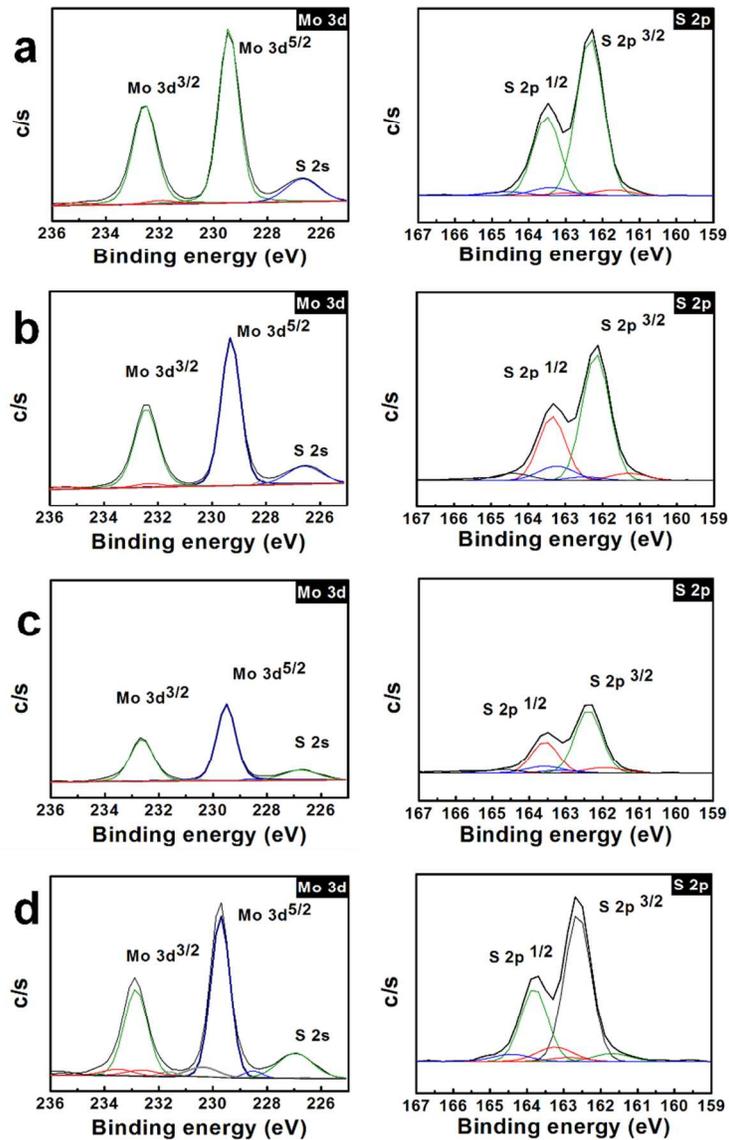
### **Characterization of the samples**

The Mo 3d, S 2s, and S 2p regions of the XPS spectra for samples  $100^{\circ}\text{C} / 2\mu\text{m}$ ,  $75^{\circ}\text{C} / 2\mu\text{m}$ ,  $75^{\circ}\text{C} / 43\mu\text{m}$  and T are shown in **Figure S4** and the deconvolution in **Figure S5**. The Mo 3d spectra consists of peaks at around 229.5 and 232.6 eV that correspond to  $\text{Mo}^{4+} 3d^{5/2}$  and  $\text{Mo}^{4+} 3d^{3/2}$  of  $\text{MoS}_2$ , respectively. Deconvolution of peaks from samples  $100^{\circ}\text{C} / 2\mu\text{m}$ ,  $75^{\circ}\text{C} / 2\mu\text{m}$ , and T revealed additional peaks that were shifted to lower and higher energies with respect to the



**Figure S4. X-ray photoelectron spectroscopy (Fig S4a and S4b) for films deposited on sapphire. For reference, the dashed lines in each figure shows the expected binding energy or Raman shift for a stoichiometric MoS<sub>2</sub>. The thinnest film (~1.3 nm, film 75°C / 43 μm) was deposited from the target with the highest density and seems to be the most stoichiometric**

actual binding energy value of the MoS<sub>2</sub> peaks. In the S 2p region of the spectra, additional peaks were obtained from the doublet peaks of MoS<sub>2</sub>, S 2p<sup>1/2</sup> and S 2p<sup>3/2</sup>, which have binding energy at 163.6 and 162.4 eV, respectively. The shifting of these additional peaks suggests the presence of molybdenum oxides, MoS<sub>X</sub> (with X≠2) phase, molybdenum oxysulfides, and free Mo. Furthermore, no peaks were observed between 233 and 236 eV for thin film 75°C / 43 μm, which indicated that Mo atoms remained unoxidized.



**Figure S5. Deconvolutions of XPS spectra for each thin film. Thin film 100°C / 2  $\mu\text{m}$  (a), Thin film 75°C / 43  $\mu\text{m}$  (b), 75°C / 43  $\mu\text{m}$ , (c), Thin film T (d). Thin film 100°C / 2  $\mu\text{m}$  showed binding energy typical of  $\text{MoS}_2$  and two bands typical of  $\text{MoO}_3$**

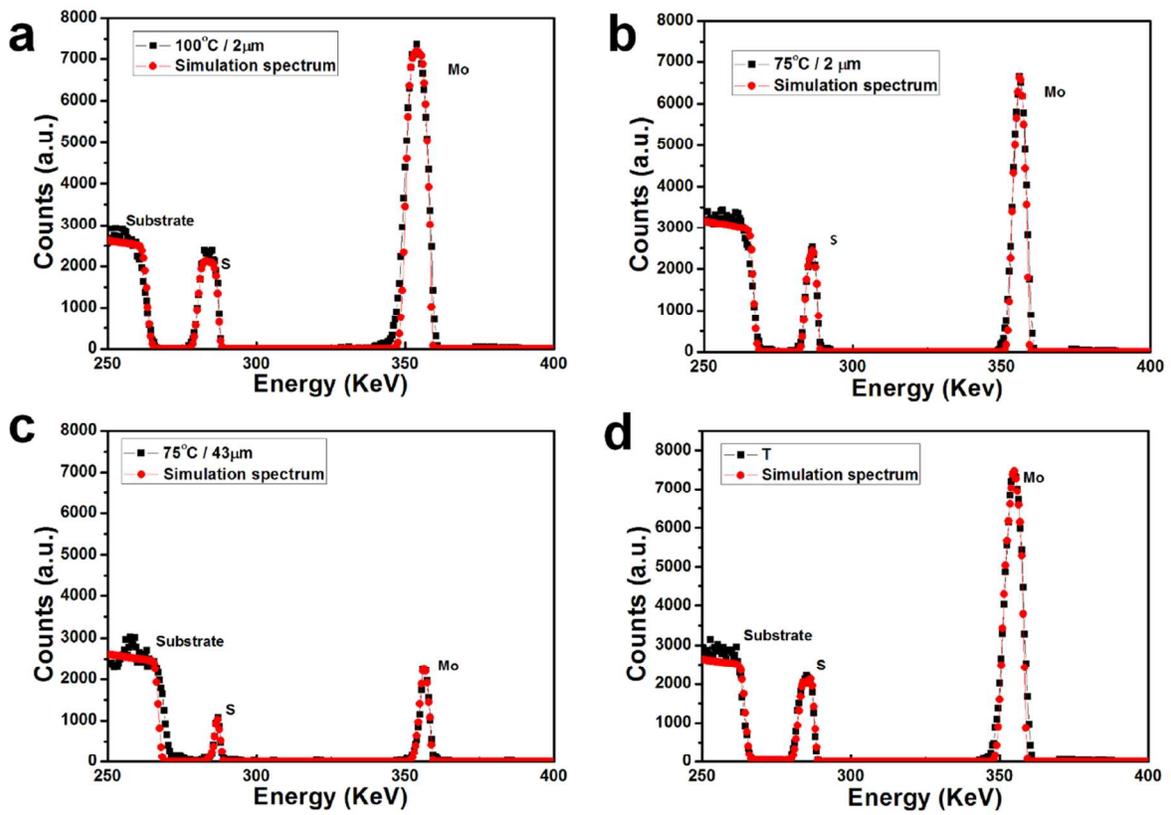
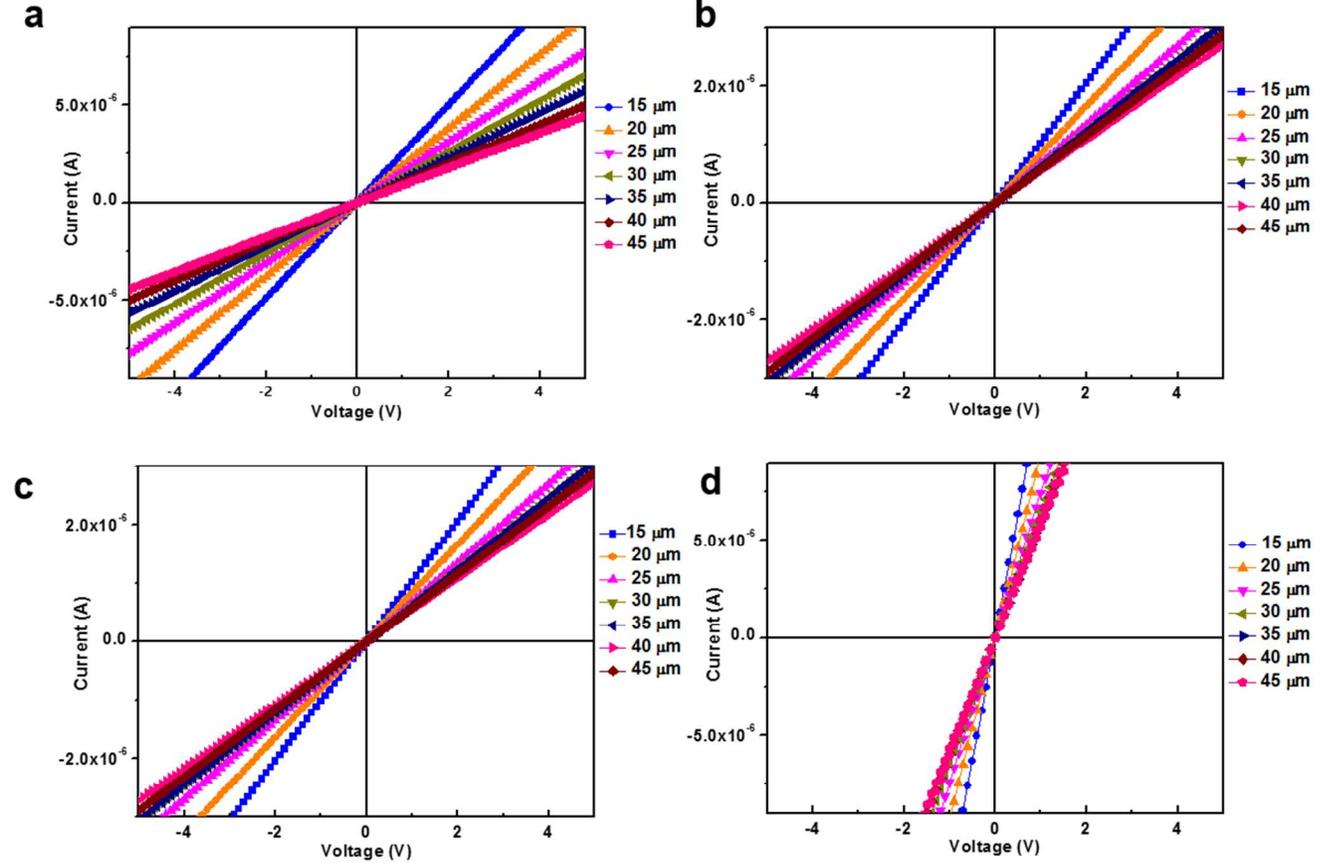


Figure S6 Rutherford Backscattering Spectrometry. RBS spectra (dark line) and simulation spectra (red line) for thin films 100°C / 2 μm (a), 75°C / 2 μm, (b), 75°C / 43 μm, (c), and T (d).

## Electrical Characterization- Circular Transfer Length Method



**Figure S7. CTLM measurements for thin films. I-V characteristics of (a) Thin film 100°C / 2  $\mu\text{m}$  (b) 75°C / 2 $\mu\text{m}$ , (c), 75°C / 43 $\mu\text{m}$ , (d) and T.**

The CTLM measurements were made on circles from of 15 to 45 microns, since those were the ones that opened easier during the lift off lithography process. **Figure S7** shows the I-V measurements for each sample. The Voltage for the evaluation varied from -4 to 4 Volts. Figure S8 shows the electron microscopy image of the CTLM pattern.

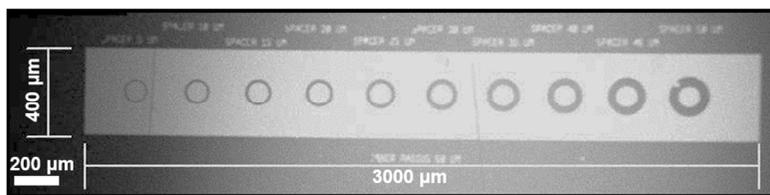


Figure S8. Scanning Electron Microscopy (SEM) image from the CTLM pattern. The light part are the Au contacts and the dark part is the MoS<sub>2</sub> thin film .

### Phase Contrast Imaging – Thin film Crystallinity

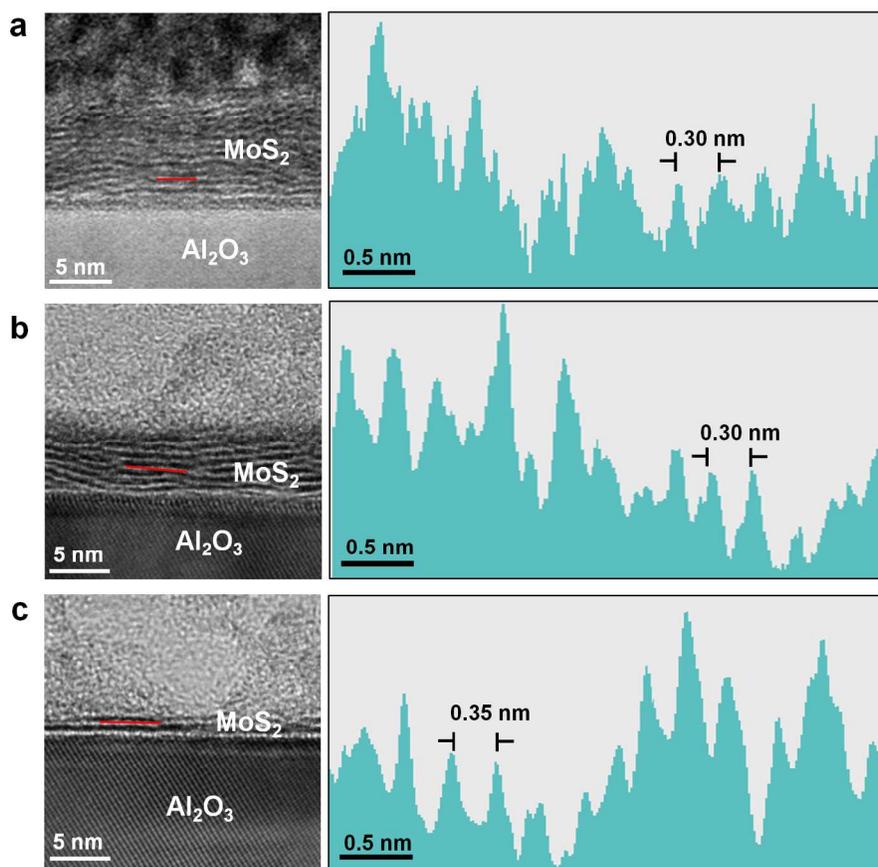
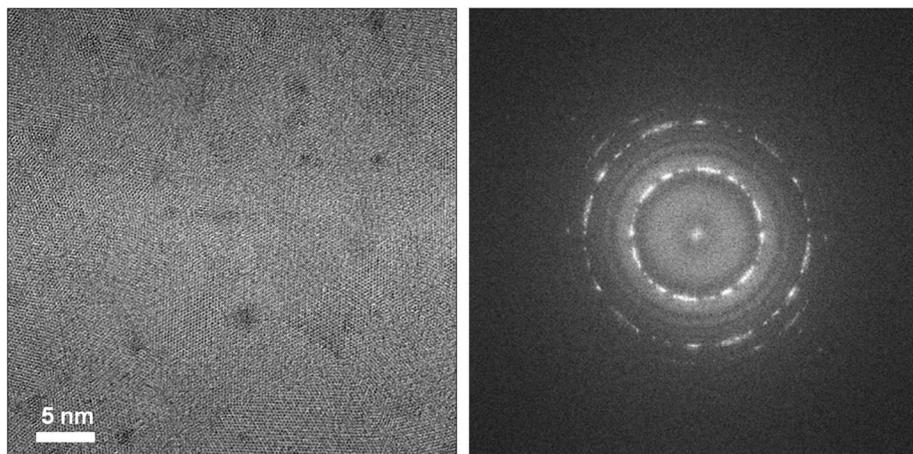


Figure S9. Phase contrast profiling. Cross section of films 100 Thin film 100°C / 2 μm (b) 75°C / 2 μm, y (c) 75°C / 43 μm. The red line shows the location of the profile.

A profile for each thin film was acquired using a Gatan Microscopy Suite software (v. 2.32.888.0). Each line in red (**Figure S9**) shows the location of the profile. From these analyses an atomic arrangement consistent with a crystalline film is evident from the atomic separation results shown in the corresponding profiles. The 0.30 to 0.35 nm interatomic distance experimentally measured match closely the theoretical interatomic separation between Sulfur in MoS<sub>2</sub>, which is approximately 0.31 nm. (Stupian and Leung 1987)Evidently, the films synthesized in this work are crystalline.

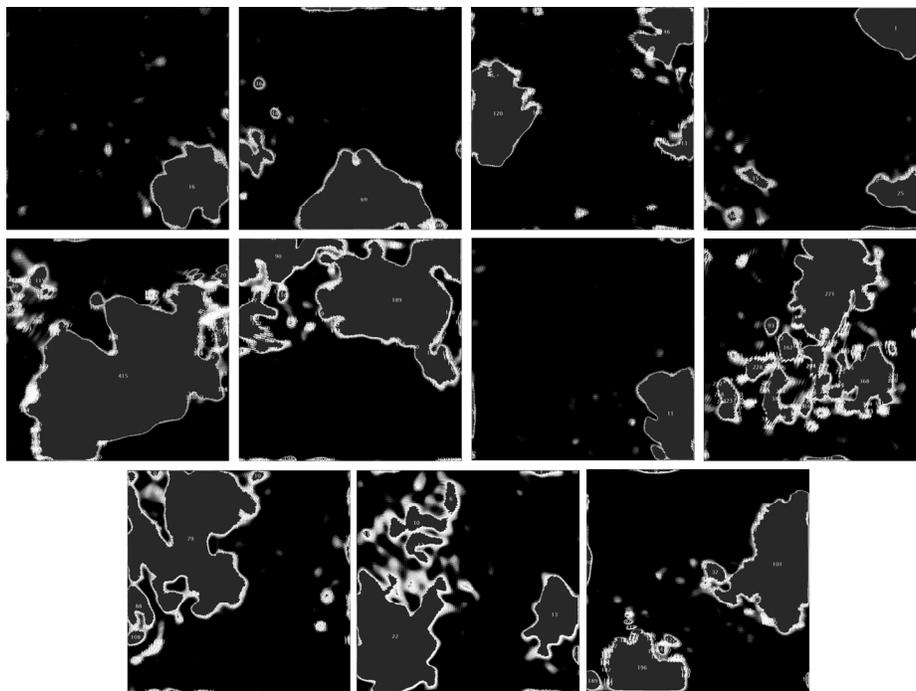
### Grain Size on Graphene



**Figure S10. Top view of MoS<sub>2</sub> Grains and MoS<sub>2</sub> on graphene FFT image simulation. The rings and multiple points in the materials shows that the film is polycrystalline.**

The grain size of thin film 75°C / 43 μm was measured on Graphene. On average the grain size was 16.80 nm. Furthermore, the FFT image shows that the film is polycrystalline (**Figure S10**).

A plain view image of the deposited film on Graphene were obtained using a JEOL JEM-ARM 200F transmission electron microscope operating at 200 kV. The FFT image show demonstrates the polycrystalline nature of the MoS<sub>2</sub> thin film (**Figure S10**). Grain orientation maps were then created (using the Gatan Microscopy Suite software) from the Fourier spectrum of these images, by moving a mask at discrete angular steps of 5 degrees along a circular path corresponding to the **G** vectors of the MoS<sub>2</sub> reciprocal lattice, and obtaining their inverse FFT. A Sobel and Smooth spatial filters were applied to highlight the grain area, after which the Particle Analysis package was used to identify the grains in the image and obtain their dimensions.(Seyring, et al. 2009) **Figure S11** shows a region of the sample where the MoS<sub>2</sub> grains were found, its grain orientation map, and grain size distribution derived from the above method. On Average the grain size was  $16.81 \pm 6.39$  nm (**Figure S12**).



**Figure S11. Grain orientation maps for 75°C / 43  $\mu$ m on Graphene. The areas in gray represent the grains of MoS<sub>2</sub>.**

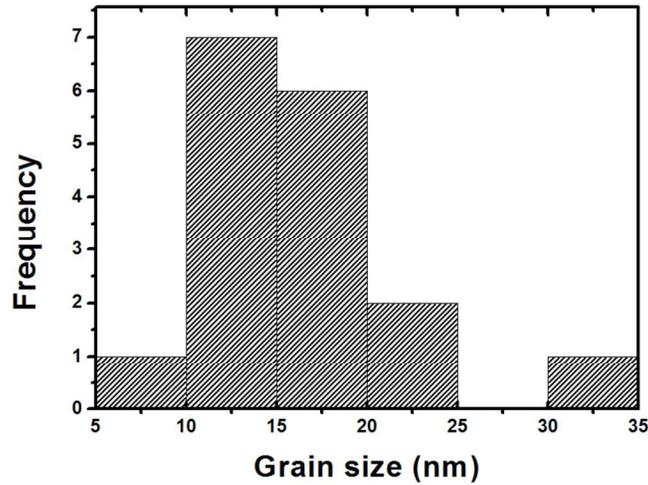


Figure S12. Grain size distribution for thin film 75°C / 43  $\mu\text{m}$  deposited on Graphene

## References

1. Stupian, G. W.; Leung, M. S., Imaging of MoS<sub>2</sub> by Scanning Tunneling Microscopy. *Appl. Phys. Lett.* **1987**, *51*, 1560-1562.
2. Seyring, M.; Song, X.; Chuvilin, A.; Kaiser, U.; Rettenmayr, M., Characterization of Grain structure in Nanocrystalline Gadolinium by High-resolution Transmission Electron Microscopy. *J. Mater. Res.* **2009**, *24*, 342-346.