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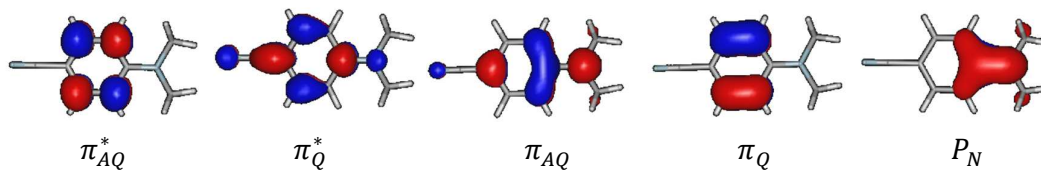
Insight into the Mechanisms of Luminescence of
Aminobenzonitrile and Dimethylaminobenzonitrile in
Polar Solvents. An Ab Initio Study

*Isabel Gómez, Pedro J. Castro and Mar Reguero**

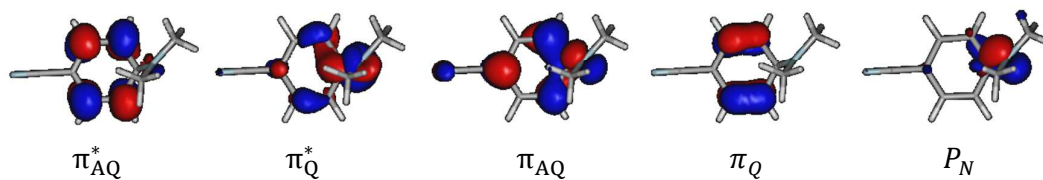
Departament de Química Física i Inorgànica.
Universitat Rovira i Virgili. Marcel·lí Domingo 1, 43007, Tarragona (Spain).

Figure S1. Molecular orbitals and excitations that describe the lowest excited states of DMABN on selected critical points. “Q” and “AQ” stand for “quinoidal” and “antiquinoidal”.

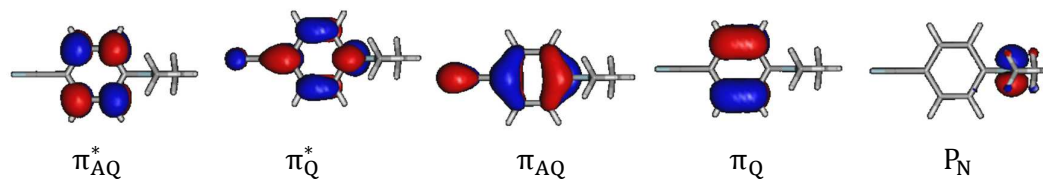
FC, LE and PICT:



TS(LE $\rightarrow$ TICT):



TICT:



$$LE(S_1) = {}^1\left(\pi_{AQ} \rightarrow \pi_{AQ}^*\right) + {}^1\left(\pi_Q \rightarrow \pi_Q^*\right)$$

$$PICT(S_2) = {}^1\left(\pi_{AQ} \rightarrow \pi_Q^*\right)$$

$$TICT(S_1) = {}^1\left(P_N \rightarrow \pi_Q^*\right)$$

Figure S2. Geometries of the ground-state, LE, PICT, RICT and TICT minima, and the transition state connecting LE and TICT species, optimized in the gas phase for ABN and DMABN. All bond lengths are in Å.

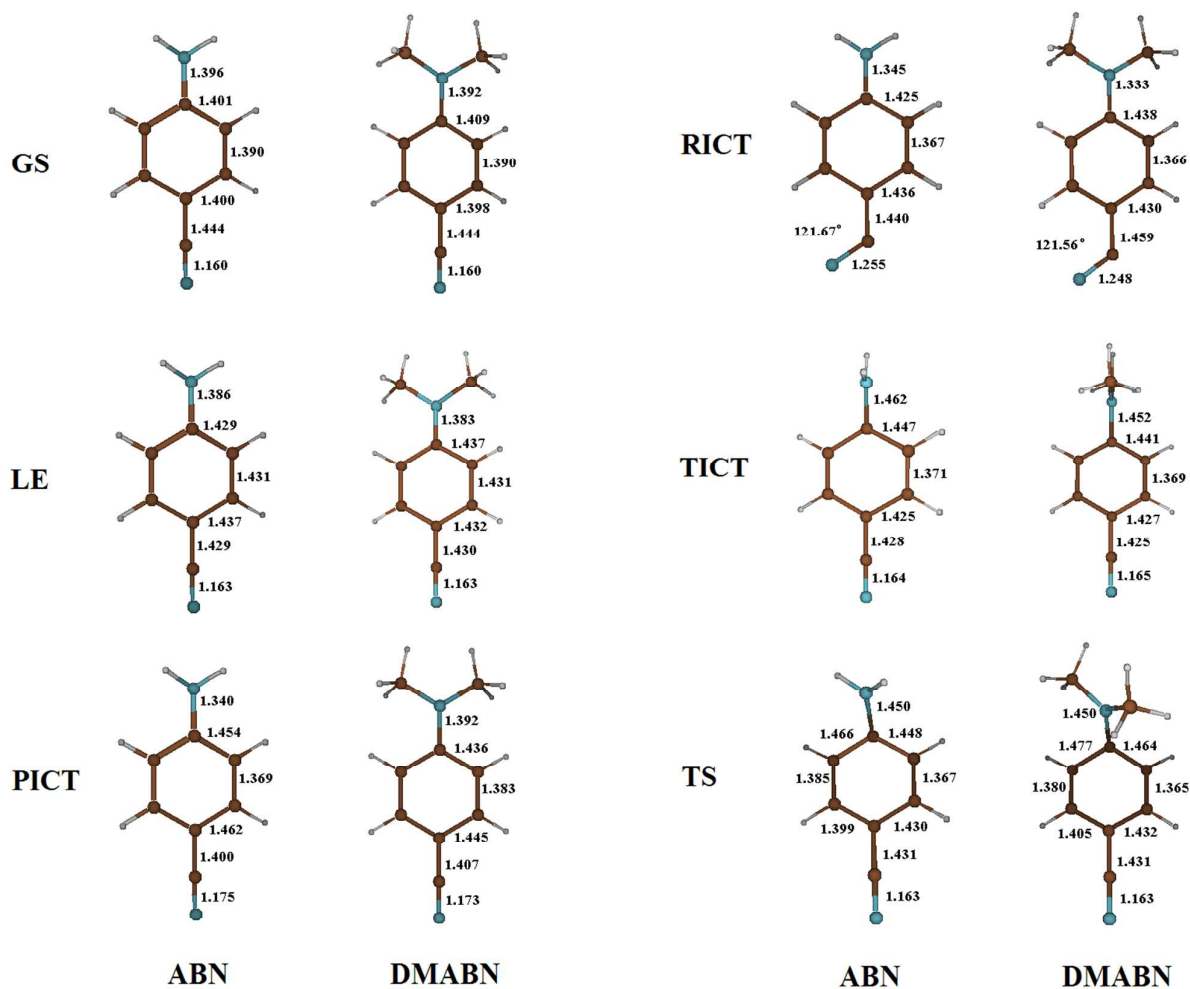


Figure S3. Geometries of the ground-state, LE, PICT, RICT and TICT minima, and the transition state connecting LE and TICT species, optimized in acetonitrile for ABN and DMABN. All bond lengths are in Å.

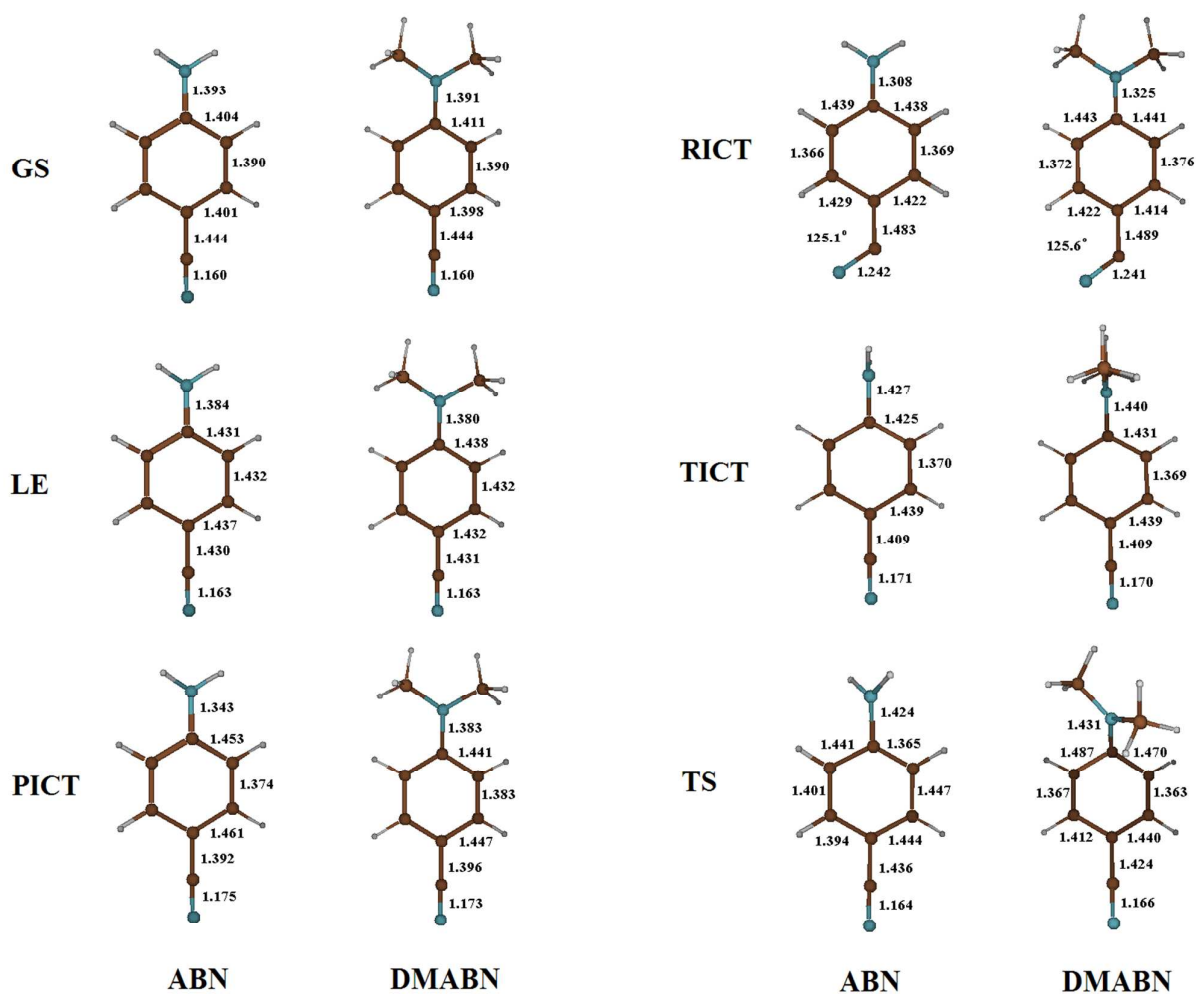


Figure S4. Forces in the potential energy surface of the S_2 (CT) state calculated at the CASSCF(12,11)/6-31G(d) level in Franck-Condon region of DMABN.

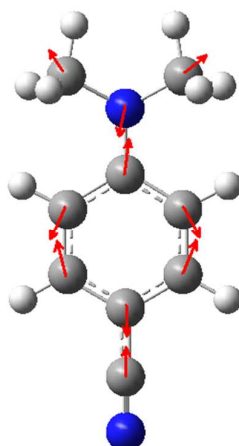
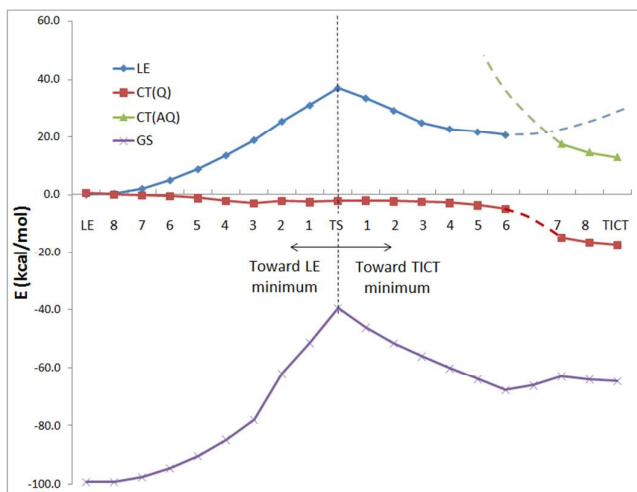


Figure S5. Energy profiles of the S_0 , S_1 and S_2 PES from the TS that connects the LE and TICT minima, and each one of these minima. The profiles were obtained by a linear interpolated internal reaction coordinate path at the SA3-CASSCF(12,11)/CASPT2/cc-pVDZ level in acetonitrile. All energies are in kcal mol⁻¹, relative to the LE minimum.

Point in the path	ΔE CASPT2 LE	ΔE CASPT2 CT(Q)	ΔE CASPT2 CT(AQ)
S_1 -LE	0.0	0.6	-
8	0.3	0.1	-
7	2.0	-0.2	-
6	4.9	-0.5	-
5	8.7	-1.2	-
4	13.5	-2.2	-
3	18.8	-3.0	-
2	25.4	-2.2	-
1	31.0	-2.6	-
S_1 -TS	36.9	-2.1	-
1	33.5	-2.0	-
2	29.2	-2.2	-
3	24.9	-2.5	-
4	22.7	-2.8	-
5	21.7	-3.7	-
6	20.6	-4.9	-
7	-	-14.9	17.4
8	-	-16.5	14.6
S_1 -TICT	-	-17.3	12.9



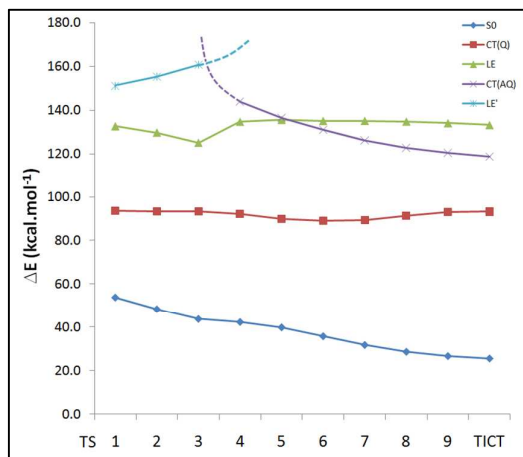
Stability of the results of the profiles of Figure 3 with the average of states.

The conditions of the state average (SA) have an influence in the results. As the number of states averaged increases, the description of each state deteriorates at the CASSCF level. At the CASPT2 level, the better the reference function, the better the results in the perturbation calculations. So with larger state averages, the CASSCF as well as the CASPT2 results are less accurate.

Obviously, the best option is to perform calculations without state average, but it can lead, in some cases, to convergence problems impossible to resolve. This was the case here, given the crossings of states present in these profiles. In Figure 3 we have used a SA-3 to calculate at the same time the profiles of states S_0 , S_1 and S_2 .

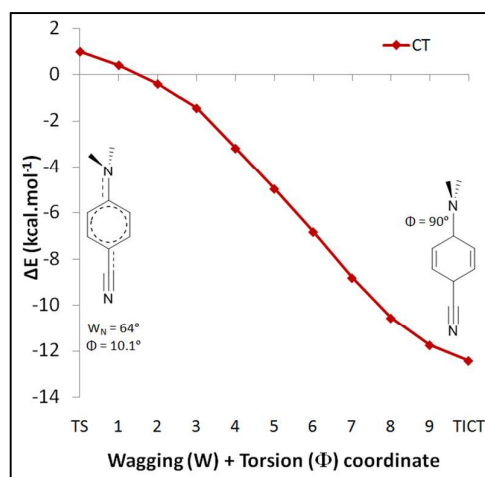
To include both LE and CT(AQ) to avoid the discontinuity in point-7, we repeated the calculation with a SA-4, but the energy of the CT(AQ) state increases very quickly when approaching the TS, producing another crossing, so the states involved in the average change again, producing another discontinuity (Figure S6).

Figure S6. Energy profiles of the S_0 , S_1 and S_2 PES from the TS to the TICT minimum. The profiles were obtained by a linear interpolated internal reaction coordinate path at the SA4-CASSCF(12,11)/CASPT2/cc-pVDZ level in acetonitrile. All energies are in kcal mol⁻¹, relative to the ground state minimum.



On top of this, the inclusion of the fourth state modifies noticeably the profile of the S_1 -CT state, where a minimum at partially rotated geometries appears. Given the controversy about the existence of this pTICT minimum, we tried to recalculate the CT profile in the most accurate way (within our methodology), i. e., without state average. Unfortunately these calculations did not converged at every point, so SA-2 was used instead, with weights of 90% for CT and 10% for S_0 . The profile obtained, show in Figure S7, show a monotonous decrease until the TICT geometry (torsion angle =90°), discarding the existence of a stable pTICT species in this system.

Figure S7. Energy profile of the S_1 -CT surface from the TS to the TICT minimum for DMABN in acetonitrile. The profile was obtained by a linear interpolated internal reaction coordinate path at the CASSCF(12,11)/CASPT2/cc-pVDZ level with a SA2 and weights of 90% for the CT state and 10% of the ground state. All energies are in kcal mol⁻¹, relative to the LE minimum.



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