

*Supporting Information*

**Enhanced Catalyst Activity and Enantioselectivity with  
Chirality-switchable Polymer Ligand PQXphos in Pd-catalyzed Asymmetric  
Silaborative Cleavage of *meso*-Methylenecyclopropanes**

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**Contents**

**1. General**

**2. Experimental Procedures and Spectral Data for the New Compounds**

**3. Determination of the Enantiomeric Excesses of the Products by HPLC with a Chiral  
Stationary Phase**

**4. References**

**5. NMR Spectra of New Compounds**

## 1. General

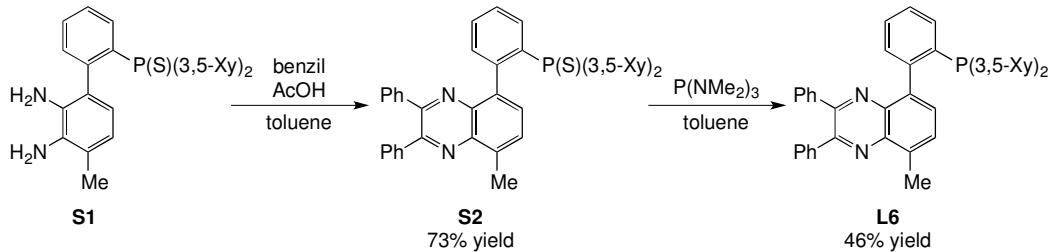
All reactions were carried out under an atmosphere of nitrogen with magnetic stirring. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian Mercury vx400 or a JEOL JNM-A500 spectrometer at ambient temperature. <sup>1</sup>H NMR data are reported as follows: chemical shift in ppm downfield from tetramethylsilane ( $\delta$  scale), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, sex = sextet, sep = septet, m = multiplet, and br = broad), coupling constant (Hz), and integration. <sup>13</sup>C NMR chemical shifts are reported in ppm downfield from tetramethylsilane ( $\delta$  scale). <sup>31</sup>P NMR chemical shifts are reported in ppm downfield from H<sub>3</sub>PO<sub>4</sub> (85%). All <sup>13</sup>C NMR and <sup>31</sup>P NMR spectra were obtained with complete proton decoupling. The GPC analysis was carried out with TSKgel G4000H<sub>HR</sub> (CHCl<sub>3</sub>, polystyrene standard).

Toluene and THF were dried and deoxygenized using an alumina/catalyst column system (GlassContour Co.). Benzil (Wako), acetic acid (Wako), tris(dimethylamino)phosphine (Wako), 0.5 M HCl/MeOH solution (Wako), 1-bromopropane (Wako), trimethylphosphine (Strem), sodium borohydride (Aldrich), tris(dibenzylideneacetone)dipalladium(0) (Strem), triphenylphosphine (Nacalai tesque), and PS-PPh<sub>3</sub> **L9** (Biotage) were used as received from the commercial sources. 1,1,2-trichloroethane (Wako) was purchased from the commercial sources and distilled before use. Sodium Hydride (Nacalai tesque) was purchased from the commercial sources and washed with hexane. Methylenecyclopropanes (MCPs) **2a-2f**<sup>S1</sup>, Silylborane **1**<sup>S2</sup>, PQXphos **L1-L3**<sup>S3</sup>, **L4**<sup>S1c</sup>, and **L5**<sup>S1</sup> were synthesized by the method reported previously. **L7**<sup>S4</sup> and **L8**<sup>S5</sup> were provided by Professor Yasushi Tsuji's group (Kyoto University).

## 2. Experimental Procedures and Spectral Data for New Compounds

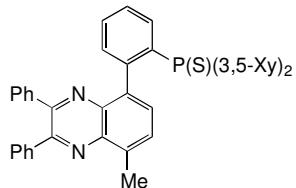
## 2.1 Synthesis of Ligand L6

Ligand **L6** was prepared according to Scheme S1.



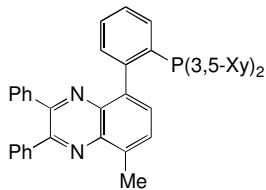
**Scheme S1.** Synthesis of Ligand L6

### 2.1.1 Synthesis of S2



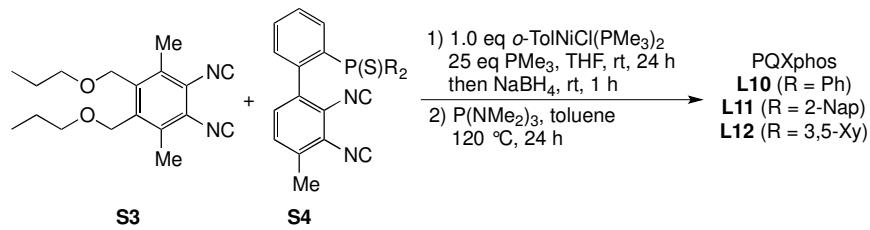
A mixture of **S1<sup>S3</sup>** (428 mg, 0.91 mmol), benzil (201 mg, 0.96 mmol), and acetic acid (1 mL) in toluene (15 mL) was heated for 2 h under reflux with azeotropic removal of water. After evaporation of volatile material under reduced pressure, the residual solid was dissolved in  $\text{CH}_2\text{Cl}_2$  and subjected to silica gel column chromatography (hexane: $\text{CH}_2\text{Cl}_2$  = 3:5), giving the title compound **S2** (429 mg, 73%).;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.03 (d,  $J$  = 7.2 Hz, 1H), 7.74 (dd,  $J$  = 15.2 Hz, 8.0 Hz, 1H), 7.53-7.57 (m, 3H), 7.49 (s, 1H), 7.40-7.46 (m, 5H), 7.31-7.40 (m, 5H), 7.24-7.29 (m, 2H), 7.07 (s, 1H), 7.04 (s, 1H), 6.94 (s, 1H), 6.53 (s, 1H), 2.69 (s, 3H), 2.21 (s, 6H), 1.91 (s, 6H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ) Due to the complexity of the spectra all the peaks are listing without take into consideration C-P couplings.  $\delta$  150.8, 150.5, 142.1, 142.0, 139.6, 139.3, 139.1, 139.0, 137.5, 137.5, 137.4, 136.6, 136.4, 136.3, 136.3, 134.5, 134.4, 134.3, 133.9, 133.7, 133.7, 133.4, 132.9, 132.7, 132.7, 132.5, 131.7, 131.6, 131.5, 130.9, 130.3, 130.3, 130.2, 129.9, 129.4, 129.3, 128.6, 128.5, 128.3, 128.2, 127.9, 127.2, 127.0, 21.3, 21.0, 17.1;  $^{31}\text{P}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  42.6 (s); HRMS (ESI)  $m/z$  calcd for  $\text{C}_{43}\text{H}_{37}\text{N}_2\text{PS}+\text{H}$  ( $\text{M}^++\text{H}$ ): 645.2493, found: 645.2471

### 2.1.2 Synthesis of L6



To a solution of **S2** (429 mg, 0.69 mmol) in toluene (5 mL) were added  $\text{P}(\text{NMe}_2)_3$  (240  $\mu\text{L}$ , 1.3 mmol) at room temperature. The mixture was stirred at 120  $^{\circ}\text{C}$  for 24 h, then water was added to the mixture. The mixture was washed with water and brine, and dried over  $\text{Na}_2\text{SO}_4$ . After evaporation of the volatiles, the residue was subjected to silica gel column chromatography (hexane:CH<sub>2</sub>Cl<sub>2</sub> = 1:1), giving the title compound **L6** (189 mg, 31% yield).; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.64 (ddd, *J* = 7.2 Hz, 2.4 Hz, 1.6 Hz, 1H), 7.57 (d, *J* = 7.2 Hz, 1H), 7.46-7.49 (m, 3H), 7.38-7.45 (m, 5H), 7.24-7.33 (m, 4H), 7.00 (s, 1H), 6.96 (s, 1H), 6.94 (s, 1H), 6.71-6.74 (m, 3H), 2.91 (s, 3H), 2.31 (s, 6H), 2.03 (s, 6H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) Due to the complexity of the spectra all the peaks are listing without take into consideration C-P couplings.  $\delta$  150.9, 150.6, 144.5, 144.2, 139.9, 139.7, 139.6, 139.2, 138.4, 138.3, 138.1, 138.0, 137.7, 137.6, 137.5, 137.5, 137.1, 137.0, 137.0, 133.1, 131.7, 131.7, 131.5, 131.5, 131.5, 131.4, 131.4, 130.2, 130.2, 130.0, 129.8, 129.7, 128.6, 128.5, 128.3, 128.1, 127.7, 127.6, 127.2, 21.3, 21.1, 17.2; <sup>31</sup>P NMR (160 MHz, CDCl<sub>3</sub>)  $\delta$ -10.7 (s); HRMS (ESI) *m/z* calcd for C<sub>43</sub>H<sub>37</sub>N<sub>2</sub>P+H (M<sup>+</sup>+H): 613.2773, found: 613.2767

## 2.2 Synthesis of PQXphos **L10-L12**



**Scheme S2.** Synthesis of PQXphos (**L10-L12**)

**[Polymerization]** To a THF (45 mL) solution of organonickel initiator (*o*-Tol) $\text{NiCl}(\text{PMe}_3)_2$ <sup>S6</sup> (0.01 mM in THF, 0.20 mL, 2.0  $\mu\text{mol}$ ) and  $\text{PMe}_3$  (0.1 M in THF, 0.50 mL, 50  $\mu\text{mol}$ ) was added a mixture of monomer **S3**<sup>S7</sup> (312 mg, 0.95 mmol) and **S4**<sup>S3</sup> (50  $\mu\text{mol}$ ) in THF (5 mL) at room temperature. The mixture was stirred for 24 h at room temperature. To the reaction mixture was added  $\text{NaBH}_4$  (12 mg, 0.32 mmol), and the mixture was stirred for 1 h. The solvent was then evaporated until approximately half of the original volume. The mixture was poured into vigorously stirred MeOH (200 mL), and precipitated polymer was collected by centrifugation

followed by washing with MeOH  $\times$  2. After drying in vacuo, fibriform polymer was obtained.

**[Reduction of Phosphine Sulfide]** A mixture of the obtained polymer and P(NMe<sub>2</sub>)<sub>3</sub> (0.618 mL, 3.40 mmol) in toluene (6.7 mL) was stirred at 120 °C for 24 h. The mixture was diluted with THF (2.2 mL) and poured into vigorously stirred MeOH (400 mL). Precipitated material was collected by filtration. Purification by centrifugation with MeOH  $\times$  2 gave PQXphos(**L10-L12**) as fibriform solid. See Table S1 for the yields and the GPC data for the obtained polymers.

**Table S1.** the yields and the GPC data for the PQXphos

entry	PQXphos	% Yield (2 steps)	$M_n^a$	$M_w/M_n^a$
1	<b>L10</b>	89	$1.6 \times 10^5$	1.30
2	<b>L11</b>	83	$1.5 \times 10^5$	1.39
3	<b>L12</b>	82	$1.9 \times 10^5$	1.34

<sup>a</sup> Determined by GPC (polystyrene standard).

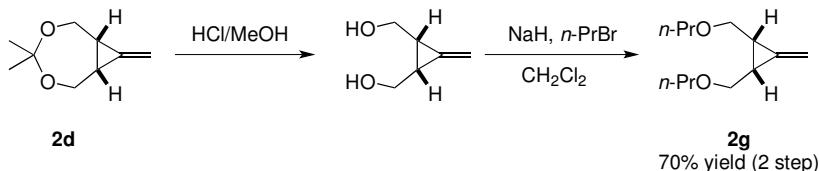
**L10:** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  0-6.0, (brm, peak top; 0.90, 1.60, 2.35, 3.47, 4.58 (24x+3y)H), 6.4-8.4 (brm, 16yH); <sup>31</sup>P NMR (160 MHz, CDCl<sub>3</sub>)  $\delta$  -15.6 (brs)

**L11:** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  0-6.0, (brm, peak top; 0.90, 1.60, 2.35, 3.46, 4.56 (24x+3y)H), 6.4-8.4 (brm, 20yH); <sup>31</sup>P NMR (160 MHz, CDCl<sub>3</sub>)  $\delta$  -14.3 (brs)

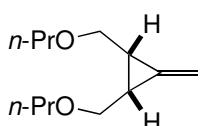
**L12:** <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  0-6.0, (brm, peak top; 0.90, 1.60, 2.35, 3.46, 4.58 (24x+15y)H), 6.4-8.4 (brm, 12yH); <sup>31</sup>P NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  -15.6 (brs)

### 2.3 Preparation of Methylenecyclopropane **2g**

Methylenecyclopropane **2g** was prepared according to *Scheme S2*.



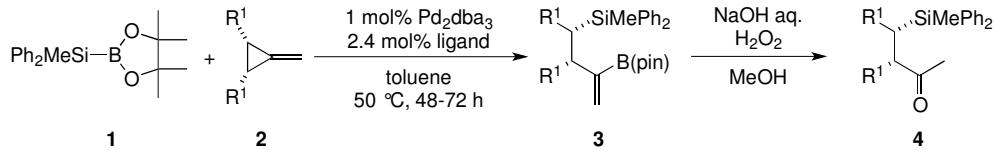
**Scheme S2. Synthesis of Methylenecyclopropane **2g****



**Synthesis of meso-1-methylene-2,3-bis(propoxymethyl)cyclopropane (**2g**):** In a 30 mL of round-bottom flask, **2d**<sup>S1</sup> (925 mg, 6 mmol) and 0.5 M HCl/MeOH solution (10 mL) were

charged. The mixture was stirred at room temperature. After stirring for 2 h, 1 mL of  $\text{Et}_3\text{N}$  was added and the solution was diluted with  $\text{Et}_2\text{O}$ . The resulting mixture was passed through a pad of Celite<sup>®</sup> to remove the solid materials. The solution was concentrated and the residue was dissolved in DMF. NaH (432 mg, 18 mmol) was added to the solution at 0 °C. The mixture was stirred at 0 °C for 30 min, then 1-bromopropane (2.21 g, 18 mmol) was added to the mixture. The mixture was stirred at room temperature for 24 h, then water was added to the mixture. The mixture was washed with water and brine, and dried over  $\text{Na}_2\text{SO}_4$ . After evaporation of the volatiles, the residue was subjected to silica gel column chromatography (hexane: $\text{Et}_2\text{O}$  = 20:1), giving the title compound **2g** (832 mg, 70% yield).: <sup>1</sup>H NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.43 (t,  $J$  = 2.4 Hz, 2H), 3.48 (dd,  $J$  = 5.2 Hz, 2.4 Hz, 4H), 3.39 (ddt,  $J$  = 7.2 Hz, 5.2 Hz, 2.4 Hz, 4H), 1.97 (sex,  $J$  = 2.4 Hz, 2H), 1.60 (sep,  $J$  = 7.2 Hz, 4H), 0.92 (t,  $J$  = 7.2 Hz, 6H); <sup>13</sup>C NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  136.1, 104.0, 72.3 (2C), 68.7 (2C), 22.9 (2C), 19.7 (2C), 10.6 (2C); IR (ATR) 2934, 2854, 1466, 1364, 1094, 1045, 889  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_{22}\text{O}_2\text{H}$  ( $\text{M}+\text{H}^+$ ): 199.1698, found: 199.1690

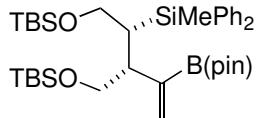
#### 2.4 Asymmetric Silaborative C-C Cleavage of meso-MCPs using PQXphos (*P*)-L1-3 as chiral ligands (Table 1)



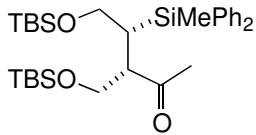
**General procedure:** To a solution of PQXphos (4.8  $\mu\text{mol}$  phosphorous atom) in toluene (200  $\mu\text{L}$ ) was added  $\text{Pd}_2\text{dba}_3$  (0.01 M in toluene, 200  $\mu\text{L}$ , 2  $\mu\text{mol}$ ). The mixture was stirred at room temperature for 5 min. To the mixture was added **2** (0.30 mmol) and **1** (0.20 mmol) in this order, and the resulting mixture was heated at 50 °C with stirring. The reaction was monitored by GC. After the reaction was completed (48-72 h), subsequent addition of MeCN (10 mL) resulted in precipitation of the PQXphos. The suspension was passed through a pad of Celite<sup>®</sup> using MeCN as an eluent. The crude product was isolated by silica gel column chromatography to give a 2-boryl-4-silyl-1-butene derivative. To determine the enantiomeric excess, the silylated alkenylborane obtained was converted to  $\beta$ -silyl ketone. To a methanol solution (2 mL) of the 2-boryl-4-silyl-1-butene derivative was added aqueous NaOH solution (3N, 2.5 mL) and the mixture was cooled to 0 °C. Aqueous  $\text{H}_2\text{O}_2$  solution (30% 1.5 mL) was slowly added to the mixture. The resulting solution was stirred at room temperature for 12 h. The resulting mixture was extracted with  $\text{Et}_2\text{O}$  and the extracts were washed with water. After drying with anhydrous  $\text{MgSO}_4$ , the concentrated mixture was purified by PTLC to give a  $\beta$ -silyl ketone. Enantiomeric excess of this compound was determined by HPLC analysis.

## 2.5 Spectral Data for New Compounds

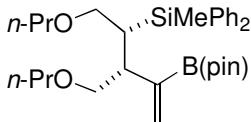
Compounds **3a-e** were characterized by comparing their spectral data with those reported earlier<sup>S1c</sup>.



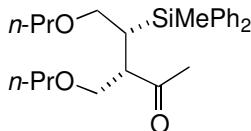
**(2R,3R)-1,4-bis[(tert-butyldimethylsilyl)oxy]-(2-methyldiphenylsilyl)-3-[1-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)ethenyl]butane (3f):** <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.61-7.65 (m, 2H), 7.57-7.61 (m, 2H), 7.27-7.38 (m, 6H), 5.93 (d, *J* = 4.0 Hz, 1H), 5.60 (d, *J* = 3.5 Hz, 1H), 3.68 (dd, *J* = 9.5 Hz, 0.5 Hz, 1H), 3.55 (dd, *J* = 9.5 Hz, 0.5 Hz, 1H), 3.45 (dd, *J* = 10.0 Hz, 5.0 Hz, 1H), 3.41 (dd, *J* = 10.0 Hz, 10.0 Hz, 1H), 2.80 (ddd, *J* = 10.0 Hz, 5.0, 5.0 Hz, 1H), 1.84 (d, *J* = 11.0 Hz, 1H), 1.27 (s, 6H), 1.27 (s, 6H), 0.86 (d, *J* = 1.5 Hz, 1H), 0.74 (d, *J* = 1.5 Hz, 1H), 0.72 (d, *J* = 1.5 Hz, 1H), -0.12 (s, 3H), -0.16 (s, 3H), -0.21 (s, 3H), -0.23 (s, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  138.7, 137.9, 135.0 (2C), 134.5 (2C), 132.6, 128.6 (2C), 127.5 (2C), 127.4 (2C), 82.9 (2C), 66.2, 62.3, 48.8, 28.5, 26.0 (3C), 25.9 (3C), 24.9 (2C), 24.7 (2C), 18.3, 18.2, -4.0, -5.4, -5.5, -5.6, -5.7. The boron-bound carbon was not detected due to quadrupolar relaxation. ; HRMS (ESI) *m/z* calcd for C<sub>37</sub>H<sub>63</sub>B<sub>1</sub>O<sub>4</sub>Si<sub>3</sub>+NH<sub>4</sub> (M<sup>+</sup>+NH<sub>4</sub>): 684.4471, found: 684.4465; [ $\alpha$ ]<sub>D</sub><sup>24</sup> +56.2 (c 0.821, CH<sub>2</sub>Cl<sub>2</sub>, 95% ee).



**(3S,4R)-5-((tert-butyldimethylsilyl)oxy)-3-((tert-butyldimethylsilyl)oxy)methyl)-4-(methyldiphenylsilyl)pentan-2-one (4f):** <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.80-7.85 (m, 4H), 7.57-7.62 (m, 6H), 3.86 (d, *J* = 3.5 Hz, 2H), 3.73 (dd, *J* = 10.0 Hz, 5.0 Hz, 1H), 3.62 (dd, *J* = 10.0 Hz, 10.0 Hz, 1H), 3.49 (ddd, *J* = 10.0 Hz, 10.0 Hz, 5.0 Hz, 1H), 2.44 (s, 3H), 2.11-2.15 (m, 1H), 1.10 (s, 9H), 0.99 (s, 9H), 0.98 (s, 3H), 0.14 (s, 3H), 0.09 (s, 3H), 0.06 (s, 3H), 0.05 (s, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  213.7, 137.1, 136.0, 134.7 (2C), 134.3 (2C), 129.2 (2C), 127.9 (2C), 127.8 (2C), 66.4, 62.9, 62.0, 53.5, 32.9, 28.8, 25.9 (3C), 25.7 (3C), 18.2, 18.1, -4.6, -5.8, -5.9, -5.9; IR (ATR) 2953, 2928, 2856, 1714, 1472, 1427, 1252, 1082, 833, 775, 698 cm<sup>-1</sup>; HRMS (FAB) *m/z* calcd for C<sub>31</sub>H<sub>53</sub>O<sub>3</sub>Si<sub>3</sub>+H (M<sup>+</sup>+H): 557.3303, found: 557.3290; [ $\alpha$ ]<sub>D</sub><sup>24</sup> +33.9 (c 0.902, CH<sub>2</sub>Cl<sub>2</sub>, 95% ee)



**(2*R*,3*R*)-1,4-diproxy-(2-methyldiphenylsilyl)-3-[1-(4,4,5,5-tetramethyl-1,3,2-dioxaborola n-2-yl)ethenyl]butane (3g):**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.63-7.66 (m, 2H), 7.56-7.59 (m, 2H), 7.31-7.36 (m, 3H), 7.27-7.31 (m, 3H), 5.88 (d,  $J$  = 3.6 Hz, 1H), 5.63 (d,  $J$  = 3.6 Hz, 1H), 3.40 (dd,  $J$  = 8.8 Hz, 2.8 Hz, 1H), 3.37 (dd,  $J$  = 9.6 Hz, 9.6 Hz, 1H), 3.30 (dd,  $J$  = 9.6 Hz, 5.2 Hz, 1H), 3.14 (dd,  $J$  = 8.8 Hz, 3.6 Hz, 1H), 3.03 (dt,  $J$  = 8.8 Hz, 1.2 Hz, 2H), 2.84-3.00 (m, 3H), 1.80 (dt,  $J$  = 10.4 Hz, 2.8 Hz, 1H), 1.50-1.56 (m, 2H), 1.30-1.35 (m, 2H), 1.27 (s, 6H), 1.27 (s, 6H), 0.92 (t,  $J$  = 7.2 Hz, 3H), 0.74 (t,  $J$  = 7.2 Hz, 6H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  138.4, 138.0, 135.1 (2C), 134.6 (2C), 131.3, 128.5 (2C), 127.4 (2C), 127.3 (2C), 82.9 (2C), 73.6, 72.2, 71.8, 69.9, 46.5, 26.9, 24.8 (2C), 24.8 (2C), 23.1, 22.5, 11.0, 10.5, -3.9. The boron-bound carbon was not detected due to quadrupolar relaxation. ; IR (ATR) 2981, 2934, 2858, 1427, 1360, 1304, 1140, 1105, 788, 698  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{37}\text{H}_{63}\text{B}_1\text{O}_4\text{Si}_3+\text{H}$  ( $\text{M}^++\text{H}$ ): 523.3415, found: 523.3402;  $[\alpha]^{24}_D$  +66.7 (c 1.1080,  $\text{CH}_2\text{Cl}_2$ , 95% ee)



**(3*S*,4*R*)-4-(methyldiphenylsilyl)-5-propoxy-3-(propoxymethyl)pentan-2-one (4g):**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.54-7.60 (m, 4H), 7.30-7.38 (m, 6H), 3.20-3.51 (m, 5H), 2.93-3.14 (m, 4H), 2.16 (s, 3H), 1.92 (sep,  $J$  = 3.2 Hz, 1H), 1.52 (sex,  $J$  = 2.8 Hz, 2H), 1.36 (sex,  $J$  = 2.8 Hz, 2H), 0.90 (t,  $J$  = 7.2 Hz, 3H), 0.77 (t,  $J$  = 7.2 Hz, 3H), 0.71 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  213.2, 136.7, 136.1, 134.8 (2C), 134.5 (2C), 129.2, 129.1, 127.8 (2C), 127.7 (2C), 72.6, 72.5, 72.4, 70.2, 51.3, 31.9, 26.7, 22.9, 22.5, 10.8, 10.4, -4.3; IR (ATR) 2960, 2934, 2862, 1714, 1472, 1427, 1105, 790, 734, 698  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{25}\text{H}_{36}\text{O}_3\text{Si}+\text{H}$  ( $\text{M}^++\text{H}$ ): 413.2512, found: 413.2506;  $[\alpha]^{25}_D$  +32.2 (c 0.982,  $\text{CH}_2\text{Cl}_2$ , 95% ee)

## 2.6 Use of Inverted PQxphos (*M*)-(R,R)-L3 as a Ligand for Asymmetric Silaboration of *meso*-MCP 2a (Scheme 1)

(*P*)-(R,R)-L3 (16.1 mg, 2.4  $\mu\text{mol}$  phosphorus atom) in 1,1,2-trichloroethane (0.3 mL) and toluene (0.1 mL) was stirred at 60  $^{\circ}\text{C}$  for 24 h. To the mixture was added  $\text{Pd}_2\text{dba}_3$  (0.01 M in 1,1,2-trichloroethane / toluene = 3/1, 100  $\mu\text{L}$ , 1  $\mu\text{mol}$ ), and the solution was stirred at room temperature for 10 min. After evaporation of the mixture, the residue was added 1,1,2-trichloroethane (30  $\mu\text{L}$ ), 2a (0.15 mmol, 16.2 mg), and 1 (0.1 mmol, 32.4 mg). The

mixture was stirred at 50 °C for 48 h. Subsequent addition of MeCN (10 mL) resulted in precipitation of the PQXphos. The suspension was passed through a pad of Celite® using MeCN as an eluent. The crude product was isolated by silica gel column chromatography to give a 2-boryl-4-silyl-1-butene derivative. To determine the enantiomeric excess, the silylated alkenylborane obtained was converted to  $\beta$ -silyl ketone. To a methanol solution (2 mL) of the 2-boryl-4-silyl-1-butene derivative was added aqueous NaOH solution (3N, 2.5 mL) and the mixture was cooled to 0 °C. Aqueous H<sub>2</sub>O<sub>2</sub> solution (30% 1.5 mL) was slowly added to the mixture. The resulting solution was stirred at room temperature for 12 h. The resulting mixture was extracted with Et<sub>2</sub>O and the extracts were washed with water. After drying with anhydrous MgSO<sub>4</sub>, the concentrated mixture was purified by PTLC to give a  $\beta$ -silyl ketone. The enantiomeric excess of this compound was determined by HPLC analysis.

## 2.7 Asymmetric Silaborative C-C Cleavage of *meso*-MCP **1f** in the Presence of Various Phosphine Ligands (Table 2)

### 2.7.1: Silaboration of MCP using low-molecular phosphine ligands (entry 1-3, Table 2)

To a solution of ligand (2.4  $\mu$ mol phosphorous atom) in toluene (100  $\mu$ L) was added Pd<sub>2</sub>dba<sub>3</sub> (0.01 M in toluene, 100  $\mu$ L, 1  $\mu$ mol). The mixture was stirred at room temperature for 5 min. To the mixture was added **2f** (0.15 mmol) and **1** (32.4 mg, 0.10 mmol) in this order, and the resulting mixture was heated at 50 °C with stirring. After 48 h, the volatile materials were evaporated. The residue was analyzed by NMR (internal standard: *o*-xylylene oxide).

### 2.7.2: Silaboration of MCP using PQXphos (entry 4-9, Table 2)

To a solution of PQXphos (2.4  $\mu$ mol phosphorous atom) in toluene (100  $\mu$ L) was added Pd<sub>2</sub>dba<sub>3</sub> (0.01 M in toluene, 100  $\mu$ L, 1  $\mu$ mol). The mixture was stirred at room temperature for 5 min. To the mixture was added **2f** (0.15 mmol) and **1** (32.4 mg, 0.10 mmol) in this order, and the resulting mixture was heated at 50 °C with stirring for 48 h. Subsequent addition of MeCN (10 mL) resulted in precipitation of the PQXphos. The suspension was passed through a pad of Celite® using MeCN as an eluent. After evaporation of the volatiles, the residue was analyzed by NMR (internal standard: *o*-xylylene oxide). To determine the enantiomeric excess, the silylated alkenylborane obtained was converted to  $\beta$ -silyl ketone. To a methanol solution (2 mL) of the 2-boryl-4-silyl-1-butene derivative was added aqueous NaOH solution (3N, 2.5 mL) and the mixture was cooled to 0 °C. Aqueous H<sub>2</sub>O<sub>2</sub> solution (30% 1.5 mL) was slowly added to the mixture. The resulting solution was stirred at room temperature for 12 h. The resulting mixture was extracted with Et<sub>2</sub>O and the extracts were washed with water. After drying with anhydrous MgSO<sub>4</sub>, the concentrated mixture was purified by PTLC to give a  $\beta$ -silyl ketone. The enantiomeric excess of this compound was determined by HPLC analysis.

## 2.8 Dependence of Time Course of Silaboration of *meso*-MCP **1a** on Phosphine Ligands (Table 3)

### 2.8.1: Silaboration of MCP using phosphine ligands (entry 1-7, Table 3)

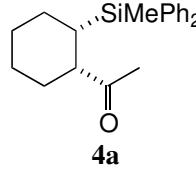
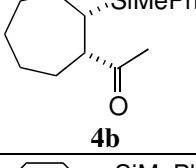
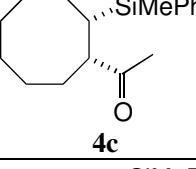
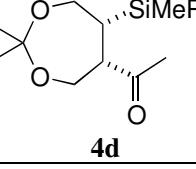
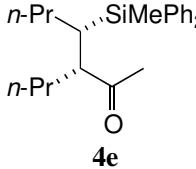
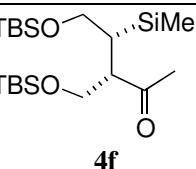
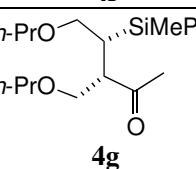
To a solution of ligand (2.4  $\mu$ mol phosphorous atom) in toluene (100  $\mu$ L) was added Pd<sub>2</sub>dba<sub>3</sub> (0.01 M in toluene, 100  $\mu$ L, 1  $\mu$ mol). The mixture was stirred at room temperature for 5 min. To the mixture was added **2a** (0.15 mmol) and **1** (32.4 mg, 0.10 mmol) in this order, and the resulting mixture was heated at 50 °C with stirring. The mixture was analyzed by GC (internal standard: *n*-tetradecane 30  $\mu$ L). After 120 h, the crude product was isolated by silica gel column chromatography to give a 2-boryl-4-silyl-1-butene derivative. To determine the enantiomeric excess, the silylated alkenylborane obtained was converted to  $\beta$ -silyl ketone. To a methanol solution (2 mL) of the 2-boryl-4-silyl-1-butene derivative was added aqueous NaOH solution (3N, 2.5 mL) and the mixture was cooled to 0 °C. Aqueous H<sub>2</sub>O<sub>2</sub> solution (30% 1.5 mL) was slowly added to the mixture. The resulting solution was stirred at room temperature for 12 h. The resulting mixture was extracted with Et<sub>2</sub>O and the extracts were washed with water. After drying with anhydrous MgSO<sub>4</sub>, the concentrated mixture was purified by PTLC to give a  $\beta$ -silyl ketone. The enantiomeric excess of this compound was determined by HPLC analysis.

### 2.8.2: Silaboration of MCP using PQXphos (entry 8-13, Table 3)

To a solution of PQXphos (2.4  $\mu$ mol phosphorous atom) in toluene (100  $\mu$ L) was added Pd<sub>2</sub>dba<sub>3</sub> (0.01 M in toluene, 100  $\mu$ L, 1  $\mu$ mol). The mixture was stirred at room temperature for 5 min. To the mixture was added **2** (0.15 mmol) and **1** (32.4 mg, 0.10 mmol) in this order, and the resulting mixture was heated at 50 °C with stirring. The mixture was analyzed by GC (internal standard: *n*-tetradecane 30  $\mu$ L). Subsequent addition of MeCN (10 mL) resulted in precipitation of the PQXphos. The suspension was passed through a pad of Celite® using MeCN as an eluent. The crude product was isolated by silica gel column chromatography to give a 2-boryl-4-silyl-1-butene derivative. To determine the enantiomeric excess, the silylated alkenylborane obtained was converted to  $\beta$ -silyl ketone. To a methanol solution (2 mL) of the 2-boryl-4-silyl-1-butene derivative was added aqueous NaOH solution (3N, 2.5 mL) and the mixture was cooled to 0 °C. Aqueous H<sub>2</sub>O<sub>2</sub> solution (30% 1.5 mL) was slowly added to the mixture. The resulting solution was stirred at room temperature for 12 h. The resulting mixture was extracted with Et<sub>2</sub>O and the extracts were washed with water. After drying with anhydrous MgSO<sub>4</sub>, the concentrated mixture was purified by PTLC to give a  $\beta$ -silyl ketone. The enantiomeric excess of this compound was determined by HPLC analysis.

**3. Determination of the Enantiomeric Excesses of the Products by HPLC with a Chiral Stationary Phase**

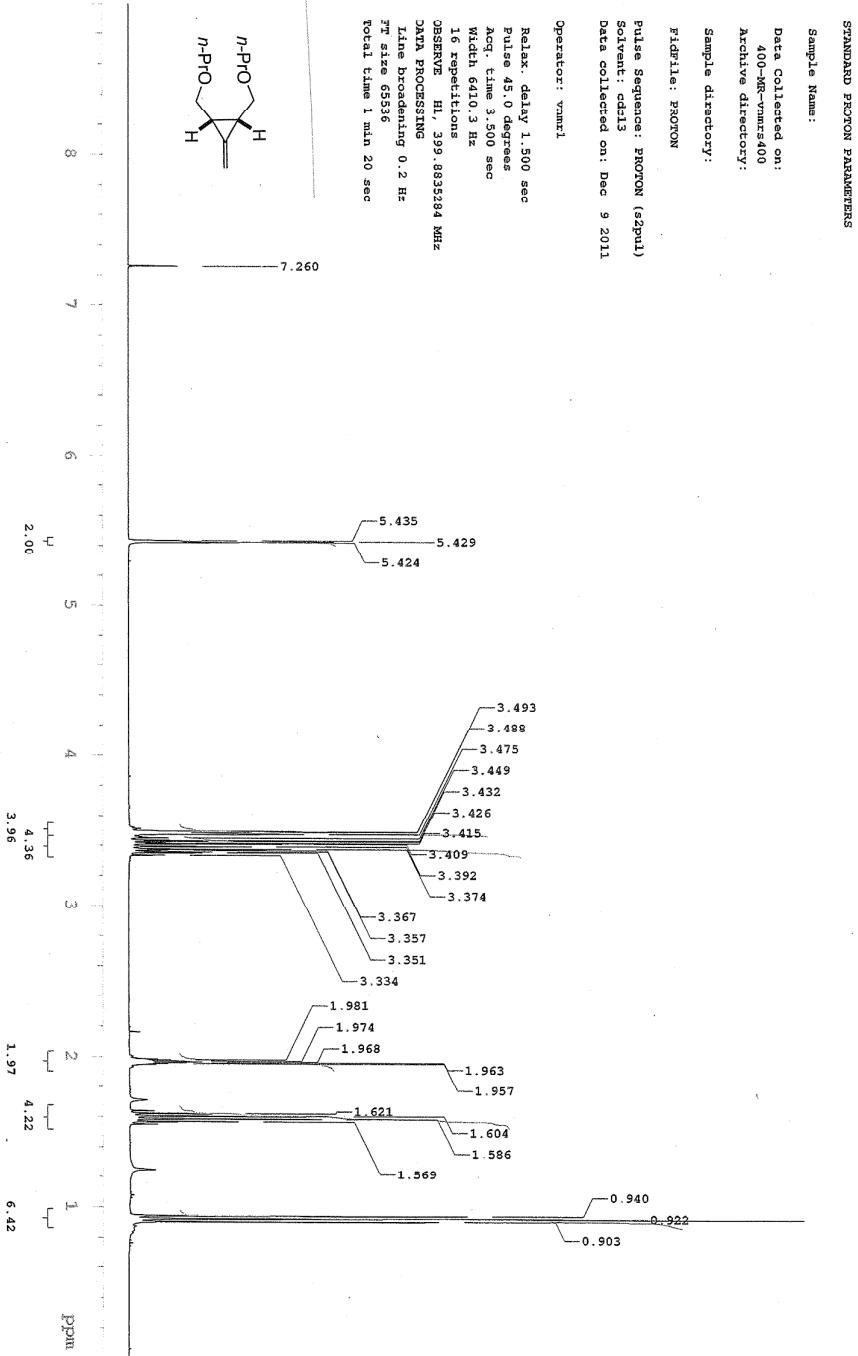
**Table S2.** HPLC separation conditions and retention times of  $\beta$ -silyl ketones

entry	Compound <b>4</b>	column	Eluent: <i>i</i> -PrOH/Hexane	Flowrate (mL/min)	<i>t</i> <sub>R</sub> of (+)-isomer (min)	<i>t</i> <sub>R</sub> of (-)-isomer (min)
1	 <b>4a</b>	OD-H	0.3%	0.6	15.7	13.2
2	 <b>4b</b>	OD-H	0.5%	0.6	9.0	10.1
3	 <b>4c</b>	OD-H	0.3%	0.6	23.6	17.9
4	 <b>4d</b>	OD-H	1.0%	0.6	11.5	14.1
5	 <b>4e</b>	OZ-H	0.3%	0.6	19.6	22.0
6	 <b>4f</b>	OZ-H	0.2%	0.6	10.2	9.4
7	 <b>4g</b>	OZ-H	0.3%	1.0	19.4	14.3

#### 4. References

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(d) Ohmura, T.; Taniguchi, H.; Suginome, M. *J. Am. Soc. Chem.* **2009**, *131*, 11298.
- [S2] Suginome, M.; Matsuda, T.; Ito, Y. *Organometallics* **2000**, *19*, 4647.
- [S3] Yamamoto, T.; Akai, Y.; Nagata, Y.; Suginome, M. *Angew. Chem., Int. Ed.* **2011**, *50*, 8844.
- [S4] Iwasawa, T.; Komano, T.; Tajima, A.; Tokunaga, M.; Obora, Y.; Fujihara, T.; Tsuji, Y. *Organometallics* **2006**, *25*, 4665.
- [S5] Ohta, H.; Tokunaga, M.; Obora, Y.; Iwai, T.; Iwasawa, T.; Fujihara, T.; Tsuji, Y. *Org. Lett.* **2007**, *9*, 89.
- [S6] Yamada, T.; Noguchi, H.; Nagata, Y.; Suginome, M. *J. Polym. Sci. Part A: Polym. Chem.* **2010**, *48*, 898.
- [S7] Yamamoto, T.; Suginome, M. *Angew. Chem., Int. Ed.* **2009**, *48*, 539.

## 5. NMR Spectra of New Compounds



<sup>1</sup>H NMR of compound **2g**

STANDARD CARBON PARAMETERS

Sample Name:

Data Collected on:

400-MR-Mainframe

Archive directory:

Sample directory:

Fidfile: CARBON

Pulse Sequence: CARBON (s2pul1)

Solvent: CDCl<sub>3</sub>

Data collected on: Dec 9 2011

Operator: vnmri

Relax. delay 0.715 sec

Pulse 45.0 degrees

Acq. time 1.285 sec

Width 2551.2 Hz

24 repetitions

OBSERVE C13, 100.5507935 MHz

DECOUPLE H1, 399.8855346 MHz

Power 42 dB

continuously on

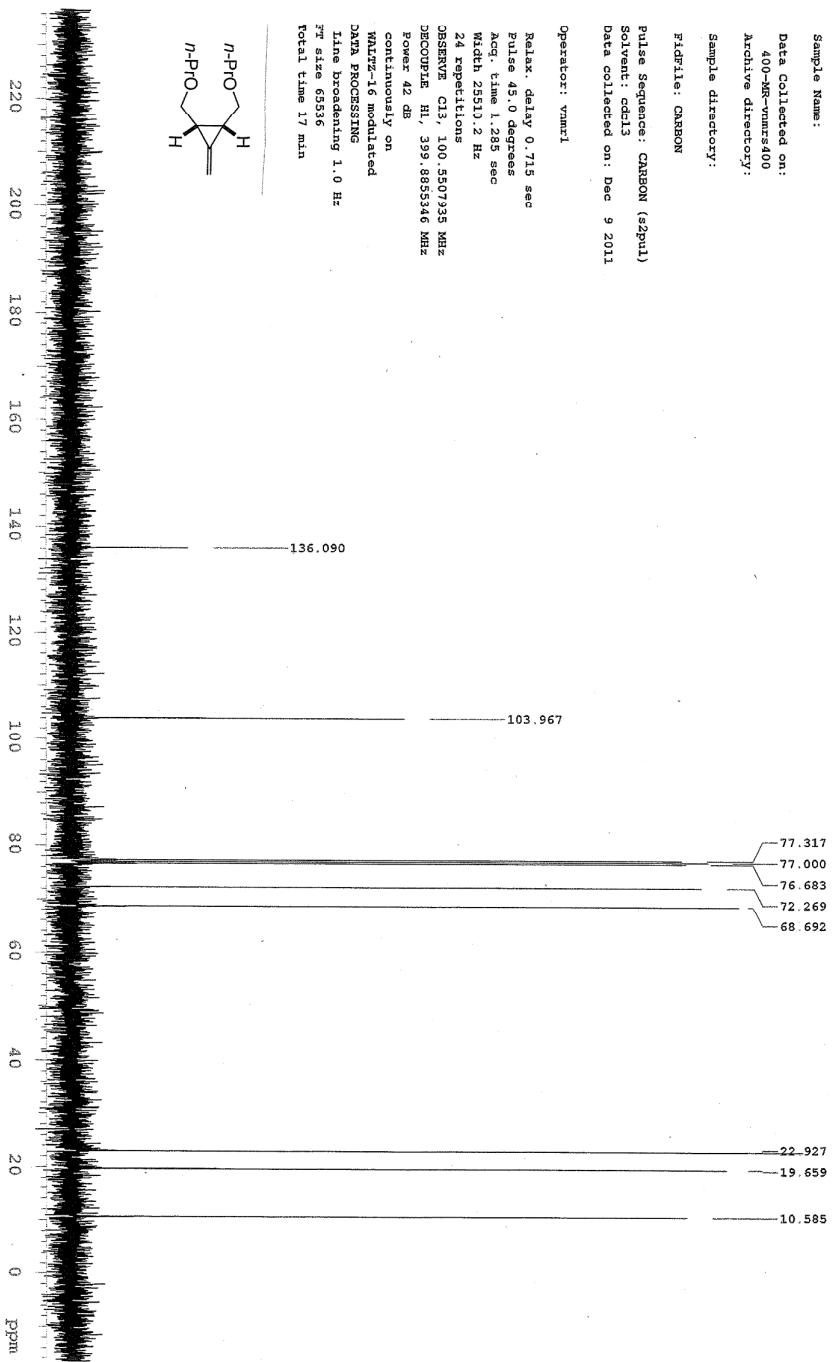
WIDET-16 modulated

DATA PROCESSING

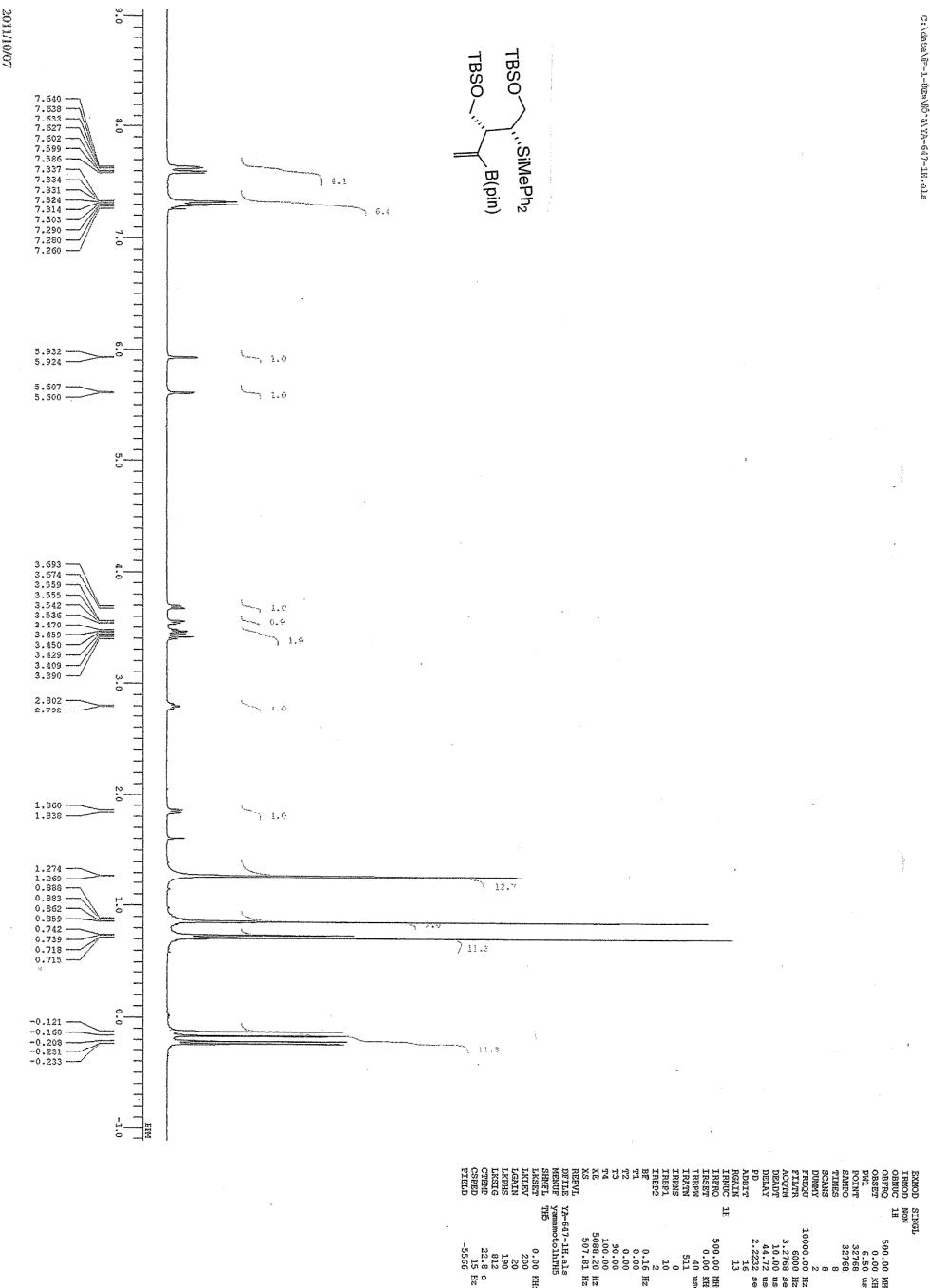
Line broadening 1.0 Hz

FT size 65536

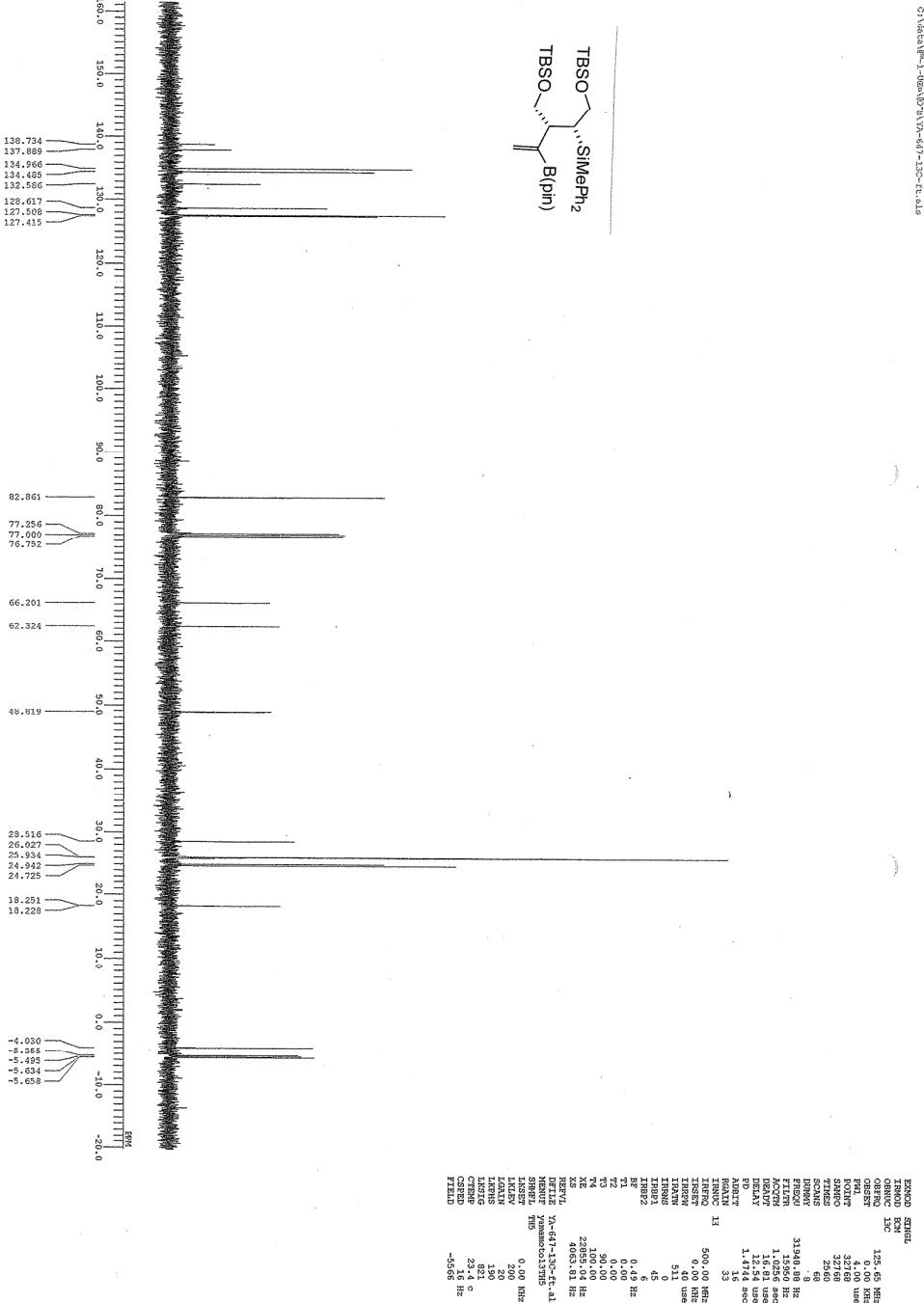
Total time 17 min



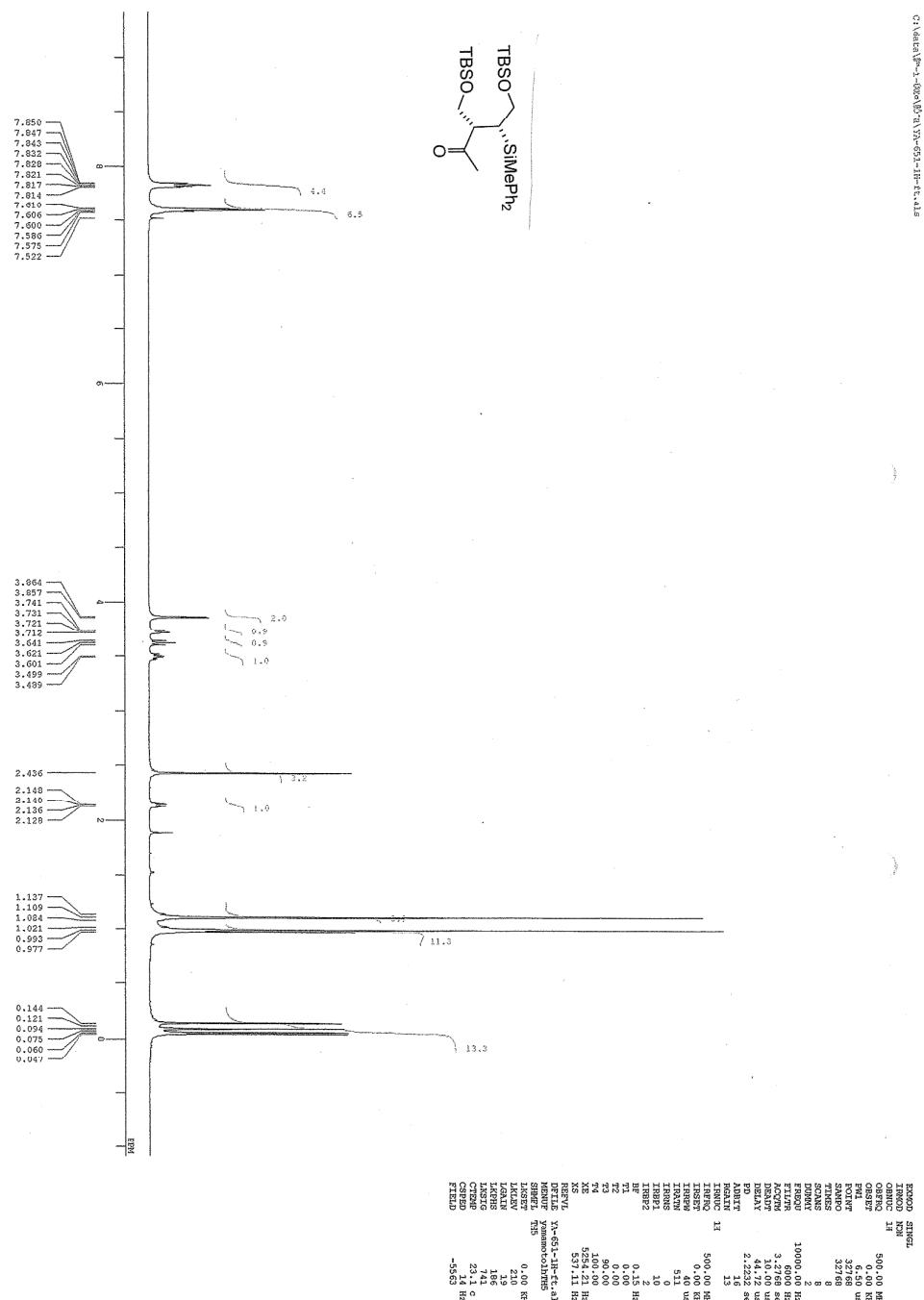
<sup>13</sup>C NMR of compound 2g



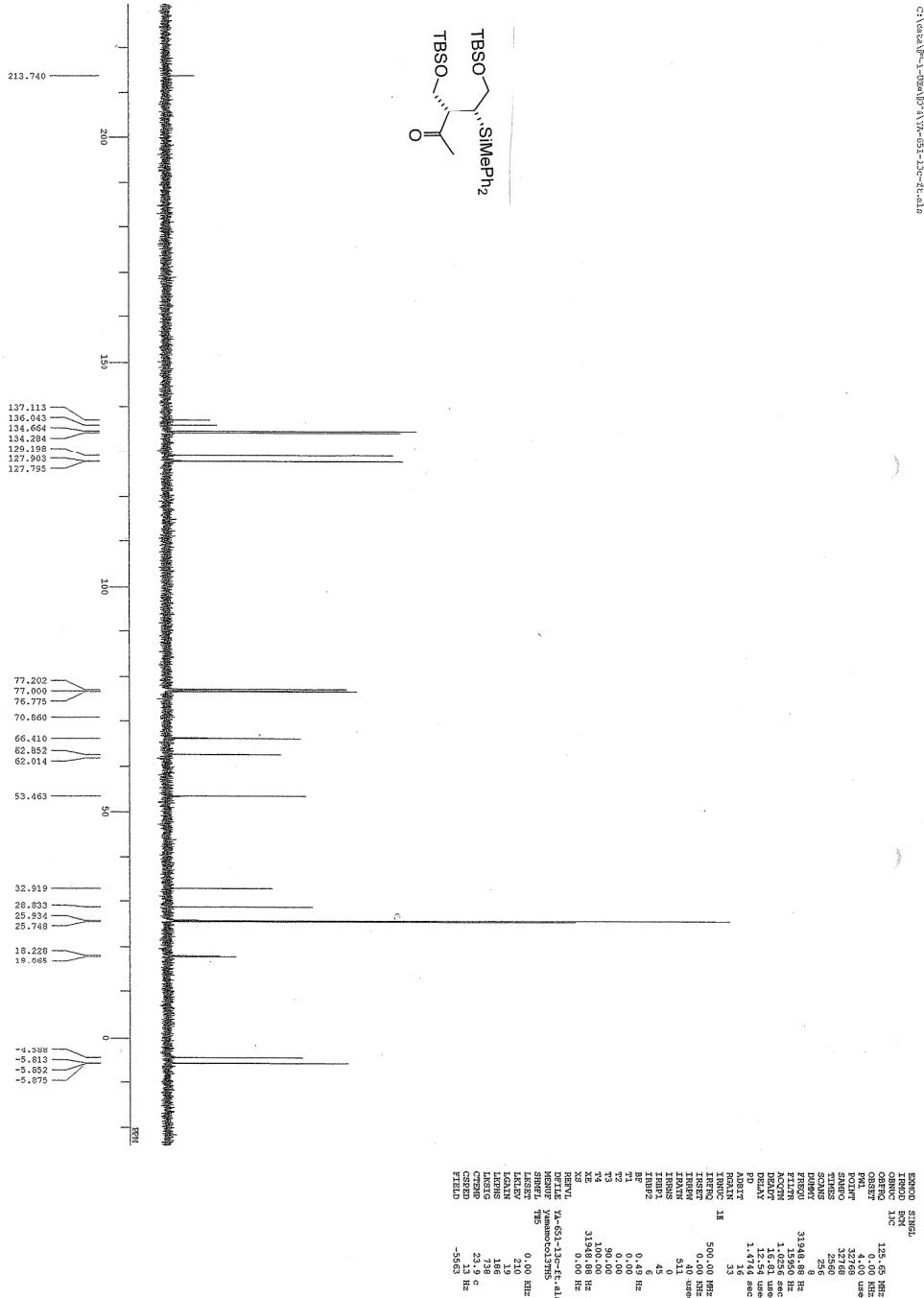
<sup>1</sup>H NMR of compound **3f**



### <sup>13</sup>C NMR of compound 3f



<sup>1</sup>H NMR of compound **4f**



<sup>13</sup>C NMR of compound 4f

STANDARD PROTON PARAMETERS

Sample Name:

Data Collected on:

400-MR-united00

Archive directory:

Sample directory:

File: PROTON

Pulse Sequences: PROTON (s2pul)

Solvent: ccl4:13

Data collected on: Dec 21 2011

Operator: vmlml

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Pulse 45.0 degrees

Acq. time 3.500 sec

Width 6410.3422 sec

16 repetitions 324

GSIMON: 1.1

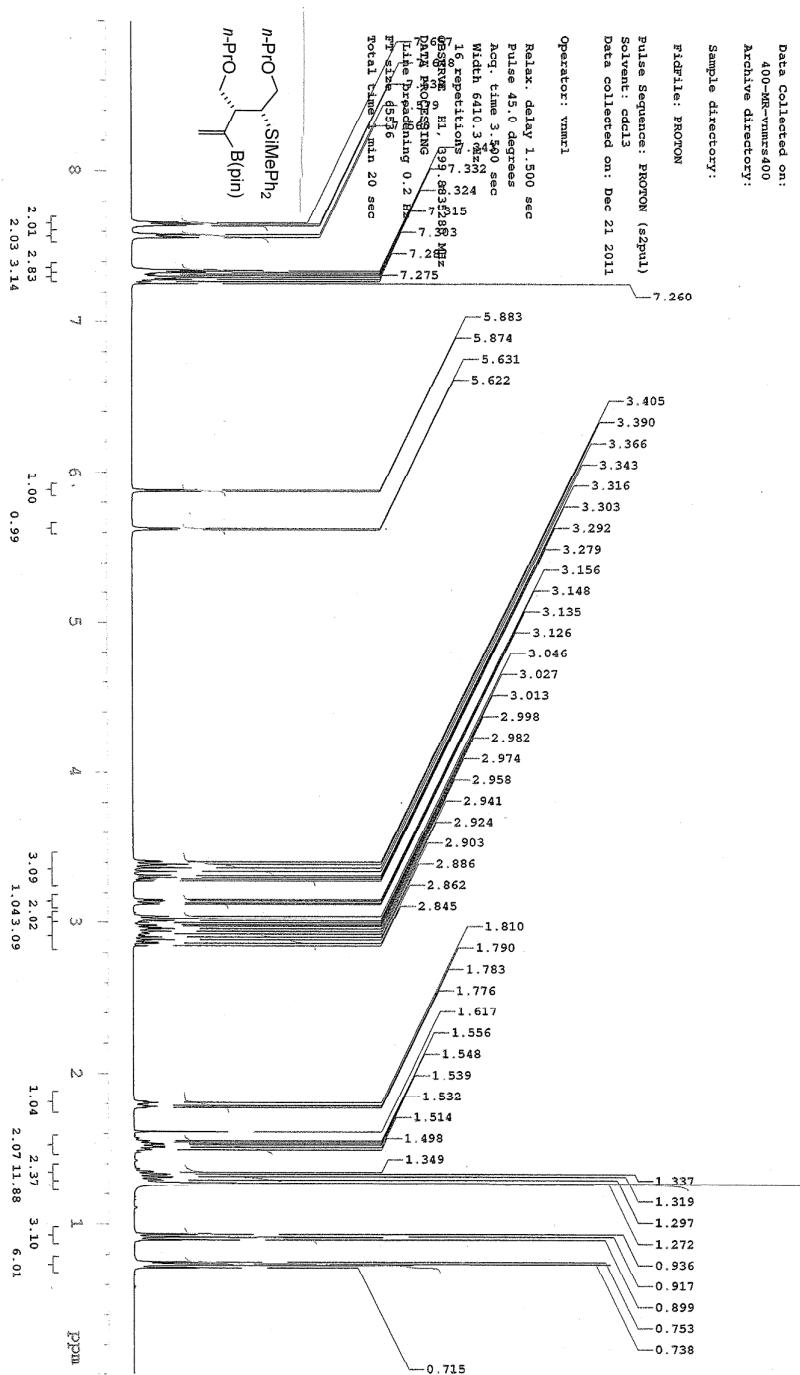
PSIMON: 1.0

RTA PROGRESS 0.2

Line broadening 0.2 Hz

Print time 0.576

Total time 20 sec



STANDARD CARBON PARAMETERS

Sample Name:

Data Collected on:

400-MR-unms100

Archive directory:

Sample directory:

FIGFILE: CARBON

Pulse Sequence: CARBON (s2pul)

Solvent: cdcl3

Data collected on: Dec 21 2011

Operator: vnmrl

Relax. delay 0.715 sec

Pulse 45.0 degrees

Acq. time 1.281 sec

Width 2350.2 Hz

56 repetitions

observe c13 100.5507951 MHz

decouple el 399.8855346 MHz

Power 42 dB

continuously on

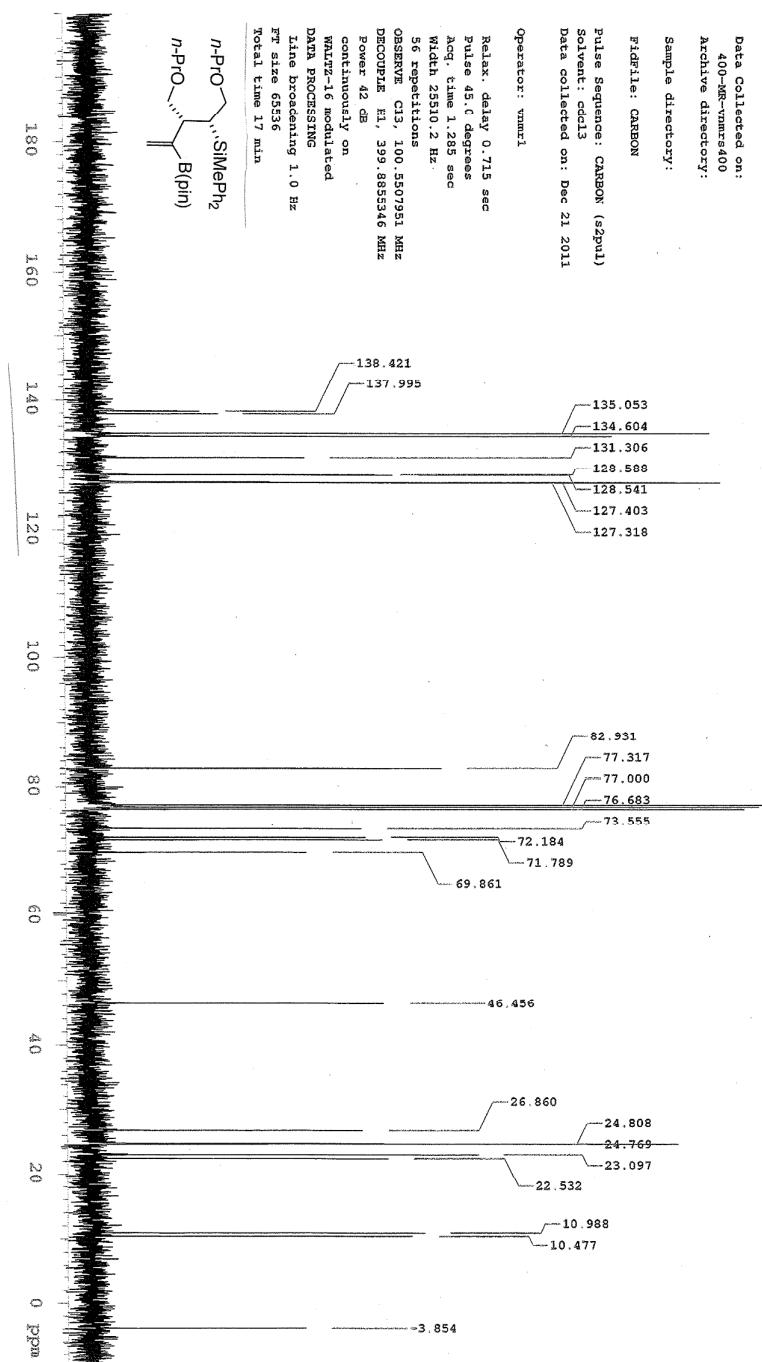
WAWW-16 modulated

DATA PROCESSING

line broadening 1.0 Hz

fft size 6536

Total time 17 min



$^{13}\text{C}$  NMR of compound 3g

## STANDARD PROTON PARAMETERS

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Data Collected on:

400-RR-vnmr400

Archive directory:

Sample directory:

Profile: PROTON

Pulse Sequence: PROTON (s2pul)

Data collected on: Dec 21 2011

Operator: vnmr1

Relax. delay 1.000 sec

Pulse 45.0 degrees

Acq. time 3.500 sec

Width 64.0 Hz

16 Scans

Observe 7.4, 7.39, 5.9, 5.28, 2.80

Data Processing 7.567

Line broadening 12 Hz

PPM

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7.562

7.361

7.358

7.351

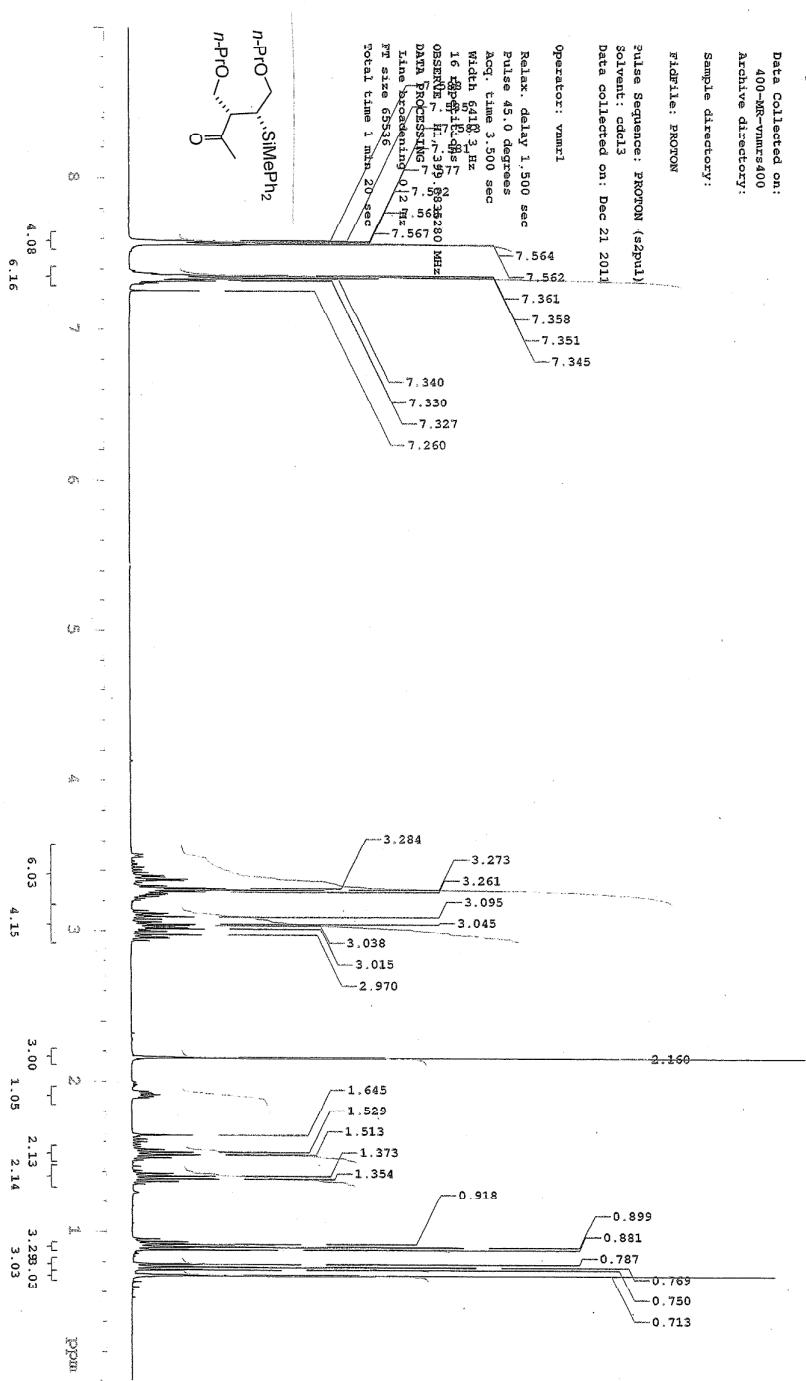
7.345

7.340

7.330

7.327

7.260



1H NMR of compound 4g

STANDARD CARBON PARAMETERS

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Data Collected on:  
400-MR-mars400

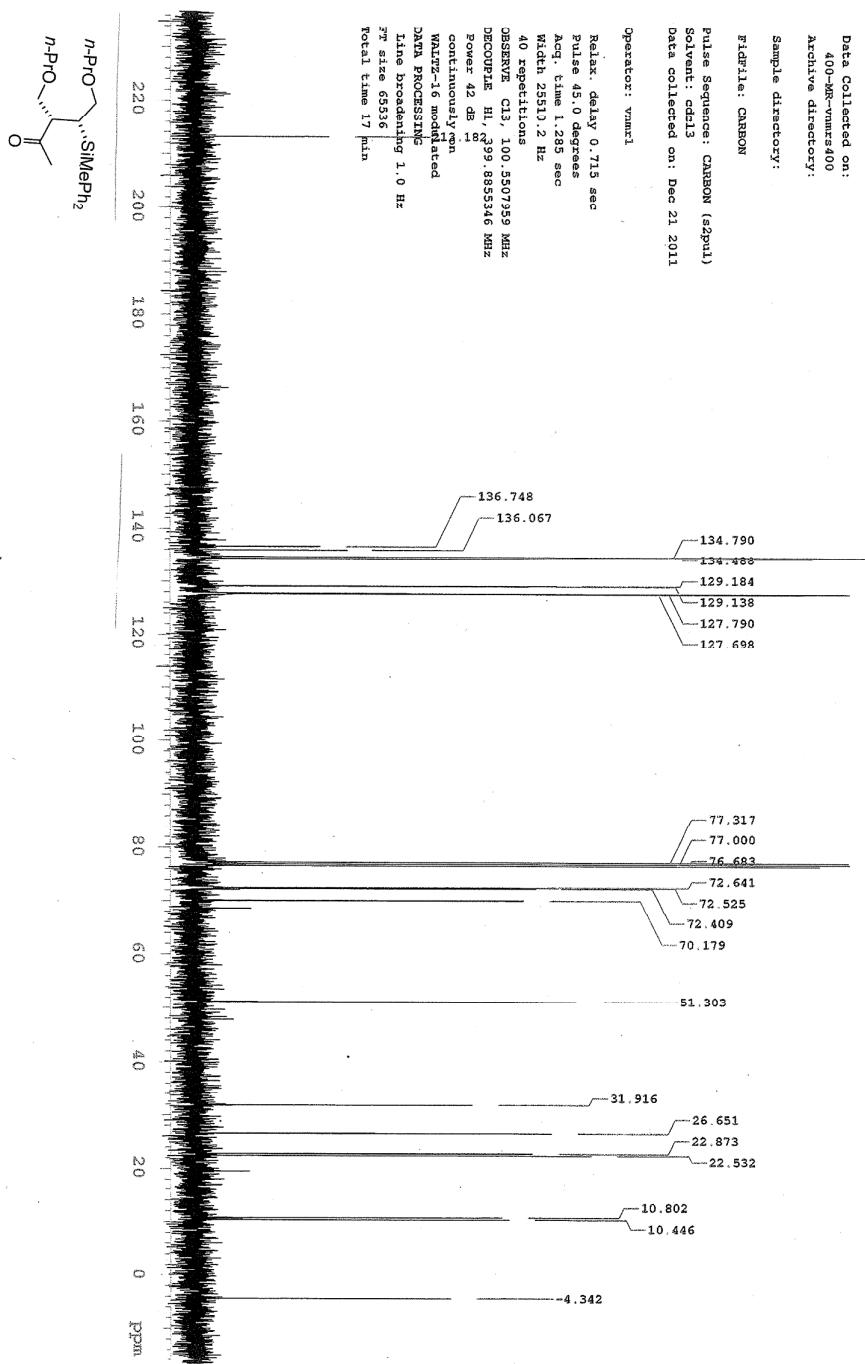
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Sample directory:  
FILE: CARBON

Pulse Sequence: CARBON (s2pml)  
Solvent: cd13  
Date collected on: Dec 21 2011

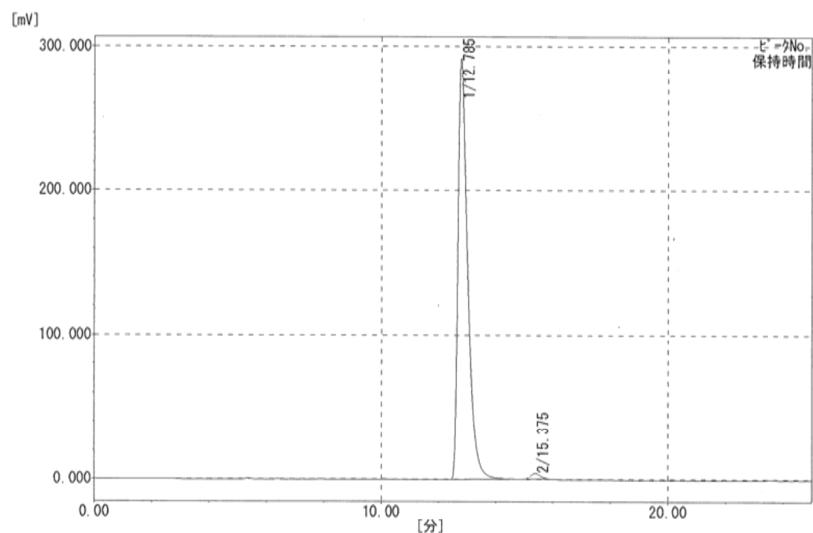
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Pulse 45.0 degrees  
Acq. time 1.285 sec  
Width 2551.2 Hz

40 repetitions  
OBSERVE C13, 100.550759 MHz  
DECOUPLE H1, 399.8855346 MHz  
Power 42 dB 18  
continuously on  
WALTZ-16 modulated  
DATA PROCESSING  
Line broadening 1.0 Hz  
PPM size 65536  
Total time 17 min



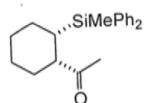
$^{13}\text{C}$  NMR of compound 4g

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 カップ番号 : 56  
 収集属性 : 独立 0.00 - 25.00 min  
 計算方法 : 百分率法

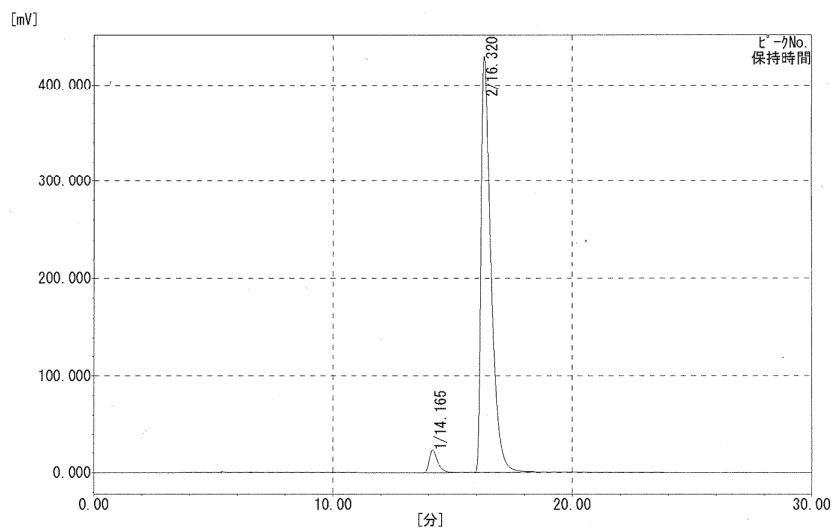


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2	1	15.375	3.87	22.58	90.96	1.32	9250	4.25	1.38

CH.1 Peak Not Found.



タイトル :  
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 計算方法 : 百分率法



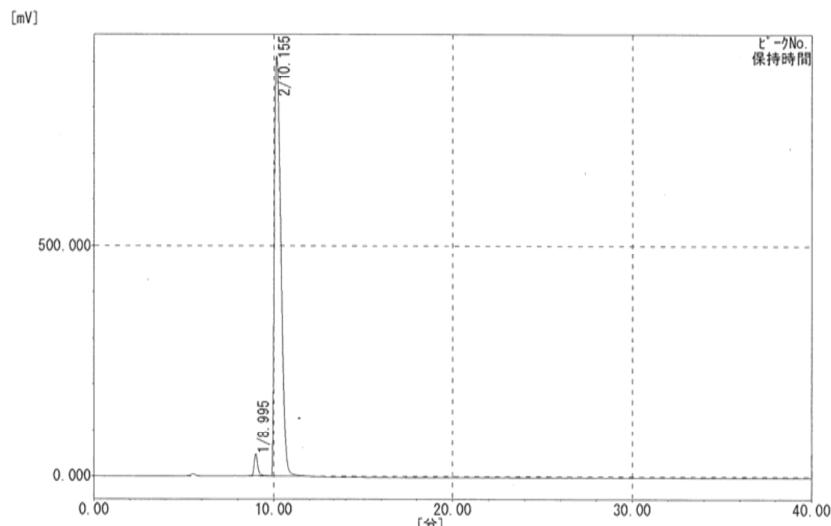
No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	14.165	22.93	21.62	566.20	4.37	8560	0.00	1.82
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CH.1 Peak Not Found.



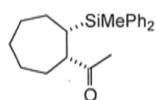
HPLC trace for *ent*-4a (Scheme 1)

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 計算方法 : 百分率法

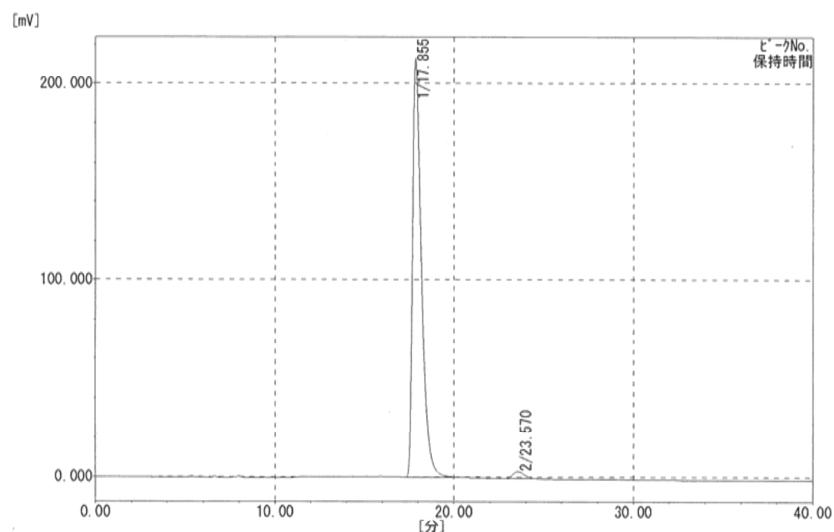


No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	8.995	47.33	11.98	652.26	2.70	11237	0.00	1.48
2	1	10.155	913.50	24.71	23525.97	97.30	3368	2.23	2.53

CH.1 Peak Not Found.

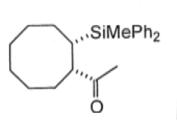


タイトル :  
 データ : C:\HPLCData\data\_akai.mdb YNO3970383 YA-560  
 日時 : 収集 2011/06/26 18:21:33 計算 2012/06/01 17:00:26  
 カップ番号 : 59  
 収集属性 : 独立 0.00 - 40.00 min  
 計算方法 : 百分率法

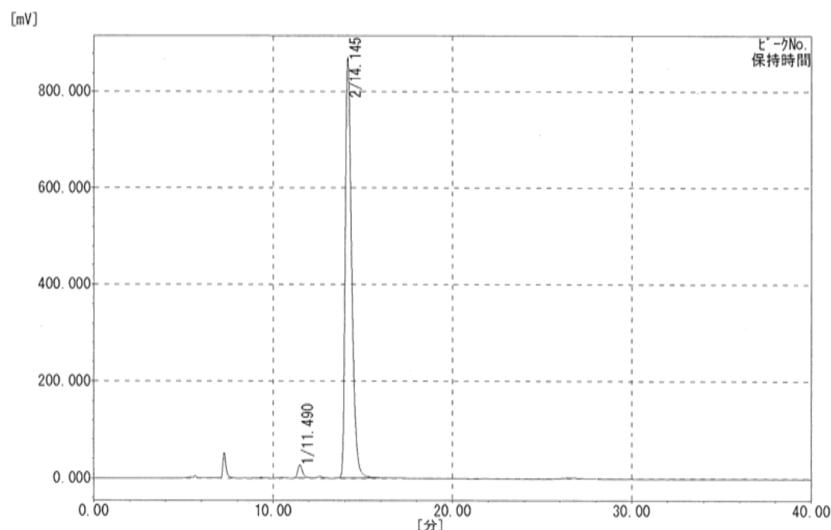


No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mV×秒]	面積%	理論段数	分離能	非対称係数
1	1	17.855	213.34	30.85	7396.49	97.92	6680	0.00	2.29
2	1	23.570	3.53	41.87	157.22	2.08	6319	5.55	1.30

CH.1 Peak Not Found.

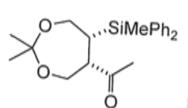


タイトル :  
 データ : C:\HPLCData\data\_akai.mdb YN03970400 YA-566  
 日時 : 収集 2011/07/06 21:45:56 計算 2012/06/01 16:56:21  
 カップ番号 : 8  
 収集属性 : 独立 0.00 - 40.00 min  
 計算方法 : 百分率法

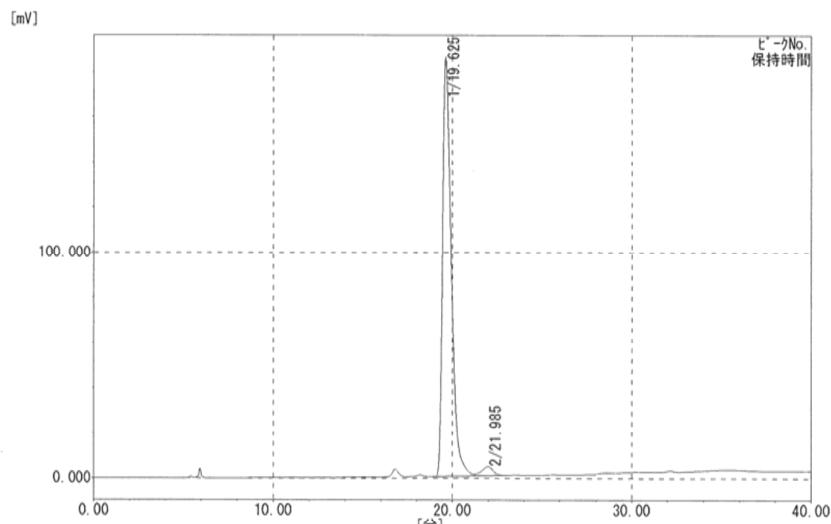


No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	11.490	26.43	16.05	460.76	2.10	10216	0.00	1.32
2	1	14.145	872.52	22.25	21497.25	97.90	8060	4.90	1.84

CH.1 Peak Not Found.

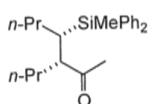


タイトル :  
 データ : C:\HPLCData\data\_akai.mdb YN03970385 YA-559  
 日時 : 収集 2011/06/26 20:54:46 計算 2012/06/01 16:59:10  
 カップ番号 : 58  
 収集属性 : 独立 0.00 - 40.00 min  
 計算方法 : 百分率法

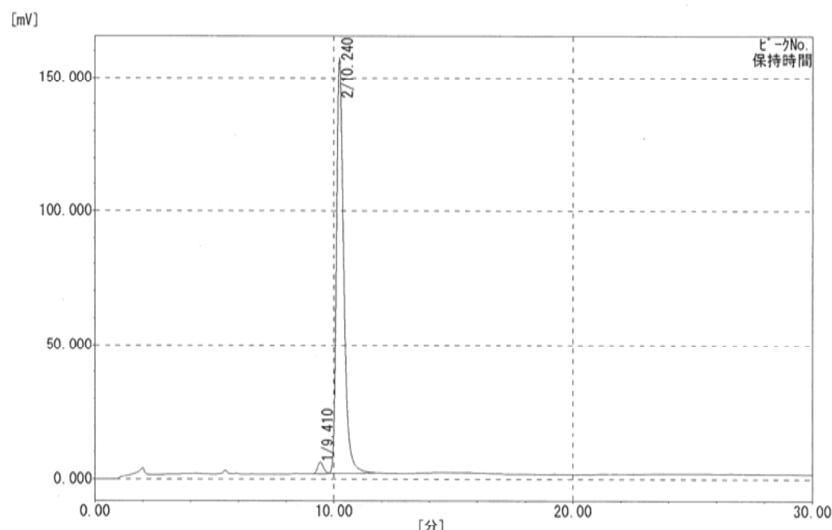


No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	19.625	185.95	30.35	6262.46	97.26	8336	0.00	2.04
2	1	21.985	3.95	40.85	176.32	2.74	5777	2.34	0.00

CH.1 Peak Not Found.

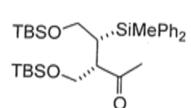


タイトル :  
 データ : C:\HPLCData\data\_akai.mdb YN03970565 b-silylketone  
 日時 : 収集 2011/12/05 14:41:45 計算 2012/06/01 17:07:52  
 カップ番号 : 86  
 収集属性 : 独立 0.00 - 30.00 min  
 計算方法 : 百分率法

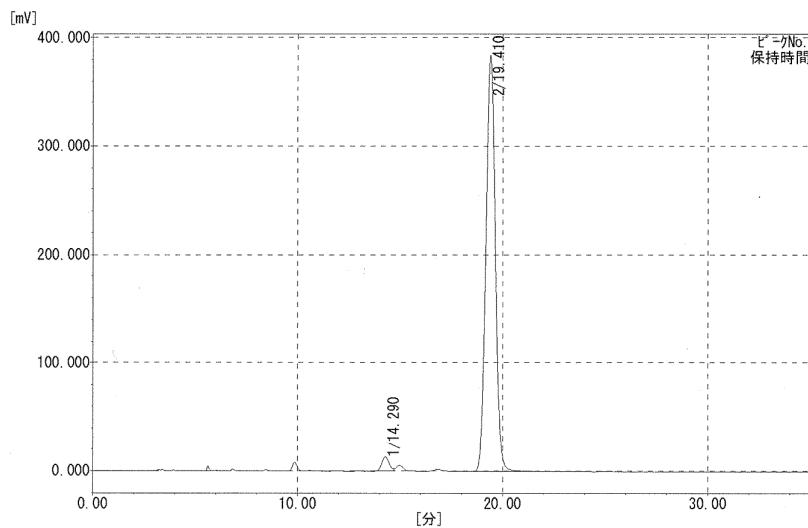


No.	Ch	時間[分]	高さ[mV]	半値幅[秒]	面積[mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	9.410	4.28	15.84	73.81	2.18	7037	0.00	1.49
2	1	10.240	155.48	18.92	3317.14	97.82	5841	1.69	1.52

CH.1 Peak Not Found.



タイトル :  
 データ : C:\HPLCData\data\_akai.mdb CLC-0692 YA  
 日時 : 収集 2012/06/05 14:08:33 計算 2012/06/05 15:21:32  
 カップ番号 : 96  
 収集属性 : 独立 0.00 - 35.00 min  
 計算方法 : 百分率法



No.	Ch	時間 [分]	高さ [mV]	半値幅 [秒]	面積 [mVx秒]	面積%	理論段数	分離能	非対称係数
1	1	14.290	12.92	23.24	319.32	2.54	7538	0.00	0.00
2	1	19.410	384.39	29.49	12236.53	97.46	8638	6.86	1.05

CH.1 Peak Not Found.

