

## Supporting Information for

### Synthesis of a Double Spiro Polyindenofluorene with Stable Blue Emission

Doojin Vak, Bogyu Lim, Soo-Hyoung Lee, and Dong-Yu Kim

**Material Synthesis.** Synthesis of 1-bromo-2-benzylbenzene was reported in our previous work.<sup>1</sup> 1-chlorododecane was purchased from TCI chemicals. Other starting materials were purchased from Aldrich Chemical Co. and used without further purification except THF, which was dried by refluxing over calcium hydride

**4,4,5,5-tetramethyl-2-(2,5-dimethyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-1,3,2-dioxaborolane (1).** 1,4-dibromoxylene (13.2 g, 50 mmol) was dissolved in 300 ml THF and cooled to -80 °C. *tert*-buthyllithium (1.7 M) (100 ml, 170 mmol) was added very slowly through a cannula under nitrogen. The solution was stirred at -40 °C for 12 h. The solution was cooled to -80 °C again and then 2-isopropoxy-4,4,5,6-tetramethyl-1,3,2-dioxaborolane (25 ml, 123 mmol) was added quickly. The solution was stirred at rt for 1 h. The solution was extracted using 500 ml of ether, dried over MgSO<sub>4</sub>, and evaporated. The crude product was dissolved in 300 ml of boiling hexane and recrystallized in the refrigerator. The product was white powder with 47 % yield. Mp: 178 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 1.34 (s, 24H), 2.48 (s, 6H), 7.54 (s, 2H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 21.37, 24.78, 83.39, 136.98, 140.62. Anal. Calcd for C<sub>20</sub>H<sub>32</sub>B<sub>2</sub>O<sub>4</sub>: C, 97.08, H, 9.01. Found: C, 97.34, H, 9.31. MS(EI): m/z 358.15 (M<sup>+</sup>)

**4,4''-Dibromo-2',5'-dimethyl-[1,1';4',1'']terphenyl (2).** **1** (7.16 g, 20 mmol), bromiodobenzene (22.6 g, 80 mmol) and Pd(PPh<sub>3</sub>)<sub>4</sub> (0.22 g, 0.2 mmol) were dissolved in 400 ml of THF and 100 ml of degassed K<sub>2</sub>CO<sub>3</sub> (2 M) solution was added to the THF solution. The reaction mixture was refluxed under nitrogen for 6 h. The reaction mixture was transferred to separatory funnel and 500 ml of ether was added. The organic solution was washed by water several times. The organic layer was dried over MgSO<sub>4</sub>, and evaporated. The crude product was recrystallized from dichloromethane and hexane. The product was yellowish white powder with 93 % yield. Mp: 202 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 2.25 (s, 6H), 7.11 (s, 2H), 7.23 (d, *J* = 6.6 Hz, 2H), 7.56 (d, *J* = 6.6 Hz, 2H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 19.73, 121.12, 130.93,

<sup>1</sup> Vak, D.; Chun, C.; Lee, C. L.; Kim, J.-J.; Kim, D.-Y. *J. Mater. Chem.* **2004**, *14*, 1342.

131.35, 131.80, 132.73, 140.03, 140.51. Anal. Calcd for C<sub>20</sub>H<sub>8</sub>Br<sub>2</sub>: C, 57.72, H, 3.88. Found: C, 57.53, H, 3.86. MS(EI): m/z 413.95 (M<sup>+</sup>)

**2,8-Dibromo-indeno[1,2-b]fluorene-6,12-dione (3).** KMnO<sub>4</sub> (15.8 g, 100 mmol) was placed in 3-neck flask and 20 ml of water was added. To the slurry, **2** (8.33 g, 20 mmol) in 500 ml of pyridine was added. The reaction mixture was stirred using mechanical stirrer and refluxed. KMnO<sub>4</sub> (7.9 g, 50 mmol) in 20 ml of water solution was added to the reaction mixture four times every 30 min. The reaction mixture was further refluxed for 6 h. The hot solution was filtered to remove the dark-brown solid (MnO<sub>2</sub>) and evaporated to remove pyridine. 250 ml of H<sub>2</sub>SO<sub>4</sub> was added to the flask and the mixture was stirred for 12 h at 80 °C. The hot reaction mixture was added to ice. Precipitates were filtered and washed by water, methanol and hot THF several times. The precipitate was dried in vacuum oven at 150 °C for 4 days. The product was violet colored insoluble powder with yield of 76 %. Anal. Calcd for C<sub>20</sub>H<sub>8</sub>Br<sub>2</sub>O<sub>2</sub>: C, 54.58, H, 1.83. Found: C, 54.51, H, 1.82. MS(DI): m/z 437.80 (M<sup>+</sup>)

**2',8'-dibromo-dispiro[[9,10]dihydro-anthracene-9,6'-[6,12]dihydro-indeno[1,2-b]fluorene-12',9''-[9,10]dihydro-anthracene] (4)** Mg (2.88 g, 12 mmol) was charged into a two-neck flask and flame-dried under nitrogen. 1-benzyl-2-bromobenzene (2.96 g, 12 mmol) in 100 ml of THF was added and refluxed for 1 h. **3** was added quickly with the protection of already formed Grignard reagent from air by the flow of nitrogen. The mixture was stirred for 12 h at rt 10 ml of HCl (2 M) solution was added and the mixture was extracted using 300 ml of ether, dried over MgSO<sub>4</sub>, and evaporated. Unreacted starting materials were removed using short liquid column chromatography and the remaining was charged into two-neck flask and 100 ml of acetic acid and 1 ml of conc. HCl solution were added. The mixture was refluxed for 6 h and white precipitates were appeared during the period. The mixture was cooled to rt and the precipitates were filtered and dried in vacuum oven. The product was recrystallized from hot chloroform and methanol twice. The product was white powder with 43 % yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 4.57 (s, 4H), 6.50 (d, *J* = 7.9 Hz, 4H), 6.99 (t, *J* = 8.2 Hz, 4H), 7.09 (s, 2H), 7.25 (t, *J* = 7.4 Hz, 4H), 7.33-7.47 (m, 10H) <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 34.75, 59.33, 117.03, 121.53, 121.73, 126.91, 127.09, 127.86, 128.50, 128.61, 130.68, 134.29, 138.35, 138.94, 140.22, 155.59, 157.91. Anal. Calcd for C<sub>46</sub>H<sub>28</sub>Br<sub>2</sub>: C, 74.61, H, 3.81. Found: C, 74.40, H, 3.80.

**2',8'-dibromo-10,10,10'',10''-tetradodecyl-dispiro[[9,10]dihydro-anthracene-9,6'-[6,12]dihydro-indeno[1,2-b]fluorene-12',9''-[9,10]dihydro-anthracene] (5)** **4** (1.4 g, 1.9 mmol), 1-chlorododecane (1.68 g, 12 mmol) and a piece of 18-crown-6 was dissolved in 150 ml of THF. To the solution, excess amount of KH, which was dispersed in mineral oil, was

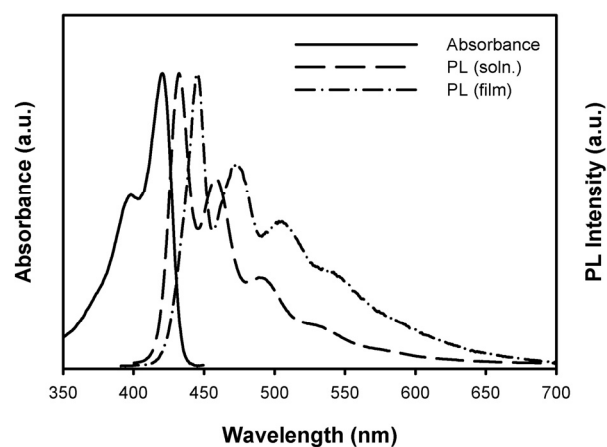
added under nitrogen. The mixture was stirred at rt for 5 h and unreacted KH was deactivated by addition of methanol under nitrogen. The solution was extracted by ether, dried over MgSO<sub>4</sub>, and evaporated. The crude product was purified by liquid chromatography using hexane and dichloromethane (9.8: 0.22) as eluents. The product was white powder with 49 % yield. (*R<sub>f</sub>* = 0.44) Mp: 152 °C. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 0.8-1.3 (m, 92H), 2.24 (br, 8H), 6.33 (d, *J* = 8.1 Hz, 4H), 6.90 (t, *J* = 7.5 Hz, 4H), 7.03 (s, 2H). 7.25 (m, *J* = 7.4 Hz, 8H), 7.32 (s, 2H), 7.52 (d, *J* = 8.1 Hz, 4H) <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 13.99, 22.58, 25.91, 26.36, 29.25, 29.55, 29.64, 29.81, 29.85 30.09, 31.87, 45.68, 46.87, 57.58, 117.12, 121.05, 121.87, 125.91, 126.25, 127.35, 128.89, 129.07, 130.20, 137.32, 139.04, 139.25, 140.20, 158.20, 160.67. Anal. Calcd for C<sub>94</sub>H<sub>124</sub>Br<sub>2</sub>: C, 79.86, H, 8.84. Found: C, 80.51, H, 9.28.

**Poly[10,10,10'',10''-tetradodecyl-dispiro[[9,10]dihydro-anthracene-9,6'-[6,12]dihydro-indeno[1,2-b]fluorene-12',9''-[9,10]dihydro-anthracene]-2',8'-diyl] P12SIF** bis(2,5-cyclooctadiene)-nickel(0) (0.36 g, 1.3 mmol), dipyriddy (0.20 g, 1.3 mmol) and 1,5-cyclooctadiene were placed in two-neck flask and 10ml of DMF was added. The solution was stirred at 80 °C for 30 min and **5** (0.71 g, 0.5 mmol) in 40 ml of toluene was added to the dark-blue solution under nitrogen. The solution was stirred for a day at 80 °C and poured into a mixture of methanol and conc. HCl solution (8 : 2). Precipitates were filtered and redissolved in 50 ml of hot chloroform. The solution was dropwised into 200 ml of methanol. The polymer was collected and purified by Soxhlet extraction using acetone for 2 days. The precipitation process was repeated again. The polymer was greenish yellow solid with 72 % yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 0.6-1.3 (br, 92H), 2.2 (br, 8H), 6.21 (br, 4H), 6.7 (br, 16H), 7.4 (br, 4H). Anal. Calcd for C<sub>94</sub>H<sub>124</sub>: C, 90.03, H, 9.97. Found: C, 90.01, H, 9.72.

**Characterization and Measurements.** <sup>1</sup>H and <sup>13</sup>C NMR spectra were measured by JEOL JNM-LA300WB 300 MHz and processed by ACD/SpecManager. Mass data were obtained by GCMS (Shimadzu, GCMS-QP2010). Thermal properties including glass transition temperature (*T<sub>g</sub>*) were determined by means of differential scanning calorimetry (DSC; TA2010) and thermo-gravimetric analysis (TGA; TA-2050) at a heating rate of 10 °C/min under nitrogen. Melting point (mp) was determined by means of DSC. Elemental analyses were performed by National Center for Inter-University Research Facilities, Seoul National University. (Elemental analyzer; CE Instruments Flash EA 1110). Absorption spectra were measured using a UV/Vis spectrophotometer (K-MAC, Spectraview-2000). Molecular weight of the polymer was measured gel permeation chromatography (GPC, Futecs, NS2001) using tetrahydrofuran (THF) as a solvent and calibrated against polystyrene standards.. PL and EL spectra were obtained using a CCD detector (Princeton Instruments, SPEC-10) with monochromator (Acton Research Co.

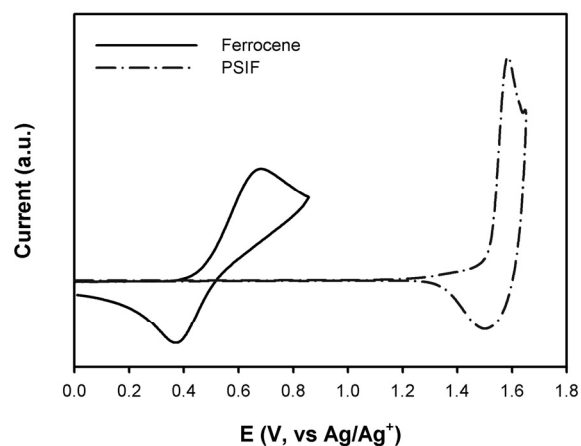
SpectraPro-300i) and excitation sources were a xenon lamp with monochromator (Acton Research Co. SpectraPro-150) and Keithley 237 Source Measurement Unit, respectively.

**Device Fabrication.** The polymer LEDs were fabricated with the structure of ITO / PEDOT (60 nm) / emitting polymer (60 nm) / LiF (5 nm) / Ca (10 nm) / Ag . The glass substrate pre-coated with indium-tin-oxide was cleaned in an ultrasonic bath with acetone, followed by boiling 2-propanol. Surface treatment was carried out by exposing the ITO to UV-ozone. The hole injecting poly(3,4-ethylene dioxythiophene) (PEDOT) layer was spin coated on the ITO with a thickness of 60 nm and baked at 200 °C for 10 min on a hot plate. A 1 wt % solution of polymer in a toluene was spin coated on the PEDOT layer as an emitting layer and baked at 200 °C for 1 hr. LiF and Ca were deposited as the cathode through a shadow mask by thermal evaporation. Finally, a Ag layer was deposited as a protecting layer.



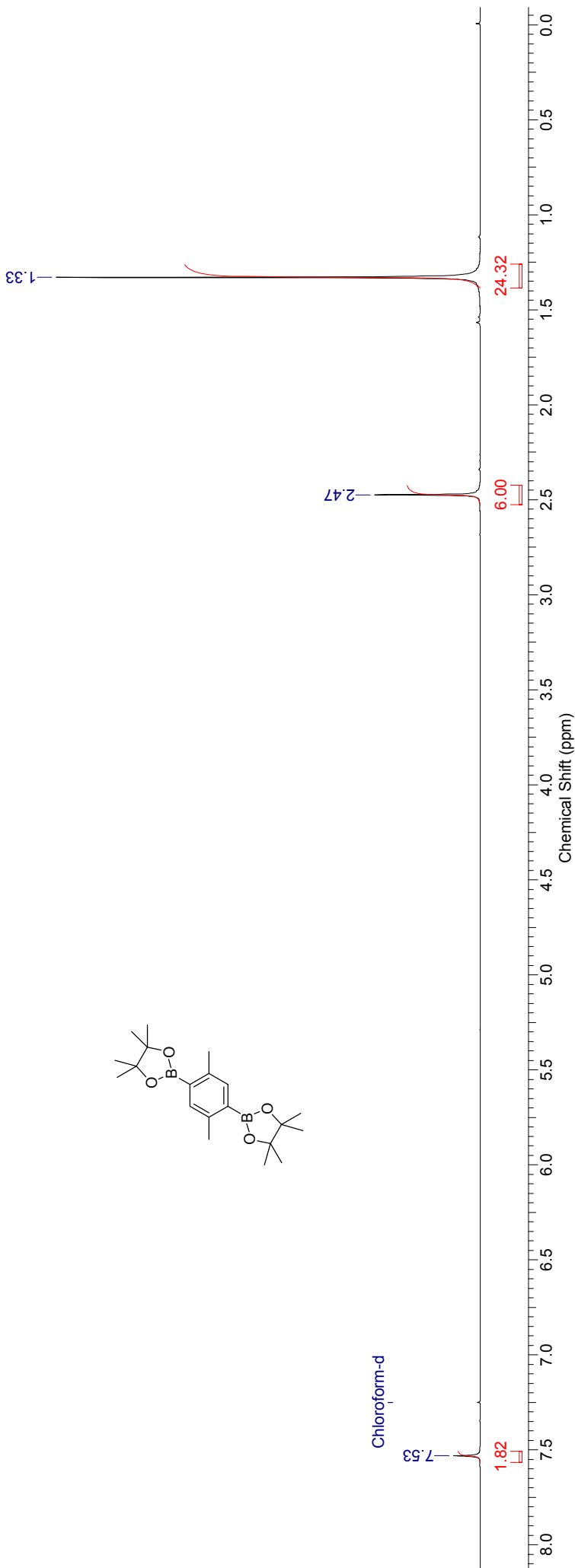
**Figure S1.** Absorption (solid line) and PL spectra of PSIF in film (long dash line) and in dichloromethane solution (dash dot line).

Cyclic voltammetry (CV) measurements were conducted in a 0.1 M  $\text{Bu}_4\text{NClO}_4$  acetonitrile solutions using a potentiostat (Eco Chemie, AUTOLAB) at a scan rate of 50 mV/sec at room temperature. To estimate the energy levels of polymers from the vacuum energy level, ferrocene/ferrocenium ( $\text{Fc}/\text{Fc}^+$ ) redox couple (4.8 eV from vacuum energy level) was used as a calibration reference.



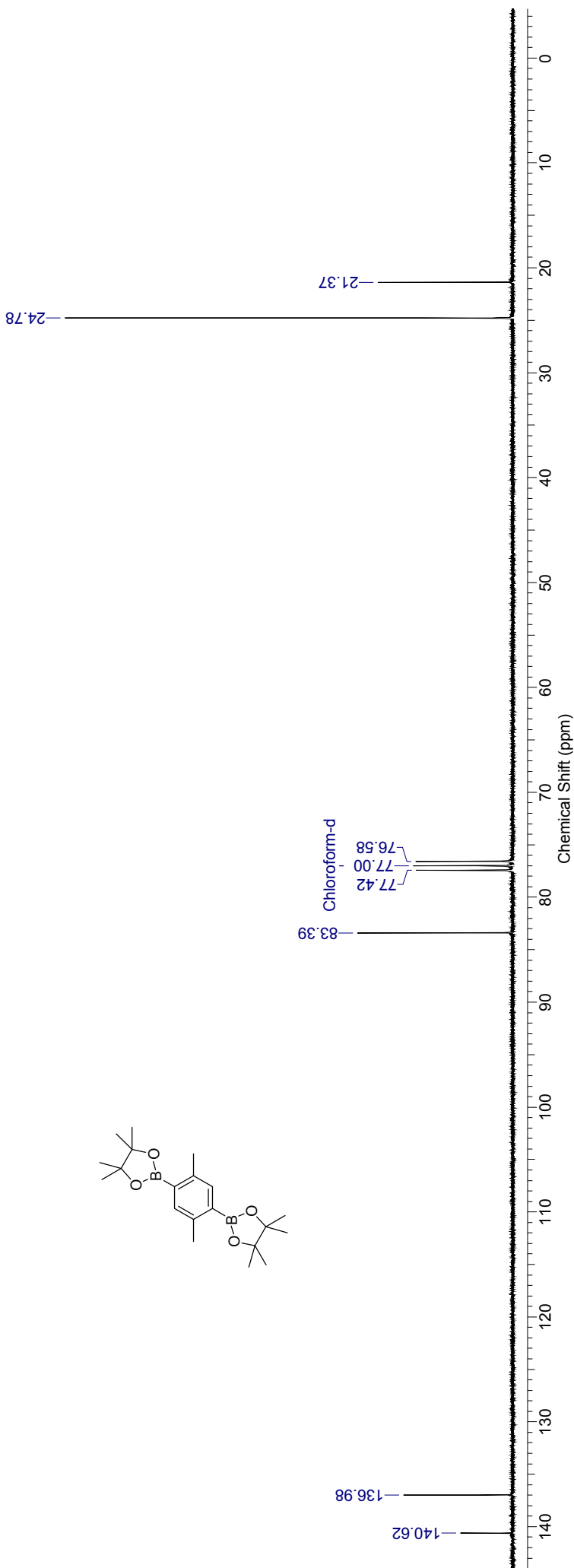
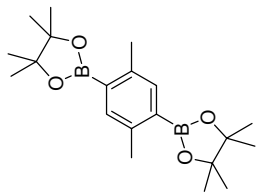
**Figure S2.** Cyclic voltammogram of PSIF. CV curve of PSIF (dash dot line) and ferrocene (solid line)

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<b>Solvent</b>	CHLOROFORM-D	<b>Number of Transients</b>	16
		<b>Sweep Width (Hz)</b>	6009.62
		<b>Temperature (degree C)</b>	0.000



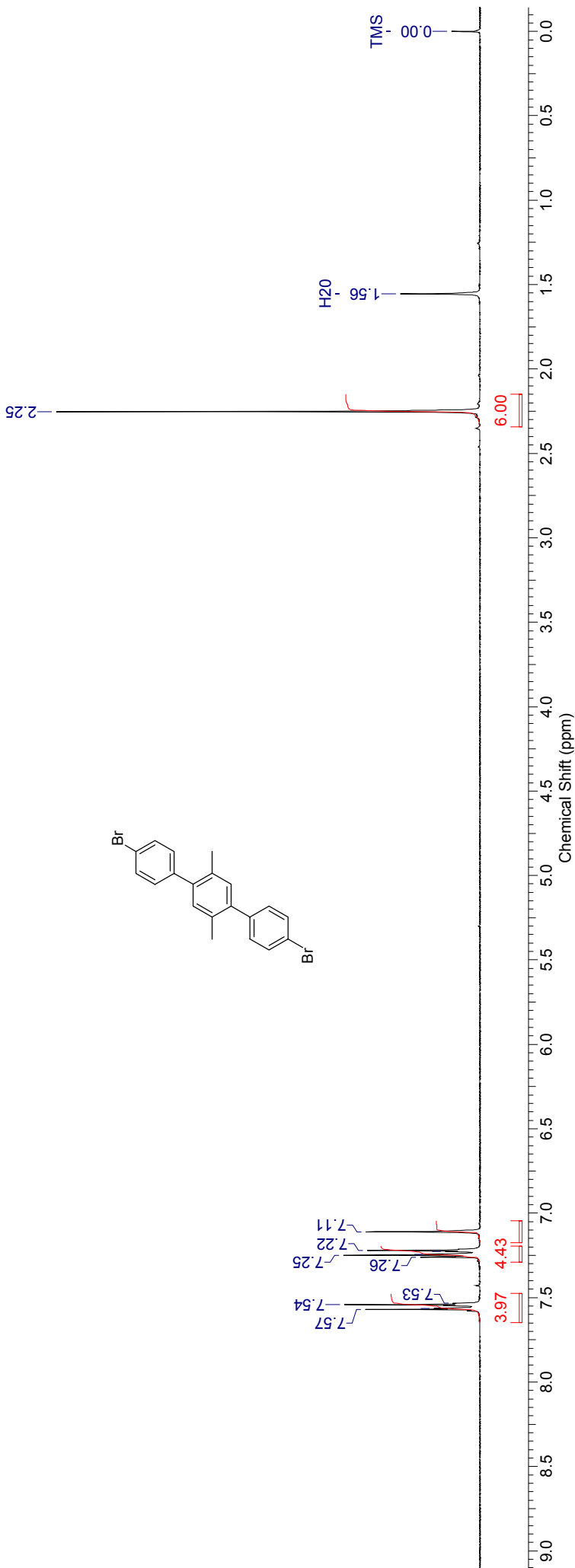
No.	(ppm)	(Hz)	Height
1	1.33	399.4	1.0000
2	2.47	743.3	0.2489
3	7.53	2262.4	0.0628

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		<b>Sweep Width (Hz)</b>	20366.60
		<b>Temperature (degree C)</b>	0.000



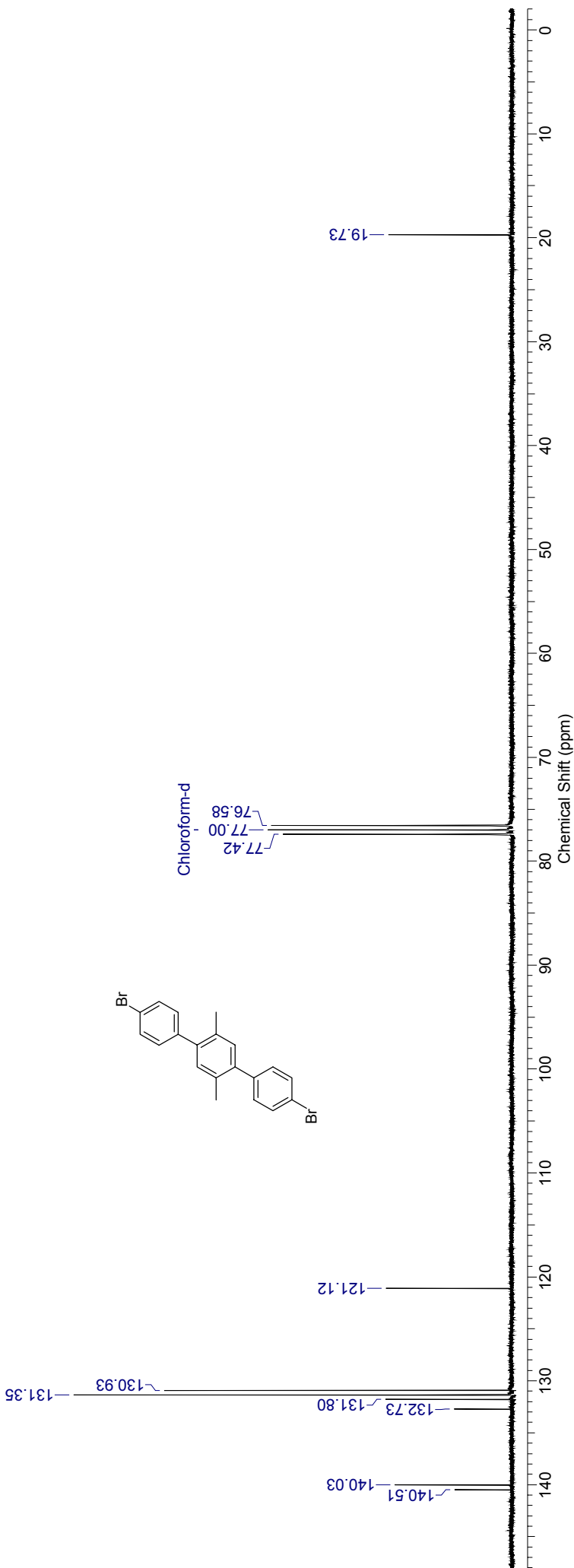
No.	(ppm)	(Hz)	Height
1	21.37	1612.6	0.3020
2	24.78	1869.7	1.0000
3	76.58	5777.6	0.2163
4	77.00	5809.7	0.2222
5	77.42	5841.7	0.2160
6	83.39	6292.0	0.3472
7	136.98	10335.5	0.2442
8	140.62	10609.9	0.1172

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		<b>Points Count</b>	32768
		<b>Temperature (degree C)</b>	0.000



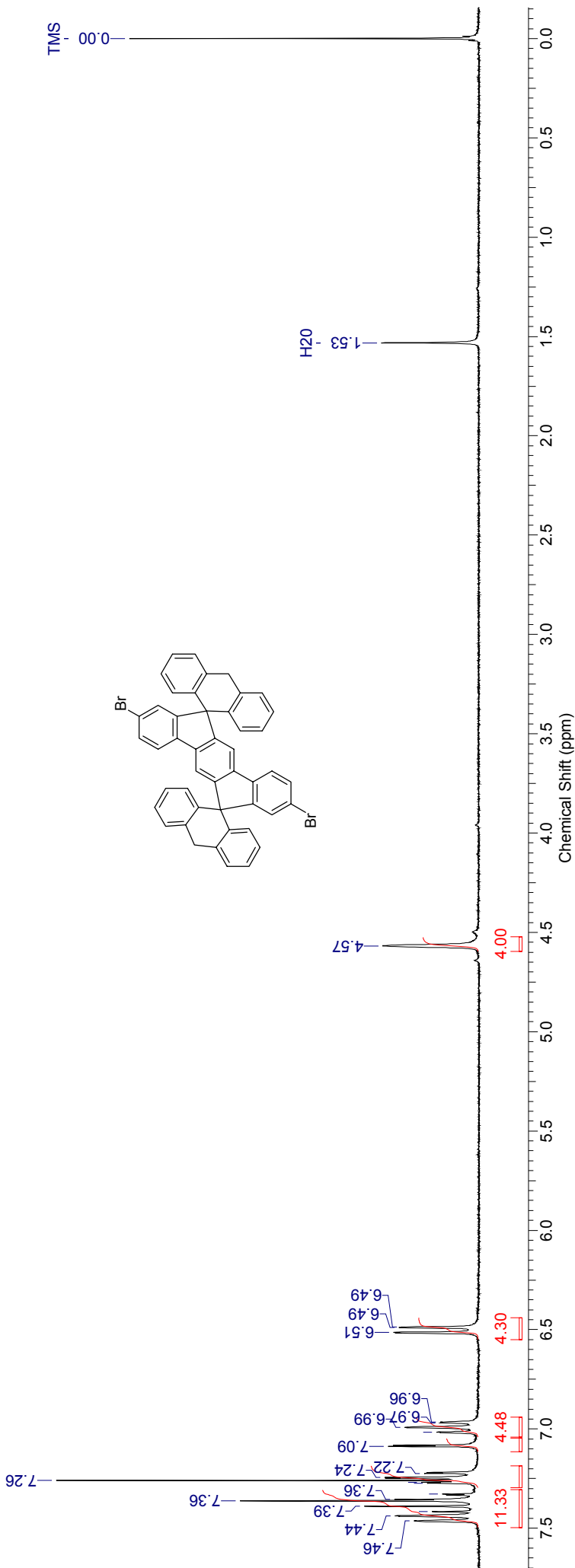
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1	-0.00	-0.0	0.0673	8	7.25	2177.7	0.3223
2	1.56	467.1	0.1878	9	7.26	2181.0	0.1410
3	2.25	676.6	1.0000	10	7.53	2263.2	0.0639
4	7.11	2136.1	0.2691	11	7.54	2265.6	0.3196
5	7.22	2169.1	0.2659	12	7.55	2267.6	0.0864
6	7.23	2171.1	0.0817	13	7.56	2272.2	0.1089
7	7.24	2175.7	0.1230	14	7.57	2274.0	0.2705

<b>Acquisition Time (sec)</b>	3.2178	<b>Comment</b>	70720-terphenyl-br-c	<b>Date</b>	Sat Jul 10 07:42:54 2004
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<b>Solvent</b>	CHLOROFORM-D	<b>Sweep Width (Hz)</b>	20366.60	<b>Temperature (degree C)</b>	0.000



No.	(ppm)	(Hz)	Height
1	19.73	1488.3	0.2807
2	76.58	5777.6	0.5484
3	77.00	5809.7	0.5562
4	77.42	5841.3	0.5223
5	121.12	9138.7	0.2869
6	130.93	9878.3	0.7931
7	131.35	9910.6	1.0000
8	131.80	9944.2	0.2884
9	132.73	10014.4	0.1316
10	140.03	10565.1	0.2667
11	140.51	10601.2	0.1298

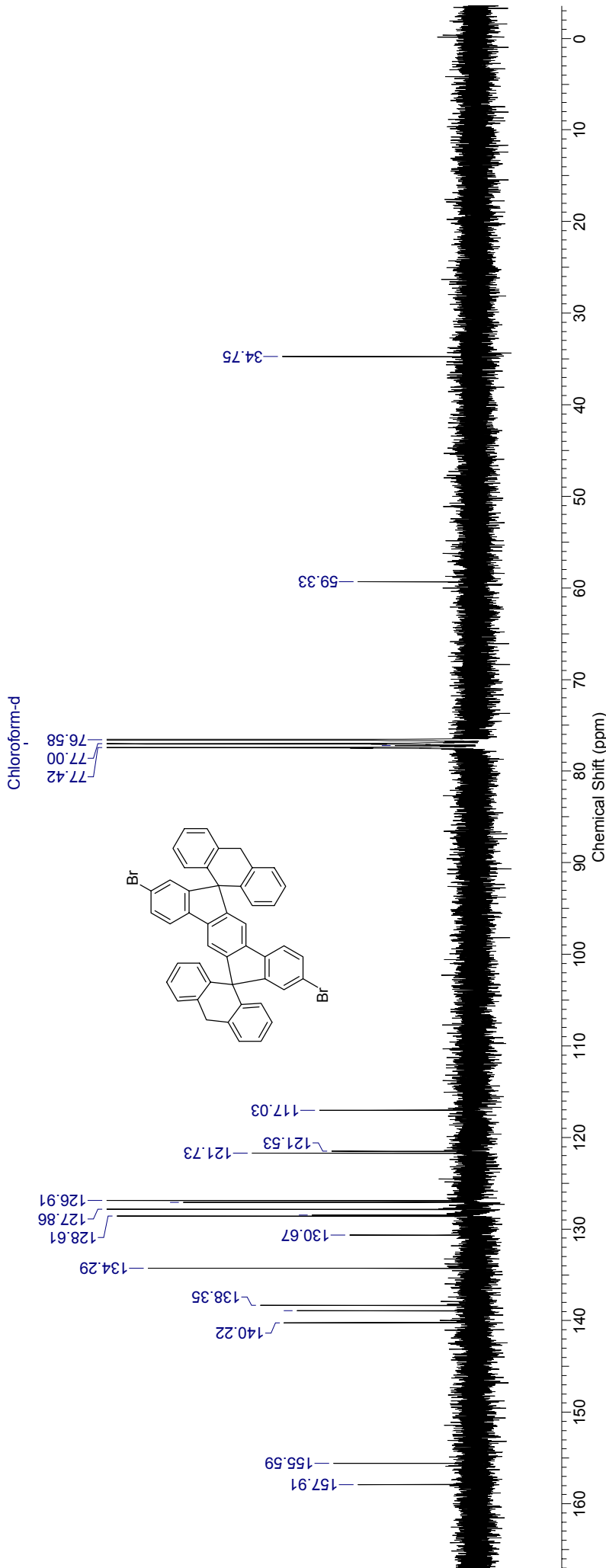
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		<b>Sweep Width (Hz)</b>	6009.62
		<b>Frequency (MHz)</b>	300.40
		<b>Points Count</b>	32768
		<b>Temperature (degree C)</b>	0.000



No.	(ppm)	Height	(Hz)	No.	(ppm)	Height	(Hz)
21	7.33	0.0984	2107.3	22	7.33	0.2009	2203.1
22	7.33	0.2009	2203.1	23	7.36	0.2128	2209.5
23	7.36	0.2128	2209.5	24	7.36	0.1215	2211.9
24	7.36	0.1215	2211.9	25	7.39	0.1270	2219.7
25	7.39	0.1270	2219.7	26	7.42	0.2208	2228.0
26	7.42	0.2208	2228.0	27	7.44	0.2196	2234.1
27	7.44	0.2196	2234.1	28	7.46	1.0000	2241.8
28	7.46	1.0000	2241.8				

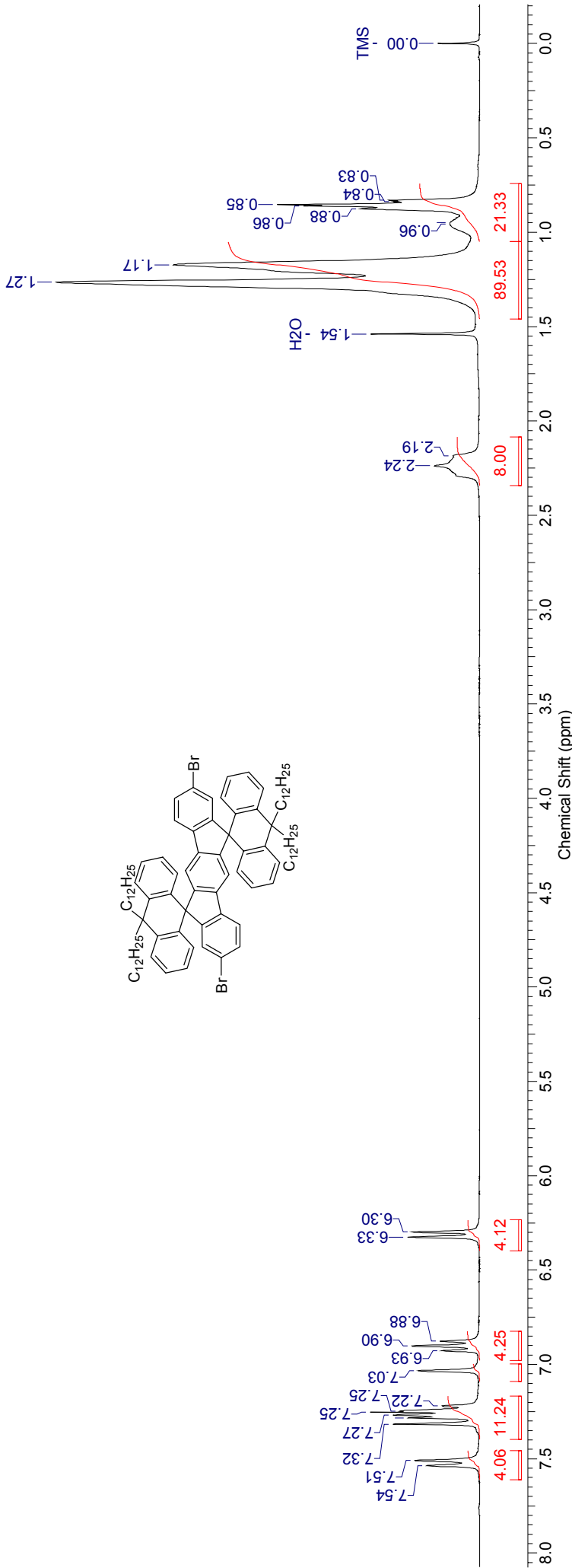
No.	(ppm)	Height	(Hz)	No.	(ppm)	Height	(Hz)
11	7.02	0.8262	2107.3	16	7.24	0.2005	1956.6
12	7.08	0.2287	2127.3	17	7.25	0.1963	1957.7
13	7.09	0.2259	2129.0	18	7.26	0.0888	2092.1
14	7.22	0.1813	2168.8	19	7.27	0.0905	2093.6
15	7.22	0.1875	2170.0	20	7.27	0.1732	2184.7
16	7.24	0.2005	2176.1				
17	7.25	0.1963	2177.4				
18	7.26	0.0888	2180.9				
19	7.27	0.0905	2183.6				
20	7.27	0.1732	2184.7				

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		<b>Points Count</b>	65536
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No.	(ppm)	(Hz)	Height	(ppm)	(Hz)	Height
1	34.75	2622.0	0.0984	12	127.86	9647.1
2	59.33	4476.4	0.0596	13	128.50	9695.3
3	76.58	5777.6	0.9574	14	128.61	9703.3
4	77.00	5809.6	1.0000	15	130.67	9859.4
5	77.21	5825.2	0.0406	16	134.29	10132.2
6	77.42	5841.7	0.9725	17	138.35	10438.3
7	117.03	8829.8	0.0792	18	138.94	10482.8
8	121.53	9169.4	0.0732	19	140.22	10579.4
9	121.73	9184.7	0.1140	20	155.59	11739.5
10	126.91	9575.0	0.1944	21	157.91	11914.5
11	127.09	9588.7	0.1492			

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2	0.83	249.6	0.2143	11	1.54	462.4	0.2577
3	0.84	251.6	0.2031	12	2.19	656.4	0.0630
4	0.85	256.6	0.4776	13	2.24	672.4	0.1061
5	0.86	258.8	0.4155	14	6.30	1892.0	0.1593
6	0.88	263.0	0.2812	15	6.33	1900.1	0.1701
7	0.95	284.8	0.0694	16	6.88	2065.9	0.0925
8	0.96	287.2	0.0704	17	6.90	2073.6	0.1598
9	1.17	352.1	0.7222	18	6.93	2080.9	0.0894
				19	7.03	2112.8	0.1445
				20	7.22	2168.8	0.0883
				21	7.25	2176.8	0.1877
				22	7.25	2179.0	0.2583
				23	7.27	2183.6	0.2044
				24	7.28	2187.8	0.1691
				25	7.32	2198.1	0.2039
				26	7.51	2255.9	0.1531
				27	7.54	2263.9	0.1248



