

# **Transition of Ionic Liquid [bmim][PF<sub>6</sub>] from liquid to High-Melting- Point Crystal When Confined in Multi-Walled Carbon Nanotubes**

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## **Supporting Information**

### **Experimental details**

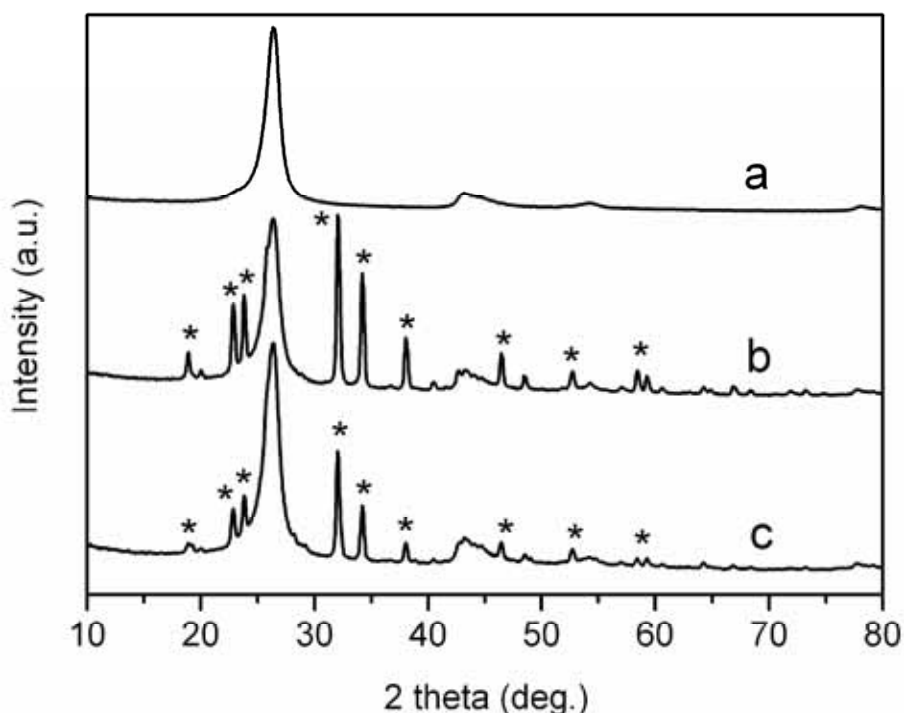
Commercially available MWNTs (purity >95%, length >5  $\mu\text{m}$ , the average internal and external diameters are about 5-10 nm and 40-60 nm, respectively. CVD method, Shenzhen Nanotech Port Co., Ltd, China) were purified and opened via sonication in a mixture of concentrated sulfuric and nitric acids (3:1, 98% and 70%, respectively) at 50 °C.<sup>1</sup> The resultant solid was washed thoroughly with deionized water until the pH was 7.0, and tip opened short nanotubes in length between 400 to 600 nm were obtained. The [bmim][PF<sub>6</sub>] was prepared by the method described previously.<sup>2,3</sup> According to the methods reported by Sloan *et al.*<sup>2,3</sup> for improving the filling yield of SWNTs incorporating species, herein, an optimum procedure including annealing opened carbon nanotubes and incubating at 90 °C under vacuum was employed. In a typical filling experiment, 186.5 mg opened MWNTs was put into a two necked flask (one of the necks was sealed by a rubber stopper and the other was

connected with a high-vacuum line). The flask was broiled by gas burner for 4 hrs under vacuum to draw out the gas inside MWNTs. Then 20 mL [bmim][PF<sub>6</sub>] was transferred into the flask through a syringe and the mixture was ultrasonically vibrated for 3 hrs at 90 °C, to fill the opened MWNTs with the ionic liquid. The resultant mixture was cooled for 3 h to room temperature. The filled samples were separated from the mixture by centrifugation and purified by six cycles of washing with methanol and filtration to completely remove the absorbed [bmim][PF<sub>6</sub>] from the nanotube surface.<sup>4</sup> The final product (named as IL@MWNTs for simplicity) was obtained by overnight drying under high vacuum. To further investigate the filling behavior of [Bmim][PF<sub>6</sub>] in the interior channel of MWNTs, the mixed solution with methanol (v/v=1:1) was introduced into the cavities of MWNTs in a similar manner and the final product was named as IL/MeOH@MWNTs.

## Characterizations

The X-ray diffraction (XRD) measurements were carried out on an X'Pert Pro diffractometer operated at 40 kV and 40 mA with Cu K $\alpha$  radiation. High resolution transmission electron microscopy (HRTEM) images were taken with a JEOL JEM2011 electron microscope operating at 200 kV. The samples for TEM observation were prepared by dispersing the filled nanotubes in ethanol and drying a few drops on a carbon-coated copper grid, then allowing them to dry in a desiccator. Melting points of the encapsulated [bmim][PF<sub>6</sub>] were determined by differential scanning calorimetry (DSC-822e, Mettler-Toledo Corp.). The sample was scanned from 223 K to 573 K at a programmed heating rate of 10 °C/min, using indium to calibrate the temperature and heat flow of the DSC device.

## XRD Patterns



**Figure S-1.** X-ray diffraction patterns of opened MWNTs (a), IL@MWNTs (b), and IL/MeOH@MWNTs (c). (\* indicates the peaks of [bmim][PF<sub>6</sub>] crystals encapsulated in MWNTs)

Figure S-1 shows the XRD patterns of the opened MWNTs and the filling products. Compared with the opened MWNTs, many new peaks appear in the IL@MWNTs and IL/MeOH@MWNTs samples (marked by asterisk). The diffraction peaks appearing at  $2\theta=18.9^\circ$ ,  $22.9^\circ$ ,  $23.8^\circ$ ,  $32.1^\circ$ ,  $34.2^\circ$ ,  $38.1^\circ$ ,  $46.5^\circ$ ,  $52.7^\circ$ , and  $58.4^\circ$  should correspond to different crystal planes of [bmim][PF<sub>6</sub>] inside MWNTs. However, the peaks calculated from the CIF file of the low-temperature crystal structure of [bmim][PF<sub>6</sub>] reported by Choudhury, et al.<sup>5</sup> are at  $2\theta = 10.41$ ,  $11.08$ ,  $11.64$ ,  $11.82$ ,  $13.11$ ,  $15.11$ ,  $16.42$ ,  $16.85$ ,  $17.82$ ,  $19.05$ ,  $19.46$ ,  $19.80$ ,  $20.78$ ,  $20.91$ ,  $21.15$ ,  $21.69$ ,  $22.26$ ,  $22.6$ ,  $23.18$ ,  $23.4$ ,  $23.84$ ,  $24.94$ ,  $25.24$ ,  $25.86$ ,  $26.2$ ,  $30.92$ ,  $36.1$ . It is indicated that a different solid phase was formed inside MWNTs in this study.

## References

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- (4) To check the efficiency of this purification method, 102 mg pristine MWNTs (length >5  $\mu\text{m}$ , the tips are closed) was mixed with 20 mL [bmim][PF<sub>6</sub>], then MWNTs were separated from the mixture by centrifugation and purified by six cycles of washing with methanol. Thermal gravity analysis of the as-obtained MWNTs showed no weight loss (Perkin-Elmer Pyris-1 series thermal analysis system, under a flowing nitrogen atmosphere at a scan rate of 10 °C/min from 50 to 800 °C). FT-IR measurement (Nicolet Avater-360) showed no existence of [bmim][PF<sub>6</sub>] in the purified sample, indicating the centrifugation method can remove the absorbed [bmim][PF<sub>6</sub>] completely.
- (5) Choudhury, A. R.; Winterton, N.; Steiner, A.; Cooper, A. I.; Johnson, K. A. *J. Am. Chem. Soc.* **2005**, 127, 16792.