

## On the vibrational spectra of the ground and the singlet excited $\pi\pi^*$ state of 6,7-dimethyl-8-ribityllumazine

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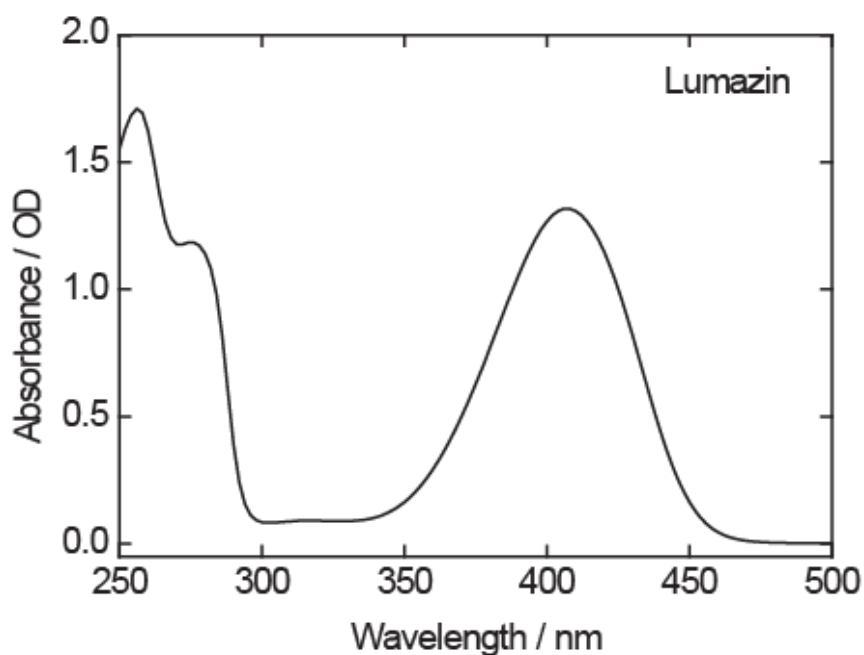
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## 1. UV/Vis absorption spectrum of 6,7-dimethyl-8-ribityl-lumazine



**Figure S1.** Absorption spectrum of 6,7-dimethyl-8-ribityl-lumazine in D<sub>2</sub>O containing 40 mM potassium phosphate (pD ~ 7).

The UV/vis absorption spectrum of 6,7-dimethyl-8-ribityl-lumazine has been discussed in the literature before <sup>1-3</sup>. Lumazine exhibits an unstructured absorption band around 400 nm with its peak centered at 408 nm. It is generally assumed that this absorption is due to the lowest singlet  $\pi\pi^*$  transition. In the short wavelength range a band at 256 nm with a shoulder at 280 nm appears.

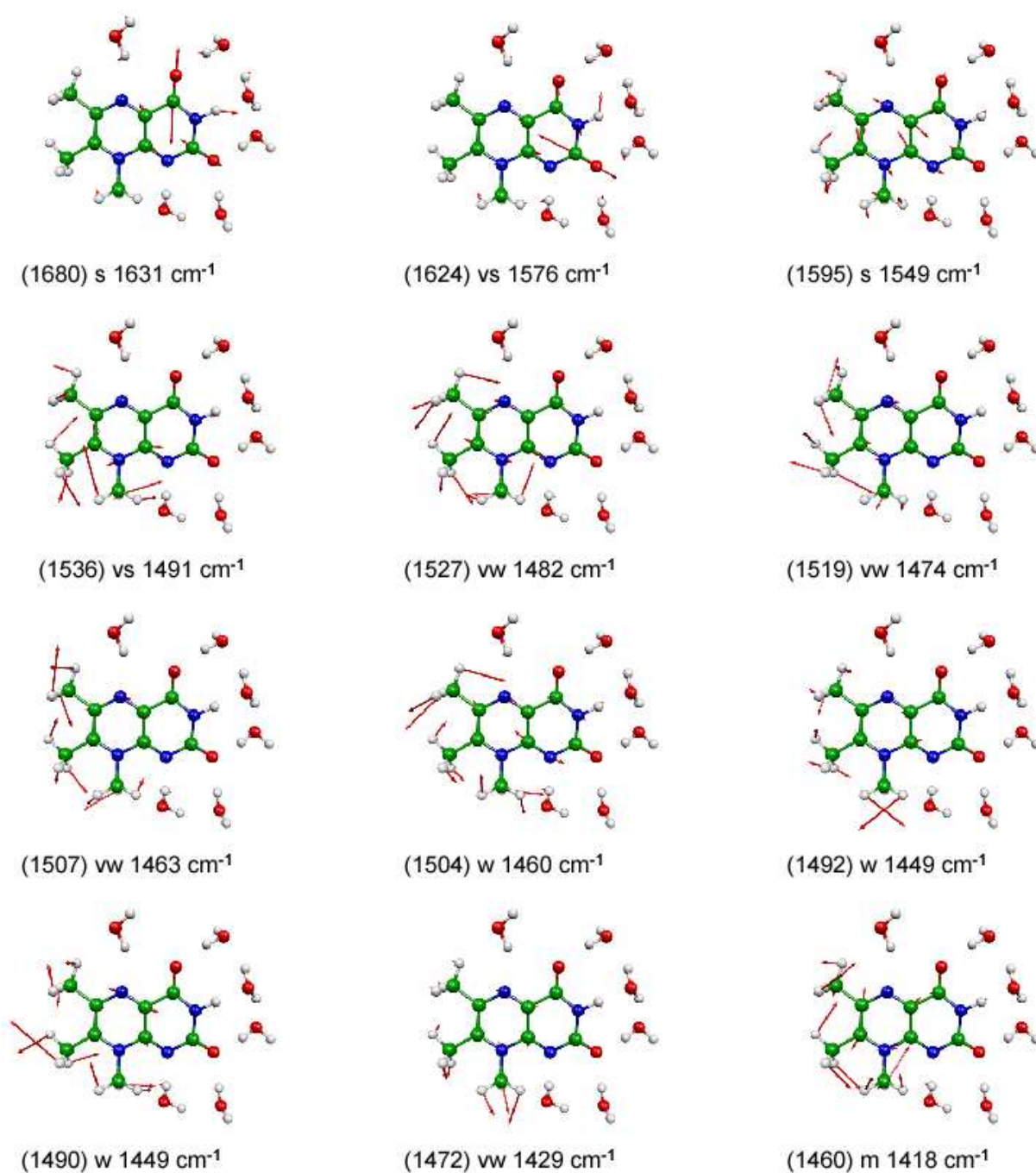
## 2. Results of global data-analysis

**Table S1.** Relaxation times of 6,7-dimethyl-8-ribityl-lumazine (RiboLu) and its isotopologues in D<sub>2</sub>O buffered solution after photoexcitation as determined by time resolved infrared spectroscopy in a global data analysis.

sample	$\tau_1$ / ps	$\tau_2$ / ps	$\tau_3$ / ns
RiboLu	2.7	88	3.2*
[U- <sup>15</sup> N <sub>4</sub> ]	1.9	81	4.9*
[2- <sup>13</sup> C <sub>1</sub> ]	2.1	91	5.2*
[4- <sup>18</sup> O <sub>1</sub> ]	2.2	84	6.1*

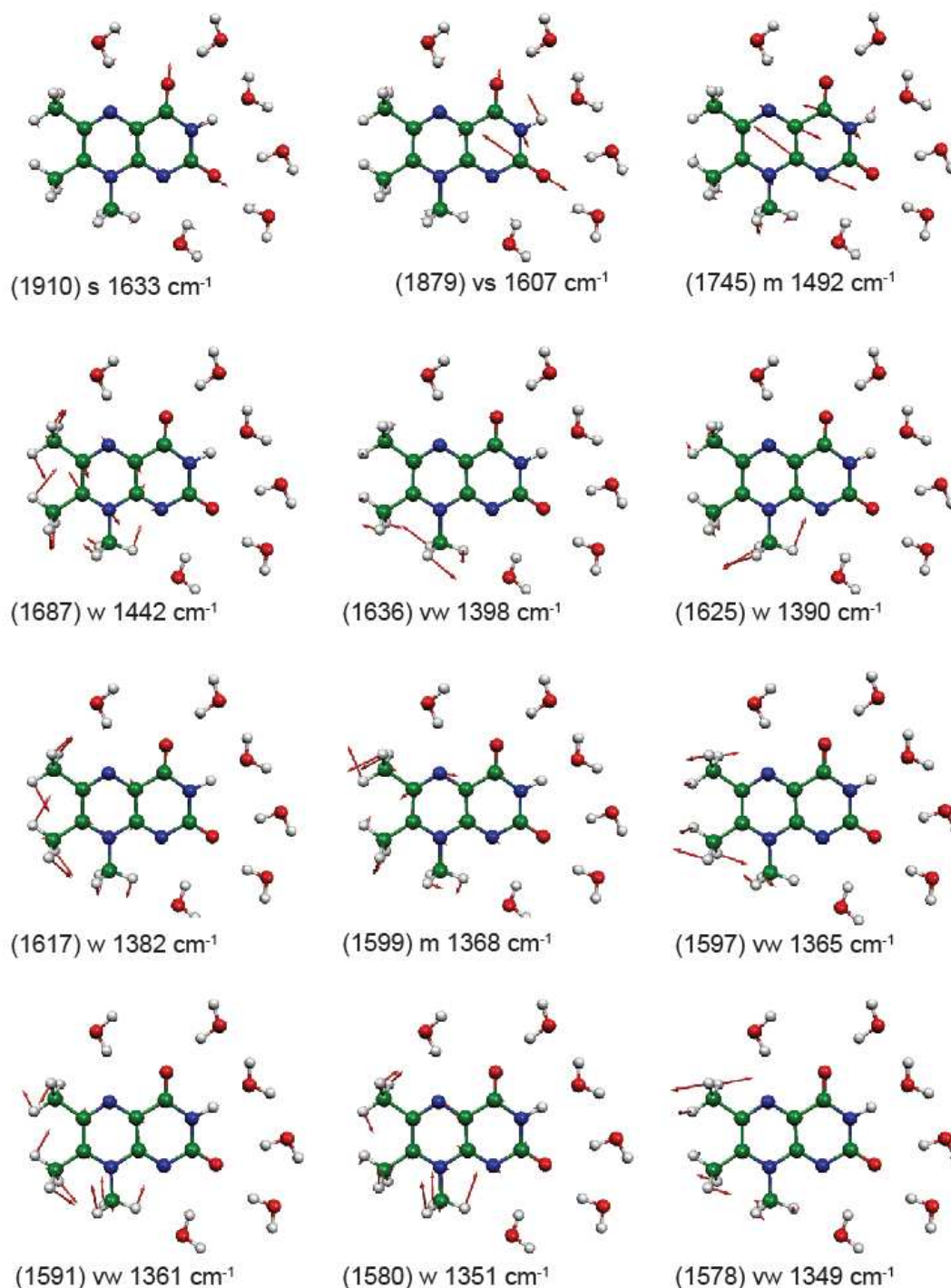
In the global data analysis the signals were modeled as sums of exponentials analytically convoluted with a Gaussian that represents the instrument response function (~ 250 fs) of the ultrafast IR spectrometer. As the time resolved experiments are limited to a maximum delay time of ~ 3 ns the values marked with \* are subject to substantial uncertainty and have to be regarded as order of magnitude approximation.

### 3. Calculated vibrational modes of MeLu in the electronic ground state



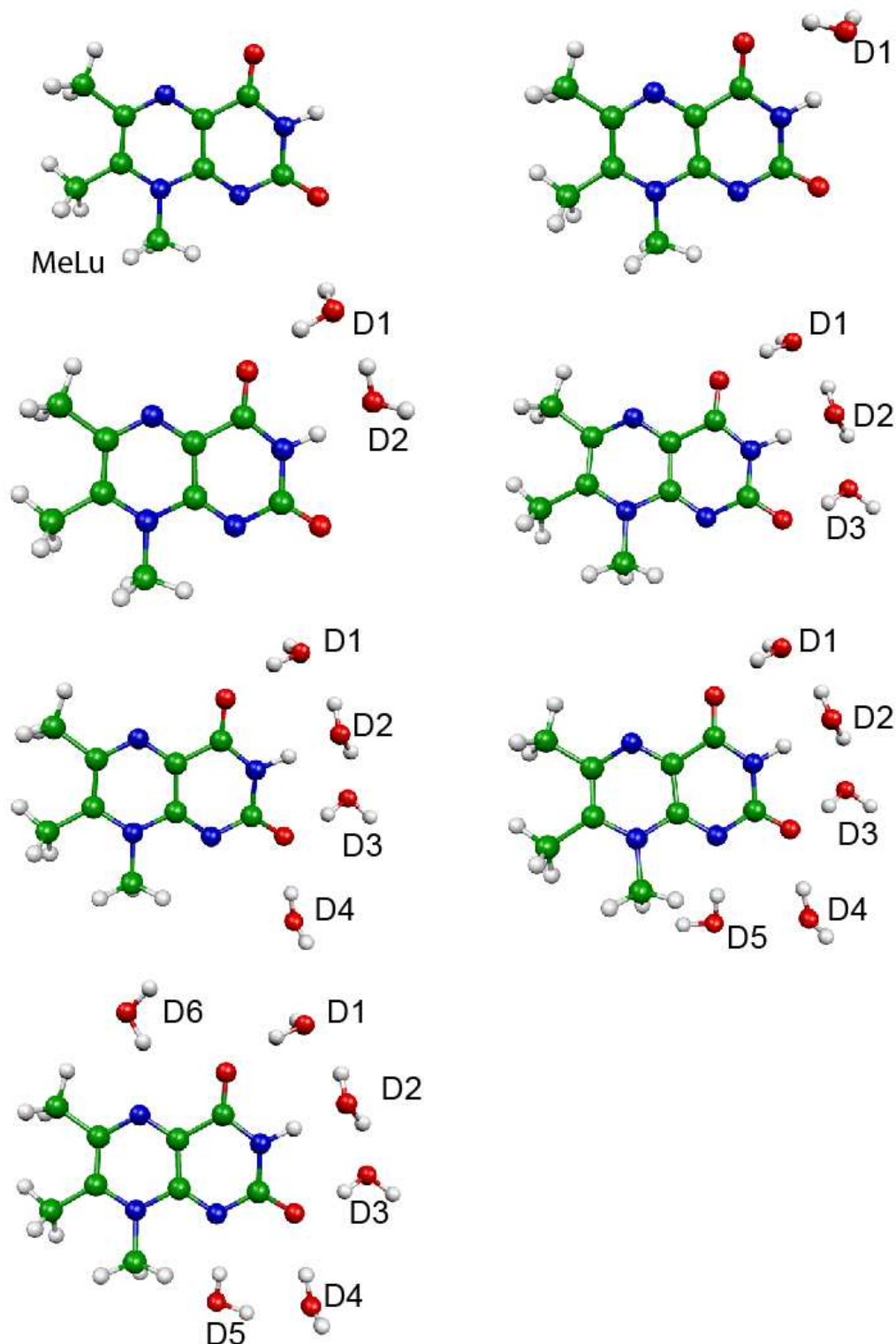
**Figure S2.** Calculated vibrational modes of N-deuterated 6,7,8-trimethyl-lumazine (BLYP/TZVP) solvated by 6  $\text{D}_2\text{O}$  molecules. The intensities of the vibrational modes are categorized as: vs, very strong; s, strong; m, medium; w, weak; vw, very weak. Scaled frequency values (scaling of factor 1.03) are given in brackets.

#### 4. Computed vibrational modes of MeLu in its electronic singlet excited $\pi\pi^*$ state



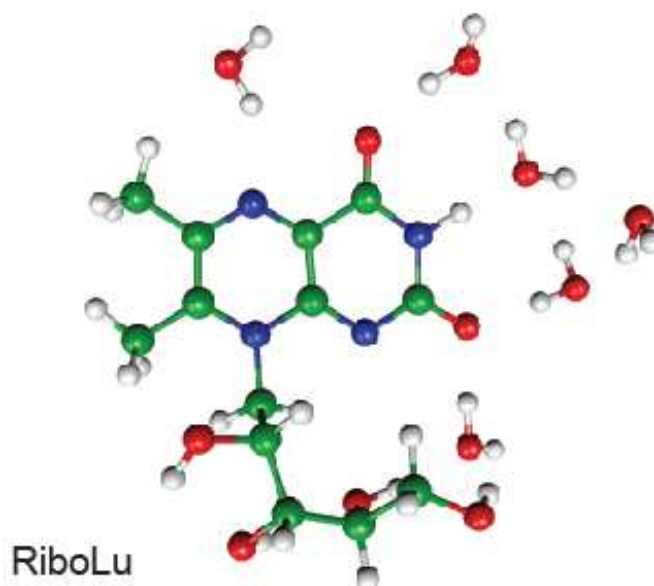
**Figure S3.** Calculated singlet  $\pi\pi^*$  excited state vibrational modes of N-deuterated 6,7,8 – trimethyl-lumazine (CIS / 6-311G\*\*) solvated by 6 molecules of deuterated water. The intensities of the vibrational modes are categorized as: vs, very strong; s, strong; m, medium; w, weak; vw, very weak. Scaled frequency values (scaling factor of 0.855) are given in brackets.

## 5. Optimized structures of 6,7,8-trimethyl-lumazine (MeLu) solvated by various numbers of D<sub>2</sub>O molecules



**Figure S4.** Fully optimized models of N-deuterated 6,7,8-trimethyl-lumazine (MeLu) solvated by various numbers of D<sub>2</sub>O molecules.

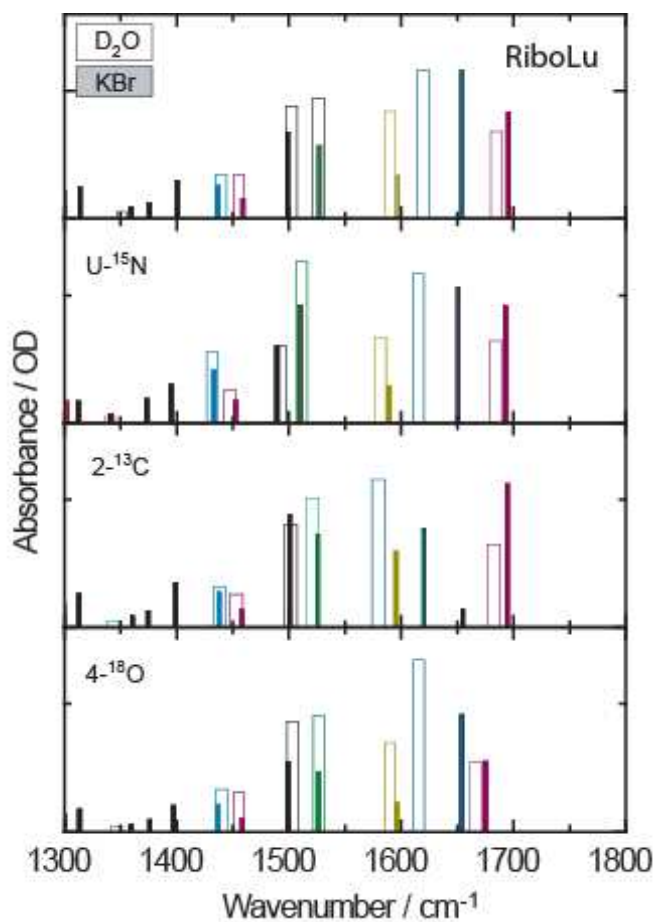
**6. Optimized structure of the 6,7-dimethyl-8-ribityl-lumazine solvated by six D<sub>2</sub>O molecules**



**Figure S5.** Fully optimized structure of N-deuterated 6,7-dimethyl-8-ribityl-lumazine (RiboLu) solvated by six D<sub>2</sub>O molecules.



## 7. Comparison between KBr and solvent surrounding



**Figure S6:** Comparison of the peak positions of the infrared absorption spectra of 6,7-dimethyl-8-ribityl-lumazine in D<sub>2</sub>O solution (open bars) and KBr-matrix (filled bars). The solvent leads to a significant downshift of the C(2)O(2) carbonyl band compared to the matrix. The positions of the other visible bands are hardly affected by the different surrounding conditions.



## 8. List of all authors for Reference 34

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