Supporting Information

Identifying and Rationalizing the Conditions for the Isomerization of 1,5-Cyclooctadiene in Iridium Complexes by Experimental and Theoretical Mechanistic Studies

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1. NMR spectra

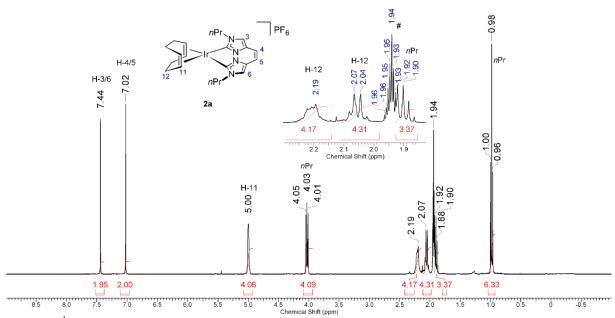


Figure S1. ¹H NMR spectrum (400.11 MHz, CD₃CN) of complex 2a (# solvent).

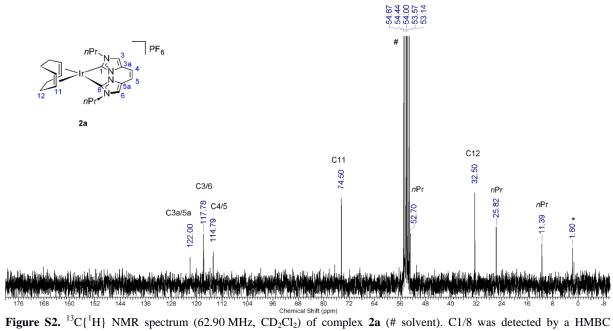


Figure S2. ¹³C{¹H} NMR spectrum (62.90 MHz, CD₂Cl₂) of complex **2a** (# solvent). C1/8 was detected by a HMBC experiment (Figure S3).

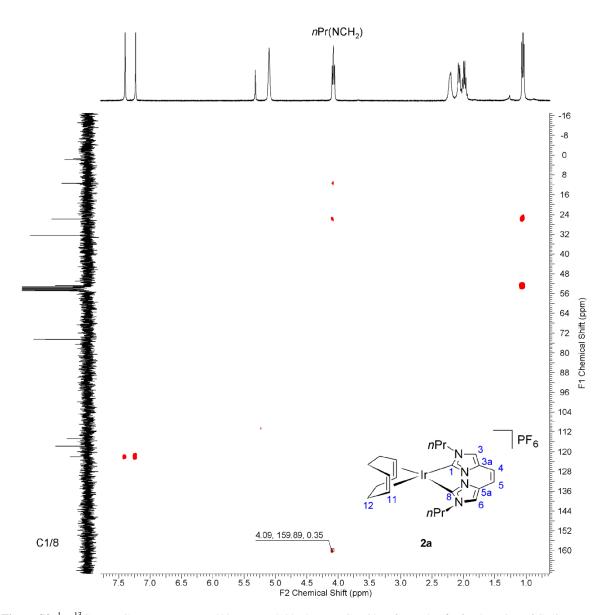


Figure S3. ¹H-¹³C-HMBC NMR spectrum (400.11 MHz/100.62 MHz, CD₂Cl₂) of complex **2a** for detection of C1/8.

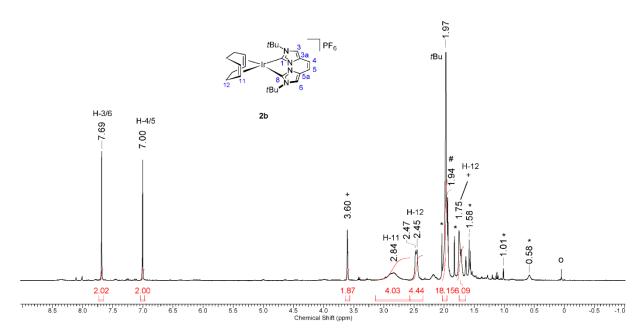


Figure S4. ¹H NMR spectrum (400.11 MHz, CD₃CN) of complex **2b** generated *in situ* from vegi^{tBu}·2HPF₆ (**1b**), *n*BuLi and $[Ir(\mu\text{-Cl})(1,5\text{-cod})]_2$ (# solvent, + thf, o grease, * unknown impurity).

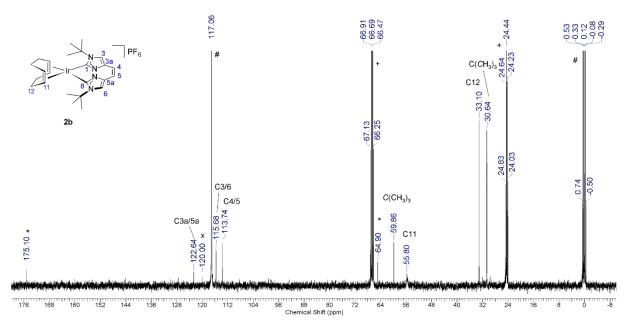


Figure S5. $^{13}C\{^{1}H\}$ NMR spectrum (100.61 MHz, CD₃CN) of complex **2b** generated *in situ* from vegi^{18u}·2HPF₆ (**1b**), $^{n}BuLi$ and $[Ir(\mu-Cl)(1,5-cod)]_2$ (# solvent, + thf- d 8, x offset, * unknown impurity). C1/8 could not be detected.

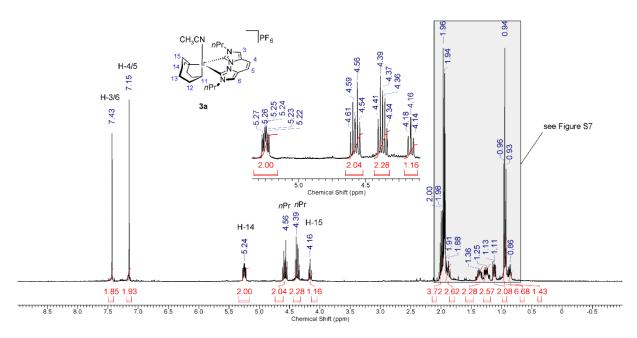


Figure S6. 1 H NMR spectrum (400.11 MHz, CD $_{3}$ CN) of complex 3a.

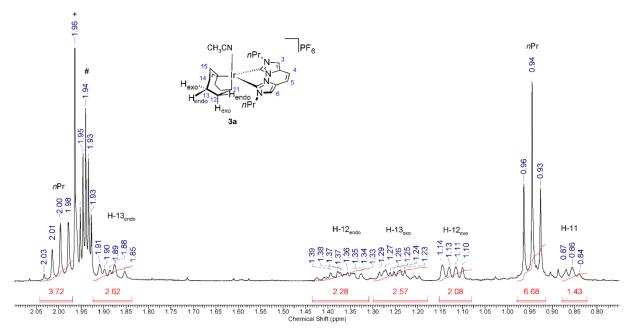


Figure S7. Excerpt from the 1 H NMR spectrum (400.11 MHz, CD₃CN) of complex **3a** highlighted in Figure S6 (# solvent, + CH₃CN). Endo and exo positions were distinguished by coupling patterns and the Karplus equation. The assignment was verified by simulation of the spectrum with MestReNova^[1]. Coordinated acetonitrile could not be detected due to its exchange with the deuterated solvent.

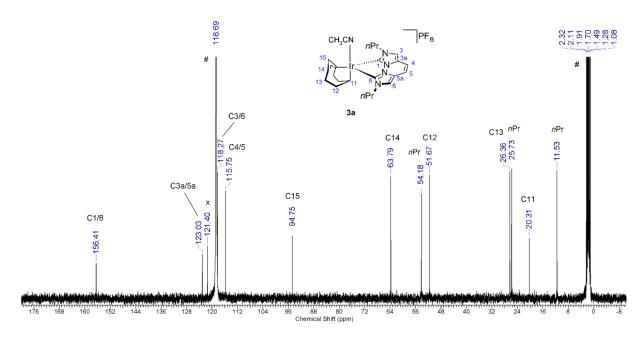


Figure S8. ¹³C{¹H} NMR spectrum (100.61 MHz, CD₃CN) of complex **3a** (# solvent, ^x offset). Coordinated acetonitrile could not be detected.

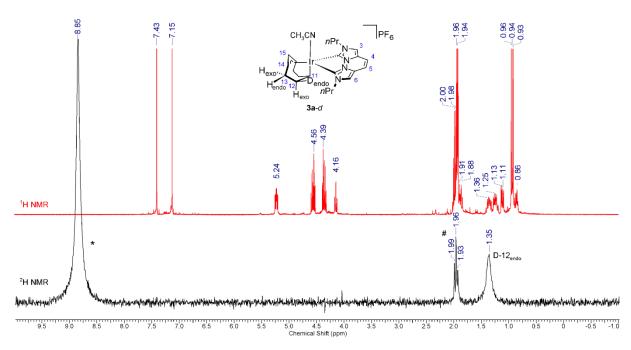


Figure S9. ²H NMR spectrum (76.77 MHz, CH₃CN) of complex **3a**-d (black), synthesized *in situ* from **2a** and stoichiometric amounts of acetic acid- d_1 in CH₃CN (# solvent (CH₂DCN, natural abundance ²H), * CH₃CO₂D), in direct comparison with the ¹H NMR spectrum of isolated **3a** (red).

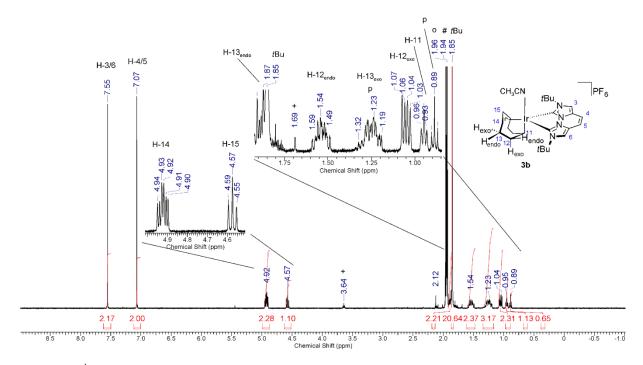


Figure S10. 1 H NMR spectrum (400.11 MHz, CD₃CN) of complex 3b (# solvent, o CH₃CN, + thf, p pentane). Coordinated acetonitrile could not be detected.

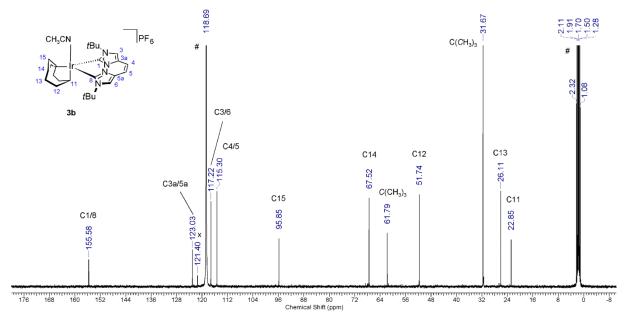


Figure S11. $^{13}C\{^{1}H\}$ NMR spectrum (100.61 MHz, CD₃CN) of complex 3b (# solvent, * offset). Coordinated acetonitrile could not be detected.

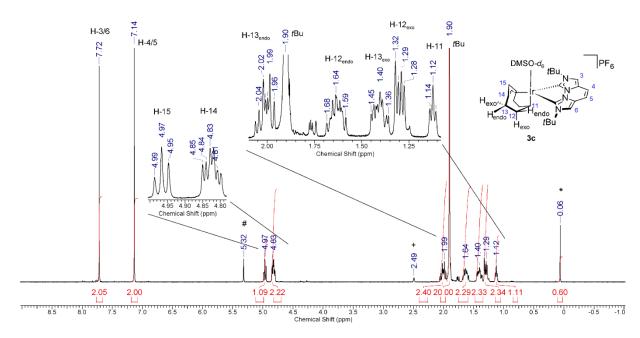
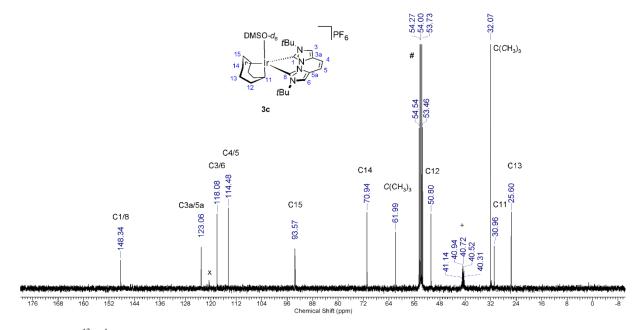


Figure S12. ¹H NMR spectrum (400.11 MHz, CD_2Cl_2) of complex 3c (# solvent, + DMSO- d_6 , * grease).



 $\textbf{Figure S13.} \ ^{13}\text{C}\{^{1}\text{H}\} \ \text{NMR spectrum (100.61 MHz, CD}_{2}\text{Cl}_{2}) \ \text{of complex 3c (\# solvent, x offset, $+$ DMSO-$d}_{6}).$

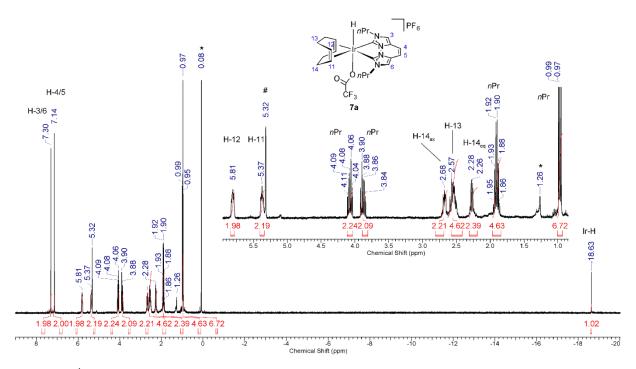
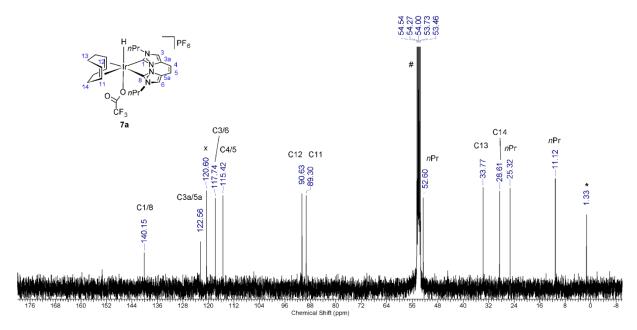


Figure S14. ¹H NMR spectrum (400.11 MHz, CD_2Cl_2) of complex **7a** (# solvent, * grease). H-12 and H-13 were assigned by observing an NOE between Ir-H and H-12/H-13 in a ¹H NOESY experiment (Figure S16). H-14_{ax} and H-14_{eq} were distinguished by observing an NOE between H-11 and H-14_{ax} but not between H-11 and H-14_{eq} (Figure S17).



 $\textbf{Figure S15.} \ ^{13}C\{^{1}H\} \ NMR \ spectrum \ (100.61 \ MHz, CD_{2}Cl_{2}) \ of \ complex \ \textbf{7a} \ (\text{\# solvent, }^{x} \ offset, \ \text{\# grease}).$

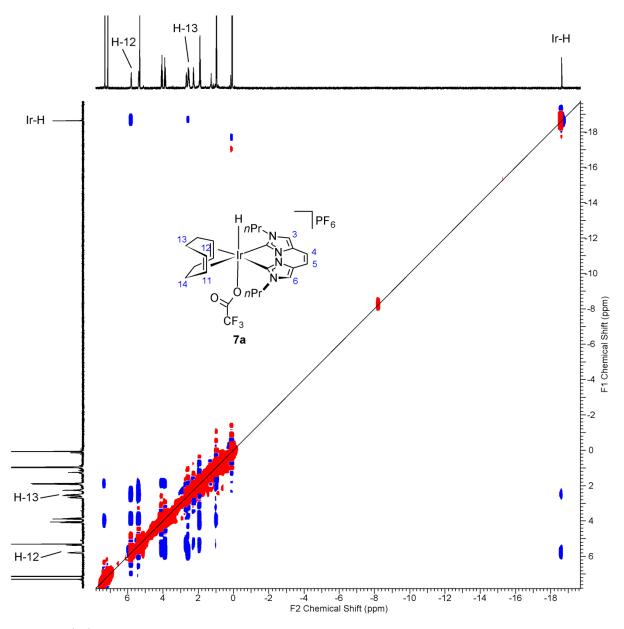


Figure S16. ¹H, ¹H NOESY spectrum (400.11 MHz, CD₂Cl₂) of complex **7a** for distinguishing H-12 and H-11 (Figure S14).

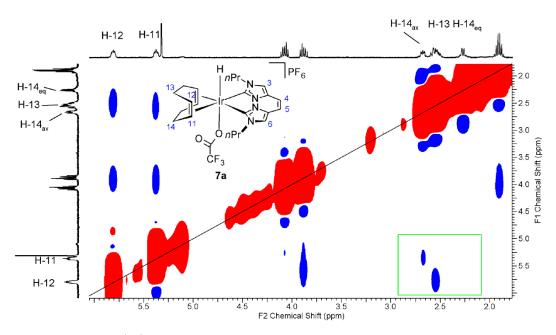


Figure S17. Excerpt from the 1 H, 1 H NOESY spectrum (400.11 MHz, $CD_{2}Cl_{2}$) of complex **7a** (Figure S16). As shown in the green area, there is a non-scalar coupling between H-11 and H-14_{ax}, but not between H-11 and H-14_{eq}. The molecular structure from X-ray diffraction shows mean distances of 2.312 Å (H-11—H-14_{ax}) and 2.691 Å (H-11—H-14_{eq}).

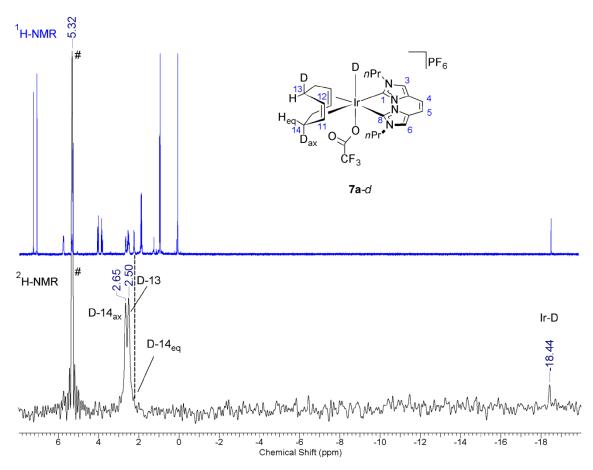


Figure S18. ²H NMR spectrum (76.77 MHz, CH_2Cl_2) of complex **7a**-d (black; # solvent/natural abundance of ²H), synthesized *in situ* from **2a** and stoichiometric amounts of trifluoroacetic acid- d_1 in CH_2Cl_2 , in direct comparison to the ¹H NMR spectrum of non-deuterated **7a** (blue, see Figure S14).

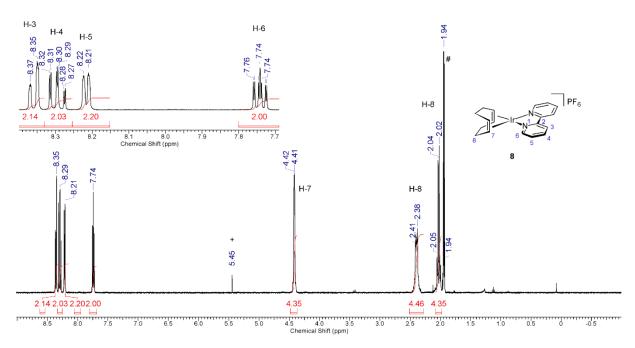


Figure S19. ¹H NMR spectrum (400.11 MHz, CD₃CN) of complex 8 (# solvent, + CH₂Cl₂).

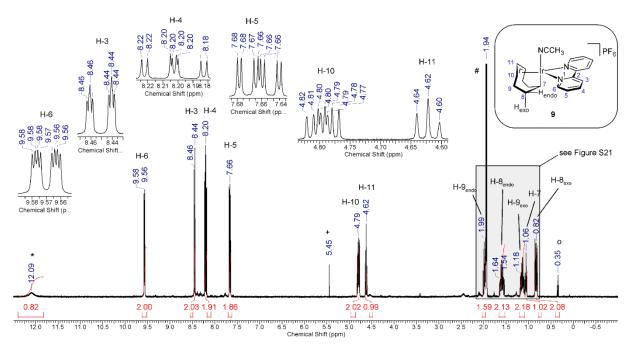


Figure S20. ¹H NMR spectrum (400.11 MHz, CD₃CN) of complex **9** (# solvent, * TFA, * CH₂Cl₂, ° impurity from TFA). Endo and exo positions were distinguished by the Karplus equation referring to the spectra of **3a**, **3b** and **3c** (Figures S6, S10, S12). For enlargement of the highlighted area see Figure S21. Coordinated acetonitrile could not be detected.

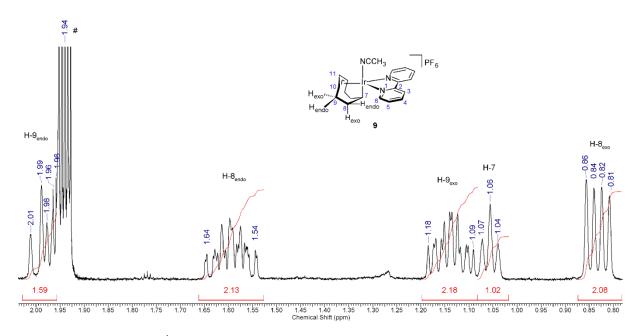


Figure S21. Excerpt from the ¹H NMR spectrum (400.11 MHz, CD₃CN) of complex **9** highlighted in Figure S20 (# solvent). Endo and exo positions were distinguished by the Karplus equation referring to the spectra of **3a**, **3b** and **3c** (Figures S6, S10, S12). Coordinated acetonitrile could not be detected.

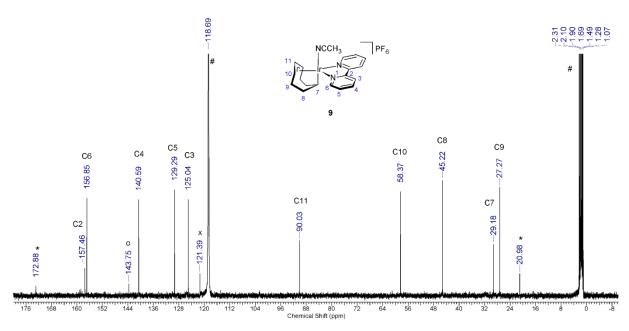


Figure S22. $^{13}C\{^{1}H\}$ NMR spectrum (100.61 MHz, CD₃CN) of complex **9**, synthesized *in situ* from [Ir(2,2'-bipyridine- $\kappa^{2}N,N')(\eta^{4}$ -cod)](PF₆) (**8**) in CD₃CN by addition of an acetic acid-trifluoroacetic acid mixture (# solvent, * offset, * AcOH, ounknown impurity). Coordinated acetonitrile could not be detected.

2. Cartesian coordinates and absolute energies

The supplemental file *Computational_data.xyz* contains the computed Cartesian coordinates of all calculated molecules reported in this study. The file may be opened as a text file to read the coordinates, or opened directly by a molecular modelling program or Mercury (version 3.3 or later, http://www.ccdc.cam.ac.uk/pages/Home.aspx) for visualization and analysis.

3. Figures of the geometrically optimized structures

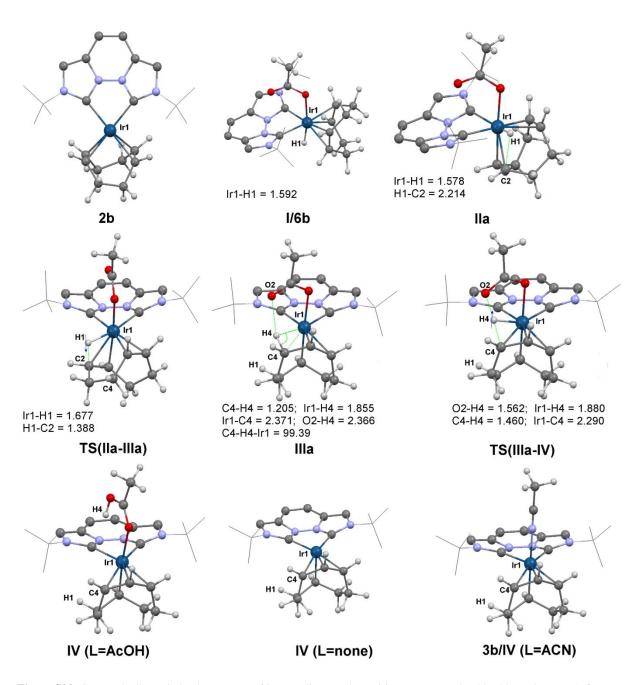


Figure S23. Geometrically optimized structures of intermediates and transition states examined in this study (Part 1; for Part 2 see Figure S24). Hydrogen atoms of the $vegi^{tBu}$ -ligand are not shown for clarity. Bond lengths are given in [Å] and angles in [$^{\circ}$]. Blue arrows in transition states indicate the normal mode with an imaginary frequency.

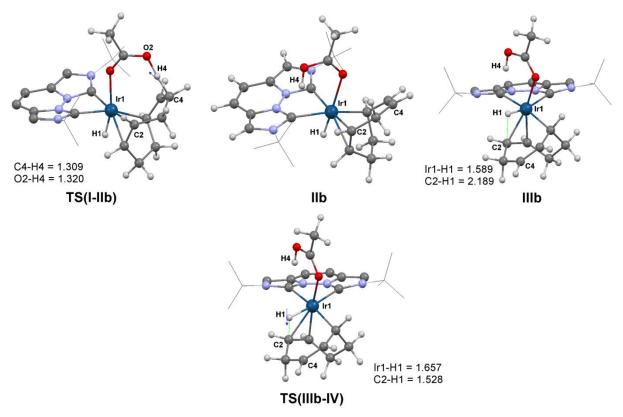


Figure S24. Geometrically optimized structures of intermediates and transition states examined in this study (Part 2; for Part 1 see Figure S23). Hydrogen atoms of the vegi^{/Bu}-ligand are not shown for clarity. Bond lengths are given in [Å] and angles in [°]. Blue arrows in transition states indicate the normal mode with an imaginary frequency.

REFERECES

 MestReNova, version 10.0.1; Mestrelab Research S.L.: Santiago de Compostela, Spain, 2015; available from http://www.mestrelab.com.