Supporting Information

2 Degradation and Metabolism of Tetrabromobisphenol A (TBBPA) in

- 3 Submerged Soil and Soil-Plant Systems
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HPLC analysis

HPLC was performed on a Nucleosil C18 column (250 mm×4 mm) at 40 °C using an Agilent HPLC Series 1100 (Agilent Technologies, Germany) coupled to a diode array detector and an online radio flow detector (RAMONA Star; Raytest, Straubenhardt, Germany). The mobile phase consisted of water (A) and acetonitrile (B) and flowed at 0.45 mL min⁻¹, using a gradient program as follows: 20% B with linear gradient to 40% B within 3 min, to 60% B within 27 min, 60% B isocratically for 8 min, to 100% B within 1 min, 100% B isocratically for 4 min, back to 20% B within 5 min, and 20% B for 2 min. The flow rate of the scintillation cocktail (Quicksafe Flow 2; Zinsser Analytic GmbH, Frankfurt, Germany) for the radiodetector was 2 mL min⁻¹. The UV signal was recorded at 254 nm.

Purification of metabolites

The extracts of day 66 were fractioned by HPLC by repeated injection of 100- μ L aliquots. The HPLC ran under the conditions described above. The eluents were collected in seven fractions according to the retention times of the radioactive peaks (Figure S2). The fractions were concentrated to dryness under a stream of nitrogen gas and re-suspended in $200~\mu$ L of acetonitrile for gas chromatography-mass spectrometry (GC-MS) analysis.

GC-MS analysis

The purified metabolites in acetonitrile (90 μ L) were derivatized using N,O-bis(trimethylsilyl)trifluoroacetamide (10 μ L) at 75 °C for 15 min. One μ L was then injected onto a ZB-5HT Inferno column (32 m × 250 μ m × 0.25 μ m; Phenomenex; Germany) in a gas chromatograph (Agilent 7890A series; USA) coupled to a mass spectrometer (Agilent 5975C series; USA), using a programmed temperature vaporizer inlet (60 °C for 0.16 min, then 60 °C/min to 300 °C for 10 min). The temperature gradient was 0 °C/min at 60 °C for 1 min, 30 °C/min to reach 150 °C, followed by 5 °C/min to reach 300 °C, which was held for 5 min. The transfer line was set at 280 °C, the MS ion source at 200 °C, and the quadrupole at 150 °C. Mass spectra were recorded in full scan mode at m/z = 60–700. The trimethylsilyl (TMS) derivatives of compounds 1–12 are shown in Figure S4–S15, respectively.

Determination of radioactivity

The radioactivity in the samples was quantitatively counted in a liquid scintillation counter (LSC; Tri-Carb 2800 TR; PerkinElmer, USA). For radioactivity in water and organic extracts, 1 mL of water or extract was mixed with 18 mL of scintillation cocktail (Hionic-Fluor, PerkinElmer). For the NaOH solution containing ¹⁴CO₂, 2 mL was mixed with 18 mL of scintillation cocktail (Ultima-Flo, PerkinElmer). For radioactivity in the plants and the soil-bound residues, dried plant material (leaves, roots, or total plants) and 150 mg of air-dried soil was combusted on a sample oxidizer (model 307; PerkinElmer). The generated CO₂ was absorbed by 18 mL of Carbo-Sorb E (PerkinElmer) and counted by LSC.

Table S1. GC-MS (after HPLC fractionation) identification of TBBPA metabolites in organic extracts prepared from ¹⁴C-TBBPA-spiked soils planted or not with rice or reed seedlings and incubated for 66 days.^a

Compound	Name	Structure	HPLC	Rt on	Identified fragments of	Soil treatment
no.			fraction	GC	TMS-derivatives [m/z (% relative	
			no. ^b	(min)	abundance) (fragment)] ^c	
1	2-(3-bromo-4-hydroxyphenyl)-2	Br Br	3, 4	33.889	606/608/610(30)/612 (M ⁺)	Rice soil
	-(3,5-dibromo-4-hydroxyphenyl)	но()()он			519/593/595(100)/597 ([M – Me] ⁺)	Reed soil
	-propane	Br			363/365(10)/367 ([Me – A] ⁺)	Unplanted soil
	(Tribromobisphenol A,				$285/287(18) ([Me - B]^{+})$	
	triBBPA)				$137/139(42) ([Me_2SiBr]^+)$	
2	2-bromo-4-[1-(3-bromo-4-hydro Br Br Br	3, 4	30.695	528/530(22)/532 (M ⁺)	Rice soil	
	xyphenyl)-1-methylethyl]phenol	но—Он			513/515(100)/517 ([M – Me] ⁺)	Reed soil
	(Dibromobisphenol A, diBBPA)				$285/287(14) ([Me - B]^{+})$	Unplanted soil
					137/139(20) ([Me ₂ SiBr] ⁺)	
3	2-bromo-4-[2-(4-hydroxyphenyl	но—Он	3	27.297	450/452(19) (M ⁺)	Rice soil
)propan-2-yl]phenol				$435/437(100) ([M - Me]^{+})$	Reed soil
	(Monobromobisphenol A,	Br			, , , ,	
	monoBBPA)					
4	4,4'-(propane-2,2-diyl)diphenol	но——————	2	23.679	372(13) (M ⁺)	Rice soil
	(Bisphenol A, BPA)				$357 (100) ([M - Me]^{+})$	Reed soil,
					207(21) ([M - D] ⁺)	
5	2-bromo-4-(2-(3,5-dibromo-4-m	Br Br O	7	32.203	548/550/552(38)/554 (M ⁺)	Rice soil
	ethoxyphenyl)propan-2-yl)phen				$533/535/537(100)/539 ([M - Me]^{+})$	Unplanted soil
	ol	Br			455/457(12)/459 ([M – Me – Br] ⁺)	•

	(Tribromobisphenol A monomethyl ether,				285/287(20) ([Me – B] ⁺) 305/307(16)/309 ([Me – C] ⁺)	
6	MeO-triBBPA) 1,3-dibromo-5-(2-(3-bromo-4-m ethoxyphenyl)propan-2-yl)-2-m ethoxybenzen (Tribromobisphenol A dimethyl	Br Br O	7	32.066	490/492/494(31)/496 (M ⁺) 475/477/479/(100)481 ([M – Me] ⁺)	Rice soil Unplanted soil
	ether; diMeO-triBBPA)					
7	2,6-dibromo-4-(2-(3,5-dibromo-4-methoxyphenyl)propan-2-yl)p henol	HO Br Br	7	35.501	626/628/630(29)/632/634 (M ⁺) 611/613/615(100)/617/619 ([M – Me] ⁺)	Rice soil Reed soil Unplanted soil
	(Tetrabromobisphenol A monomethyl ether; MeO-TBBPA)				305/307(30)/309 ([M – C] ⁺), 363/365(23)/367 ([M – A] ⁺) 137/139(59) ([Me ₂ SiBr] ⁺)	
8	2,2-bis(3,5-dibromo-4-methoxy phenyl)propane (Tetrabromobisphenol A dimethyl ether; diMeO-TBBPA)	Br Br O	7	34.029	568/570/572(38)/574/576 (M ⁺) 553/555/557(100)/559/561 ([M – Me] ⁺) 459/461/463(6)/465 ([M – Br – OMe] ⁺) 305/307(26)/309 ([M – C] ⁺),	Rice soil Unplanted soil
9	2-bromo-4(propen-2-yl)-phenol	OH Br	2	12.268	284/286(69) (M ⁺) 269/271(100) ([M – Me] ⁺) 137/139(73) ([Me ₂ SiBr] ⁺)	Unplanted soil

10	1,3-dibromo-2-methoxy-5-vinyl benzene	Br Br	2	14.215	290/292(100)/294 (M ⁺) 275/277(27)/279 ([M – Me] ⁺) 196/198(12) ([M – Me – Br] ⁺)	Unplanted soil
11	3,5-dibromo-4-methoxyphenol, or 2,6-dibromo-4-methoxyphenol	Br Br Br Or OH	2	14.259	352/354(63)/356 (M ⁺) 337/339(100)/341 ([M – Me] ⁺)	Unplanted soil
12	2,6-dibromo-4(propen-2-yl)-phe nol	OH Br Br	2	16.424	362/364(46)/366 (M ⁺) 347/349(100)/351 ([M – Me] ⁺) 268/270(34) ([M – Me – Br] ⁺) 137/139(35) ([Me ₂ SiBr] ⁺)	Unplanted soil

definition and the characteristic mass fragments that are similar to the methyl ethers of TBBPA (compounds 7–8); compounds 9–12 were identified according to the bromine according to the bromine isotope pattern and the corresponding bromine-free counterpart molecules.

^b Fractions on radio-HPLC; Fraction 1: 1–5 min, 2: 5–15 min, 3: 15–25 min, 4: 25–31 min, 5: 31–34 min, 6: 34–42 min, and 7: 42–50 min. For details, see Figure S2A

 c [M – A] $^{+}$, [M – B] $^{+}$, [M – C] $^{+}$, and [M – D] $^{+}$ represent fragments 2,6-dibromo-4-(2-propyl)-trimethylsilyloxybenzene,

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2-bromo-4-(2-propyl)-trimethylsilyloxybenzene, 2,6-dibromo-4-(2-propyl)-methoxybenzene, and 4-(2-propyl)-trimethylsilyloxybenzene, respectively.

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2,6-dibromo-4-(2-propyl)-trimethylsilyloxybenzene

2,6-dibromo-4-(2-propyl)-methoxybenzene

2-bromo-4-(2-propyl)-trimethylsilyloxybenzene

4-(2-propyl)-trimethylsilyloxybenzene

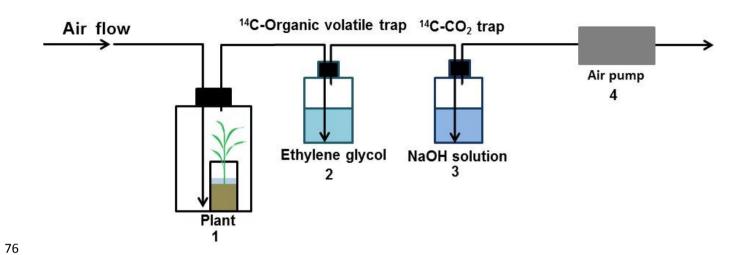


FIGURE S1. Flow-through system for the degradation of TBBPA in saturated soil with and without plant growth. Fresh air was introduced into the system at a rate of 500 mL h^{-1} by an air pump. Legend: 1, planted flask; 2, monoethylene glycol vessel for trapping volatile organic compounds; 3, NaOH solution vessel for trapping CO₂; 4, air pump.

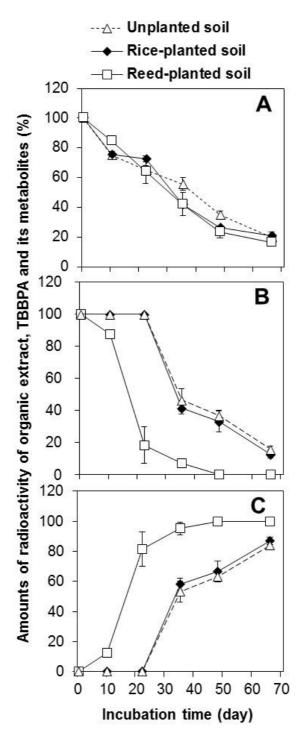


FIGURE S2. Radioactivity recovered from total organic extracts (A), TBBPA (B), and metabolites (C) in submerged soil during incubation with 14 C-TBBPA with or without growth of rice or reed seedlings. Amounts of radioactivity in TBBPA and its metabolites are shown as relative values to the total radioactivity in the extracts. Data are the means of three individual experiments \pm one standard deviation.

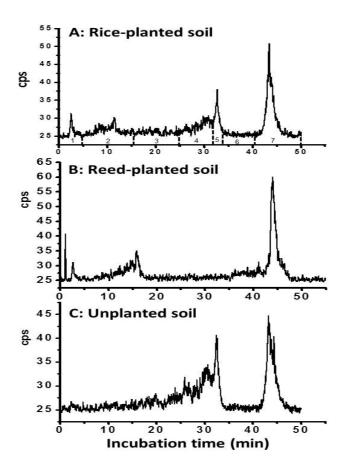


FIGURE S3. HPLC Radio-chromatograms of organic extracts of ¹⁴C-TBBPA-spiked soils planted with rice (A) or reed (B) seedlings or left unplanted (C) and incubated for 66 days. The extracts were fractionated into seven parts according to their retention time on HPLC (shown in A) for further GC-MS analysis.

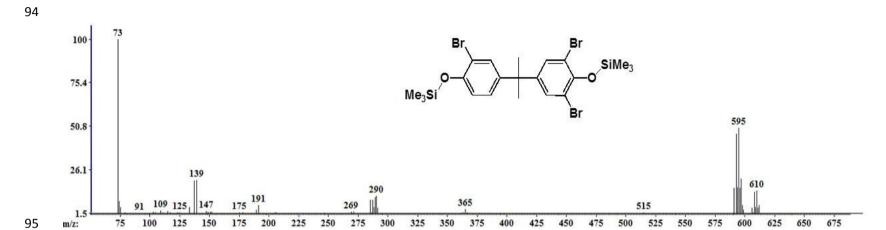


FIGURE S4. Mass spectra of the TMS derivative of tribromobisphenol A (triBBPA) (compound 1 in Table S1)

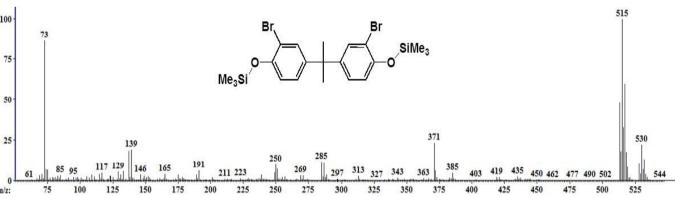


FIGURE S5. Mass spectrum of the TMS derivative of dibromobisphenol A (diBBPA) (compound 2 in Table S1)



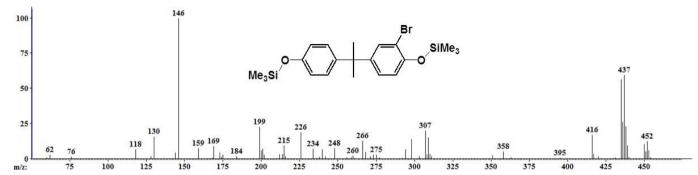


FIGURE S6. Mass spectrum of the TMS derivative of monobromobisphenol A (monoBBPA) (compound 3 in Table S1)



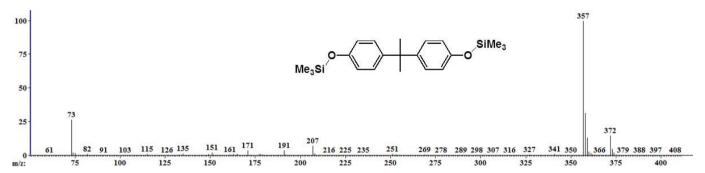


FIGURE S7. Mass spectrum of the TMS derivative of bisphenol A (BPA) (compound 4 in Table S1)

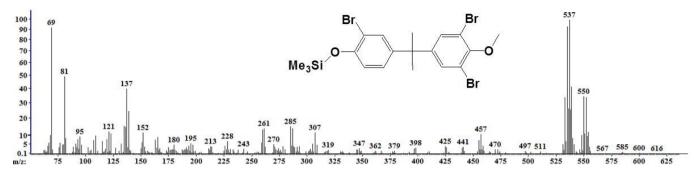


FIGURE S8. Mass spectrum of the TMS derivative of tribromobisphenol A monoethyl ether (MeO-triBBPA) (compound 5 in Table S1)

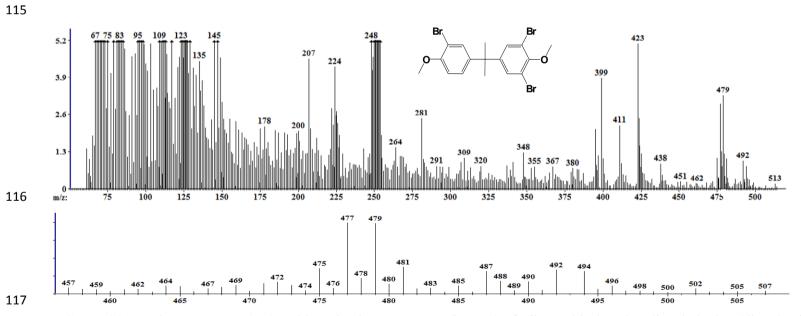


FIGURE S9. Total mass spectra (top) and bromine isotope patters (bottom) of tribromobisphenol A dimethyl ether (diMeO-triBBPA) (compound 6 in Table S1)

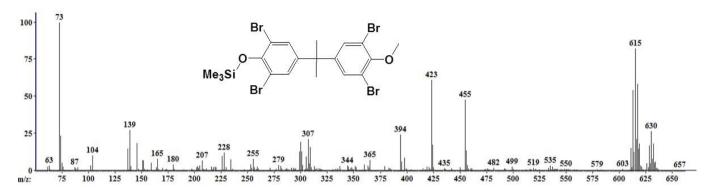


FIGURE S10. Mass spectrum of the TMS derivative of tetrabromobisphenol A monoethyl ether (MeO-TBBPA) (compound 7 in Table S1)

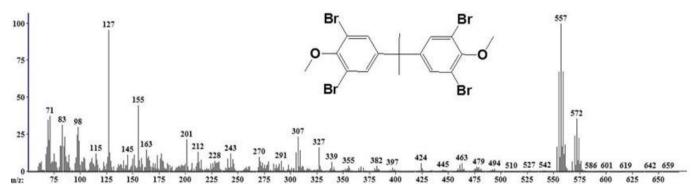


FIGURE S11. Mass spectrum of TBBPA dimethyl ether (diMeO-TBBPA) (compound 8, Table S1)

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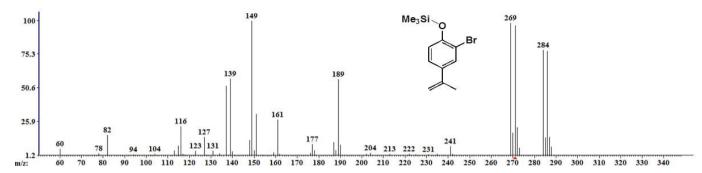


FIGURE S12. Mass spectrum of the TMS derivative of 2-bromo-4(propen-2-yl)-phenol (compound 9, Table S1)

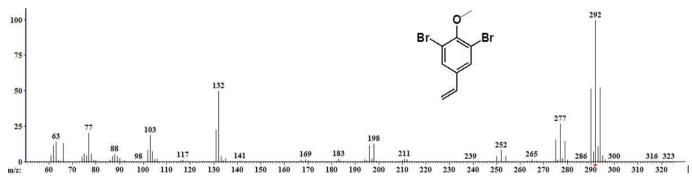


FIGURE S13. Mass spectrum of 1,3-dibromo-2-methoxy-5-vinylbenzene (compound 10, Table S1)

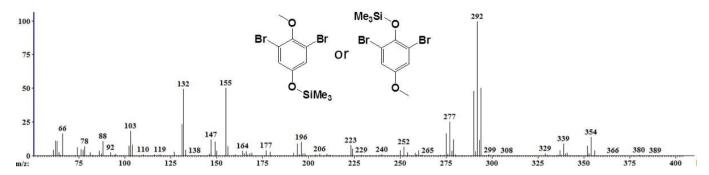


FIGURE S14. Mass spectrum of the TMS derivative of 3,5-dibromo-4-methoxyphenol or 2,6-dibromo-4-methoxyphenol (compound 11, Table S1)

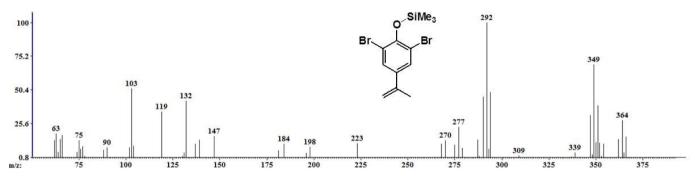


FIGURE S15. Mass spectrum of the TMS derivative of 2,6-dibromo-4(propen-2-yl)-phenol (compound 12, Table S1)

Figure S16. Proposed pathways for the degradation of TBBPA and MeO-TBBPA in submerged soil via oxidative skeletal rearrangements followed by reduction and dehydration.

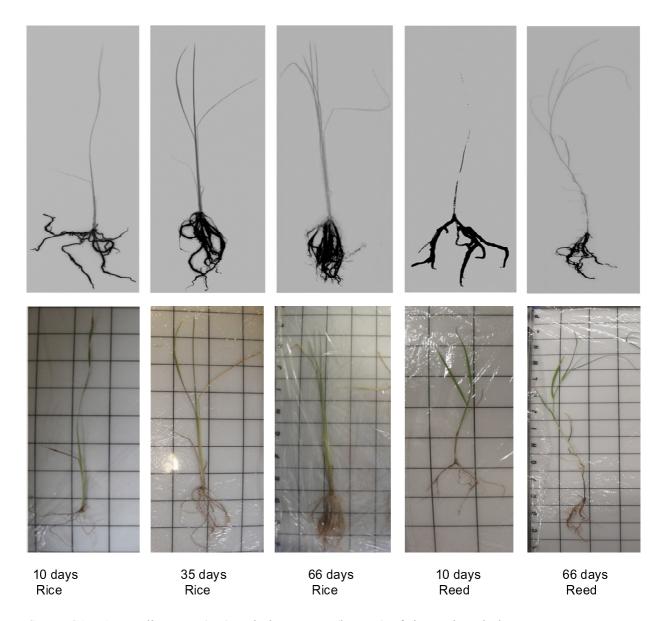


FIGURE S17. Autoradiograms (top) and photograms (bottom) of rice and reed plants grown for 10, 35, and 66 days in submerged soil containing ¹⁴C-TBBPA.

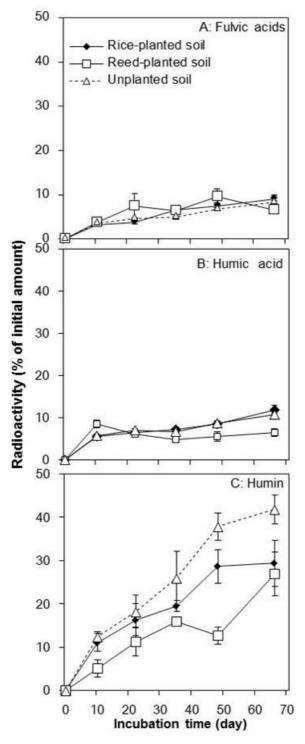


FIGURE S18. Fractionation of bound residues of 14 C-TBBPA in submerged soil planted or not with rice or reed seedlings during 66 days of incubation. The bound residues were fractionated into fulvic acids (A), humic acids (B), and humin (C) fractions. The values are the means of three individual experiments \pm one standard deviation.