

The Conformational Manifold of α -Aminoisobutyric Acid (Aib) Containing Alanine-Based Tripeptides in Aqueous Solution Explored by Vibrational Spectroscopy, Electronic Circular Dichroism Spectroscopy, and Molecular Dynamics Simulations.

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Supporting Information

Materials and Method.

NMR Spectroscopy. Nuclear Magnetic Resonance spectra were recorded at 300 MHz (^1H)/75 MHz (^{13}C) on a Varian Gemini-300 and NMR spectrometers. ^1H and ^{13}C chemical shifts (δ) are given in parts per million (ppm) using residual protonated solvent as an internal standard. Coupling constants are given in Hertz (Hz). The following abbreviations are used: s = singlet, d = doublet, t = triplet, m = multiplet. Low resolution mass spectral data were recorded on a Micromass LCT (TOF MS ES+) instrument.

Melting points were determined on a Bausch and Lomb hot stage. Analytical reversed-phase HPLC was performed on a Phenomenex C_{18} Jupiter column (4.6 x 250 mm). Preparative reversed phase HPLC was performed on a Phenomenex Gemini C_{18} column (22 x 250 mm). Separations were achieved using linear gradients of buffer B in A (A = aqueous 0.045 % HCl; B = 90 % CH_3CN , 10 % H_2O , 0.045 % aqueous HCl) at a flow rate of 1 mL/min (analytical) and 15 mL/min (preparative). Abbreviations: MeCN, acetonitrile; HCl, hydrogen chloride; DCM, dichloromethane; petrol, petroleum spirit (bp 40 – 60 °C); EtOAc, ethyl acetate; DIEA, diisopropylethyl amine; AcOH, acetic acid; MeOH, methanol; Et_2O , diethyl ether; DMSO, dimethylsulfoxide; DMF, *N,N*-dimethylformamide; HBTU, *O*-benzotriazol-1-yl-*N,N,N',N'*-tetramethyluronium hexafluorophosphate; HATU, *O*-(7-azabenzotriazol-1-yl)-*N,N,N',N'*-tetramethyluronium hexafluorophosphate.

Synthesis of Ac-Aib-OH. H-Aib-OH (1.03 g, 10 mmol) was dissolved with stirring in boiling acetic acid to give a 10% w/v solution. After allowing the solution to cool slightly, acetic anhydride (1.53 g, 15 mmol, 1.5 equiv.) was added in portions over a 10-minute period. The mixture was refluxed for 5-10 min, allowed to cool overnight, and the solvent evaporated to give a syrup, to which water (50 mL) was added and the solution

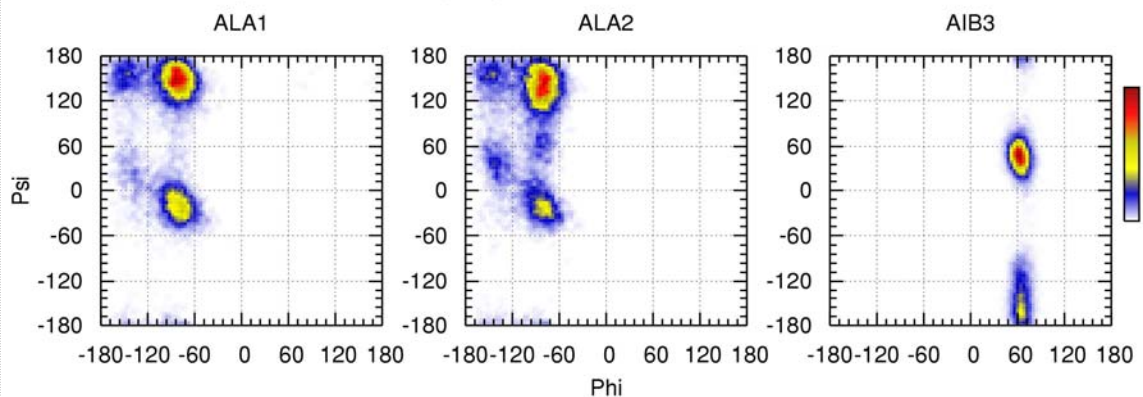
was re-evaporated. This process was repeated until a solid was obtained. The solid was washed with ice-water, then dried under vacuum. The solid, free from acetic acid was recrystallized from H₂O to give the desired product. (0.25 g, 40 %). ¹H NMR (300 MHz, d⁶-DMSO) δ 1.22 (s, 6H, C(CH₃)₂), 1.68 (s, 3H, CH₃CO), 7.92 (s, NH) 11.98(br s, 1H, OH); ¹³C NMR (75 MHz, d⁶-DMSO) δ 23.1, 25.6, 55.2, 169.2, 176.2. ES-MS *Mr* 145.2, calcd for C₆H₁₁NO₃, 145.1 (monoisotopic).¹

Synthesis of linear peptides. All linear peptides were chemically synthesised on chlorotrityl resin stepwise using Fmoc protecting groups and *in situ* HBTU activation protocols as previously described.^{2,3} Coupling efficiencies were determined by the quantitative ninhydrin test⁴ and recoupled where necessary to obtain >99.5% efficiency. The linear peptides were cleaved from resin using 95% TFA with 5% H₂O. The product was purified by reverse-phase HPLC. No TFA was used in the preparative HPLC. The procedure yielded the following results. Ac-Ala-Ala-Aib-OH: Yield after cleavage and purification was 142 mg (71%); ES-MS *Mr* 287.2, calculated for C₁₂H₂₁N₃O₅, 287.2 (monoisotopic); Ac-Ala-Aib-Ala-OH: Yield after cleavage and purification was 156 mg (78%); ES-MS *Mr* 287.3, calculated for C₁₂H₂₁N₃O₅, 287.2 (monoisotopic); Ac-Aib-Ala-Ala-OH: Yield after cleavage and purification was 171 mg (86%); ES-MS *Mr* 287.3, calculated for C₁₂H₂₁N₃O₅, 287.2 (monoisotopic).

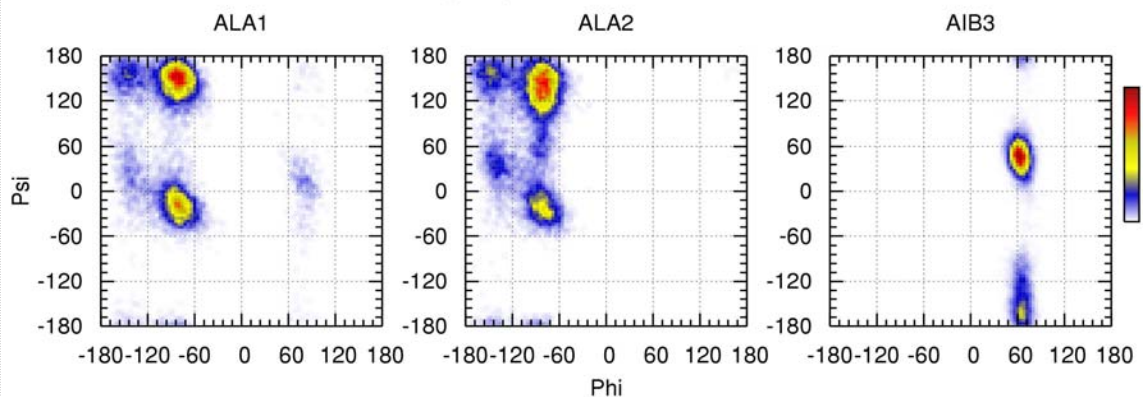
General method for methyl ester formation. The linear peptides were dissolved in 2 mL EtOH and diazomethane in ether (20 mL of 0.4 M solution) was added dropwise at room temperature. The solution was left standing for 60 minutes, 1 mL of HOAc added to

neutralize the excess diazomethane and solvents removed under vacuum. The residue was taken up in 3 mL aqueous acetonitrile (50%) and lyophilized. Preparative HPLC gave the desired products. No TFA was used in the preparative HPLC. The procedure yielded the following results. Ac-Ala-Ala-Aib-OCH₃: Yield after cleavage and purification was 103 mg (80%); ES-MS *Mr* 301.2, calcd for C₁₃H₂₃N₃O₅, 301.2 (monoisotopic); Ac-Ala-Aib-Ala-OCH₃: Yield after cleavage and purification was 122 mg (87%); ES-MS *Mr* 301.3, calculated for C₁₃H₂₃N₃O₅, 301.2 (monoisotopic); Ac-Aib-Ala-Ala-OCH₃: Yield after cleavage and purification was 147 mg (91%); ES-MS *Mr* 301.3, calculated for C₁₃H₂₃N₃O₅, 301.2 (monoisotopic).

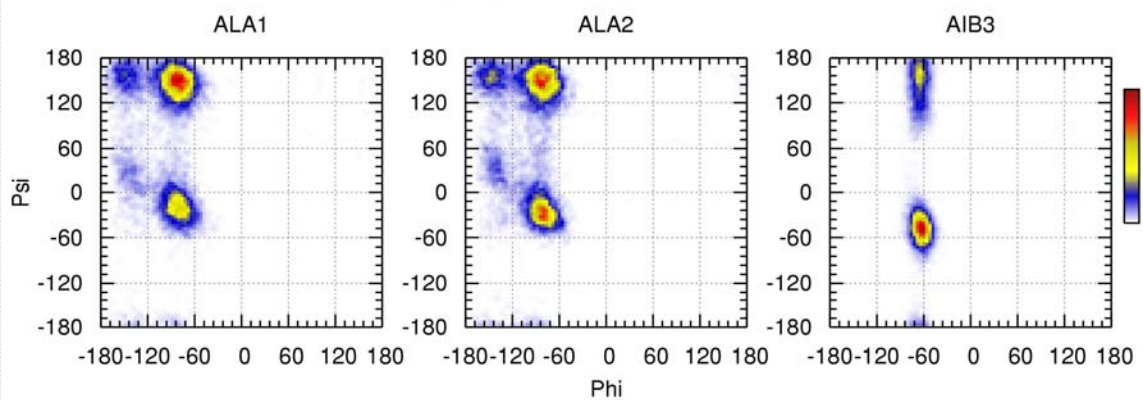
AAB, 300K, Alpha helix, $\Phi(\text{AIB})=+58$



AAB, 300K, Beta strand, $\Phi(\text{AIB})=+58$



AAB, 300K, Alpha helix, $\Phi(\text{AIB})=-58$



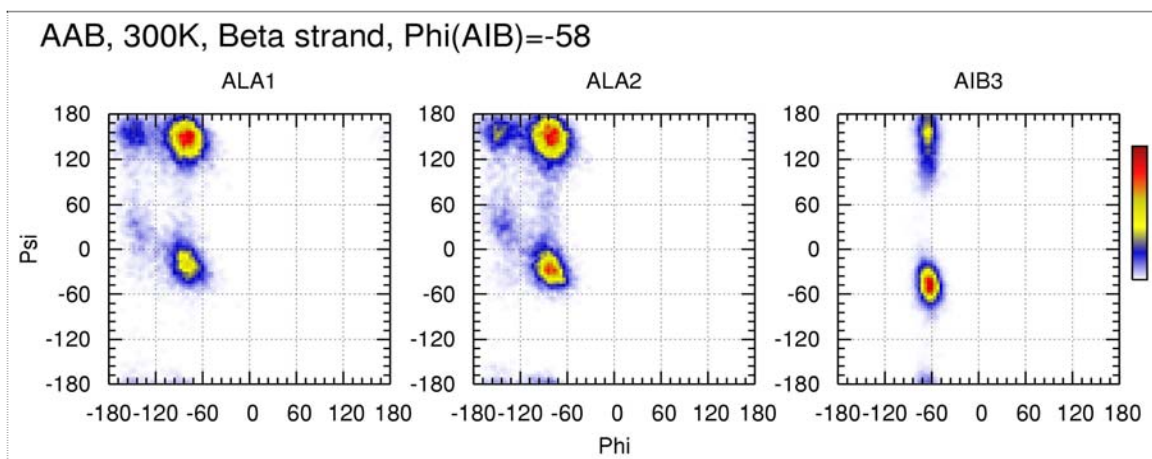
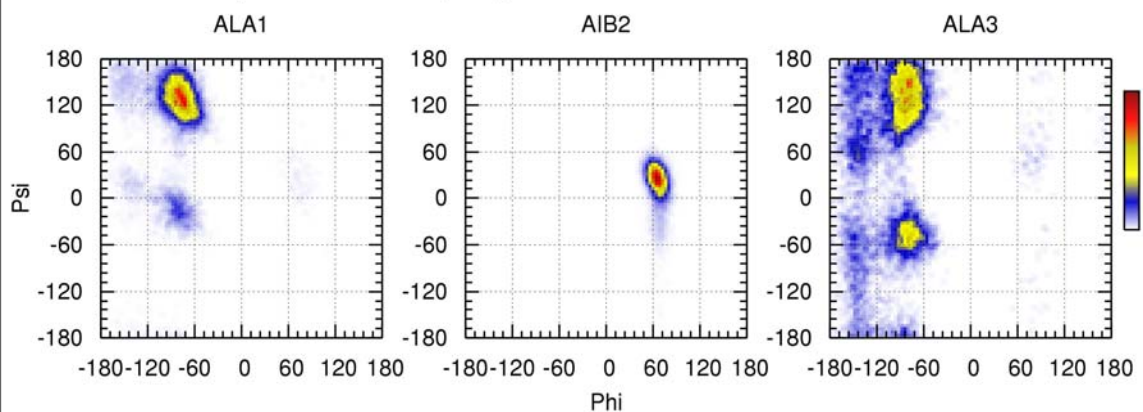
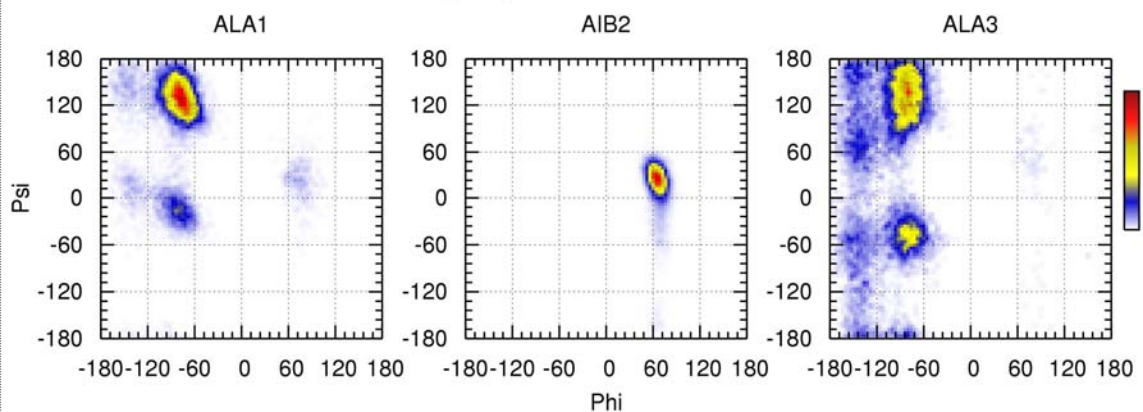


Figure S1: Summary Ramachandran plots of four Ac-AlaAlaAib-OMe (AAB) simulations at 300° K.

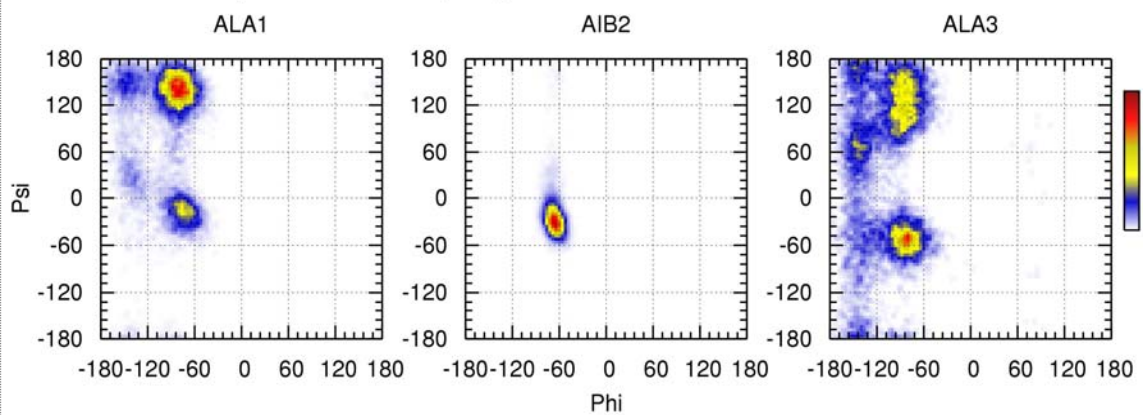
ABA, 300K, Alpha helix, $\Phi(\text{AIB})=+58$



ABA, 300K, Beta strand, $\Phi(\text{AIB})=+58$



ABA, 300K, Alpha helix, $\Phi(\text{AIB})=-58$



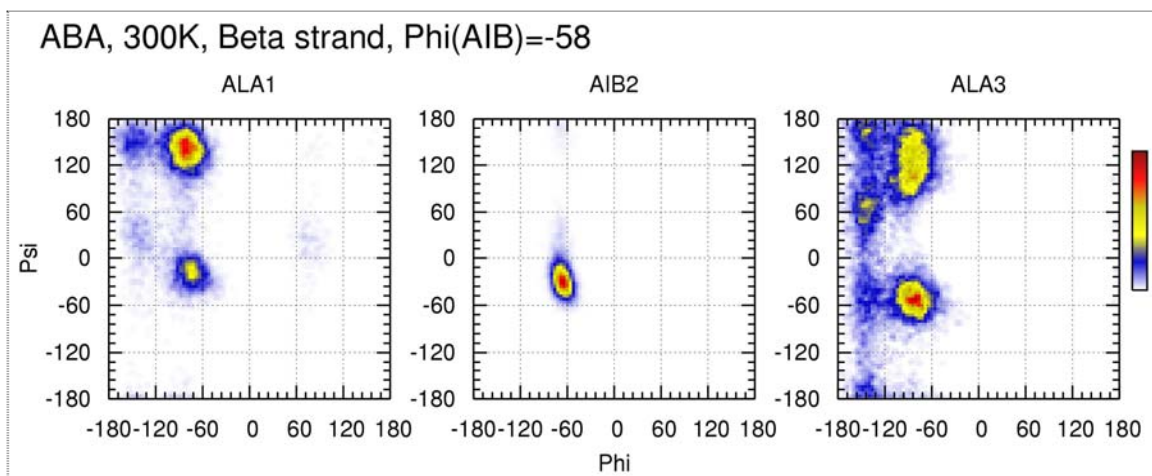
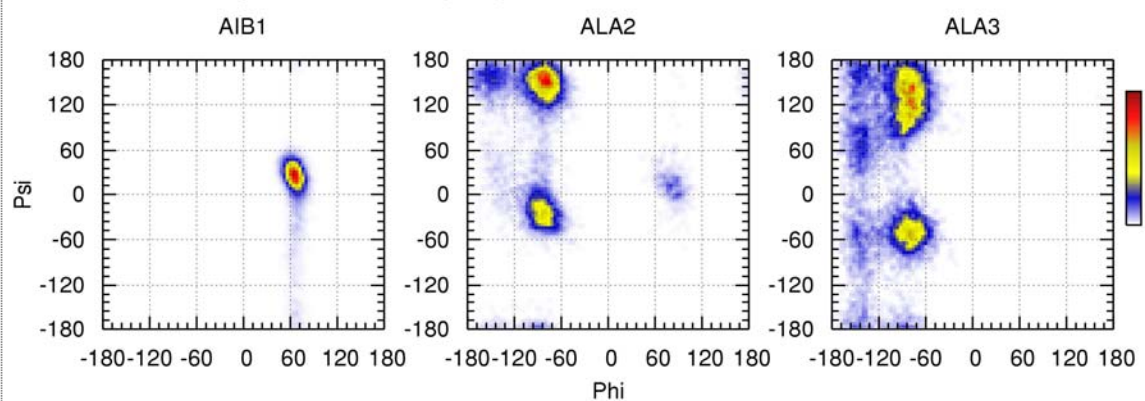
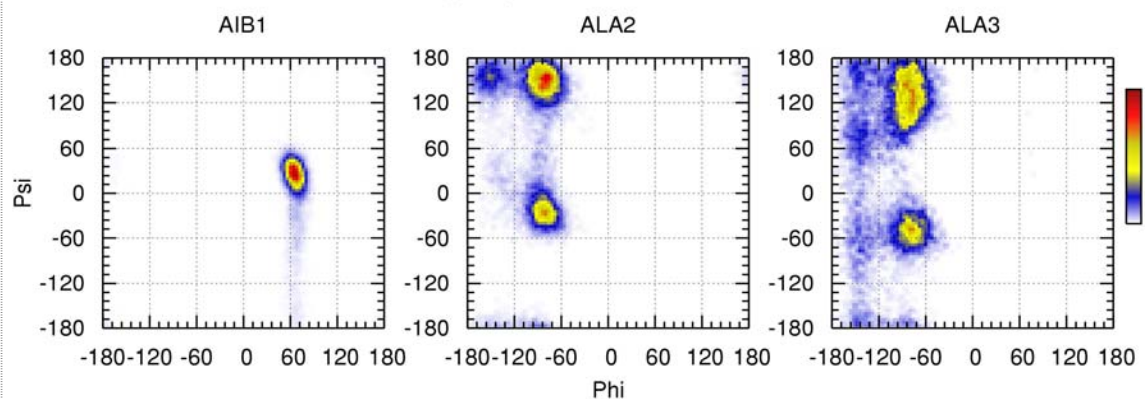


Figure I2: Summary Ramachandran plots of four Ac-AlaAibAla-OMe (ABA) simulations at 300° K.

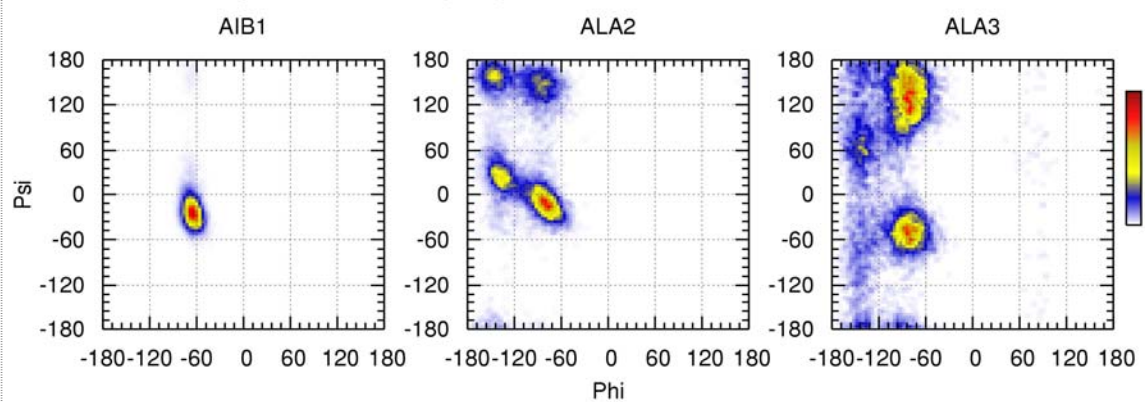
BAA, 300K, Alpha helix, $\Phi(\text{AIB})=+58$



BAA, 300K, Beta strand, $\Phi(\text{AIB})=+58$



BAA, 300K, Alpha helix, $\Phi(\text{AIB})=-58$



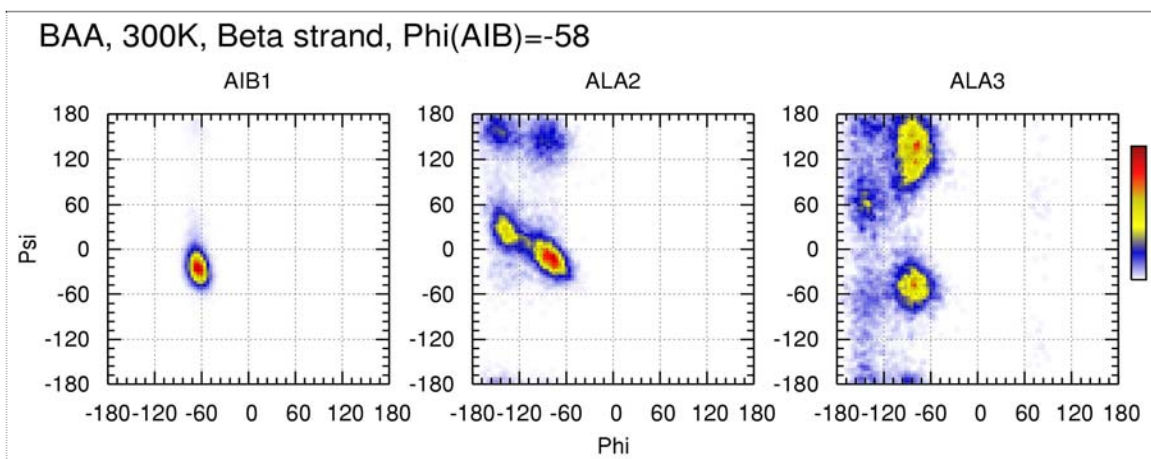


Figure 13: Summary Ramachandran plots of four Ac-AibAlaAla-OMe (BAA) simulations at 300° K.

References

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