

ACYL TRANSFER CATALYSIS WITH 1,2,4-TRIAZOLE ANION

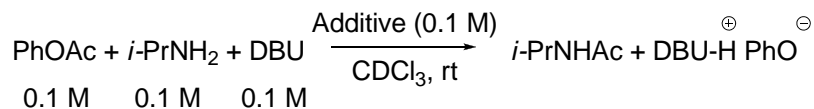
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SUPPORTING INFORMATION.

1. General. All reagents and solvents were obtained commercially and used as received. Stock solutions were prepared in volumetric flasks of specified volume by dissolving reagents in a solvent and bringing the volume to the mark. ^1H NMR and ^{13}C NMR spectra were recorded on a Unity 300 MHz Varian spectrometer. Flash column chromatography was performed over ICN Ecohrom silica gel (32-63 μm).

2. Catalytic activity screen

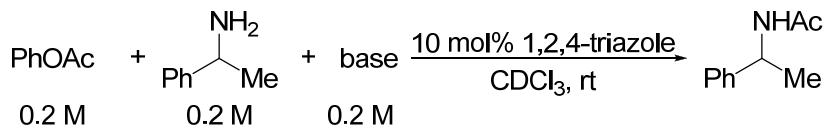


Stock solution A: 0.20 M phenyl acetate (126 μL , 1.00 mmol) and 0.20 M isopropylamine (86 μL , 1.00 mmol) in 5.0 mL of CDCl_3 .

Stock solution B: 0.20 M DBU (60 μL , 0.40 mmol) and 0.20 M additive (0.40 mmol **1**, **8-21**, or none for the blank) in 2.0 mL of CDCl_3 .

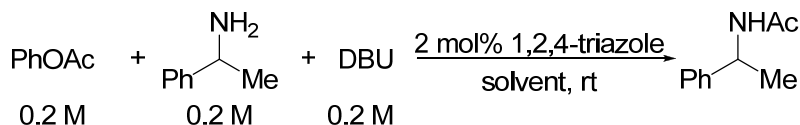
The test was carried out by mixing 0.50 mL aliquots of stock solutions **A** and **B** at room temperature and monitoring reaction progress by ^1H NMR. The time required to reach 50% conversion of isopropylamine into N-isopropylacetamide was recorded as $t_{1/2}$ in Table 1. In those cases when the conversion was lower than 50% after 3 days, $t_{1/2}$ was not determined.

3. Influence of base



Phenyl acetate (25 μL , 0.20 mmol), α -phenethylamine (25 μL , 0.20 mmol) and 0.20 mmol of a base (30 μL of DBU, or 28 μL of NEt_3 , or none for the blank) were added successively to 1.0 mL of 0.02 M stock solution of 1,2,4-triazole in CDCl_3 at room temperature. The reaction was progress monitored by ^1H NMR. The % conversion of α -phenethylamine into N-phenethylacetamide was recorded in Figure 3.

4. Influence of solvents on the aminolysis of phenyl acetate

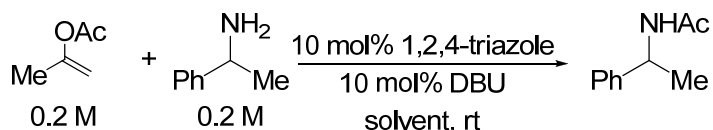


Stock solution C: 0.20 M 1,2,4-triazole (0.40 mmol, 28 mg) and 0.20 M DBU (0.40 mmol, 60 μ L) in 2 mL of CDCl_3 .

Stock solution D: 0.20 M phenyl acetate (50 μ L, 0.40 mmol), 0.20 M α -phenethylamine (51 μ L, 0.40 mmol) and 0.20 M DBU (0.40 mmol, 60 μ L) in 2.0 mL of a deuterated solvent (chloroform- d_3 , acetonitrile- d_3 , benzene- d_6 , DMSO- d_6 or THF- d_8).

The test was carried out by adding 20 μ L of Stock solution C into 1.0 mL of Stock solution D and monitoring the conversion of α -phenethylamine into N-phenethylacetamide by ^1H NMR at 10 min, 30 min, 1 h, 2 h, 6 h, and 24 h. The data were plotted in Figure 4.

5. Influence of solvents on the aminolysis of isopropenyl acetate

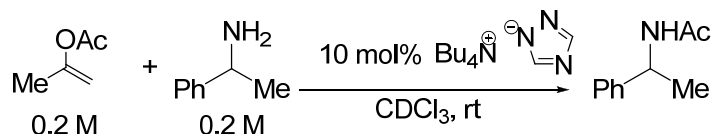


Stock solution E: 0.40 M 1,2,4-triazole (0.80 mmol, 55 mg) and 0.40 M DBU (0.80 mmol, 120 μ L) in 2 mL of CDCl_3 .

Stock solution F: 0.20 M isopropenyl acetate (44 μ L, 0.40 mmol) and 0.20 M α -phenethylamine (51 μ L, 0.40 mmol) in 2.0 mL of a deuterated solvent (chloroform- d_3 , acetonitrile- d_3 , benzene- d_6 , DMSO- d_6 or THF- d_8).

The test was carried out by adding 50 μ L of Stock solution E into 1.0 mL of Stock solution F and monitoring the conversion of α -phenethylamine into N-phenethylacetamide by ^1H NMR at 10 min, 30 min, 1 h, 2 h, 6 h, and 24 h. The data were plotted in Figure 5.

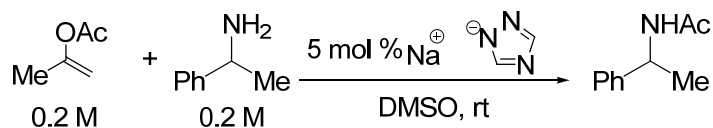
6. Catalytic activity of tetrabutylammonium triazolidine.



Catalyst reparation: A modified procedure found in a patent¹ was followed. A mixture of powdered solid NaOH (0.40 g, 10 mmol) and triazole (0.69 g, 10 mmol) in 5 mL of methanol was stirred for 30 min at room temperature. Solid Bu_4NBr (3.22 g, 10 mmol) was added and the stirring continued for 2 h. After addition of 100 mL of anhydrous Et_2O , NaBr precipitated and was filtered off. The filtrate was concentrated to dryness and held under high vacuum overnight.

Catalytic activity test: A 50 μ L aliquot of 0.40 M solution of tetrabutylammonium triazolidine in CDCl_3 was added to 1.0 mL of stock solution F (in CDCl_3) and the reaction progress was monitored by ^1H NMR. until 50% conversion (4.0 h). A control experiment with DBU triazolidine reached 50% conversion in 5.5 h.

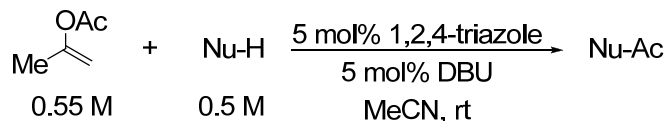
7. Catalytic activity of sodium triazolidine.



Catalyst preparation: A mixture of NaOH (10 mmol, 0.40 g) and 1,2,4-triazole (10 mmol, 0.69 g) in 5 mL of MeOH was stirred overnight and then evaporated to dryness. The evaporation residue was dried azeotropically with toluene and then kept overnight under high vacuum.

Catalytic activity test: A 50 μ L aliquot of 0.20 M solution of sodium triazolide in DMSO- d_6 was added to 1.0 mL of stock solution F (in DMSO- d_6) and the reaction progress was monitored by ^1H NMR. until 50% conversion (12 min). A control experiment with DBU triazolide reached 50% conversion in 45 min.

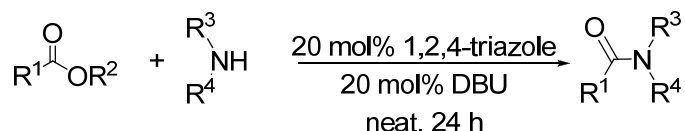
8. Preparative scale acetylation of substrates with isopropenyl acetate



General procedure: 2.0 mL of freshly prepared stock solution of 1,2,4-triazole (0.025 M) and DBU (0.025 M) in acetonitrile was added to 1.0 mmol of a substrate, followed by isopropenyl acetate (1.1 mmol, 121 μ L). After 10 h at room temperature, the solvent was removed under reduced pressure and the residue was diluted with Et₂O and passed through a plug of silica gel to give pure product upon concentration. Purification of acetanilide required the use of 2:1 Et₂O-hexane mixture as eluent. The results are summarized in Figure 7.

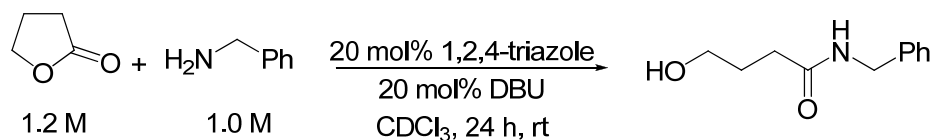
Enantiomeric purity of amides obtained from (*R*)-phenylglycine methyl ester and (*S*)-phenylalanine methyl ester was determined by chiral stationary phase HPLC using CHIRALCEL OD-H and CHIRALPAK AD-H analytical columns, respectively, (4.6x150 mm, Chiral Technologies, Inc.) with isopropanol/hexane eluent at 1 mL/min flow rate.

9. Aminolysis of unactivated acyclic esters²



General procedure: A reaction mixture consisting of an ester (1.2 mmol), an amine (1.0 mmol), 1,2,4-triazole (14 mg, 0.20 mmol) and DBU (30 μ L, 0.20 mmol) was stirred at a specified temperature for 24 h and then applied directly to a silica gel chromatographic column and eluted with Et₂O.

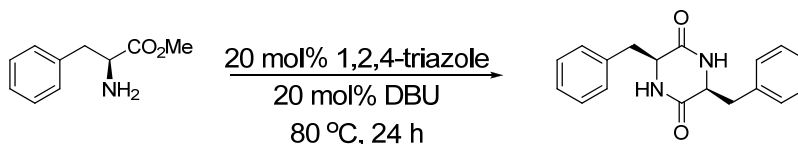
10. Aminolysis of γ -butyrolactone



A solution of γ -butyrolactone (1.2 mmol), benzylamine (1.0 mmol), 1,2,4-triazole (14 mg, 0.20 mmol) and DBU (30 μ L, 0.20 mmol) in 1.0 mL of CDCl₃ was stirred at room temperature for 24

h, concentrated and chromatographed with EtOAc-Et₂O 1:3 to give 125 mg (84% yield) of N-benzyl- γ -hydroxybutyramide.

11. Cyclocondensation of (L)-Phenylalanine methyl ester.



Commercially available (L)-phenylalanine methyl ester hydrochloride was converted into the free base by treating it with aqueous K₂CO₃ and extracting the mixture with CH₂Cl₂. A mixture of this material (180 mg, 1.0 mmol), 1,2,4-triazole (14 mg, 0.20 mmol) and DBU (30 μ L, 0.20 mmol) was heated without stirring in a vial placed in an 90 °C oil bath. After 3 h, the initially liquid mixture solidified completely. After 24 h, ¹H NMR analysis of the crude reaction mixture (in CF₃CO₂D) indicated essentially complete conversion to the expected product as a 95:5 mixture of *cis*- and *trans*-diastereomers. The solid mixture was broken up and suspended in 25 mL of hot ethanol. After cooling to rt, the suspension was filtered and the white powdery precipitate was rinsed with ethanol to give 116 mg (79% yield) of the product (97:3 *cis-trans* ratio).

¹ Sturm, E.; Eckhardt, W.; Gloor, B.; Nyfeler, R. US Patent 4,259,505, 1981.

² All reaction products obtained in this study have been previously reported. References to commercially unavailable compounds are provided below.

(a) **Methyl N-acetylphenylglycinate:** Fadnavis, N. W.; Reddy, N. P.; Bhalerao, U. T. *J. Org. Chem.* **1989**, *54*, 3218.

(b) **N-benzyl-octanamide:** Rolfe, A.; Probst, D. A.; Volp, K. A.; Omar, I.; Flynn, D. L.; Hanson, P. R. *J. Org. Chem.* **2008**, *73*, 8785.

(c) **N-benzyl-isobutyramide:** Bosch, I.; González, A.; Urpí, F.; Vilarrasa, J. *J. Org. Chem.* **1996**, *61*, 5638.

(d) **N-benzyl-lactamide:** Savinov, S. N.; Austin, D. J. *Org. Lett.* **2002**, *4*, 1415.

(e) **N-benzyl-4-hydroxybutanamide:** Decker, M.; Nguyen, T. T. H.; Lehmann, J. *Tetrahedron* **2004**, *60*, 4567.

(f) **N-cyclohexyl-octanamide:** Lücking, U.; Tucci, F. C.; Rudkevich, D. M.; Rebek, J. Jr. *J. Am. Chem. Soc.* **2000**, *122*, 8880.

(g) **N- α -phenethyl-octanamide:** Souto-Bachiller, F.; Bates, G. S.; Masamune, S. *J. Chem. Soc., Chem. Commun.* **1976**, 719.

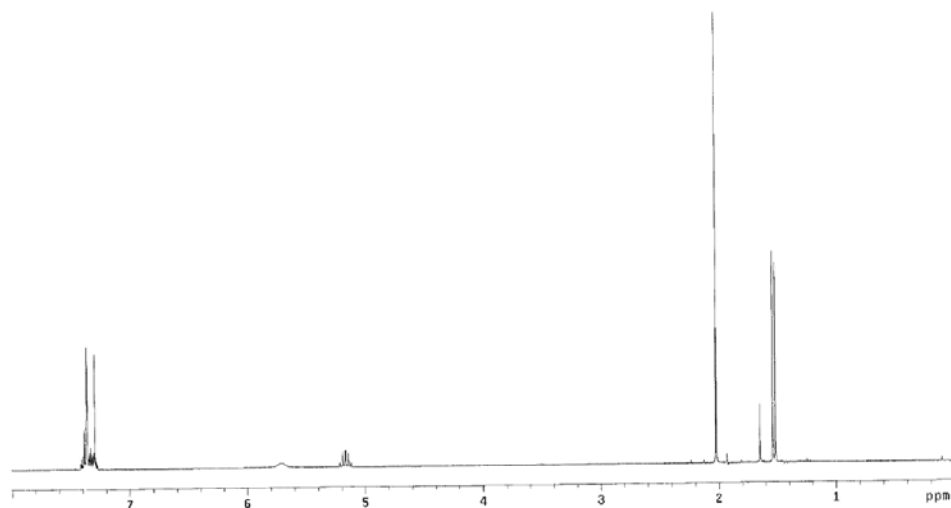
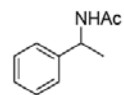
(h) **N-octanoylpiperidine:** Jensen, A. E.; Knochel, P. *J. Org. Chem.*, **2002**, *67*, 79.

(i) ***cis*-3,6-dibenzylpiperazine-2,5-dione (cyclo-Phe-Phe):** Donkor, I. O.; Sanders, M. L. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 2647.

STANDARD IN OBSERVE

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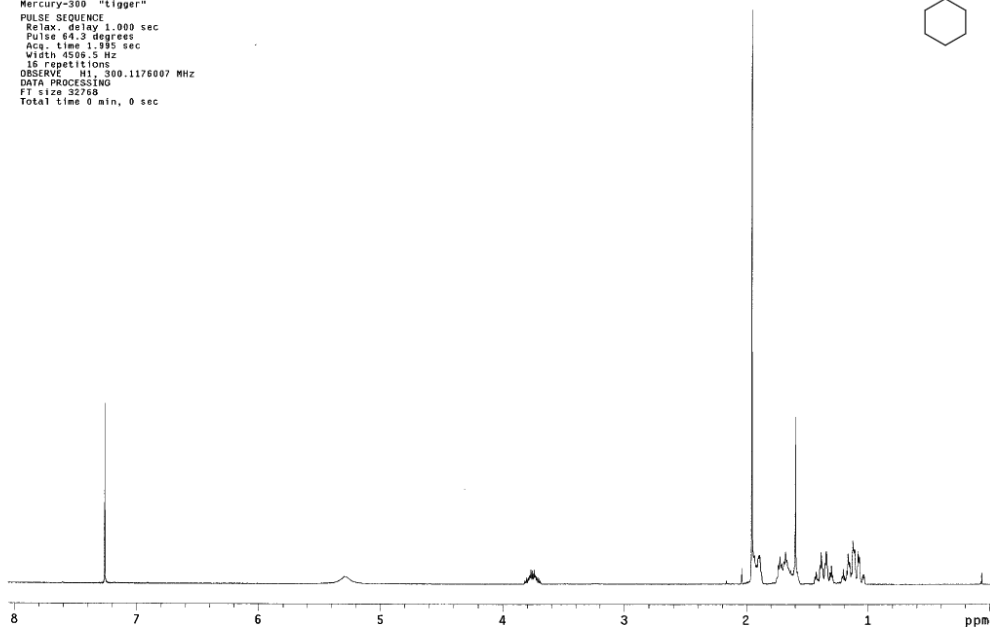
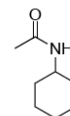
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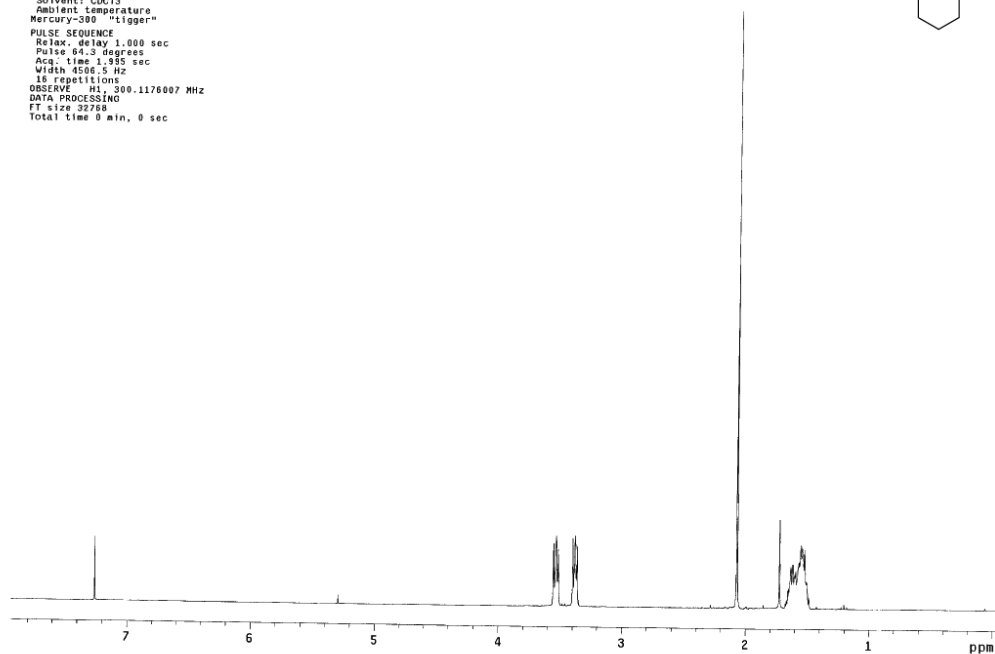
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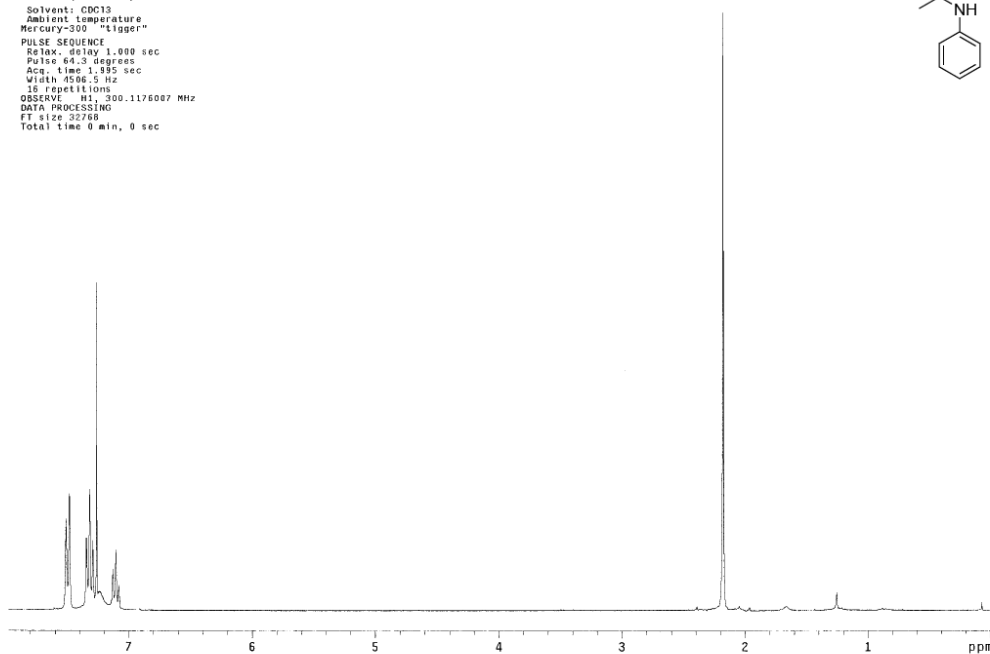
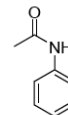
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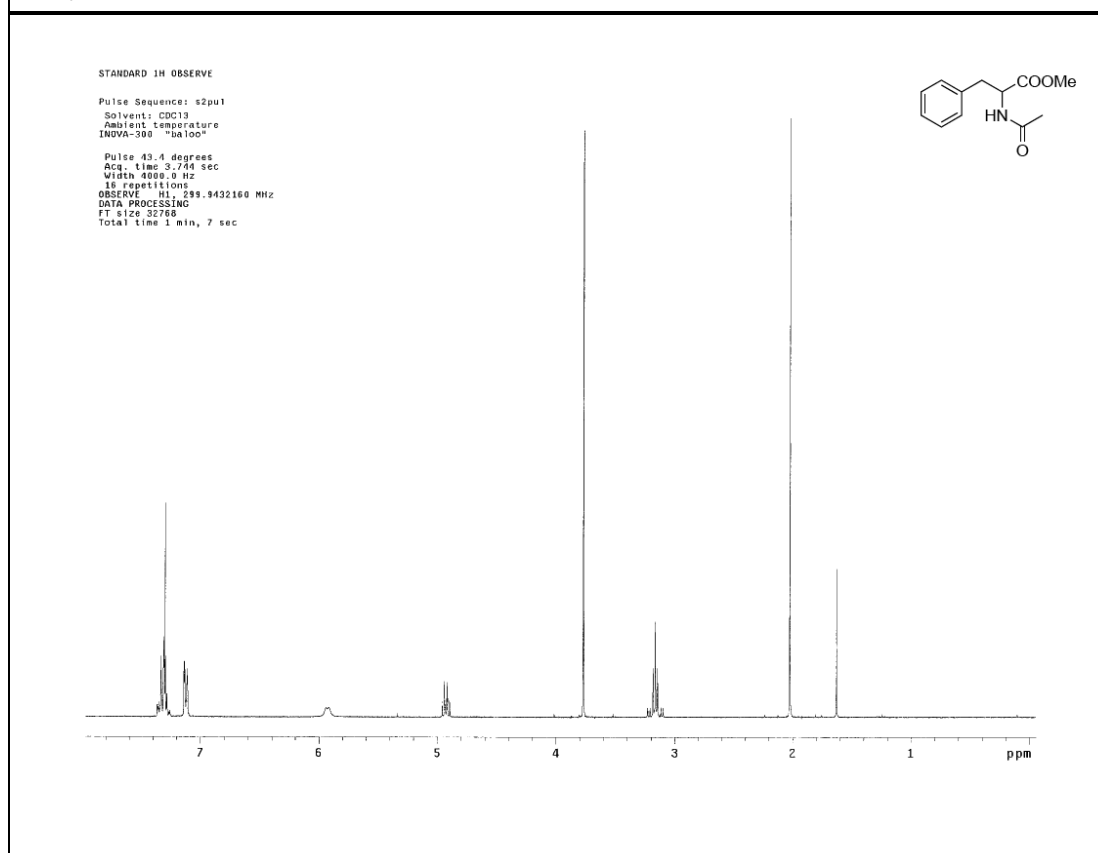
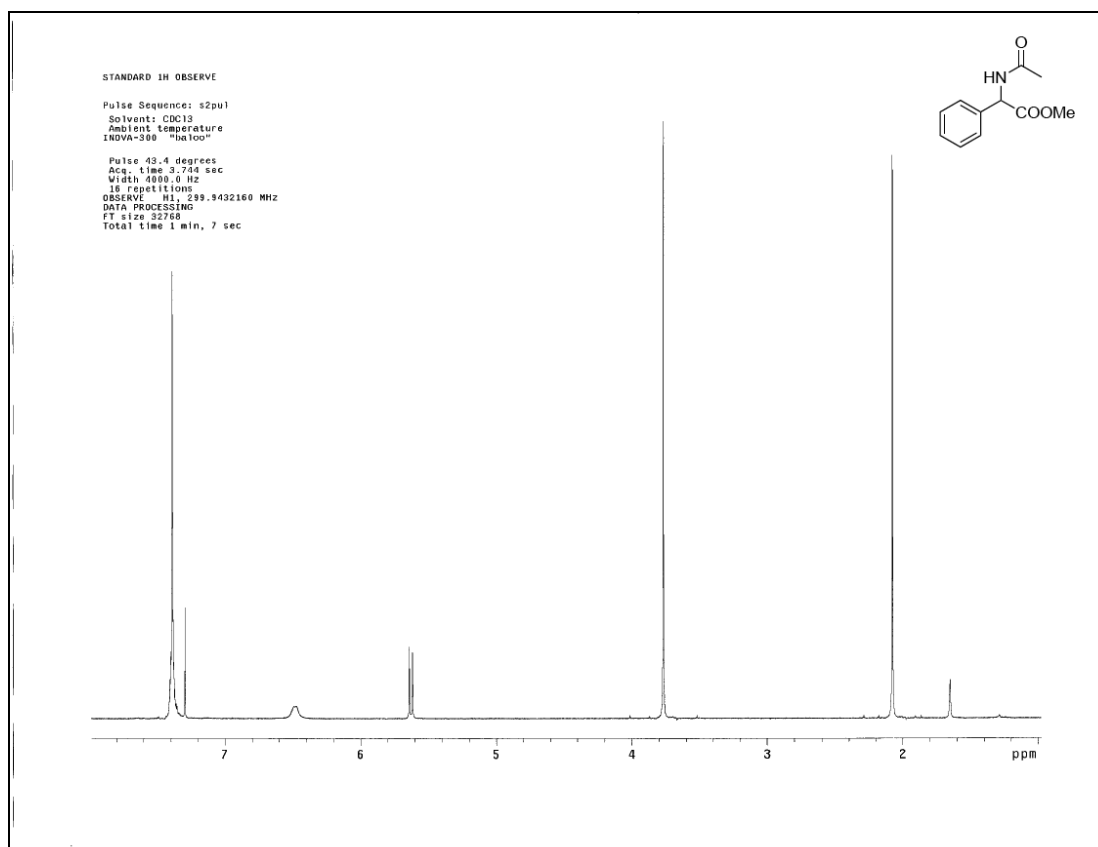
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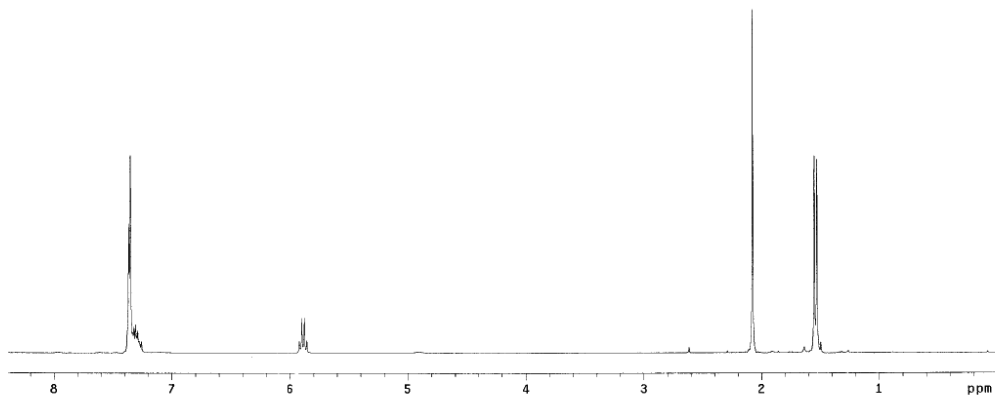
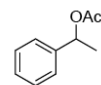
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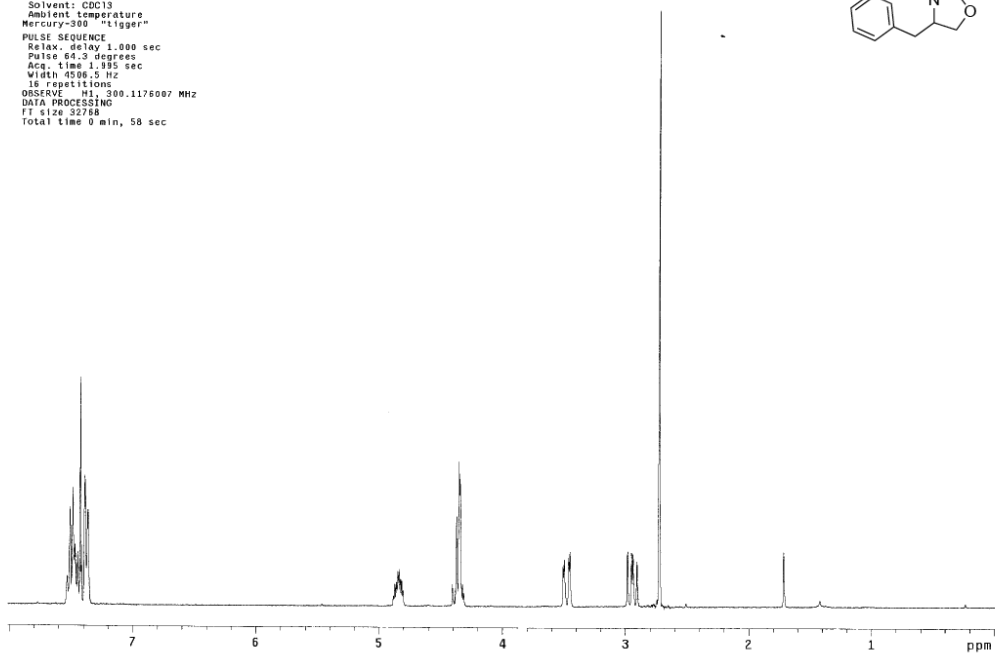
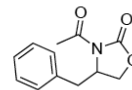
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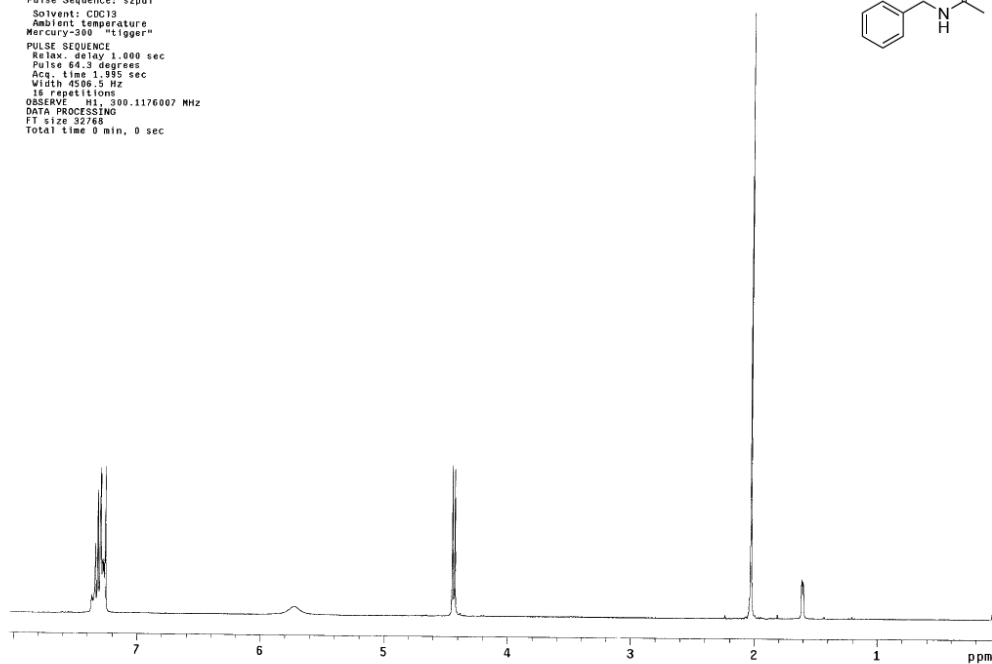
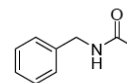
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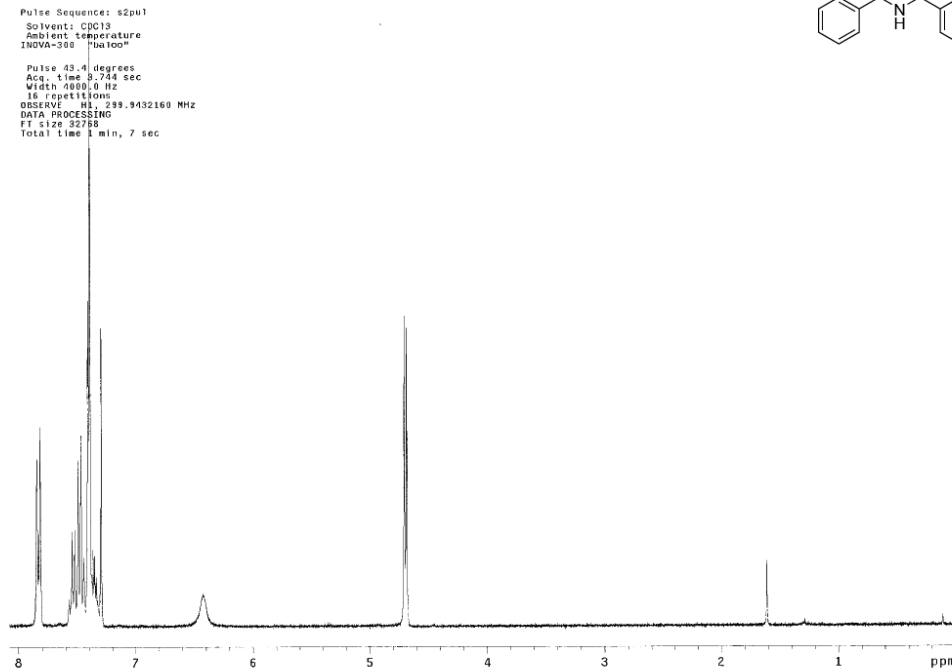
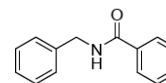
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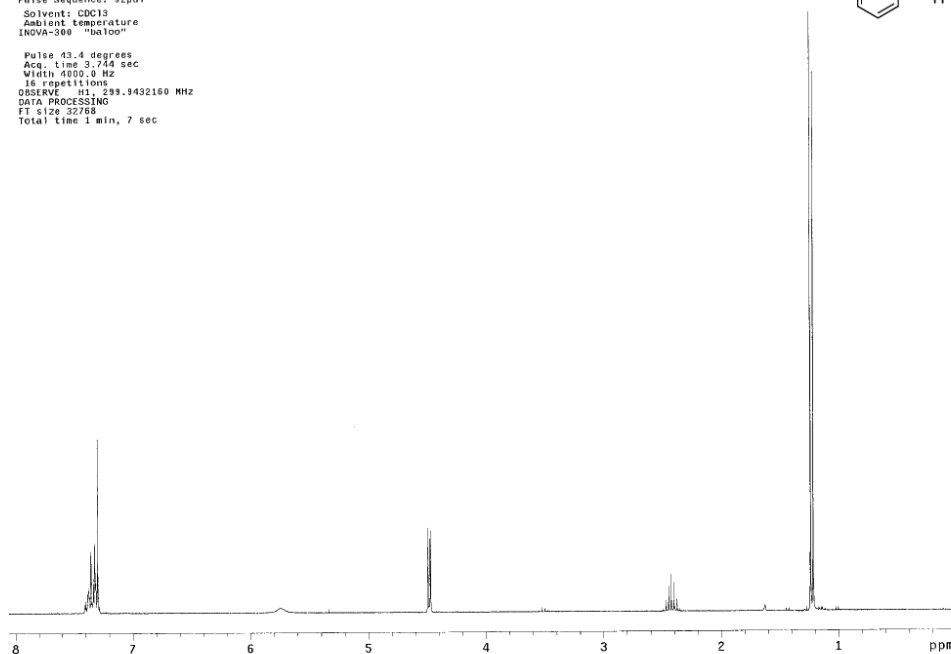
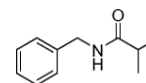
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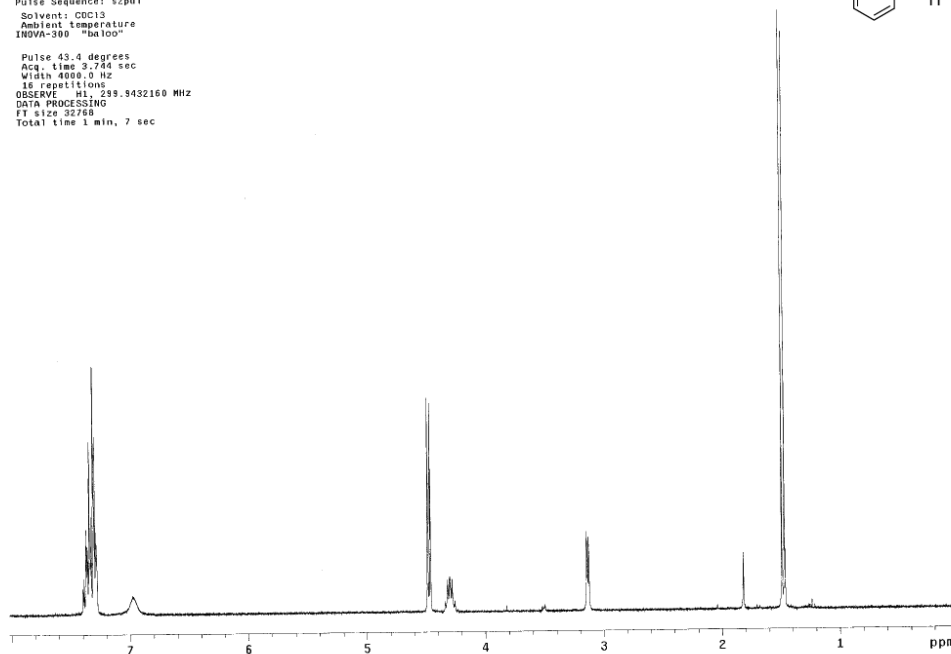
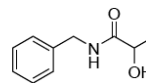
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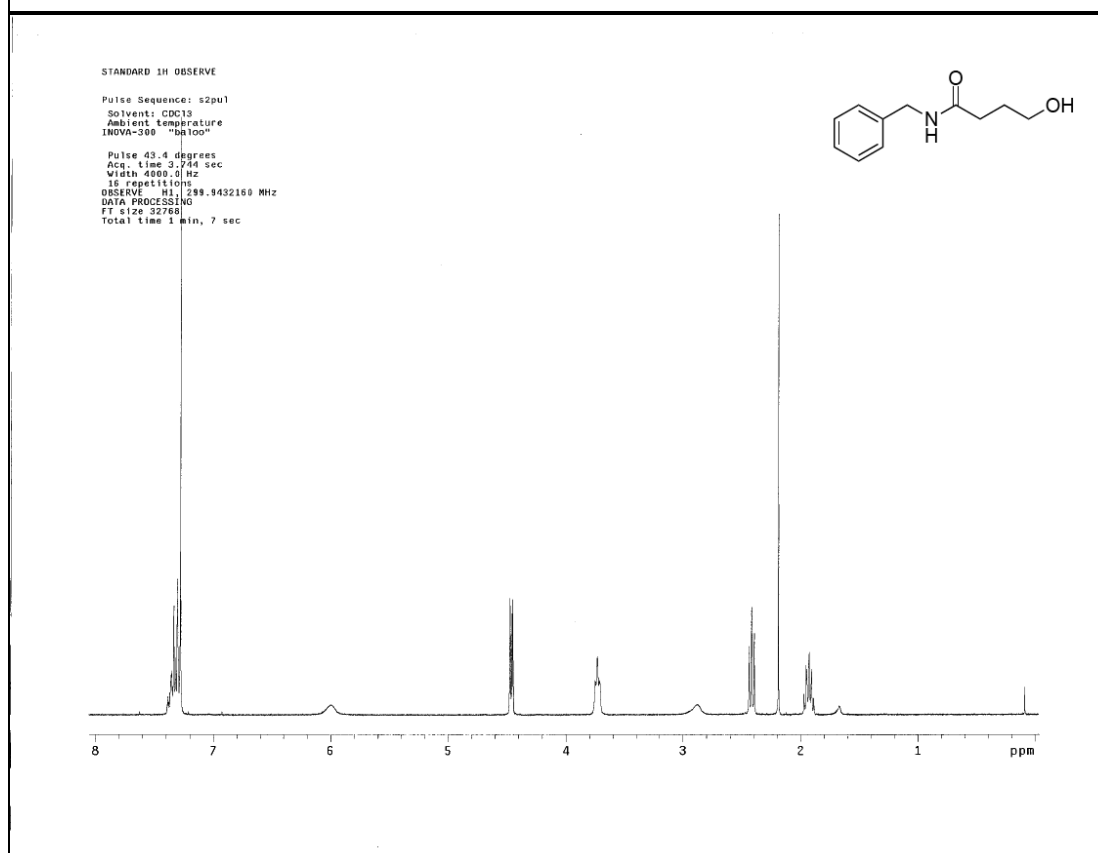
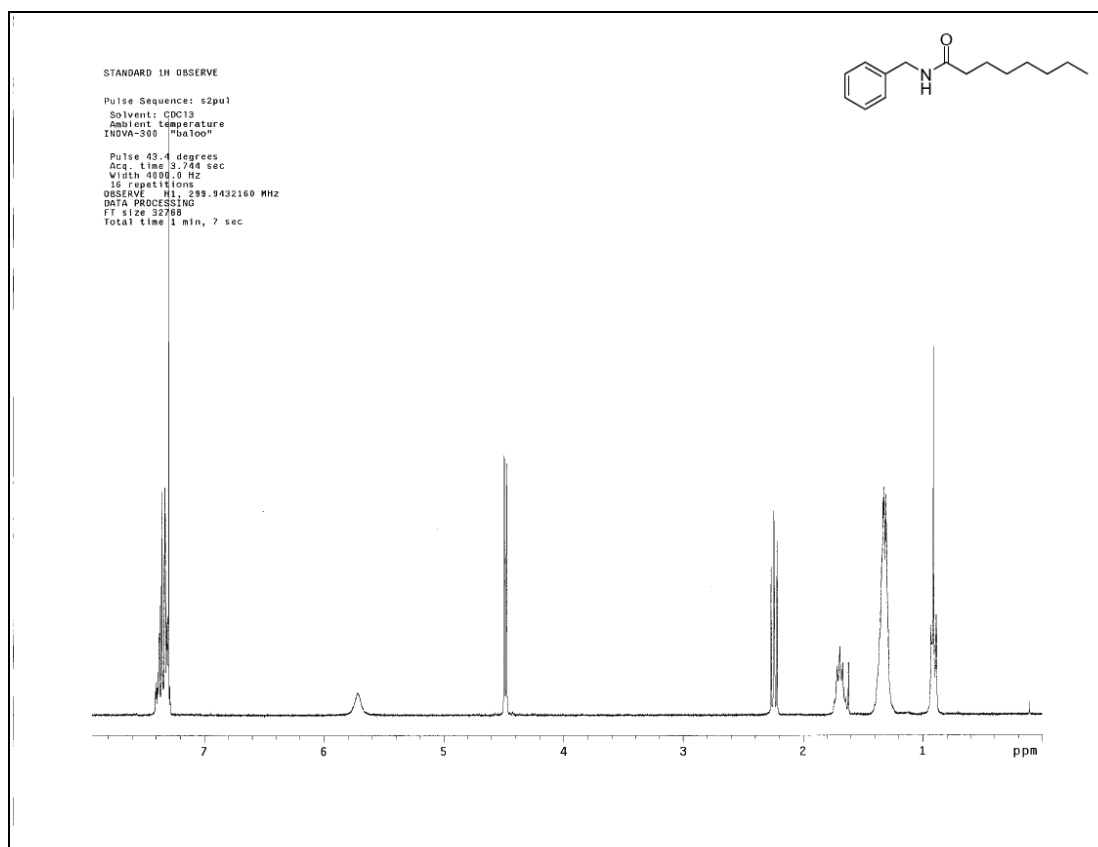
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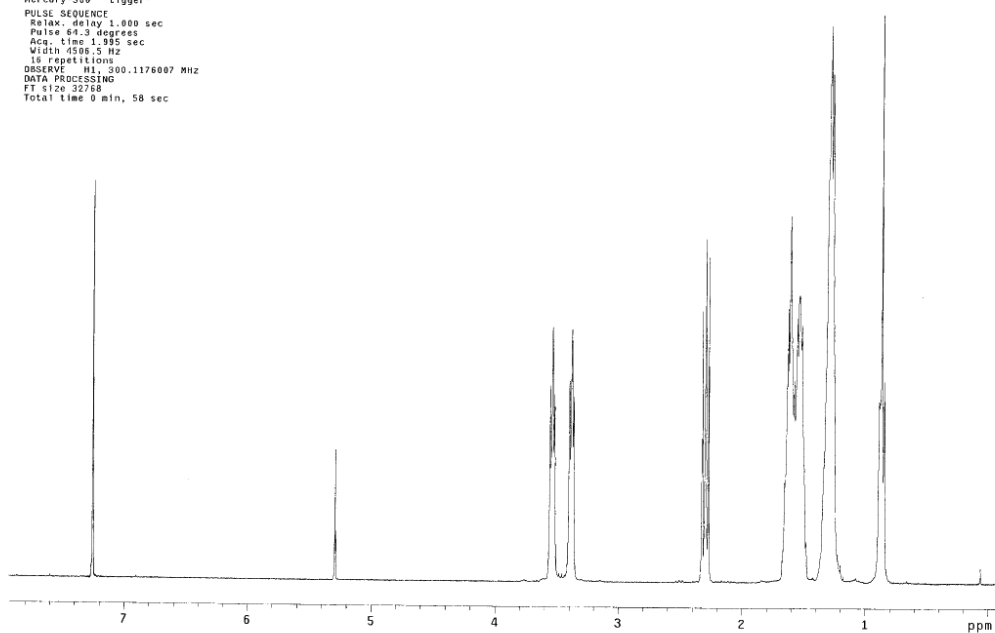
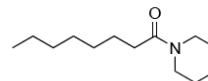
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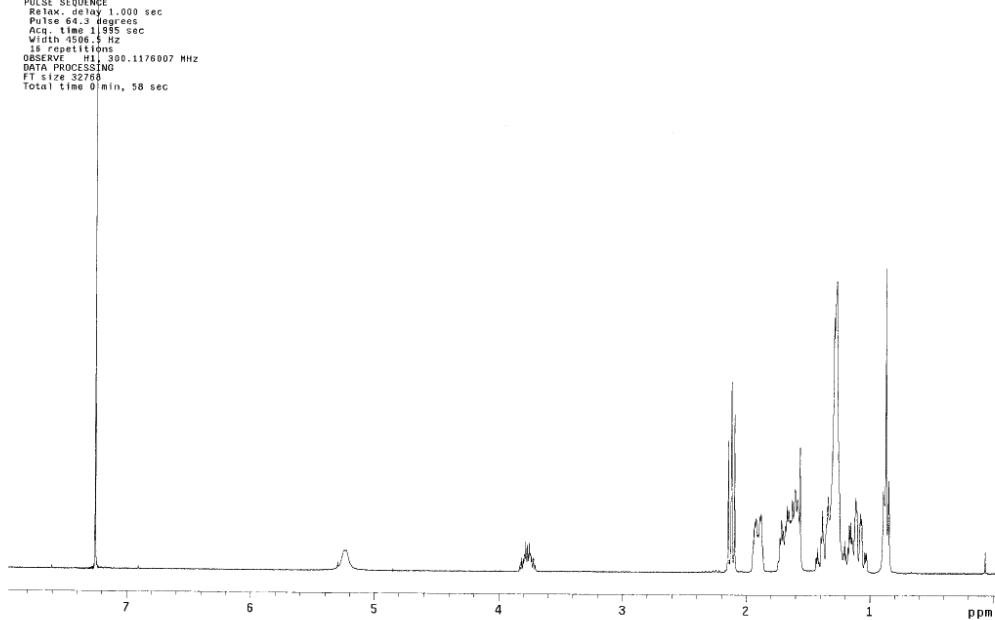
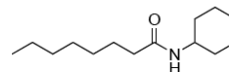
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DATA PROCESSING
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STANDARD 1H OBSERVE

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STANDARD 1H OBSERVE

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Ambient Temperature
Mercury-300 "Hgger"

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