- Supporting Information -

Sequential Oxidative α -Cyanation/anti-Markovnikov Hydroalkoxylation of Allylamines

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Table of Contents

1. General	S2
2. Oxidative α -cyanation/hydroalkoxylation of tertiary allylamines	S3
3. Preparation of tertiary allylamines 1a-d, 6a,b, and 11	S12
4. Theoretical studies	S15
5. NMR and IR spectra of products 2–5 , 7 , 8 , 10 , and 12	S21
6. NMR spectra of allylanilines and allylamines 1a-d, 6a,b, and 11	S70

1. General

All reactions were carried out under an atmosphere of dry nitrogen. 1 H (300 or 400 MHz) and 13 C (75.5 or 100.6 MHz) NMR spectra of solutions in CDCl₃ were recorded on 300 or 400 MHz NMR spectrometers. Chemical shifts are expressed in parts per million (ppm) downfield from tetramethylsilane and refer to the solvent signals ($\delta_{\rm H}$ 7.26 and $\delta_{\rm C}$ 77.16 ppm). Abbreviations for signal couplings are: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet. Signal assignments are based on gCOSY and gHSQC experiments. HRMS was performed on a Finnigan MAT 95Q mass spectrometer. Infrared spectra of neat substances were recorded on a Perkin-Elmer Spectrum BX II FT-IR spectrometer equipped with an ATR probe (diamond).

Materials. Triallylamine (Aldrich), iron(II) chloride (98 %, Aldrich), trimethylsilyl cyanide (98 %, Acros), *tert.*-butyl hydroperoxide (5.5 M solution in decane, purum, Aldrich) and d_4 -methanol (99.80 % D, Euriso-Top) were purchased.

Allylanilines and allylamines were synthesized according to literature procedures. S2,S3

⁽S1) Fulmer, G. R.; Miller, A. J. M.; Sherden, N. H.; Gottlieb, H. E.; Nudelman, A.; Stoltz, B. M.; Bercaw, J. E.; Goldberg, K. I. *Organometallics* **2010**, *29*, 2176-2179.

⁽S2) Li, L.; Jones, W. D. J. Am. Chem. Soc. 2007, 129, 10707-10713.

⁽S3) Bongartz, N. A.; Goodby, J. W. Chem. Commun. 2010, 46, 6452-6454.

2. Oxidative α -cyanation/hydroalkoxylation of tertiary allylamines

General Procedure. Under an atmosphere of dry N_2 , a Schlenk flask was charged with iron(II) chloride (10 mol %, 13 mg). The tertiary allylamine (1.0 mmol), trimethylsilyl cyanide (2.0 mmol, 0.27 mL), and MeOH (2.0 mL) were added successively by syringe. To the mixture was added dropwise tBuOOH (2.5 mmol, 0.46 mL, 5.5 M solution in decane) over a period of 5 min. The mixture was stirred at room temperature for the indicated time. At the end of the reaction, the reaction mixture was poured into a saturated aqueous Na_2CO_3 solution (20 mL) and extracted with CH_2CI_2 (3 × 20 mL). The organic phases were combined, dried (MgSO₄), and the volatile components were evaporated in a rotary evaporator. The crude product was purified by column chromatography on silica gel.

2-(Allyl(phenyl)amino)but-3-enenitrile (2a)

Following the *General Procedure*, N,N-diallylaniline **1a** (173 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 4 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **2a** (42 mg, 21 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 3.77–3.90 (m, 2 H), 5.09–5.27 (m, 3 H), 5.41–5.45 (m, 1 H), 5.64–5.70 (m, 1 H), 5.75–5.90 (m, 2 H), 6.87–6.93 (m, 3 H), 7.20–7.26 (m, 2 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 52.2 (CH₂), 55.3 (CH), 116.4 (C), 117.56 (CH), 117.61 (CH₂), 120.4 (CH₂), 121.3 (CH), 129.4 (CH), 131.0 (CH), 134.5 (CH),

147.4 (C). **HRMS** (EI, 70 eV): m/z [M]⁺⁻⁻ Calcd for [C₁₃H₁₄N₂]⁺⁻⁻ 198.1152; Found 198.1147.

2-(Allyl(phenyl)amino)-4-methoxybutanenitrile (3a)

Following the *General Procedure*, N,N-diallylaniline **1a** (173 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 16 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **3a** (198 mg, 86 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 2.03–2.18 (m, 2 H, 3-H), 3.36 (s, 3 H, 5-H), 3.51–3.55 (m, 2 H, 4-H), 3.85–4.00 (m, 2 H, 6-H), 4.76 (t, J = 7.7 Hz, 1 H, 2-H), 5.19–5.33 (m, 2 H, 8-H), 5.83–5.95 (m, 1 H, 7-H), 6.93–7.01 (m, 10-H and 12-H), 7.26–7.32 (m, 11-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 32.6 (CH₂, C-3), 49.9 (CH, C-2), 52.8 (CH₂, C-6), 58.9 (CH₃, C-5), 67.7 (CH₂, C-4), 117.5 (CH₂, C-8), 118.0 (CH, C-10), 118.6 (C, C-1), 121.3 (CH, C-12), 129.3 (CH, C-11), 134.6 (CH, C-7), 147.7 (C, C-9). ν (neat/ATR probe): 2959, 2929, 2874, 1495, 1453, 1363, 1189, 1116, 913, 731, 698 cm⁻¹. HRMS (EI, 70 eV): m/z [M]⁺⁻ Calcd for [C₁₄H₁₈N₂O]⁺⁻ 230.1414; Found 230.1414.

2-(Allyl(phenyl)amino)-4-(d₃-methoxy)butanenitrile (3a')

Following the *General Procedure* using CD₃OD as solvent, *N*,*N*-diallylaniline **1a** (173 mg, 1.00 mmol) reacted with Me₃SiCN and *t*BuOOH for 16 h. The crude product was

purified by column chromatography (SiO_2 , pentane/ CH_2CI_2 = 3:2) to give **3a'** (205 mg, 88 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 2.04–2.20 (m, 1.6 H, 3-H), 3.50–3.55 (m, 2 H, 4-H), 3.84–4.00 (m, 2 H, 6-H), 4.73–4.78 (m, 1 H, 2-H), 5.18–5.33 (m, 2 H, 8-H), 5.82–5.95 (m, 1 H, 7-H), 6.92–7.01 (m, 10-H and 12-H), 7.26–7.31 (m, 11-H). ²H NMR (61.4 MHz, CDCl₃): δ = 3.33 (s, 3 D, 5-D), ca. 2.13 (m, 0.4 D, 3-D). ¹³C NMR (100.6 MHz, CDCl₃): δ = 32.3 (t, ${}^{1}J_{C,D}$ = 20.0 Hz, CHD, C-3) and 32.6 (CH₂, C-3), 49.86/49.92 (CH, C-2), 52.79/52.80 (CH₂, C-6), 58.1 (sept, ${}^{1}J_{C,D}$ = 21.5 Hz, CD₃, C-5), 67.55/67.60 (CH₂, C-4), 117.5 (CH₂, C-8), 118.0 (CH, C-10), 118.6 (C, C-1), 121.3 (CH, C-12), 129.4 (CH, C-11), 134.6 (CH, C-7), 147.8 (C, C-9). HRMS (EI, 70 eV): m/z [M]⁺⁻⁻ Calcd for [C₁₄H₁₅²H₃N₂O]⁺⁻⁻ 233.1602; Found 233.1617.

2-(Allyl(p-tolyl)amino)-4-methoxybutanenitrile (3b)

Following the *General Procedure*, N,N-diallyl-4-methylaniline **1b** (187 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH in decane for 14 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **3b** (208 mg, 85 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 1.98–2.20 (m, 2 H, 3-H), 2.30 (s, 3 H, 13-H), 3.36 (s, 3 H, 5-H), 3.47–3.60 (m, 2 H, 4-H), 3.80–3.95 (m, 2 H, 6-H), 4.64 (t, J = 7.7 Hz, 1 H, 2-H), 5.18–5.33 (m, 2 H, 8-H), 5.81–5.93 (m, 1 H, 7-H), 6.94–6.97 (m, 2 H, 10-H), 7.09–7.13 (m, 2 H, 11-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.6 (CH₃, C-13), 32.6 (CH₂, C-3), 50.6 (CH, C-2), 53.4 (CH₂, C-6), 58.9 (CH₃, C-5), 67.8 (CH₂, C-4), 117.6 (CH₂, C-8), 118.6 (C, C-1), 119.6 (CH, C-10), 129.9 (CH, C-11), 131.7 (C, C-12), 134.8 (CH, C-7), 145.4 (C, C-9). ν (neat/ATR probe): 2925, 2875, 1616, 1514, 1455, 1385, 1239, 1219, 1183, 1117, 990, 921, 866, 806, 727 cm⁻¹. HRMS (EI, 70 eV): m/z [M]⁺⁻ Calcd for [C₁₅H₂₀N₂O]⁺⁻ 244.1571; Found 244.1576.

2-(Allyl(4-methoxyphenyl)amino)-4-methoxybutanenitrile (3c)

Following the *General Procedure*, N,N-diallyl-4-methoxyaniline **1c** (203 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 14 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **3c** (242 mg, 93 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 1.88–2.14 (m, 2 H, 3-H), 3.35 (s, 3 H, 5-H), 3.45–3.60 (m, 2 H, 4-H), 3.68–3.85 (m, 2 H, 6-H), 3.78 (s superimposed with resonances of 6-H, 3 H, 13-H), 4.43 (t, J = 7.8 Hz, 1 H, 2-H), 5.14–5.31 (m, 2 H, 8-H), 5.75–5.88 (m, 1 H, 7-H), 6.83–6.86 and 7.06–7.09 (2 m, 2 × 2 H, 10-H and 11-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 32.6 (CH₂, C-3), 51.9 (CH, C-2), 54.9 (CH₂, C-6), 55.6 (CH₃, C-13), 58.9 (CH₃, C-5), 67.9 (CH₂, C-4), 114.5 (CH, C-10 or C-11), 118.0 (CH₂, C-8), 118.7 (C, C-1), 123.8 (CH, C-10 or C-11), 134.8 (CH, C-7), 141.1 (C, C-9), 156.3 (C,

C-12). ν (neat/ATR probe): 2931, 2834, 1509, 1462, 1244, 1214, 1181, 1117, 1036, 917, 866, 834, 731 cm⁻¹. **HRMS** (EI, 70 eV): m/z [M]⁺⁻⁻ Calcd for [C₁₅H₂₀N₂O₂]⁺⁻⁻ 260.1520; Found 260.1504.

2-(Allyl(4-bromophenyl)amino)-4-methoxybutanenitrile (3d)

Following the *General Procedure*, N,N-diallyl-4-bromoaniline **1d** (252 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 19 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **3d** (207 mg, 67 %) as a yellow liquid.

¹H NMR (300 MHz, CDCl₃): δ = 2.02–2.20 (m, 2 H, 3-H), 3.35 (s, 3 H, 5-H), 3.49–3.53 (m, 2 H, 4-H), 3.82–3.97 (m, 2 H, 6-H), 4.73 (t, J = 7.7 Hz, 1 H, 2-H), 5.19–5.30 (m, 2 H, 8-H), 5.78–5.91 (m, 1 H, 7-H), 6.83–6.86 (m, 2 H, 10-H), 7.35–7.38 (m, 2 H, 11-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 32.4 (CH₂, C-3), 49.7 (CH, C-2), 52.4 (CH₂, C-6), 58.9 (CH₃, C-5), 67.4 (CH₂, C-4), 113.5 (C, C-12), 117.8 (CH₂, C-8), 118.2 (C, C-1), 119.2 (CH, C-10), 132.1 (CH, C-11), 133.9 (CH, C-7), 146.6 (C, C-9). ν (neat/ATR probe): 2929, 2877, 2831, 1590, 1493, 1458, 1385, 1236, 1183, 1116, 1081, 998, 918, 866, 808, 731, 648 cm⁻¹. HRMS (EI, 70 eV): m/z [M]⁺⁺ Calcd for [C₁₄H₁₇⁷⁹BrN₂O]⁺⁺ 308.0519; Found 308.0522.

2-(Allyl(phenyl)amino)-4-ethoxybutanenitrile (4) and 2-(allyl(phenyl)amino)-propanenitrile (5)

Following the *General Procedure* using EtOH as solvent, N,N-diallylaniline **1a** (173 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 16 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 3:2) to give **4** (166 mg, 68 %) as a colorless liquid and **5** (58 mg, 31 %) as a colorless liquid.

Analytical data for **4**: ¹**H NMR** (300 MHz, CDCl₃): δ = 1.21 (t, J = 7.0 Hz, 3 H, 5-CH₃), 2.03–2.23 (m, 2 H, 3-H), 3.45–3.59 (m, 4 H, 4-H and 5-H), 3.85–4.01 (m, 2 H, 6-H), 4.79 (t, J = 7.7 Hz, 1 H, 2-H), 5.19–5.34 (m, 2 H, 8-H), 5.83–5.95 (m, 1 H, 7-H), 6.92–7.01 (m, 10-H and 12-H), 7.26–7.32 (m, 2 H, 11-H). ¹³**C NMR** (75.5 MHz, CDCl₃): δ = 15.1 (CH₃, 5-CH₃), 32.6 (CH₂, C-3), 49.8 (CH, C-2), 52.6 (CH₂, C-6), 65.3 (CH₂, C-4), 66.5 (CH₂, C-5), 117.4 (CH₂, C-8), 117.9 (CH, C-10), 118.6 (C, C-1), 121.1 (CH, C-12), 129.2 (CH, C-11), 134.5 (CH, C-7), 147.7 (C, C-9). ν (neat/ATR probe): 2975, 2930, 2867, 1598, 1502, 1377, 1236, 1172, 1110, 990, 915, 749, 732, 692 cm⁻¹. **HRMS** (EI, 70 eV): m/z [M]⁺⁻ Calcd for [C₁₅H₂₀N₂O]⁺⁻ 244.1570; Found 244.1569.

Analytical data for **5**: ¹**H NMR** (300 MHz, CDCl₃): δ = 1.58 (d, J = 7.2, 3 H, 3-H), 3.85–4.00 (m, 2 H, 4-H), 4.56 (q, J = 7.2 Hz, 1 H, 2-H), 5.20–5.35 (m, 2 H, 6-H), 5.85–5.97 (m, 1 H, 5-H), 6.96–7.00 (m, 3 H, 8-H and 10-H), 7.26–7.33 (m, 2 H, 9-H). ¹³**C NMR** (75.5 MHz, CDCl₃): δ = 18.6 (CH₃, C-3), 47.4 (CH, C-2), 53.0 (CH₂, C-4), 117.4 (CH₂, C-6), 118.4 (CH, C-8), 119.3 (C, C-1), 121.7 (CH, C-10), 129.4 (CH, C-10),

9), 134.9 (CH, C-5), 147.7 (C, C-7). **HRMS** (EI, 70 eV): m/z [M]⁺⁻ Calcd for $[C_{12}H_{14}N_2]^{+-}$ 186.1152; Found 186.1163.

2-(Ethyl(phenyl)amino)-4-methoxybutanenitrile (7)

Following the *General Procedure*, *N*-allyl-*N*-ethylaniline **6a** (161 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 18 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 2:1) to give **7** (181 mg, 83 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 1.19 (t, J = 7.1 Hz, 3 H, 7-H), 2.02–2.19 (m, 2 H, 3-H), 3.26–3.43 (m, 2 H, 6-H), 3.36 (s superimposed with resonances of 6-H, 3 H, 5-H), 3.49–3.59 (m, 2 H, 4-H), 4.67 (t, J = 7.7 Hz, 1 H, 2-H), 6.94–7.02 (m, 3 H, 9-H and 11-H), 7.26–7.33 (m, 2 H, 10-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 13.7 (CH₃, C-7), 32.6 (CH₂, C-3), 43.7 (CH₂, C-6), 50.7 (CH, C-2), 59.0 (CH₃, C-5), 67.7 (CH₂, C-4), 118.7 (CH, C-9), 118.9 (C, C-1), 121.5 (CH, C-11), 129.5 (CH, C-10), 147.2 (C, C-8). ν (neat/ATR probe): 2977, 2929, 2875, 1598, 1502, 1385, 1248, 1190, 1116, 1014, 910, 862, 750, 730, 693, 647 cm⁻¹. HRMS (EI, 70 eV): m/z [M]⁺⁺ Calcd for [C₁₃H₁₈N₂Ol]⁺⁺ 218.1414; Found 218.1411.

2-(Allyl(phenyl)amino)acetonitrile (8)

Following the *General Procedure*, *N*-allyl-*N*-methylaniline **6b** (147 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 18 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 2:1) to give **8** (128 mg, 74 %) as a

colorless liquid. Known compound, the NMR spectroscopic data agree with those given in ref^{S4}.

¹H NMR (400 MHz, CDCl₃): δ = 3.97 (d, J = 5.6 Hz, 2 H, 3-H), 4.15 (s, 2 H, 2-H), 5.28–5.38 (m, 2 H, 5-H), 5.85–5.95 (m, 1 H, 4-H), 6.89–6.95 (m, 3 H, 7-H and 9-H), 7.30–7.34 (m, 2 H, 8-H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 39.4 (CH₂, C-3), 54.6 (CH₂, C-2), 115.3 (CH, C-7), 116.1 (C, C-1), 118.6 (CH₂, C-5), 120.3 (CH, C-9), 129.6 (C, C-8), 133.0 (CH, C-4), 147.5 (C, C-6).

2-(Diallylamino)-4-methoxybutanenitrile (10)

Following the *General Procedure*, triallylamine **9** (137 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 40 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 2:3) to give **10** (161 mg, 83 %) as a colorless liquid.

¹H NMR (300 MHz, CDCl₃): δ = 1.95–2.04 (m, 2 H, 3-H), 2.88–2.96 (m, 2 H, 6-H^a), 3.32–3.40 (m, 2 H, 6-H^b), 3.32 (s superimposed with resonances of 6-H^b, 3 H, 5-H), 3.42–3.55 (m, 2 H, 4-H), 3.98 (t, J = 7.8 Hz, 1 H, 2-H), 5.16–5.30 (m, 4 H, 8-H), 5.70–5.83 (m, 2 H, 7-H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 31.9 (CH₂, C-3), 50.3 (CH,

⁽S4) Reddy, K. H. V.; Satish, G.; Reddy, V. P.; Kumar, B. S. P. A.; Nageswar, Y. V. D. *RSC Adv.* **2012**, *2*, 11084-11088.

C-2), 54.5 (CH₂, C-6), 58.9 (CH₃, C-5), 68.0 (CH₂, C-4), 117.8 (C, C-1), 118.5 (CH₂, C-8), 134.9 (CH, C-7). ν (neat/ATR probe): 3082, 2981, 2931, 2814, 1644, 1450, 1420, 1189, 1116, 994, 920, 732 cm⁻¹. **HRMS** (EI, 70 eV): m/z [M-H]⁻ Calcd for [C₁₁H₁₇N₂O]⁻ 193.1346; Found 193.1333.

2-(Allyl(benzyl)amino)-4-methoxybutanenitrile (12)

Following the *General Procedure*, N,N-diallyl-benzylamine **11** (187 mg, 1.00 mmol) reacted with Me₃SiCN and tBuOOH for 40 h. The crude product was purified by column chromatography (SiO₂, pentane/CH₂Cl₂ = 2:3) to give **12** (188 mg, 77 %) as a colorless liquid.

¹H NMR (400 MHz, CDCl₃): δ = 1.91–1.97 (m, 2 H, 3-H), 2.87–2.93 (m, 1 H, 6-H^a), 3.19 (s, 3 H, 5-H), 3.20–3.43 (m, 4 H, 4-H, 6-H^b and 9-H^a), 3.86–3.91 (m, 2 H, 2-H and 9-H^b), 5.12–5.15 (m, 2 H, 8-H), 5.21–5.26 (m, 1 H, 7-H), 7.18–7.29 (m, 5 H, 11-H, 12-H and 13-H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 31.9 (CH₂, C-3), 50.3 (CH, C-2), 54.6 (CH₂, C-6), 55.5 (CH₂, C-9), 58.8 (CH₃, C-5), 68.0 (CH₂, C-4), 117.7 (C, C-1), 118.8 (CH₂, C-8), 127.6, 128.6, 128.7 (3 × CH, C-11, C-12 and C-13), 134.8 (CH, C-7), 138.1 (C, C-10). HRMS (EI, 70 eV): m/z [M-H]⁻ Calcd for [C₁₅H₁₉N₂O]⁻ 243.1503; Found 243.1490.

3. Preparation of tertiary allylamines 1a-d, 6a,b, and 11

According to a procedure reported in ref^{S2}: In a 500 mL round-bottom flask equipped with a reflux condenser and a stir bar, aniline (50 mmol), allyl bromide (12.1 g, 100 mmol), and Na₂CO₃ (5.40 g, 51 mmol) were added to aqueous ethanol (200 mL, EtOH/H₂O = 4/1). The reaction mixture was refluxed overnight. The crude product was poured on saturated aq. NaHCO₃ (50 mL) and extracted with Et₂O (3 × 50 mL). The combined organic layers were washed with H₂O (2 × 50 mL), dried over MgSO₄, and the solvent was evaporated in the vacuum. The crude product was distilled over KOH to provide the corresponding diallylaniline.

N,*N*-Diallylaniline (1a): 6.2 g (71 %). Known compound, the NMR spectroscopic data agree with those given in lit.^{S5}: ¹H NMR (300 MHz, CDCl₃): δ = 3.99-4.02 (m, 4 H), 5.22-5.30 (m, 4 H), 5.89-6.01 (m, 2 H), 6.77-6.81 (m, 3 H), 7.28-7.31 (m, 2 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 52.9, 112.5, 116.1, 116.4, 129.2, 134.2, 148.8.

N,N-Diallyl-4-methylaniline (1b): 6.7 g (71 %). Known compound, the NMR spectroscopic data agree with those given in lit. S5: ¹H NMR (300 MHz, CDCl₃): δ = 2.29 (s, 3 H), 3.93-3.95 (m, 4 H), 5.17-5.26 (m, 4 H), 5.84-5.97 (m, 2 H), 6.67-6.70 (m, 2 H), 7.05-7.08 (m, 2 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.3, 53.1, 112.8, 116.0, 125.6, 129.7, 134.4, 146.8.

N,N-Diallyl-4-methoxyaniline (1c): 3.5 g (34 %). Known compound, the NMR spectroscopic data agree with those given in lit. S5: ¹H NMR (300 MHz, CDCl₃): δ = 3.77 (s, 3 H), 3.88-3.89 (m, 4 H), 5.16-5.24 (m, 4 H), 5.82-5.93 (m, 2 H), 6.71-6.74

⁽S5) Chen, W.; Wang, J. Organometallics 2013, 32, 1958-1963.

(m, 2 H), 6.82-6.85 (m, 2 H). ¹³**C NMR** (75.5 MHz, CDCl₃): δ = 53.7, 55.9, 114.6, 114.8, 116.2, 134.7, 143.6, 151.7.

N,N-Diallyl-4-bromoaniline (1d): 3.5 g (28 %). Known compound, the NMR spectroscopic data agree with those given in lit. S5: ¹H NMR (300 MHz, CDCl₃): δ = 3.89-3.92 (m, 4 H), 5.14-5.20 (m, 4 H), 5.78-5.90 (m, 2 H), 6.56-6.59 (m, 2 H), 7.25-7.28 (m, 2 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 53.0, 108.2, 114.1, 116.3, 131.8, 133.5, 147.7.

N-Allyl-*N*-ethylaniline (6a) from *N*-ethylaniline (6.1 g, 50 mmol), allyl bromide (4.8 mL, 40 mmol), and Na₂CO₃ (4.2 g, 40 mmol) as described above to the preparation of **1a**-d. The crude product was purified by vacuum distillation from CaH₂: 4.8 g (74 %). Known compound, the NMR spectroscopic data agree with those given in lit. S6: ¹H NMR (300 MHz, CDCl₃): δ = 1.25 (t, *J* = 7.1 Hz, 3 H), 3.46 (q, *J* = 7.1 Hz, 2 H), 3.96-3.99 (m, 2 H), 5.19-5.29 (m, 2 H), 5.88-6.00 (m, 1 H), 6.74-6.79 (m, 3 H), 7.26-7.31 (m, 2 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 12.4, 44.8, 52.8, 112.2, 115.9, 116.0, 129.3, 134.6, 148.3.

N-Allyl-*N*-methylaniline (6b) from *N*-methylaniline (5.4 g, 50 mmol), allyl bromide (4.8 mL, 40 mmol), and Na₂CO₃ (4.2 g, 40 mmol) as described above for the preparation of **1a**-**d**. The crude product was purified by vacuum distillation from CaH₂: 4.0 g (68 %). Known compound, the NMR spectroscopic data agree with those given in lit. S6: 1 H NMR (400 MHz, CDCl₃): δ = 2.97 (s, 3 H), 3.94-3.96 (m, 2 H),

S13

⁽S6) Yang, S.-H.; Hung, C.-W. Synthesis 1999, 1747-1752.

5.16-5.23 (m, 2 H), 5.83-5.93 (m, 1 H), 6.73-6.78 (m, 3 H), 7.24-7.27 (m, 2 H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 38.1, 55.4, 112.6, 116.3, 116.5, 129.2, 134.0, 149.6.

N,N-Diallyl-benzylamine (11) from benzylamine as described above for the preparation of 1a-d. The crude product was purified by vacuum distillation from CaH₂: 5.7 g (61 %). Known compound, the NMR spectroscopic data agree with those given in lit. S5: ¹H NMR (300 MHz, CDCl₃, Me₄Si): δ = 3.00 (dt, J = 6.3 Hz, J = 1.4 Hz, 4 H), 3.49 (s, 2 H), 5.03-5.14 (m, 4 H), 5.73-5.87 (m, 2 H), 7.14-7.26 (m, 5 H). ¹³C NMR (75.5 MHz, CDCl₃): δ = 56.6, 57.7, 117.4, 126.9, 128.3, 129.0, 136.0, 139.6.

4. Theoretical studies

Following earlier theoretical work on the stabilitiy of N- and C-centered radicals^{S7_S11} DFT calculations are employed for geometry optimizations and frequency calculations for open- and closed-shell systems at the unrestricted UB3LYP/6-31G(d) level and restricted B3LYP/6-31G(d) level, respectively. S12,S13All energies are reported for the structures in gas-phase at 298.15 K where thermal corrections to enthalpies have been calculated at the same level of theory using the rigid rotor/harmonic oscillator model and a scale factor of 0.9806 (in kJ/mol). Improved relative energies were obtained using single point calculations at the ROB2-PLYP/cc-pVTZ level of theory, S14,S15 and using the G3(MP2)-RAD method developed by Radom et al. for open shell systems. S16 For selected (smaller) systems more accurate energies were calculated using the G3B3 compound scheme. S17 All the calculations were performed using the Gaussian 09 software package. S18

⁽S7) Hioe, J.; Sakic, D.; Vrcek, V.; Zipse, H. Org. Biomol. Chem. 2015, 13, 157-169.

⁽S8) Achrainer, F.; Zipse, H. *Molecules* **2014**, *19*, 21489-21505.

⁽S9) Hioe, J.; Mosch, M.; Smith, D. M.; Zipse, H. RSC Adv. 2013, 3, 12403-12408.

⁽S10) Hioe, J.; Zipse, H. Chem. Eur. J. 2012, 18, 16463-16472.

⁽S11) Hioe, J.; Zipse, H. Org. Biomol. Chem. 2010, 8, 3609-3617.

⁽S12) Becke, A. D. J. Chem. Phys. 1993, 98, 5648-5652.

⁽S13) Hariharan, P. C.; Pople, J. A. Theor. Chem. Acc. 1973, 28, 213-222.

⁽S14) (a) Grimme, S. *J. Chem. Phys.* **2006**, *124*, 034108. (b) Graham, D. C.; Menon, A. S.; Goerigk, L.; Grimme, S.; Radom, L. *J. Phys. Chem. A* **2009**, *113*, 9861-9873.

⁽S15) Dunning, T. H., Jr. J. Chem. Phys. 1989, 90, 1007-1023.

⁽S16) (a) Henry, D. J.; Parkinson, C. J.; Radom, L. J. Phys. Chem. A 2002, 106, 7927-7936. (b) Henry, D. J.; Sullivan, M. B.; Radom, L. J. Chem. Phys. 2003, 118, 4849-4860.

⁽S17) Baboul, A. G.; Curtiss, L. A.; Redfern, P. C.; Raghavachari, K. *J. Chem. Phys.* **1999**, *110*, 7650-7657.

⁽S18) Gaussian 09, Revision C.01, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria,
G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.;
Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.;

The stability of radical **2a-rad** can be quantified through the reaction enthalpy for the hydrogen transfer reaction with methane/methyl radical shown in Figure S1.

Figure S1. Isodesmic hydrogen transfer reaction between methane and radical **2a-rad**.

The most stable conformation of radical **2a-rad** points the cyano and vinyl substituents away from the phenyl group and contains a fully planar radical center. The alternative conformation **2a-rad-02** with the cyano group pointing towards the phenyl substituent is less stable by 8.6 kJ/mol (UB3LYP/6-31G(d)) or 6.8 kJ/mol (RO-B2PLYP/cc-pVTZ).

Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J., Gaussian, Inc., Wallingford CT, 2009.

Conformational preferences in the closed-shell parent 2a are less clear and the conformations 2a-01 and 2a-02 are seperated by only 3.5 kJ/mol (both at B3LYP/6-31G(d) or B2PLYP/cc-pVTZ level). However, in both conformations significant stereoelectronic effects appear to exist between the aniline nitrogen lone pair electrons and the $\sigma^*(C-C(N))$ bond. Reaction energies for the isodesmic reaction (1) and thus the radical stabilization energy (RSE) of radical 2a-rad amount to -165.1 kJ/mol (UB3LYP/6-31G(d)), -146.7 kJ/mol (ROB2-PLYP/cc-pVTZ), or -129.1 kJ/mol (G3(MP2)-RAD). Important reference systems calculated at this latter level of theory include the allyl radical with RSE = -72.0 kJ/mol and the captodatively stabilized glycyl radical (${}^{\bullet}CH(NHCOCH_3)CONHCH_3$) with RSE = -74.1 kJ/mol. This comparison illustrates that radical 2a-rad is exceedingly well stabilized through the donor amino and the acceptor cyano groups as well as the additional vinyl substituent. Together with the BDE(C-H) energy of +439.3 kJ/mol in methane, S19 the value of RSE(2a-rad) = -129.1 kJ/mol translates into a BDE(C-H) in parent 2a of +310.2 kJ/mol. Aminonitrile 2a is thus a more powerful hydrogen atom donor than thiophenol (BDE(S-H) = $+349.4 \pm 4.5 \text{ kJ/mol}$) or tri-n-butyltin hydride (HSnBu₃, BDE(Sn-H) = +328.9 kJ/mol). S19,S20

Radical **3a-rad** is significantly more flexible than **2a**, making the full exploration of conformational space rather time consuming. Starting conformations for geometry optimizations for **3a-rad** were therefore generated in two steps: for the best seven conformations of **2a-rad** all possible conformations for transforming the C-C double bond into the methoxyethyl side chain were generated and optimized; for

⁽S19) Luo, Y.-R. *Comprehensive Handbook of Chemical Bond Energies*; CRC Press: Boca Raton (FL), 2007.

⁽S20) Hioe, J.; Zipse, H. In *Encyclopedia of Radicals in Chemistry, Biology and Materials*; Chatgilialoglu, C., Studer, A., Eds.; Wiley: Chichester, 2012, pp 449-476.

all remaining conformations of **2a-rad** the methoxyethyl side chain was generated in an *all-trans* conformation and optimized. In a completely analogous fashion conformational searches were performed for the closed-shell parent **3a**.

Figure S2. Isodesmic hydrogen transfer reaction between methane and radical **3a-rad**.

This latter system again orients the aniline nitrogen and cyano groups such as to allow for efficient donor/acceptor interactions. Conformations pointing the C-H bond at the chiral center away from the phenyl ring are 3.1 kJ/mol (UB3LYP/6-31G(d)) or 2.3 kJ/mol (RO-B2PLYP/cc-pVTZ) more favorable than those pointing the C-H bond downward. All low-energy conformations orient the methyoxyethyl side chain such as to allow for hydrogen-bonding interactions between the methoxy group oxygen and the chiral center C-H bond. Conformational preferences in radical **3a-rad** are similar to those on **2a-rad** in that upward orientation of the cyano substituent is preferred by 9.2 kJ/mol (UB3LYP/6-31G(d)) or 7.0 kJ/mol (RO-B2PLYP/cc-pVTZ) over the downward orientation (as drawn in Figure S2). Reaction energies for isodesmic reaction (2) and thus the radical stabilization energy (RSE) of radical **3a-rad** amount

to -123.6 kJ/mol (UB3LYP/6-31G(d)), -109.4 kJ/mol (ROB2-PLYP/cc-pVTZ), or -91.6 kJ/mol (G3(MP2)-RAD).

Reaction energies for isodesmic reactions (1) and (2) can be combined to predict the reaction energy for hydrogen transfer reaction (3) between radical **3a-rad** and **2a** (Figure S3). This reaction is always exothermic, the reaction energy amounting to –41.4 kJ/mol (UB3LYP/6-31G(d)), –37.4 kJ/mol (ROB2-PLYP/cc-pVTZ), or –37.5 kJ/mol (G3(MP2)-RAD).

Figure S3. Isodesmic hydrogen transfer reaction between 2a and radical 3a-rad.

In order to elucidate further how substituents at the amine nitrogen and the radical center impact radical stability and C-H bond energies, additional smaller systems have been studied (Figure S4). The parent aminomethyl radical is stabilized by 46.7 kJ/mol, a value not influenced by the introduction of additional methyl substituents at the nitrogen atom. Higher stabilities are observed for the secondary 1-dimethylaminoethyl radical (RSE = -54.6 kJ/mol) and the 1-aminoallyl radical 13a-rad (RSE = -111.3 kJ/mol). Introduction of an N-phenyl substituent in this latter system as in 14a-rad remains practically without consequence (RSE = -110.0 kJ/mol). Compared to this latter system, the donor-acceptor substituted substrate radical 2a-rad is significantly more stable with RSE = -129.1 kJ/mol, while product radical 3a-rad is less stable with RSE = -91.6 kJ/mol. All RSE values can, of course, be translated into BDE(C-H) values by addition of the experimentally measured C-H

bond energy in methane (BDE = +439.3 kJ/mol). In the reaction system used here two additional bond energy values merit note. The first is that of solvent methanol with BDE(C-H) = +405.8 kJ/mol (G3B3) or +401.9 \pm 0.6 kJ/mol (exp.). The second is that for tBuO₂H with BDE(O-H) = +356.9 kJ/mol (G3B3) or +352.8 \pm 8.3 kJ/mol (exp.). This latter value appears in Figure S4 as a large grey bar due to the large error margin of \pm 9 kJ/mol.

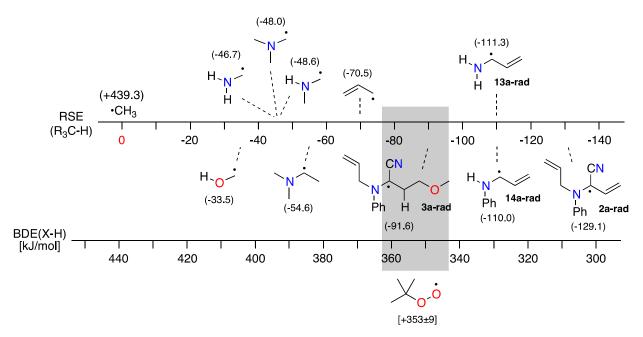
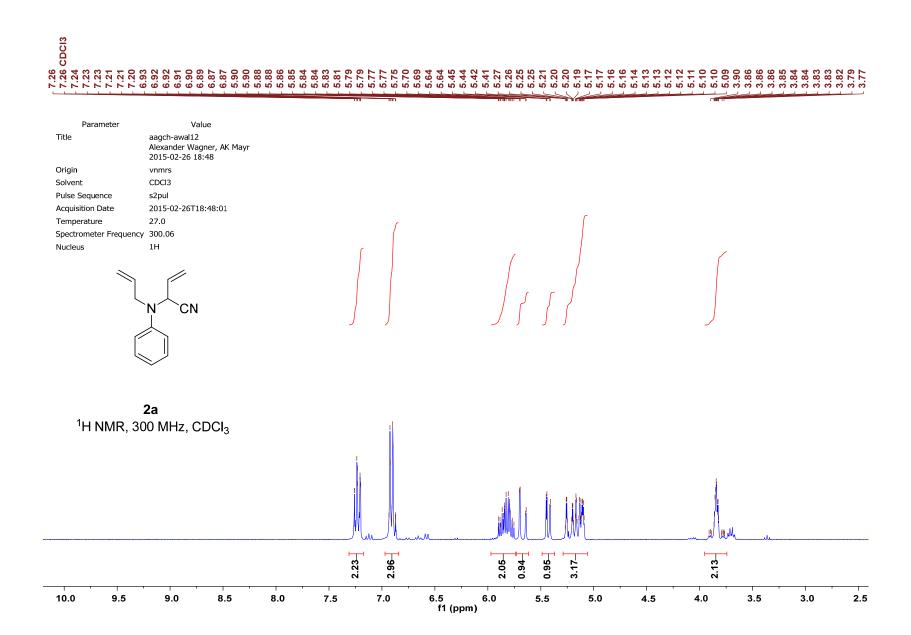
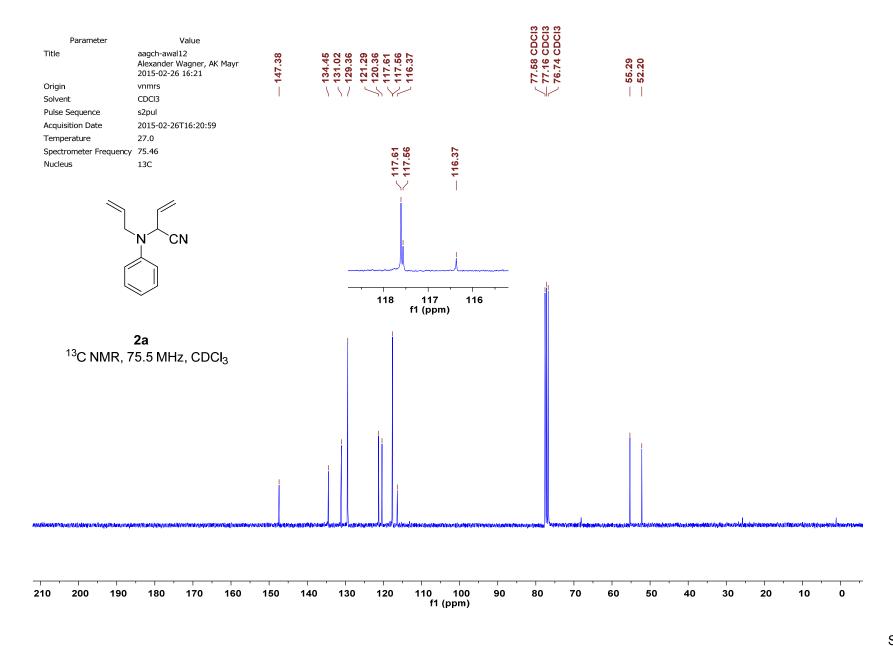
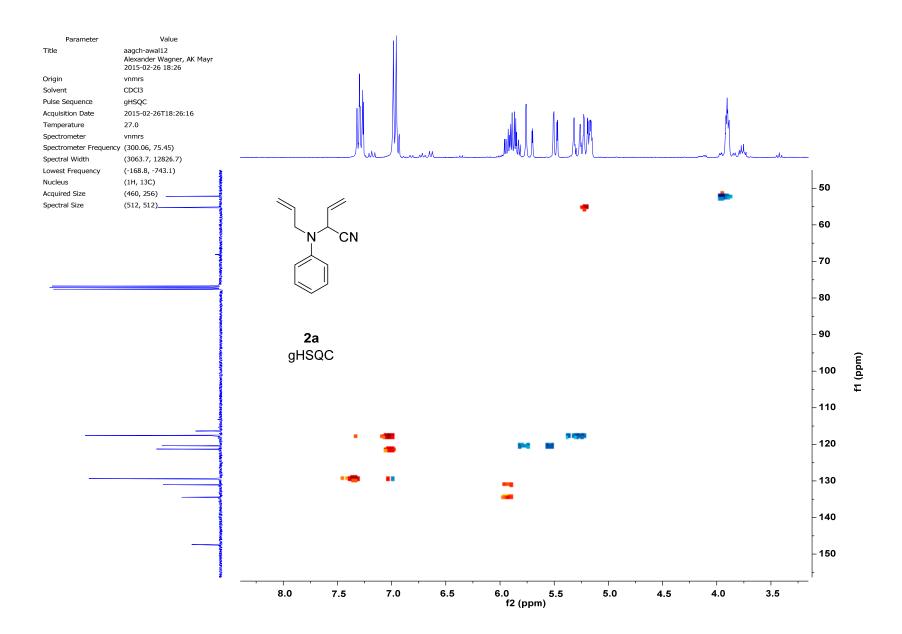


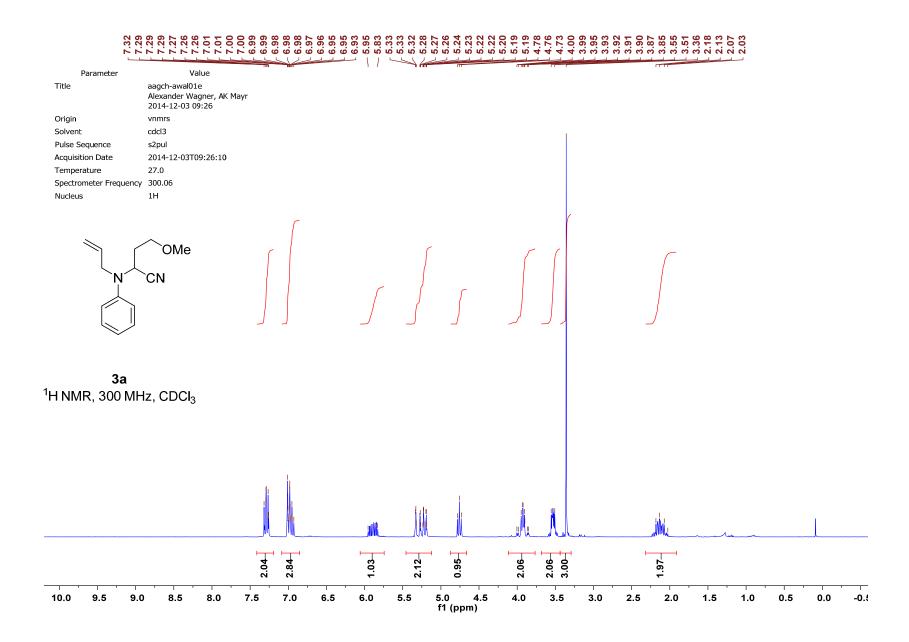
Figure S4. Radical stabilization energies (RSE) relative to methyl radical (•CH₃) and bond dissociation energies (BDE) of amino-substituted radicals and selected additional systems calculated at G3, G3B3, or G3(MP2)-RAD level of theory.

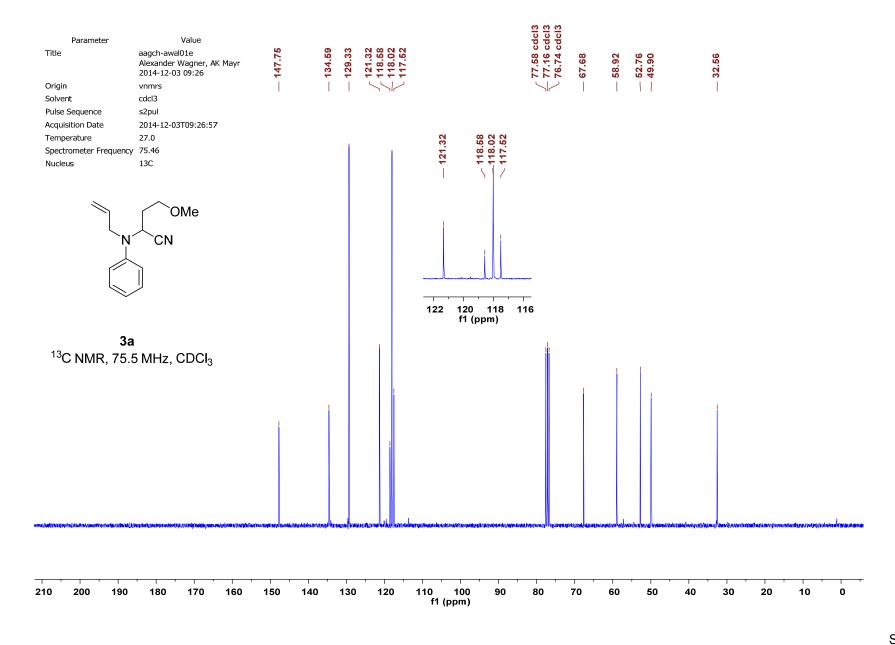
5. NMR and IR spectra of products 2-5, 7, 8, 10, and 12

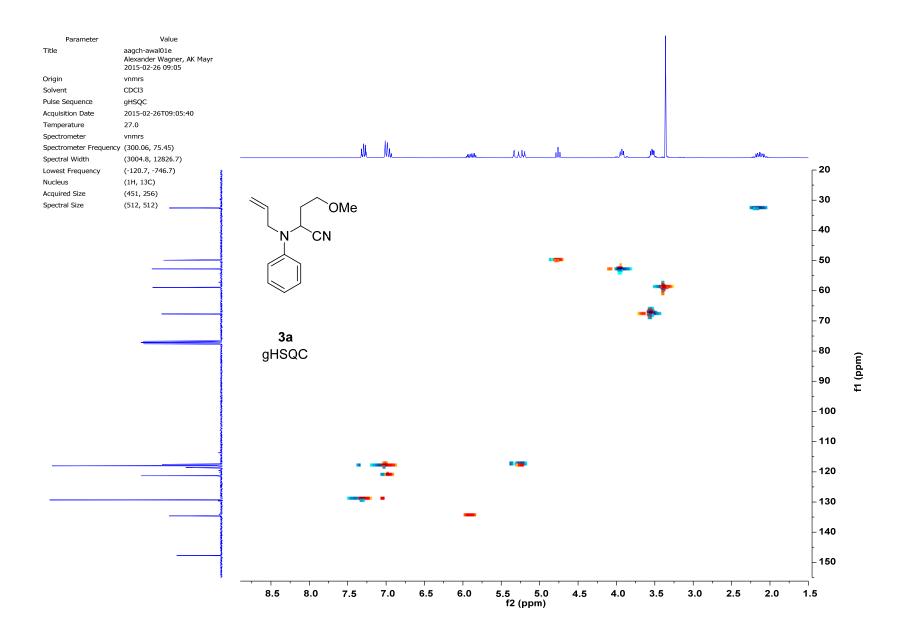


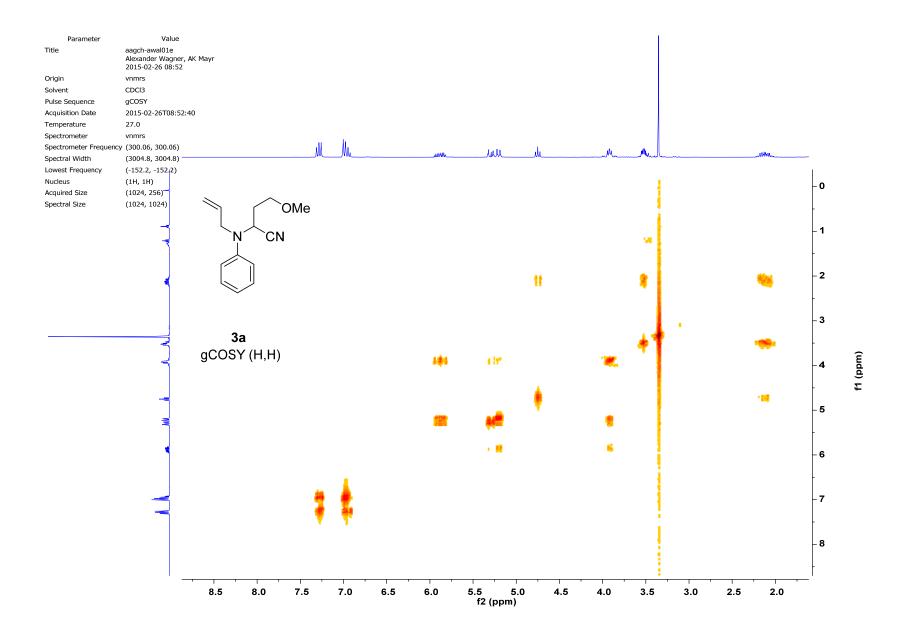


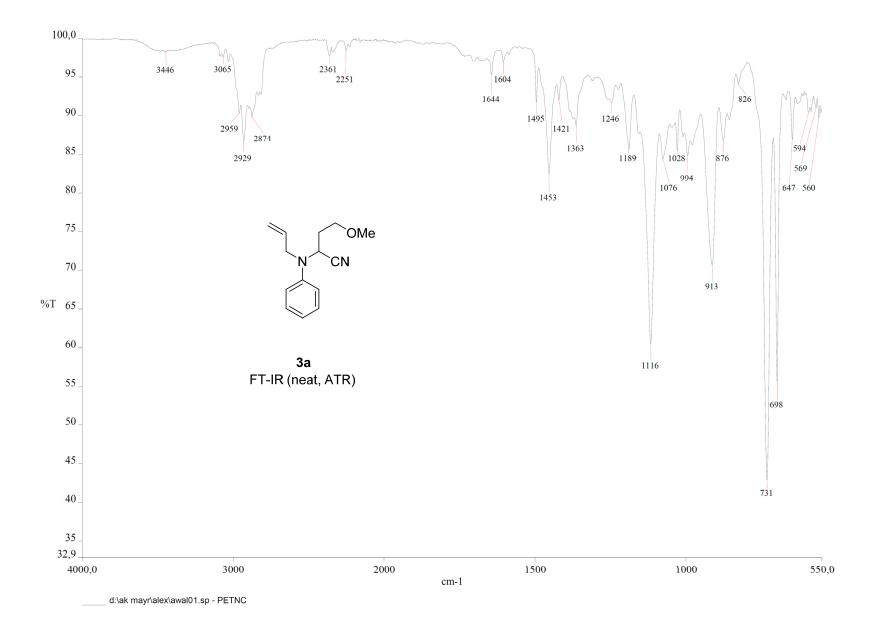


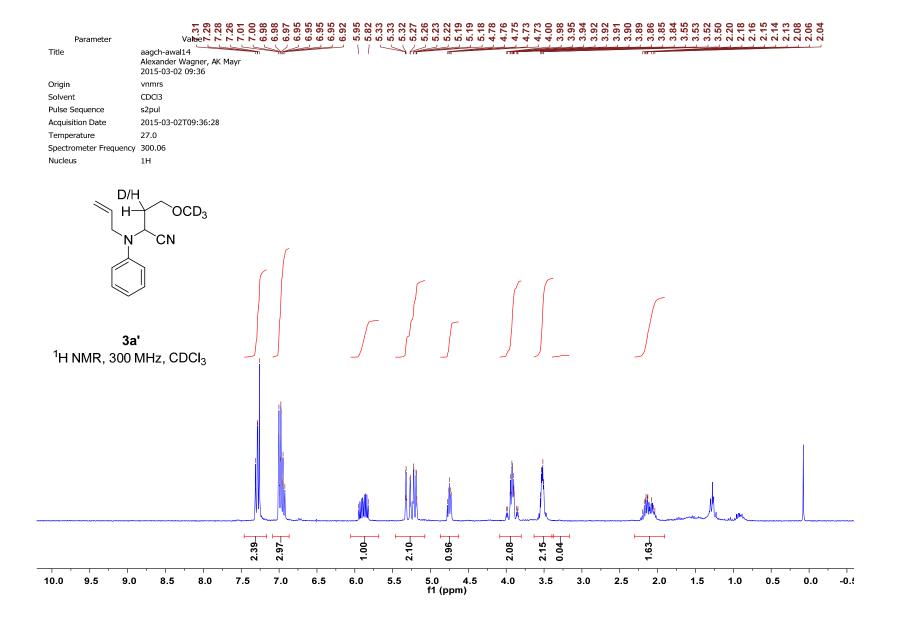


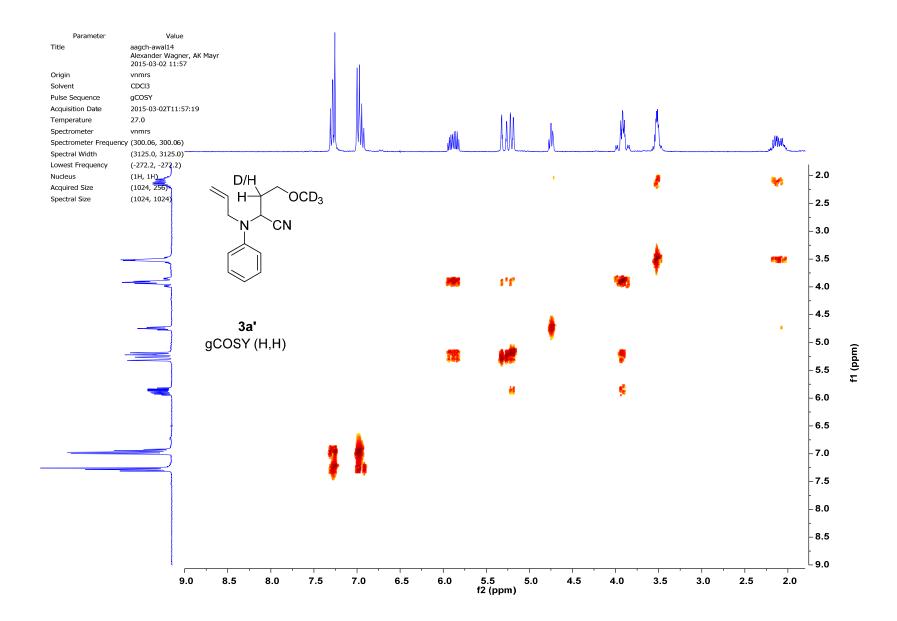


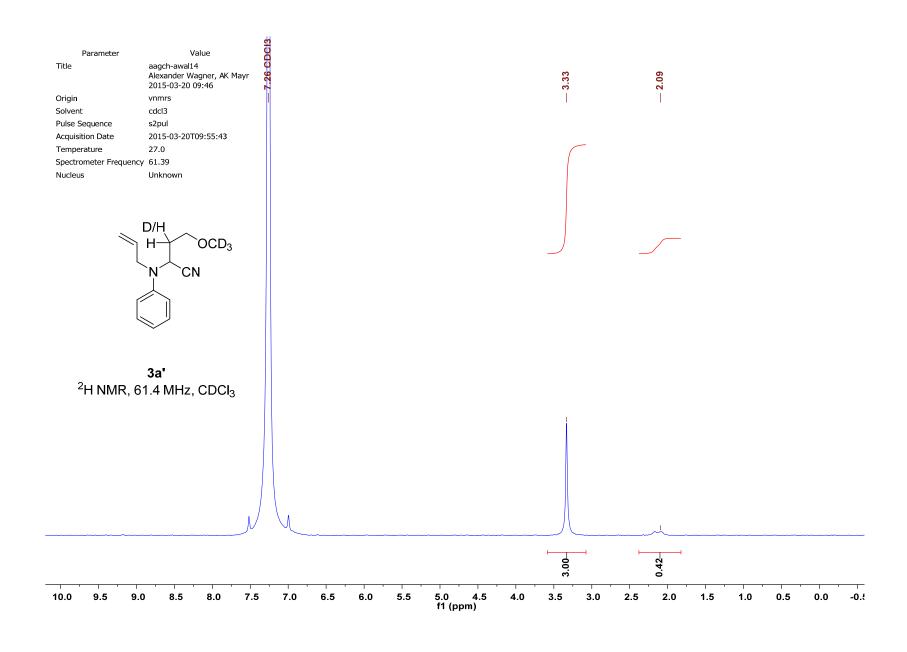


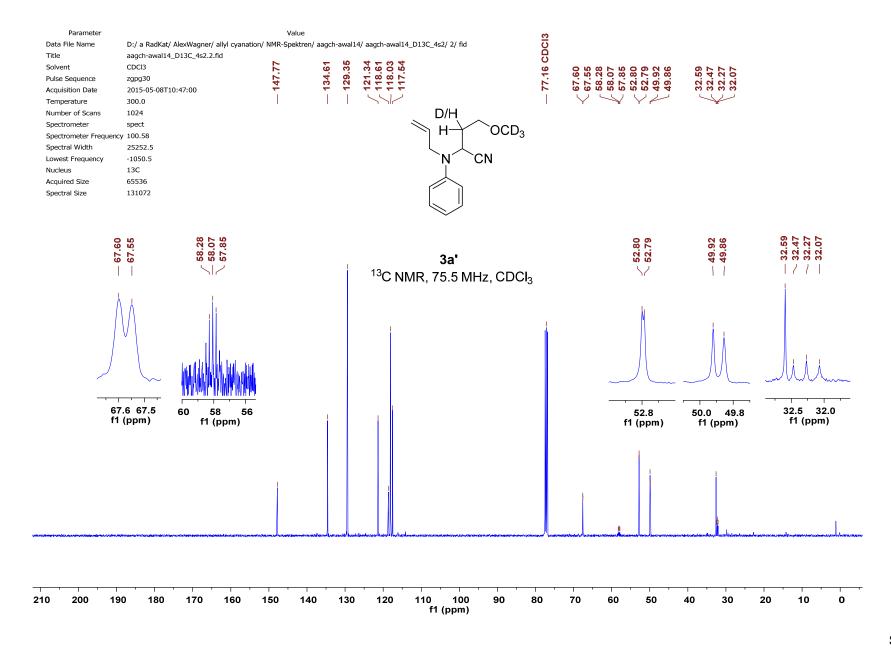


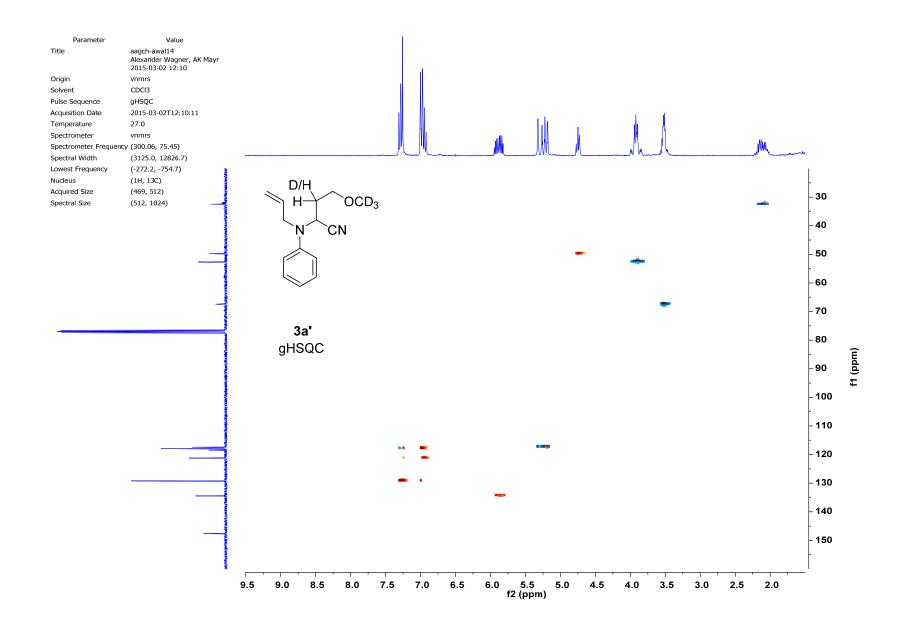


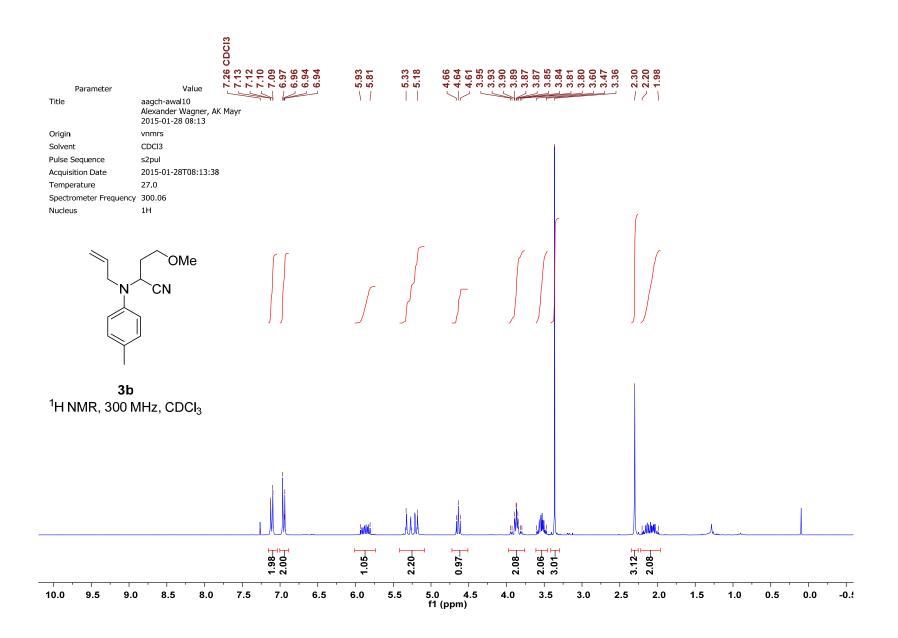


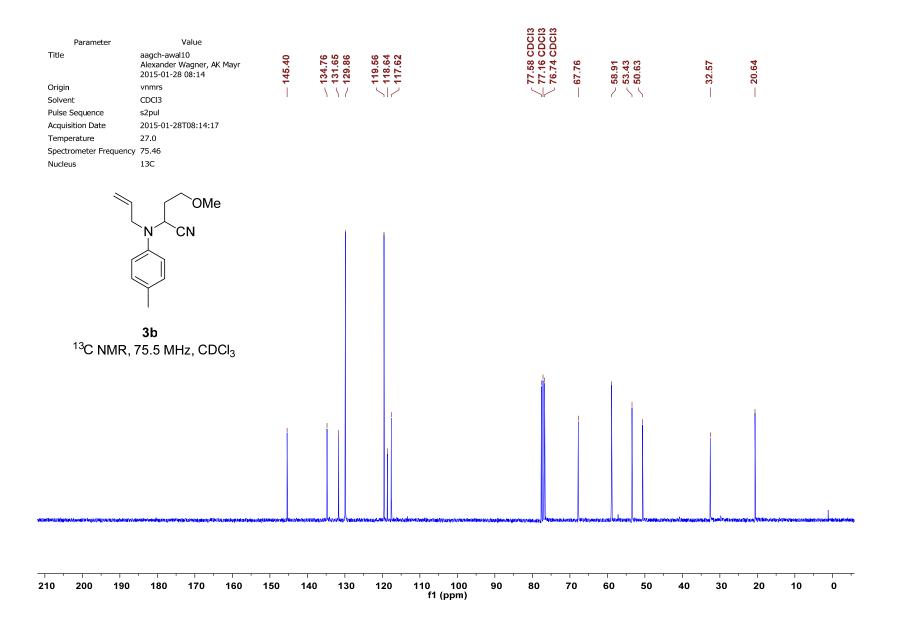


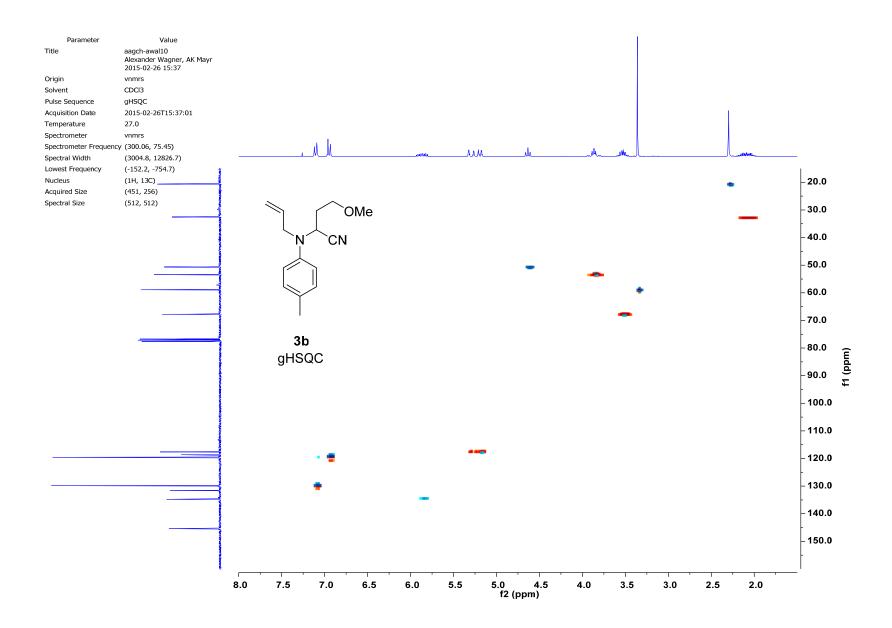


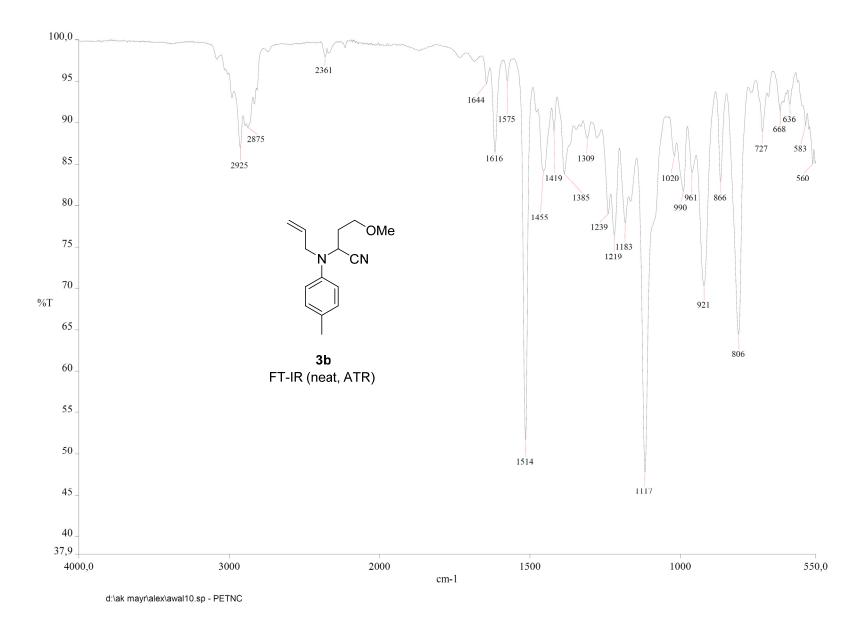


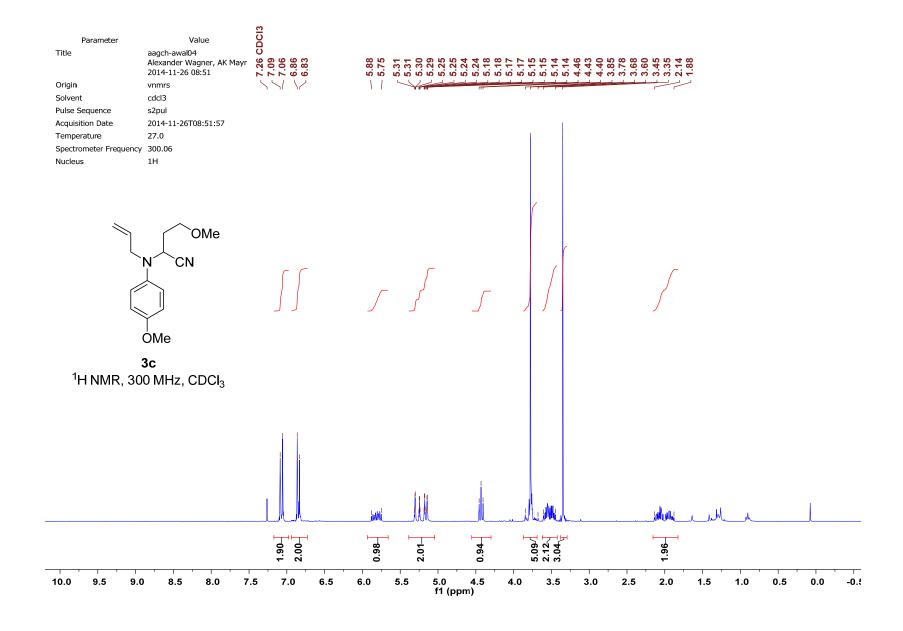


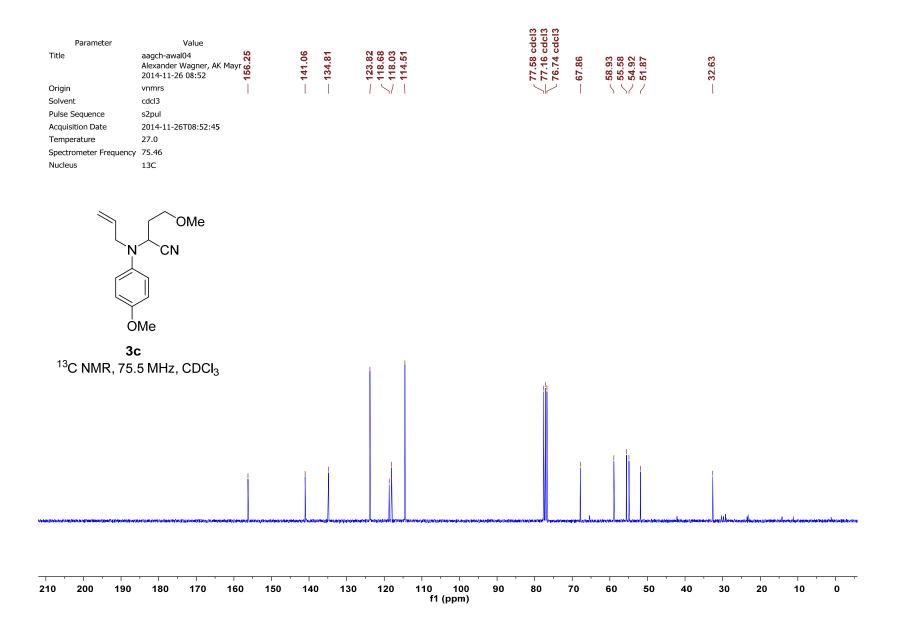


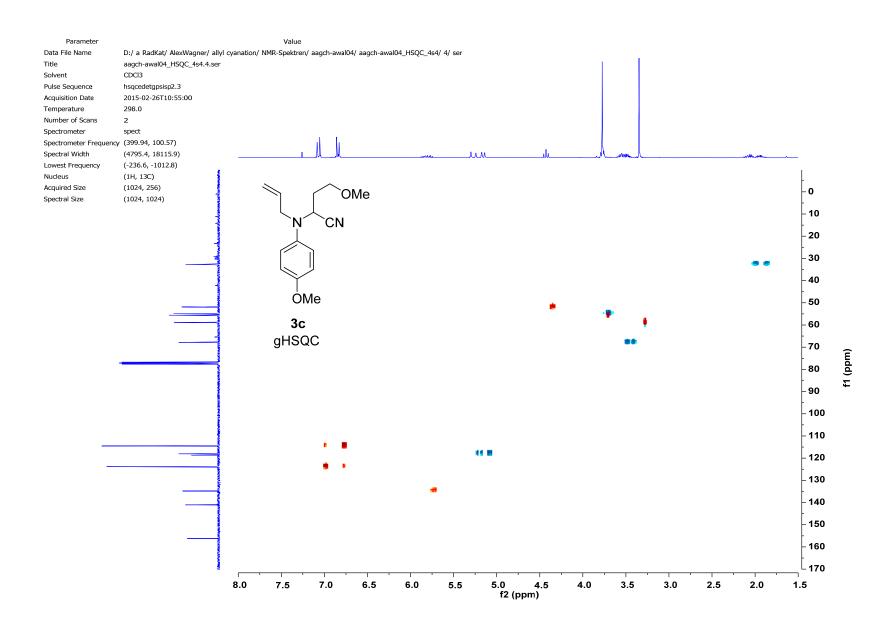


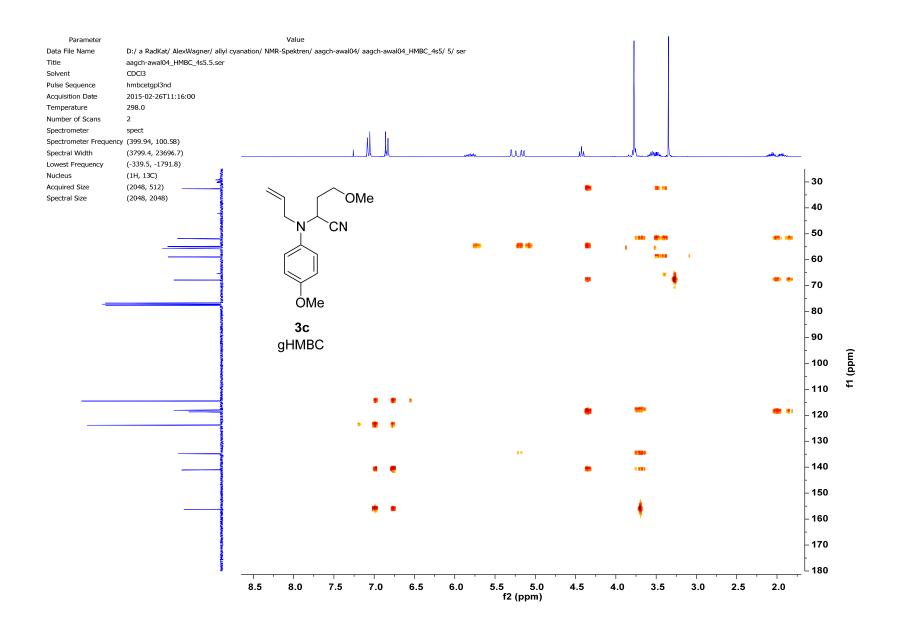


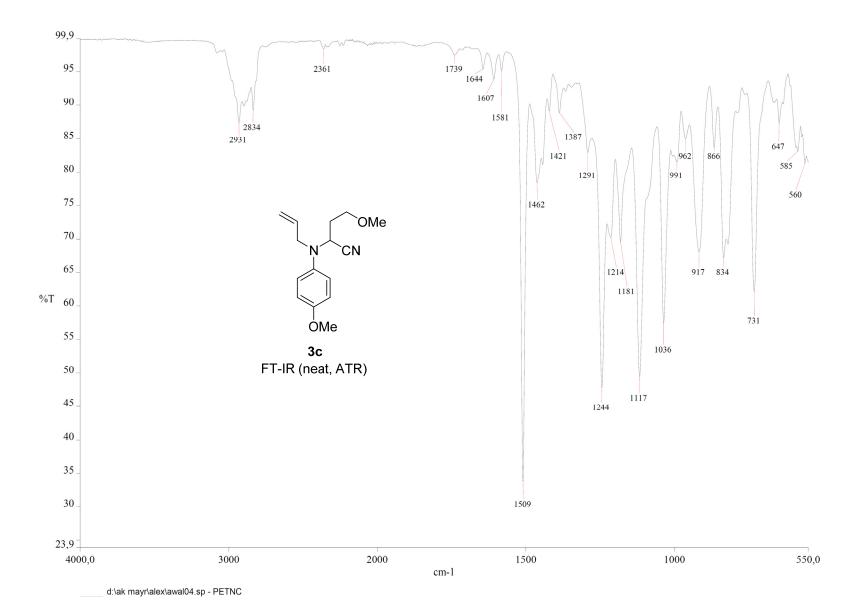


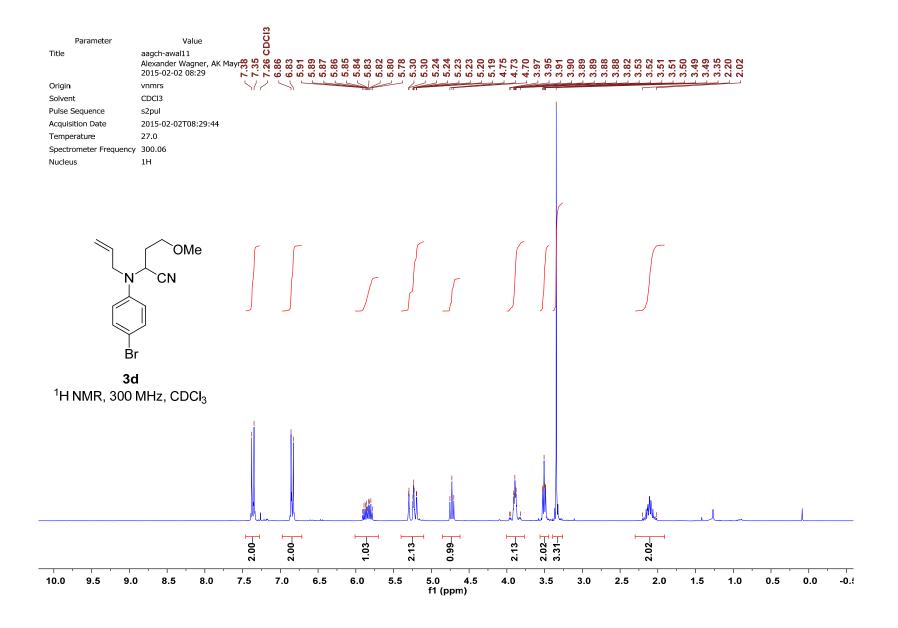


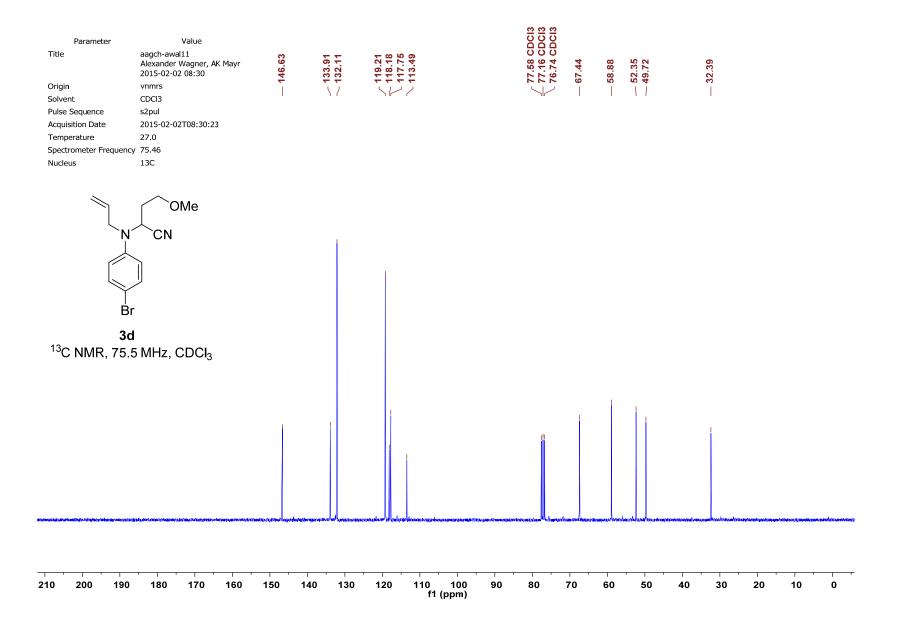


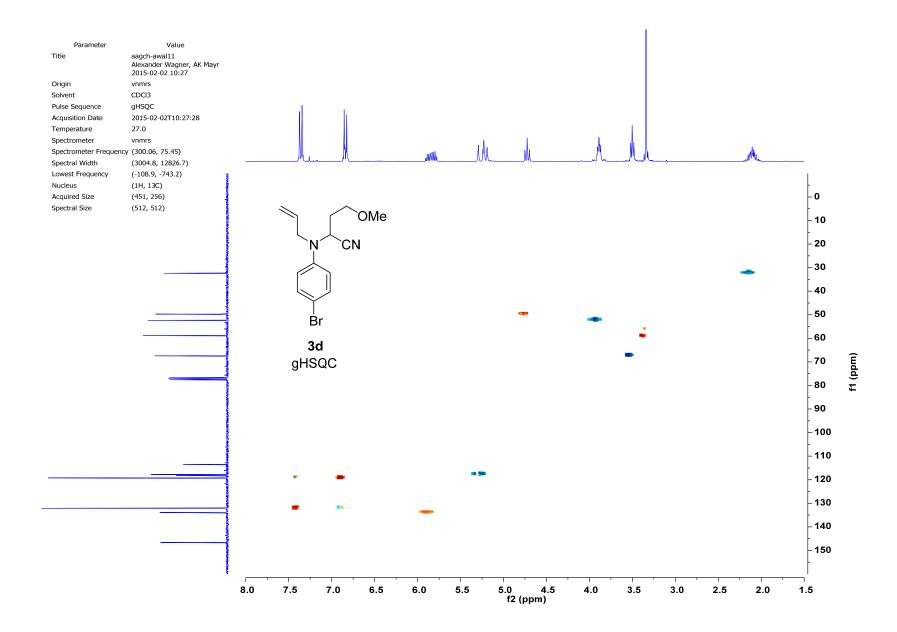


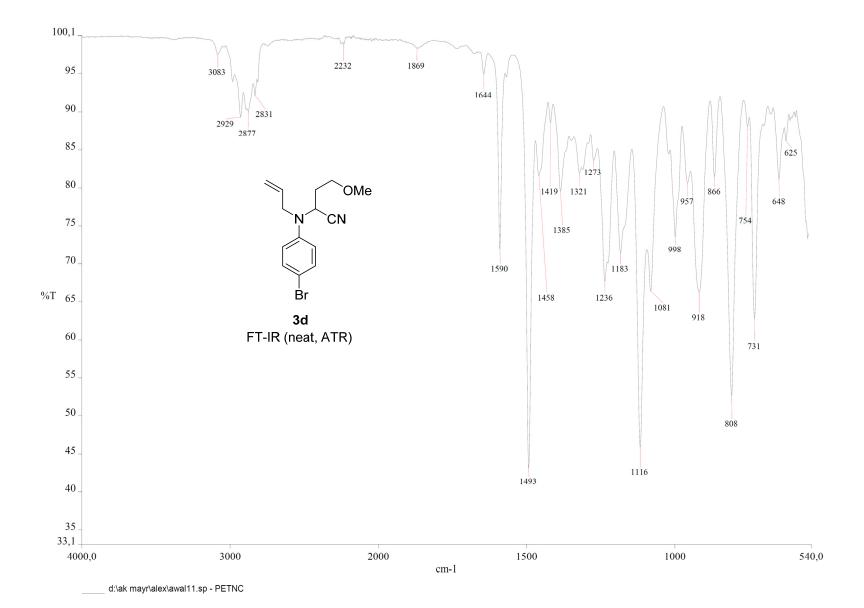


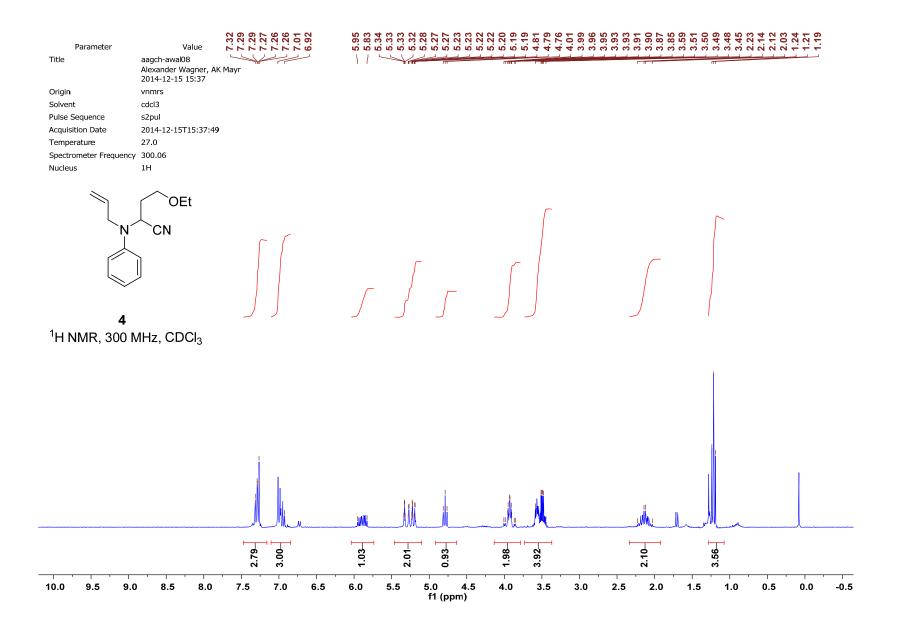


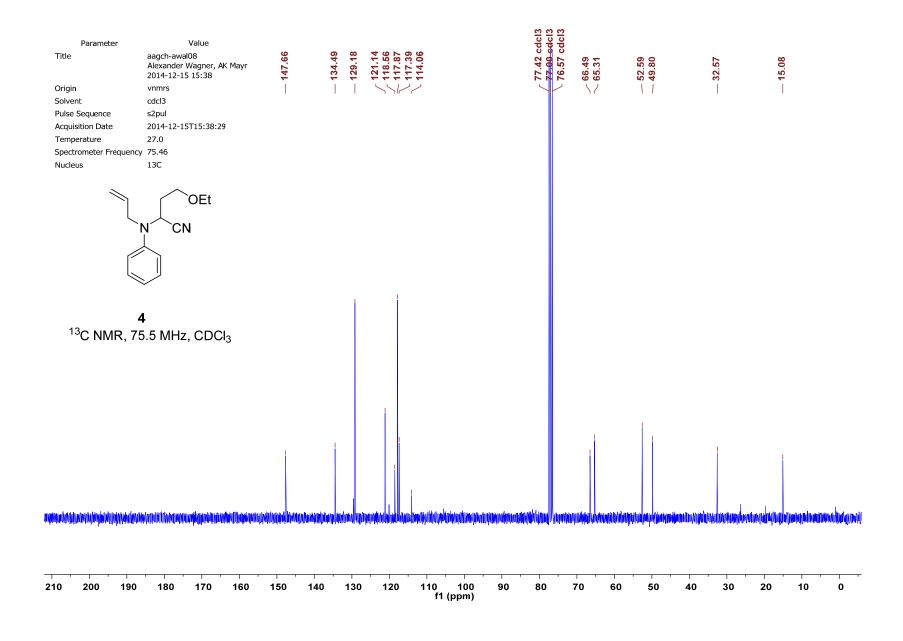


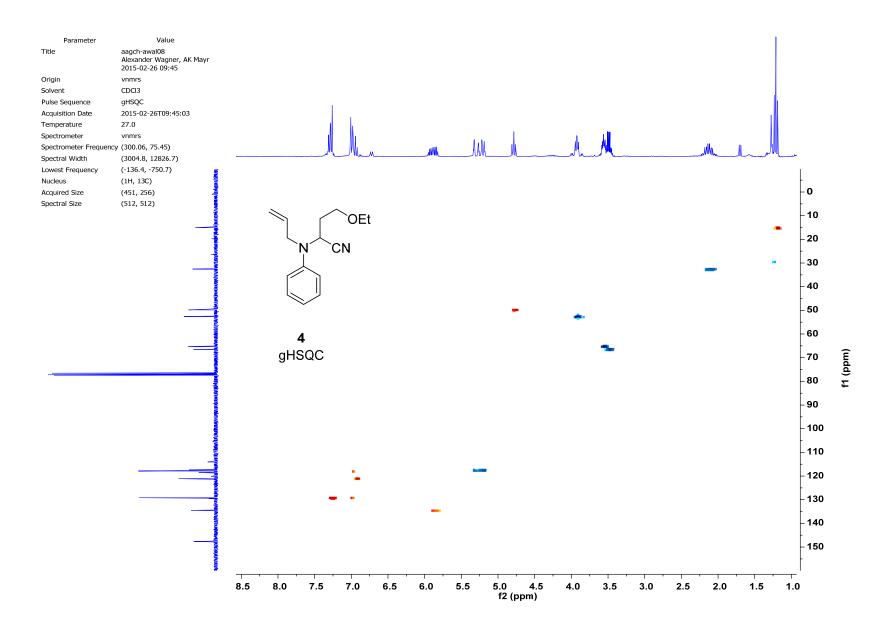


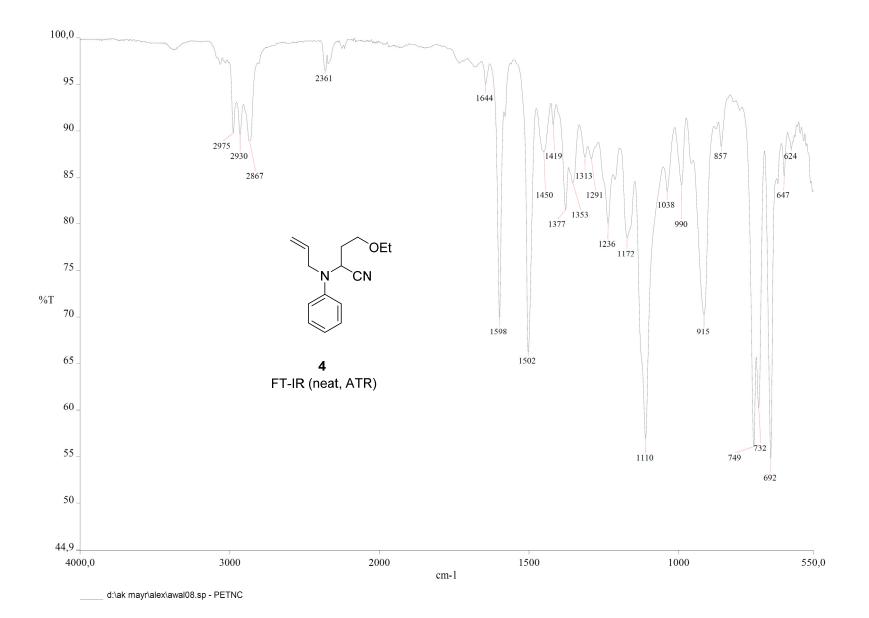


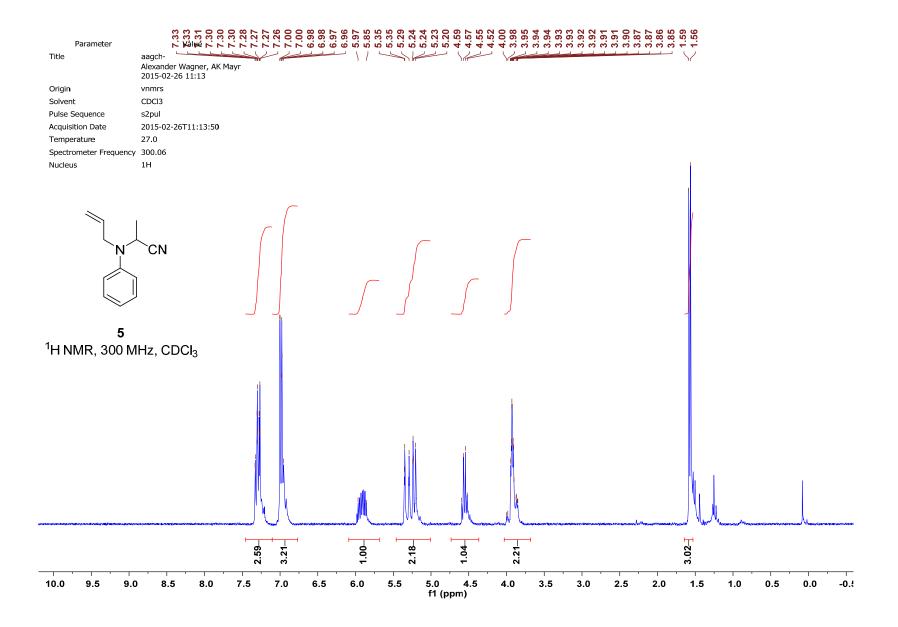


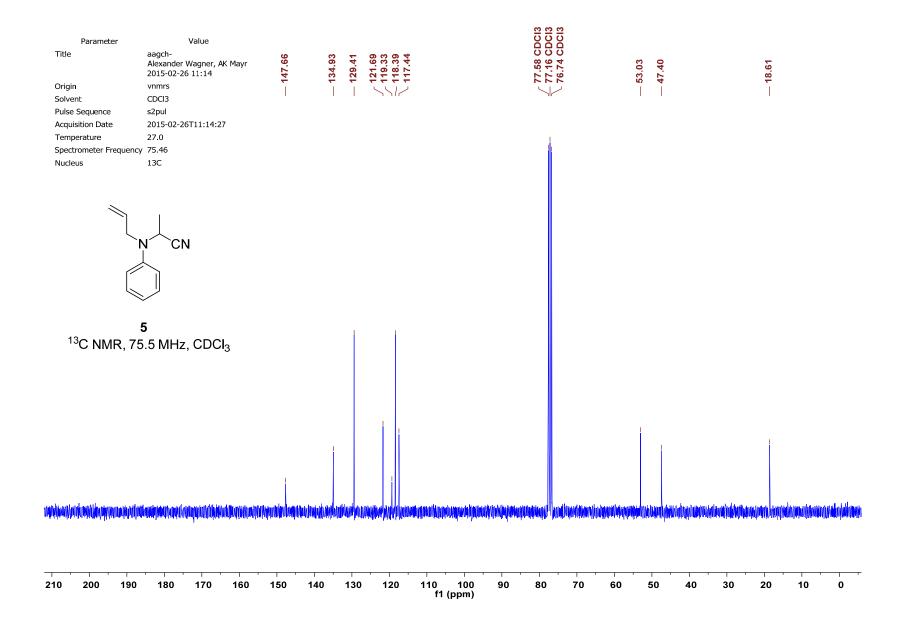


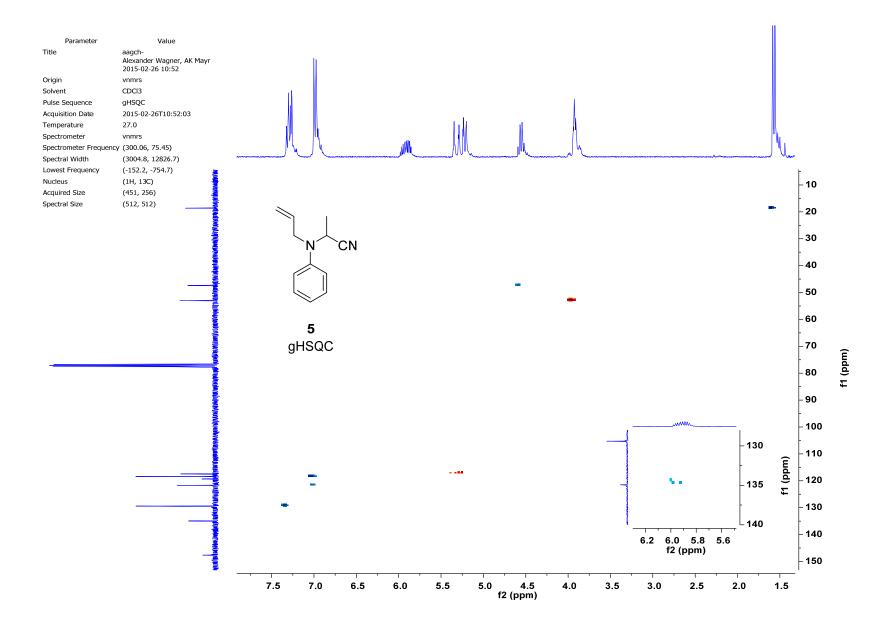


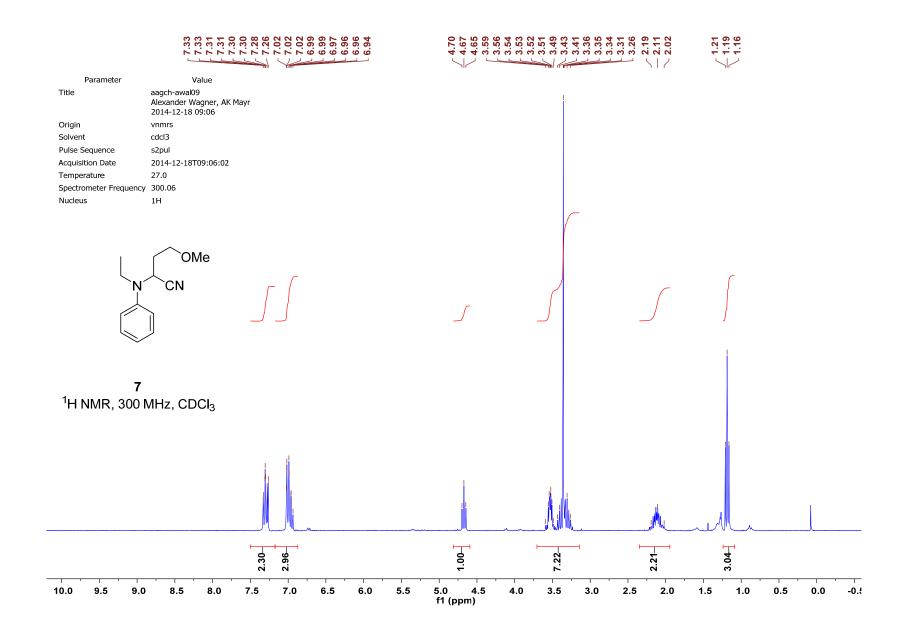


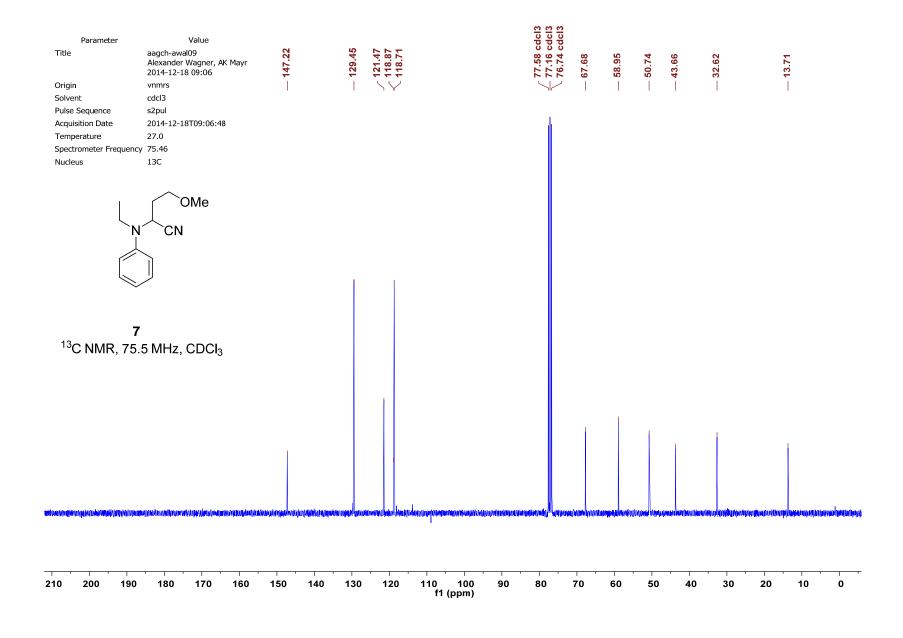


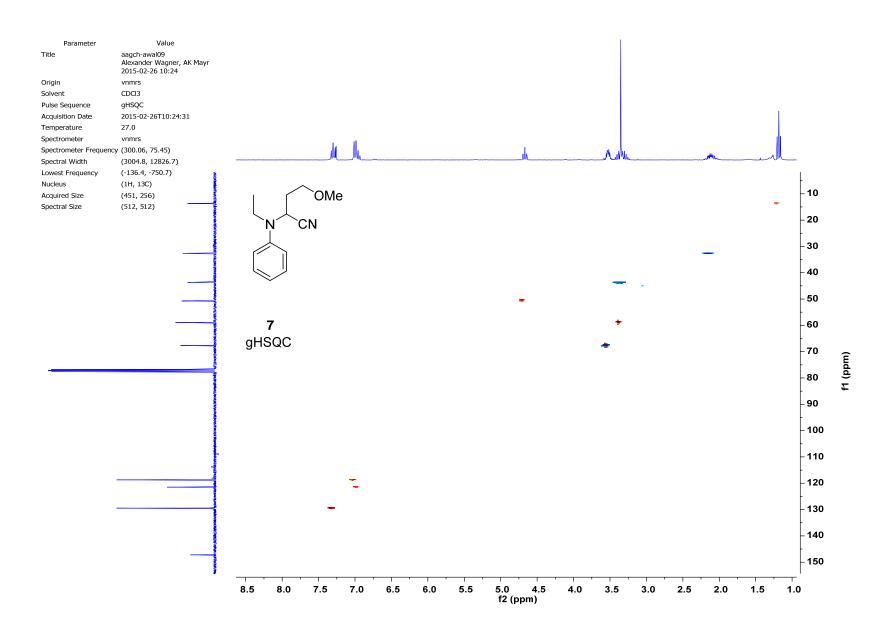


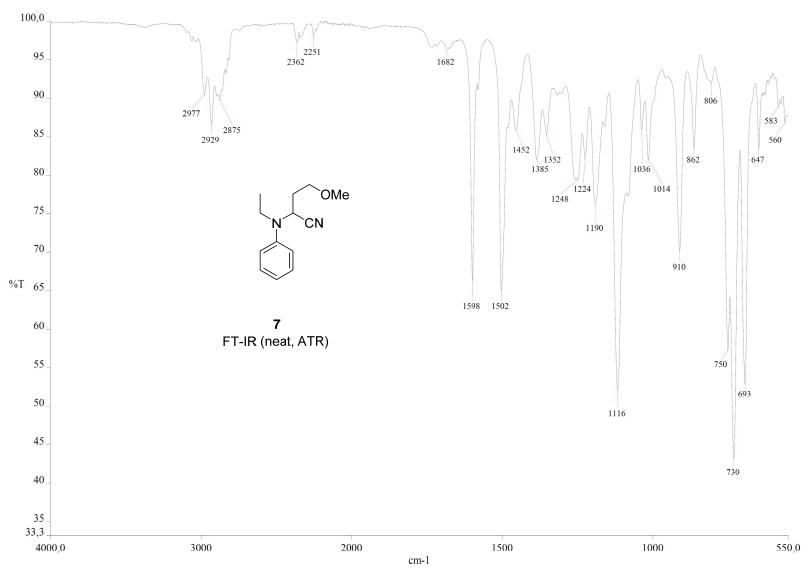




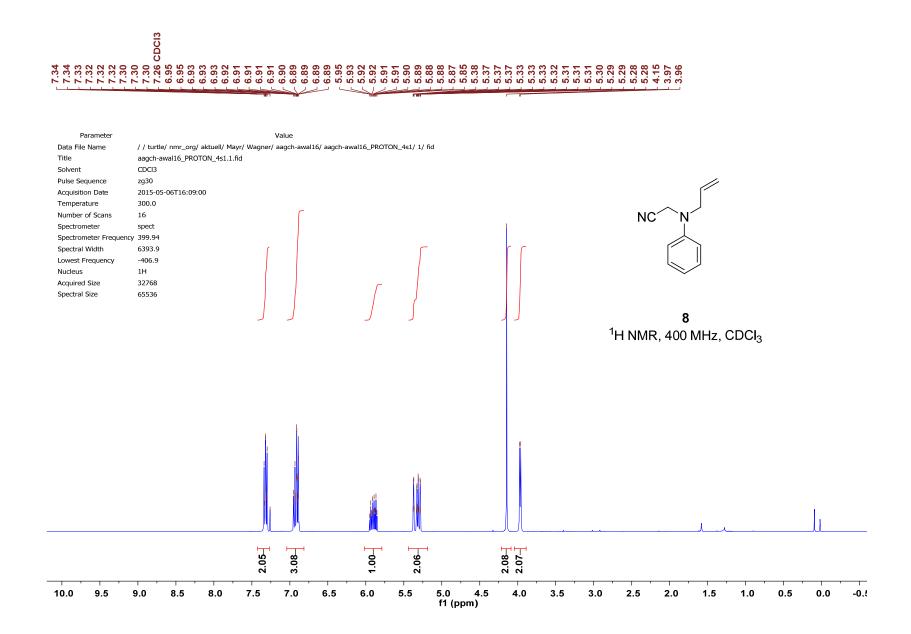


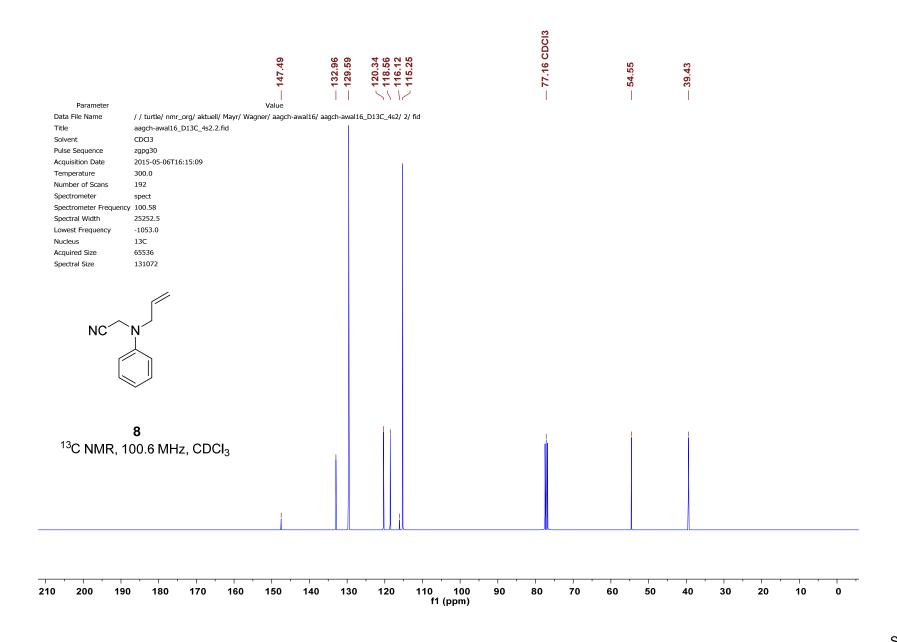


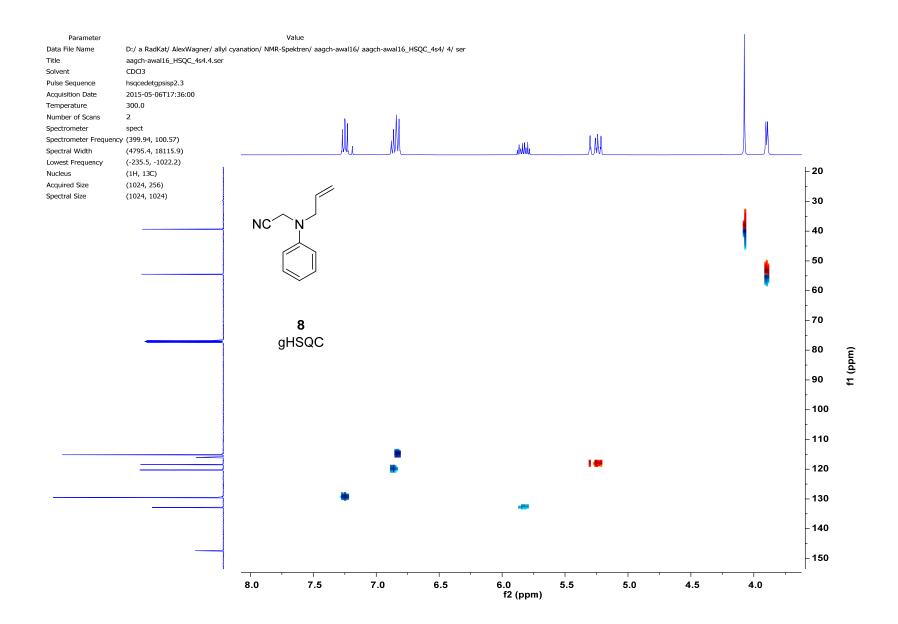


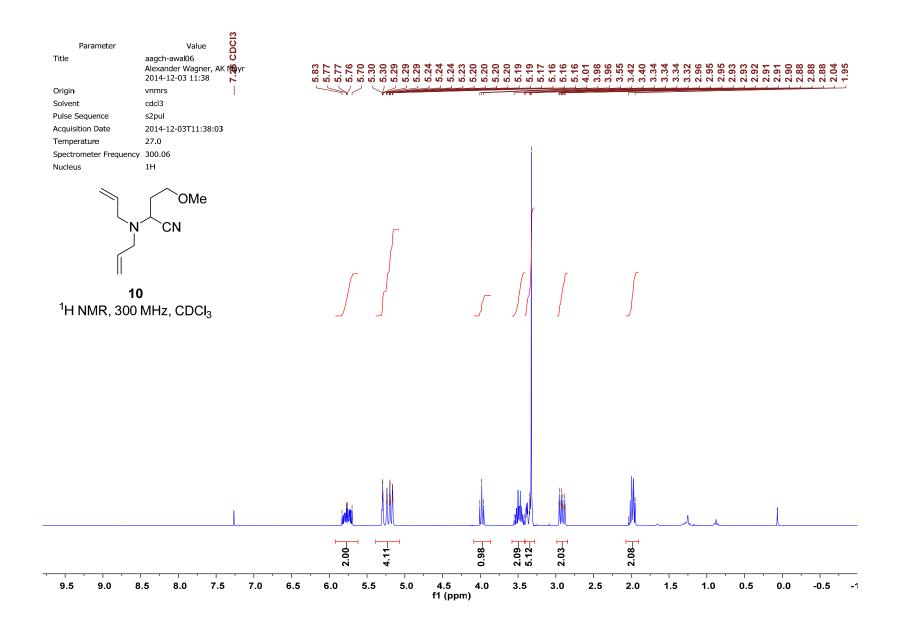


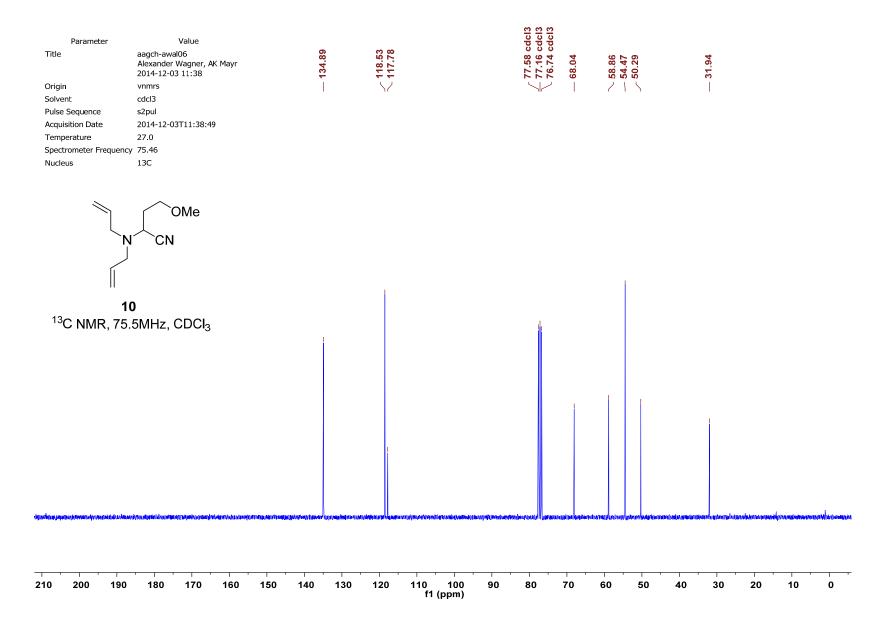
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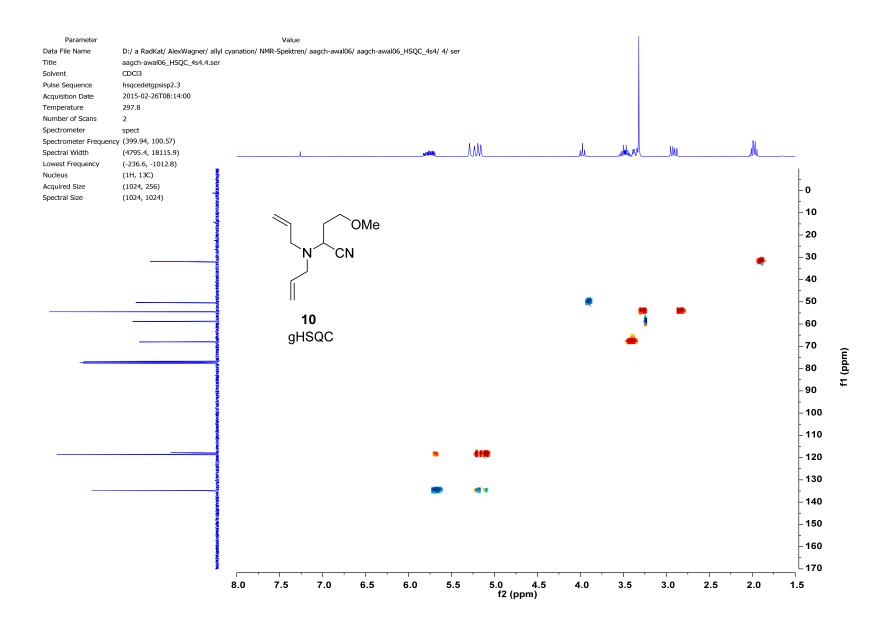


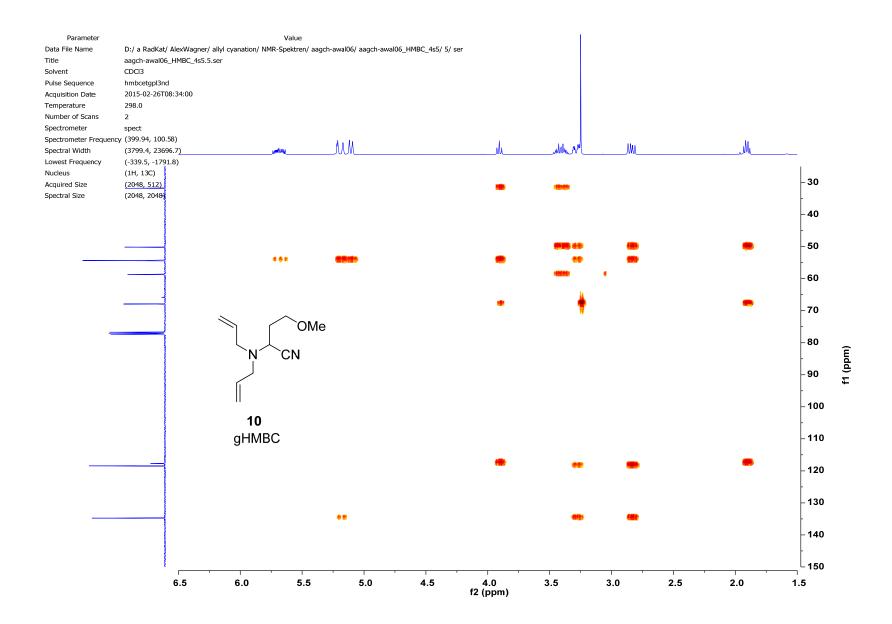


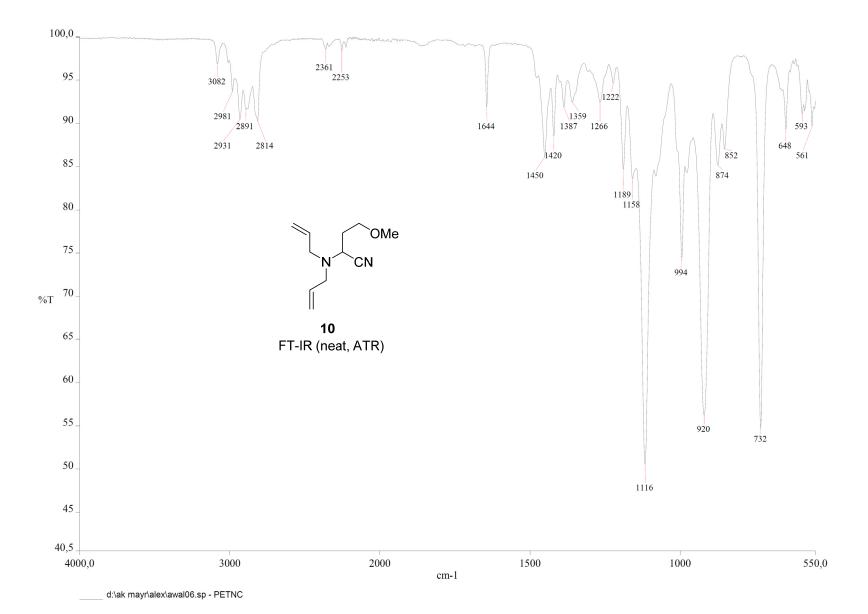


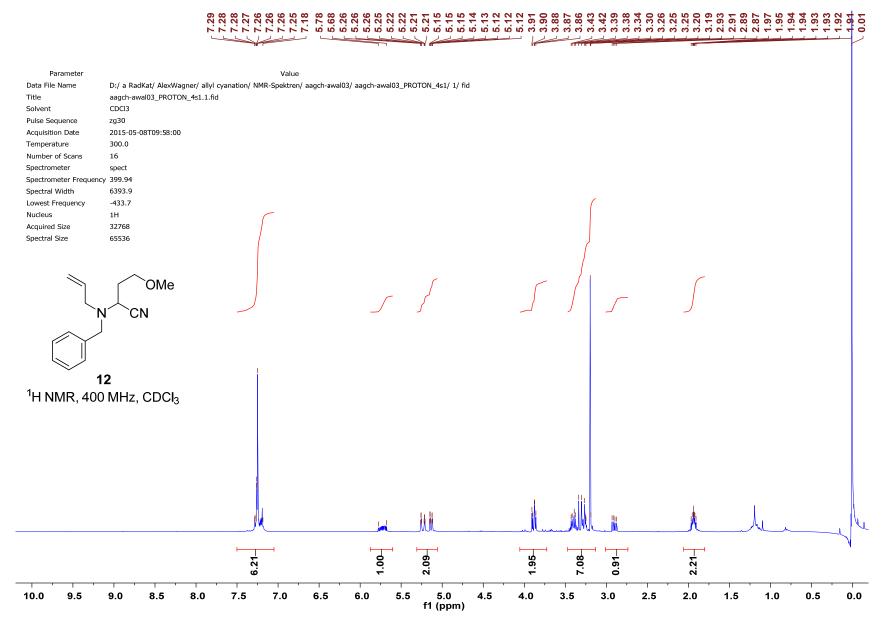


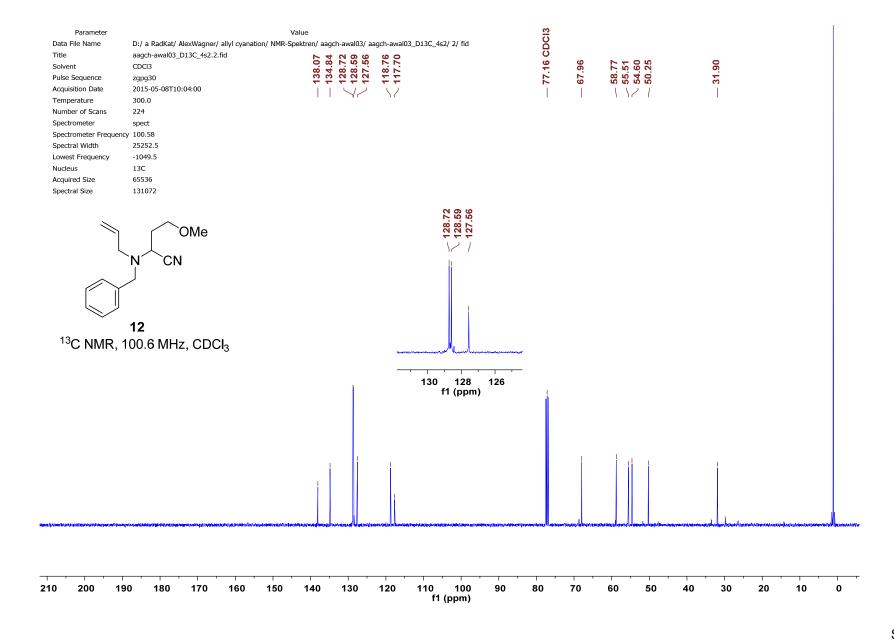


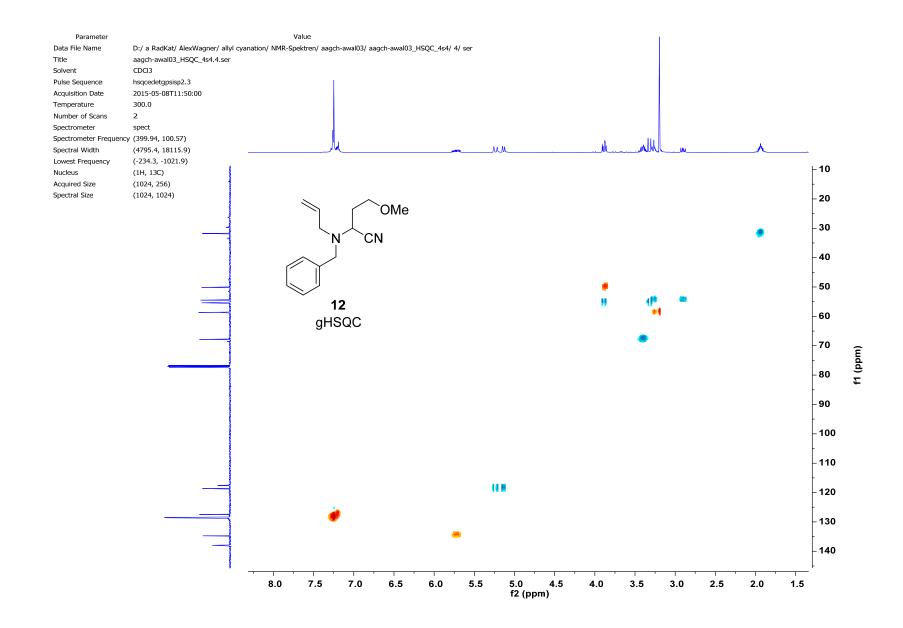












6. NMR spectra of allylanilines and allylamines 1a-d, 6a,b, and 11

