

Supporting Information

Alkaline-Earth Metal Bis[bis(trimethylsilyl)amide] Complexes with Weakly Coordinating 2,2,5,5-Tetramethyltetrahydrofuran Ligands

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Dissociation equilibria

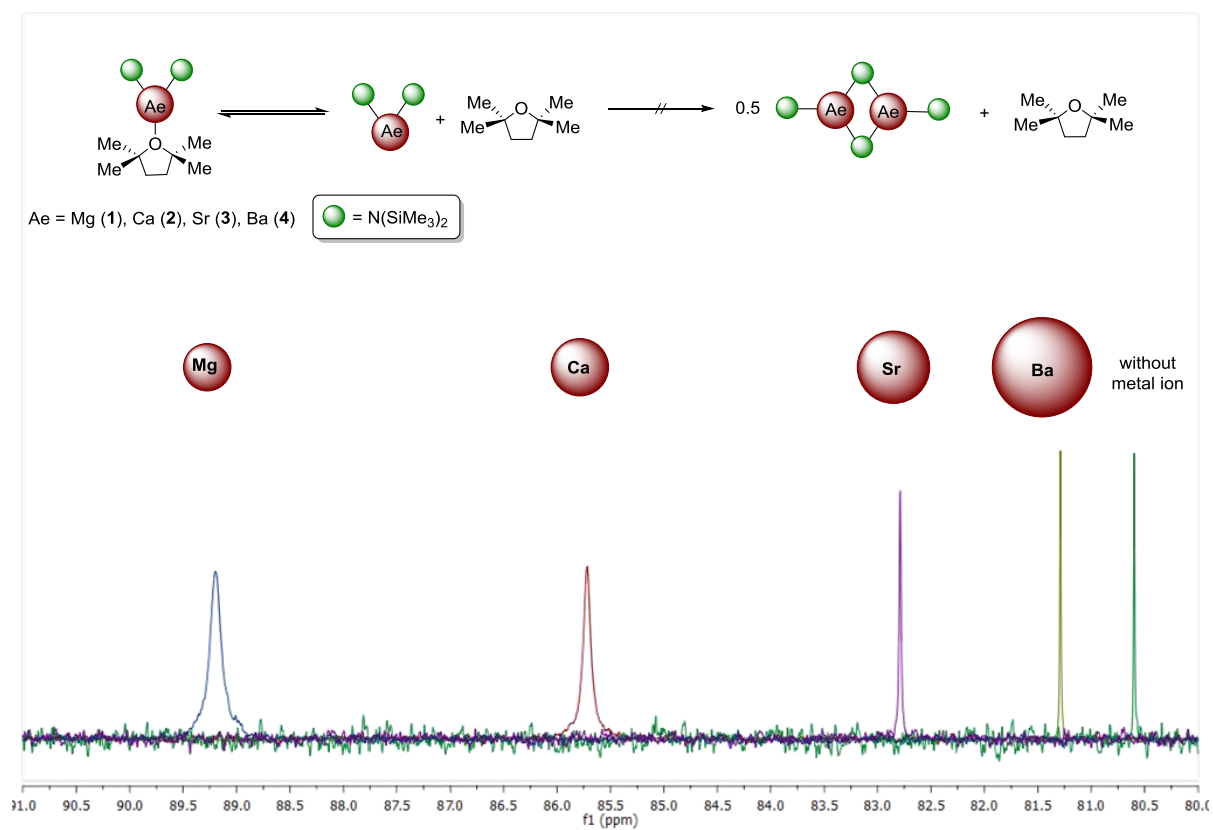


Figure S1: $^{13}\text{C}\{^1\text{H}\}$ NMR signals of quaternary carbon atom of Me_4thf (400 MHz, C_6D_6 , 296 K) from the adducts $[(\text{Me}_4\text{thf})\text{Ae}(\text{hmnds})_2]$ ($\text{Ae} = \text{Mg} - \text{Ba}$).

I. Analytical data

$[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ (**1**)

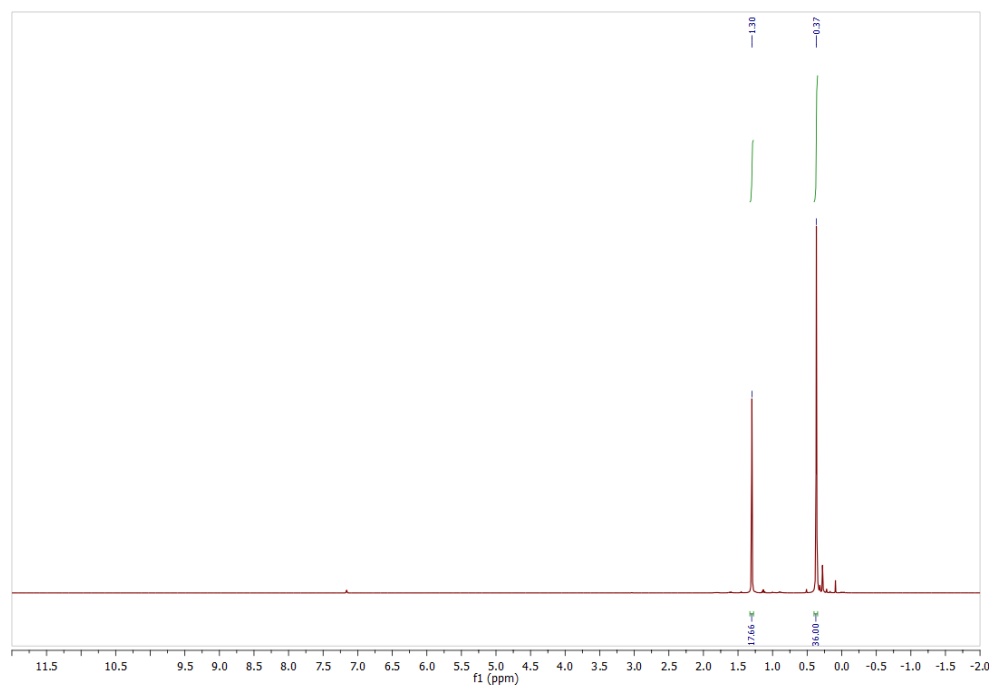


Figure S2: ^1H -NMR (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ **1**.

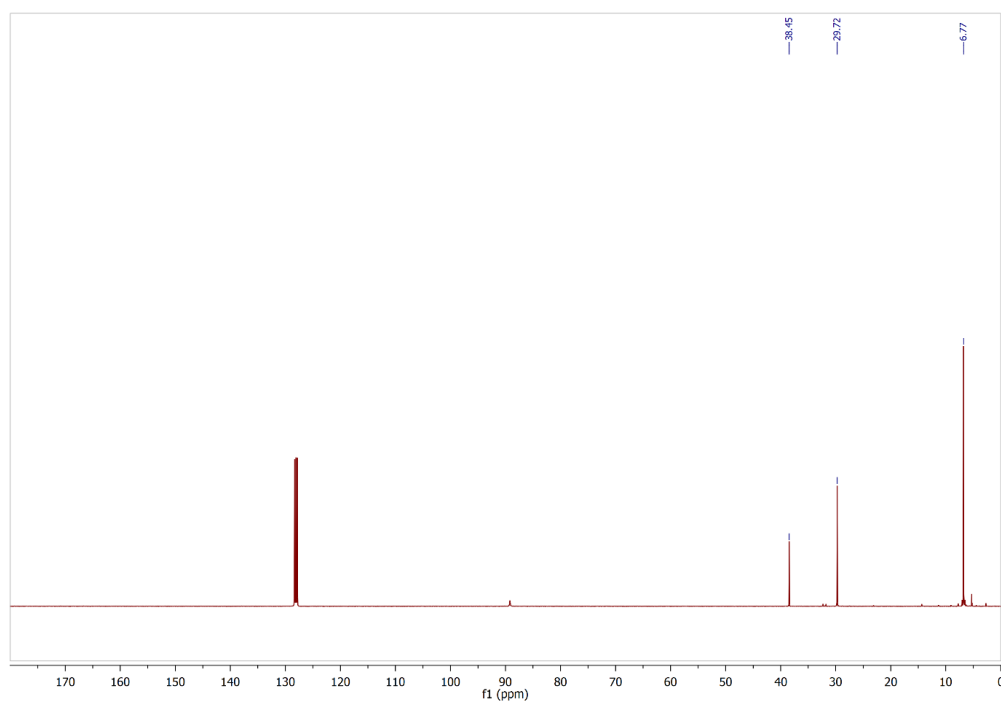


Figure S3: $^{13}\text{C}\{^1\text{H}\}$ -NMR (101 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ **1**.

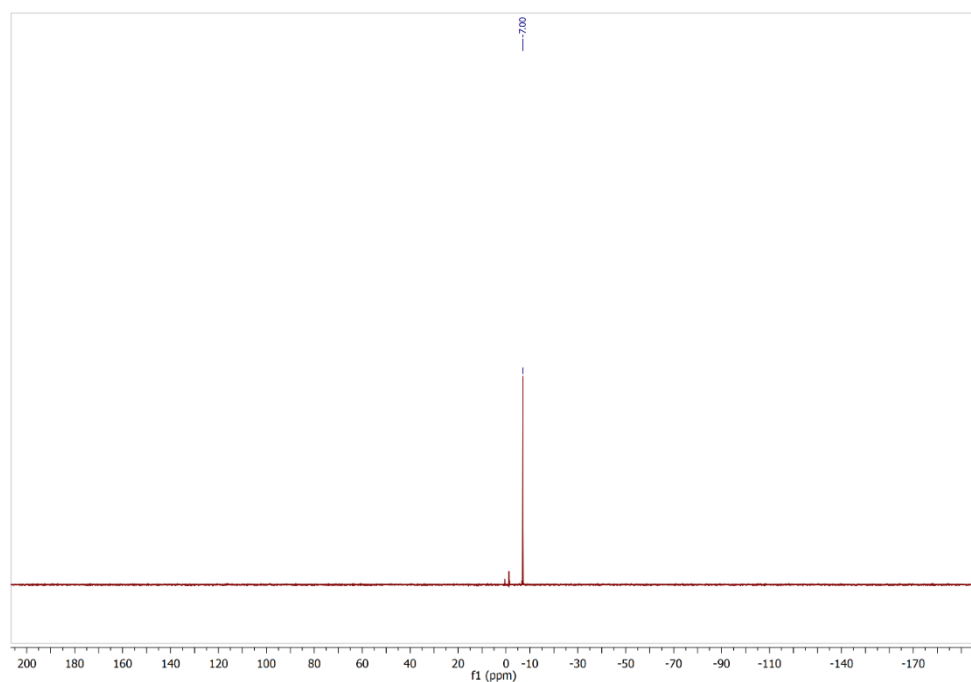


Figure S4: ^{29}Si - ^1H -DEPT-NMR (79,5 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ **1**.

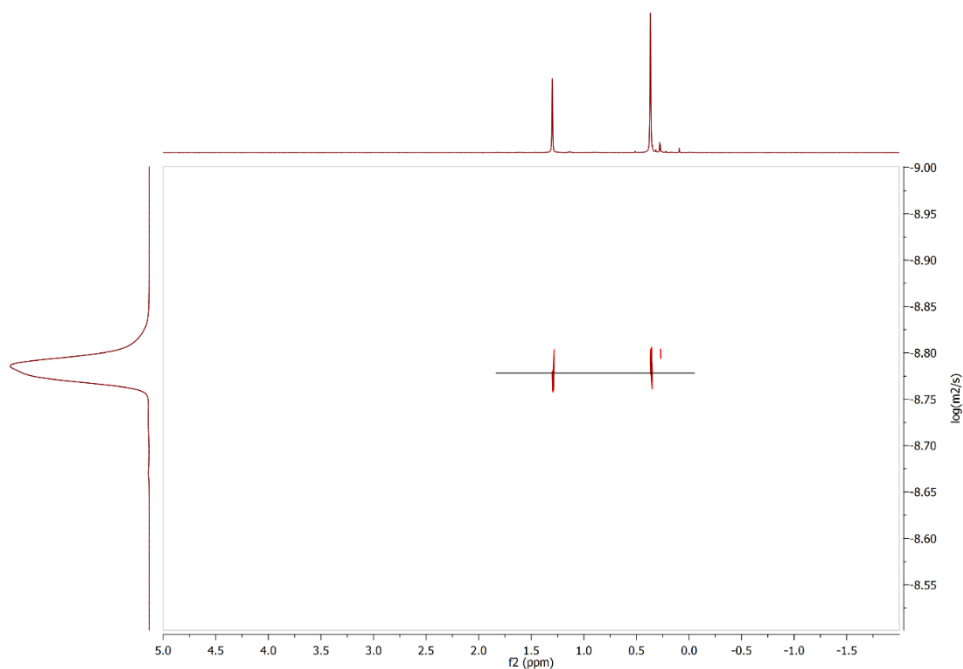


Figure S5: ^1H -DOSY (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ **1**.

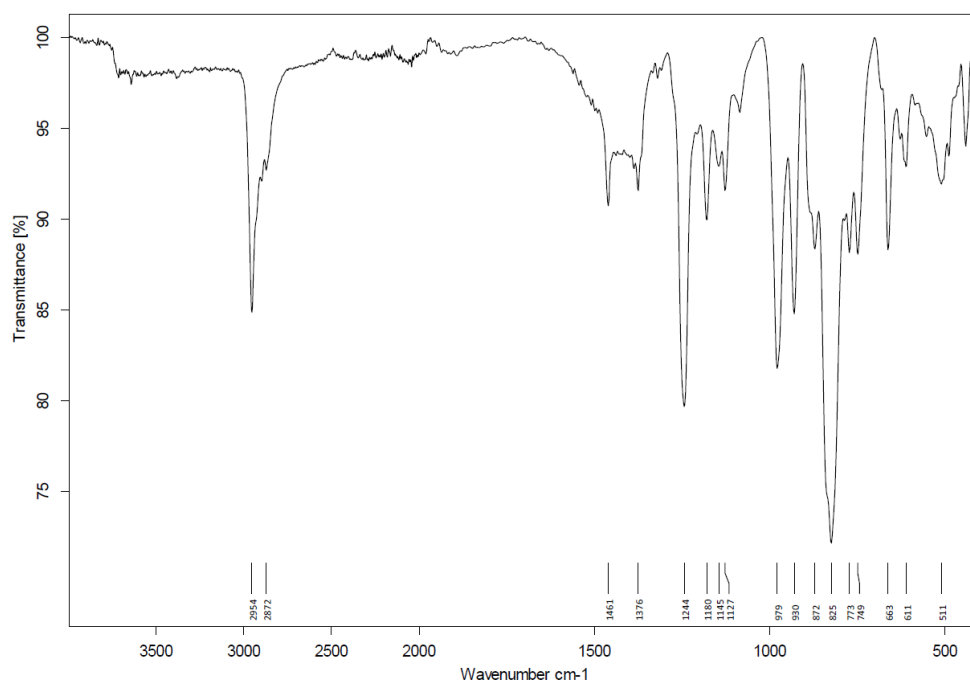


Figure S6: IR (ATR) of $[(\text{Me}_4\text{thf})\text{Mg}(\text{hmds})_2]$ **1**.

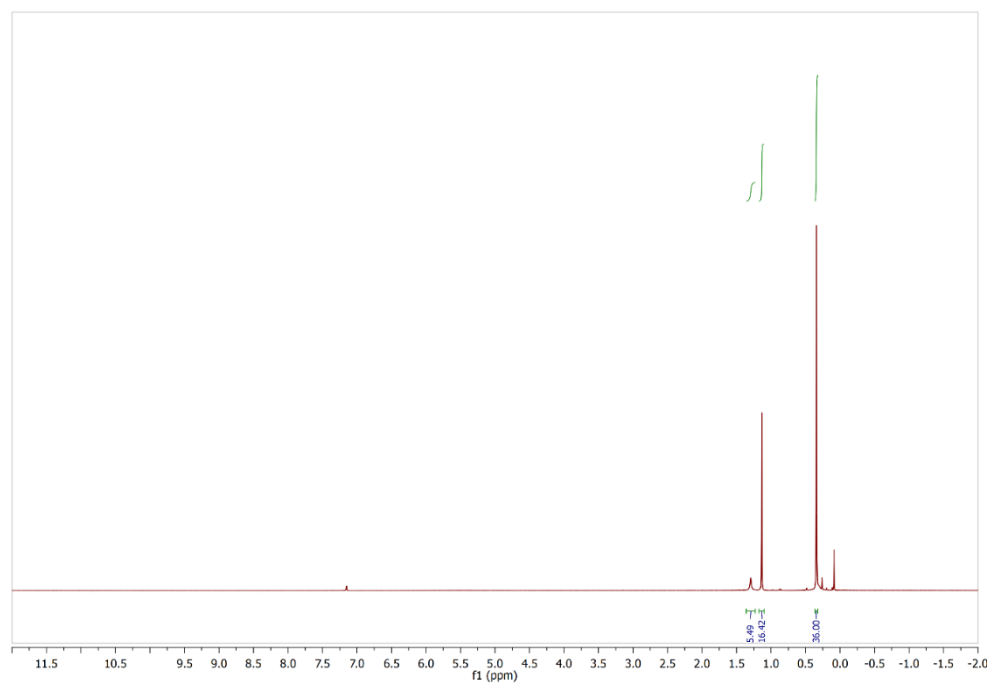
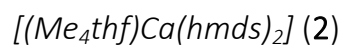


Figure S7: ^1H -NMR (400 MHz, C_6D_6 , 297K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ca}(\text{hmds})_2]$ **2**.

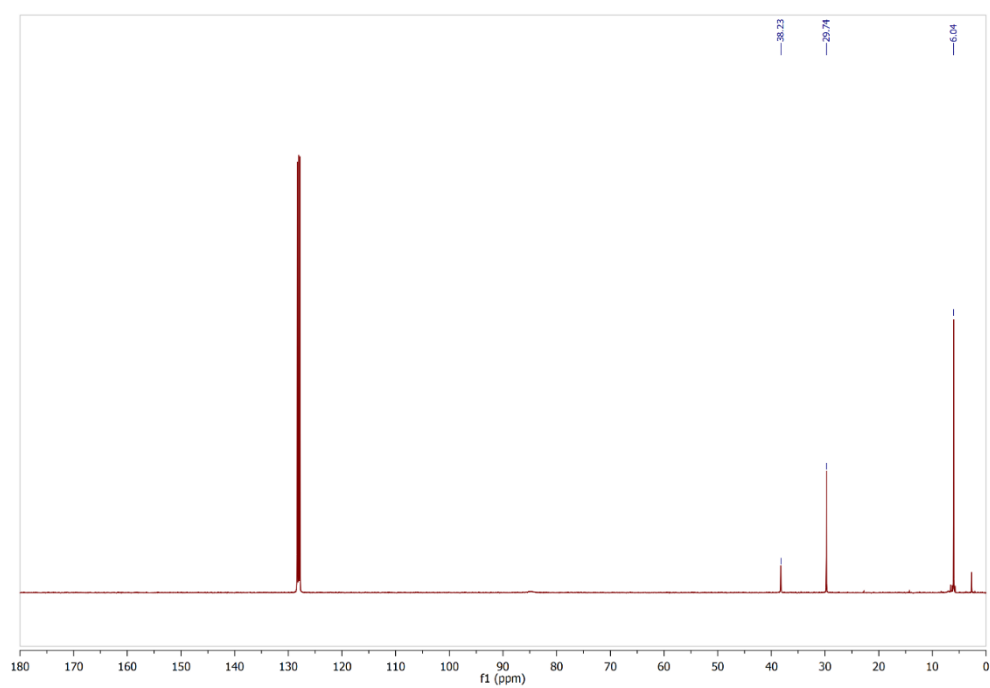


Figure S8: $^{13}\text{C}\{^1\text{H}\}$ -NMR (101 MHz, C_6D_6 , 297K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ca}(\text{hmds})_2]$ **2**.

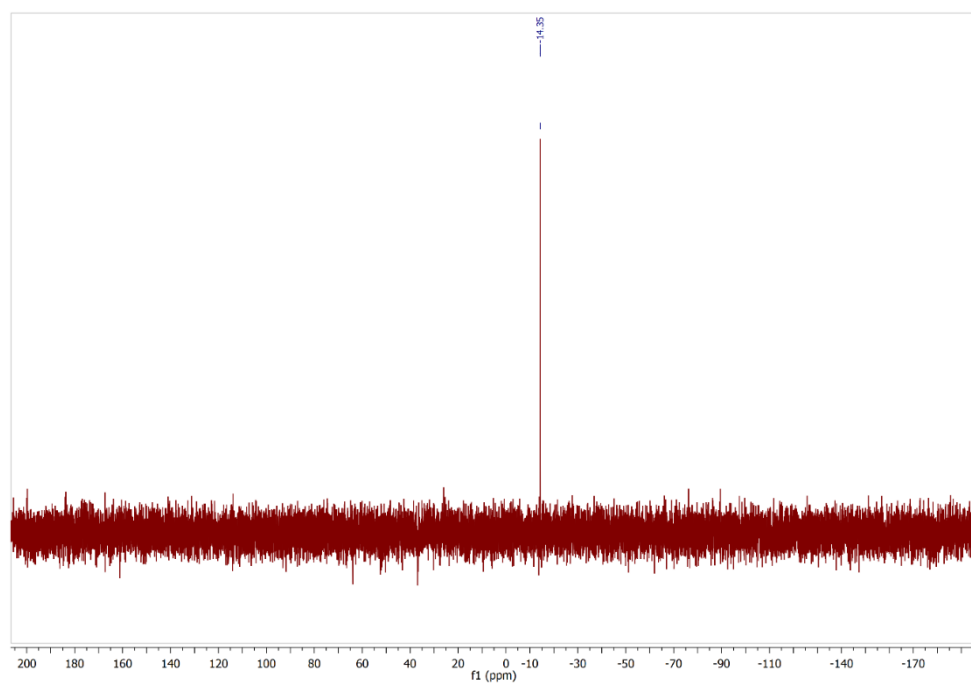


Figure S9: ^{29}Si - ^1H -DEPT- NMR (79 MHz, C_6D_6 , 297K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ca}(\text{hmds})_2]$ **2**.

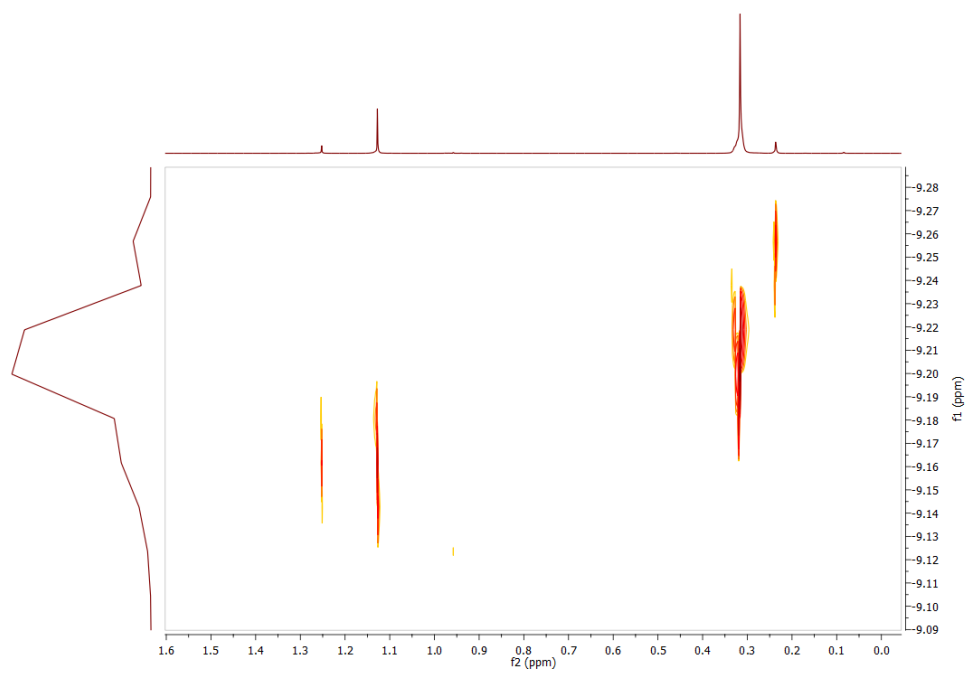


Figure S10: ^1H -DOSY-NMR (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ca}(\text{hmds})_2]$ **2**.

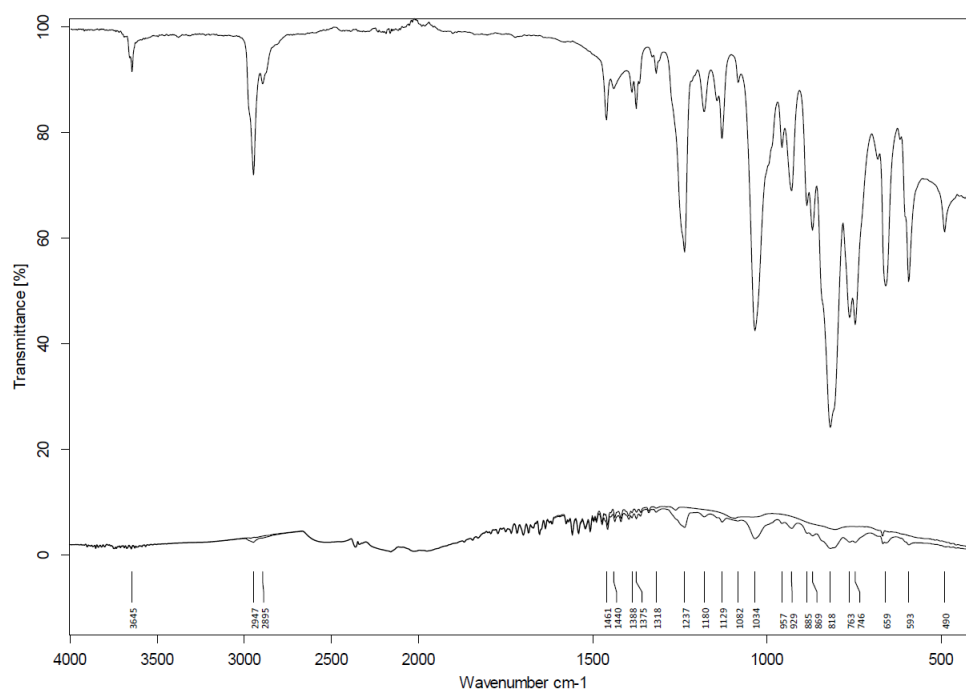


Figure S11: IR (ATR) of $[(\text{Me}_4\text{thf})\text{Ca}(\text{hmds})_2]$ **2**.

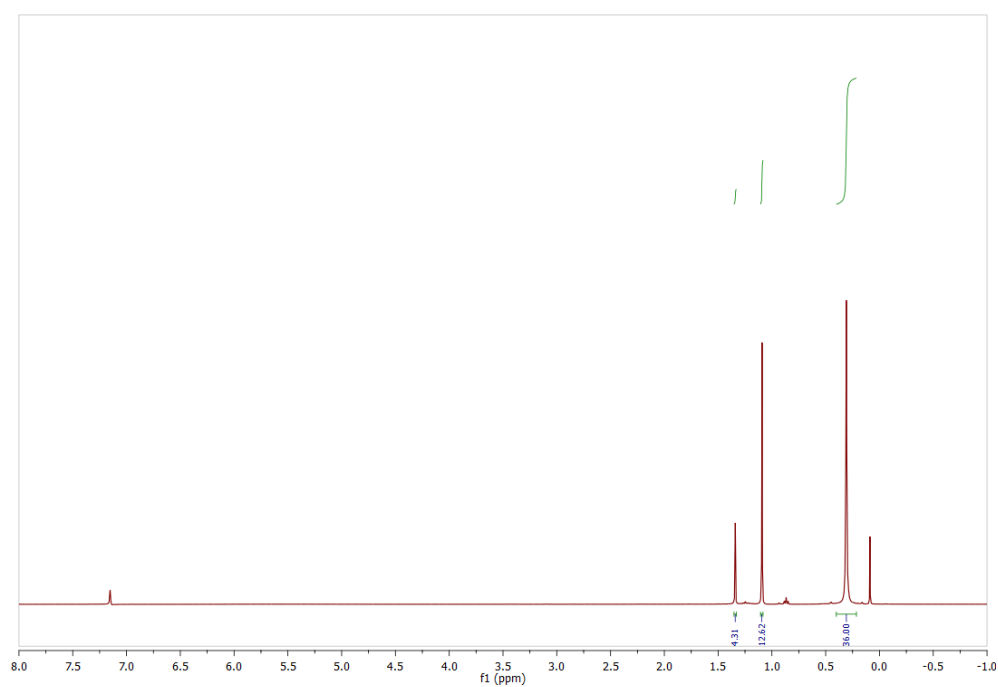
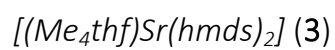


Figure S12: ^1H -NMR (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Sr}(\text{hmds})_2] \text{ (3)}$.

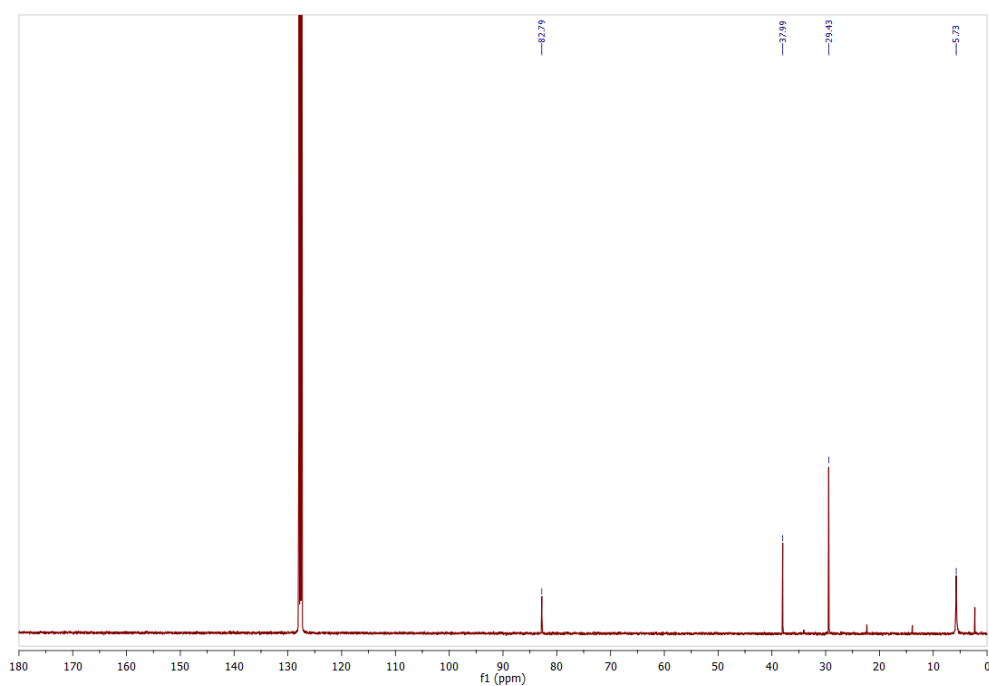


Figure S13: $^{13}\text{C}\{^1\text{H}\}$ -NMR (101 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Sr}(\text{hmds})_2] \text{ (3)}$.

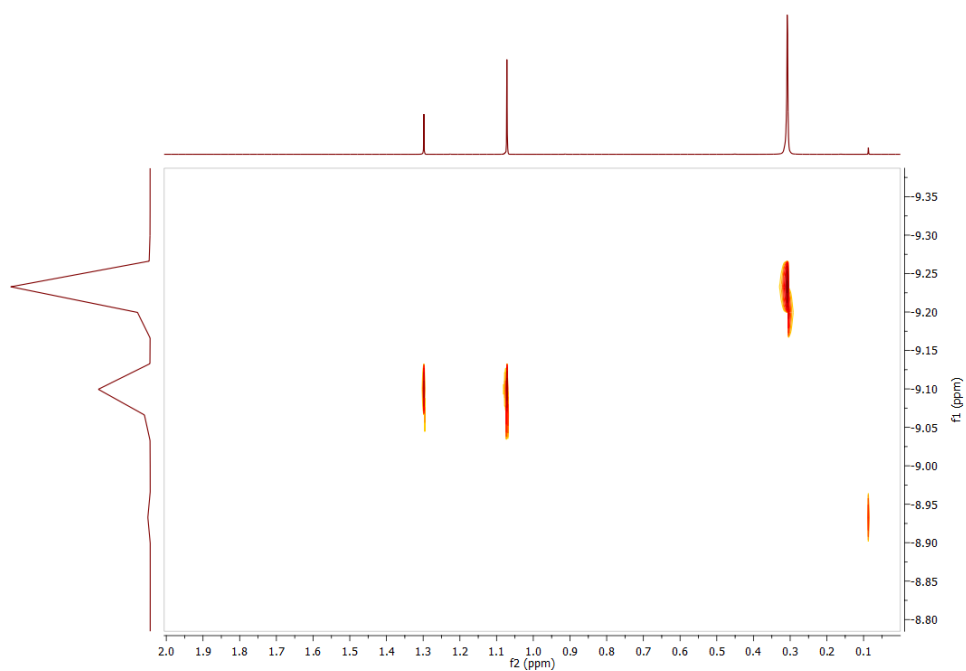


Figure S14: ^1H -DOSY-NMR (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Sr}(\text{hmds})_2] \mathbf{3}$.

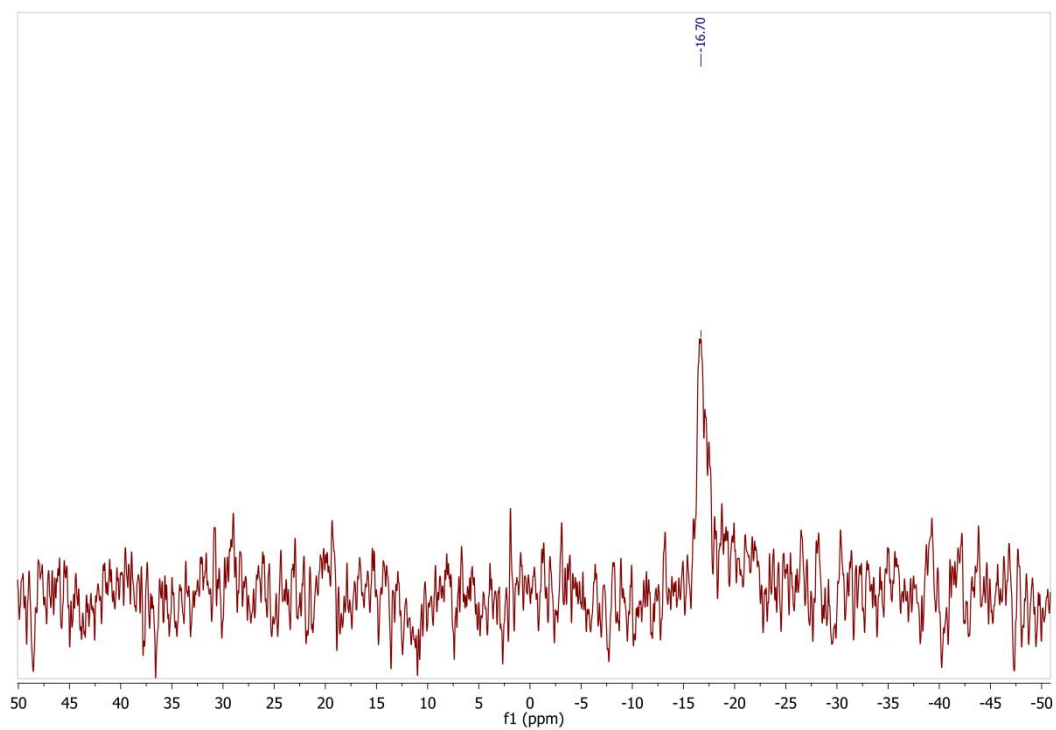


Figure S15: $^{29}\text{Si}\{^1\text{H}\}$ -NMR (79.5 MHz, C_6D_6 , 296 K) of $[(\text{Me}_4\text{thf})\text{Sr}(\text{hmds})_2]$.

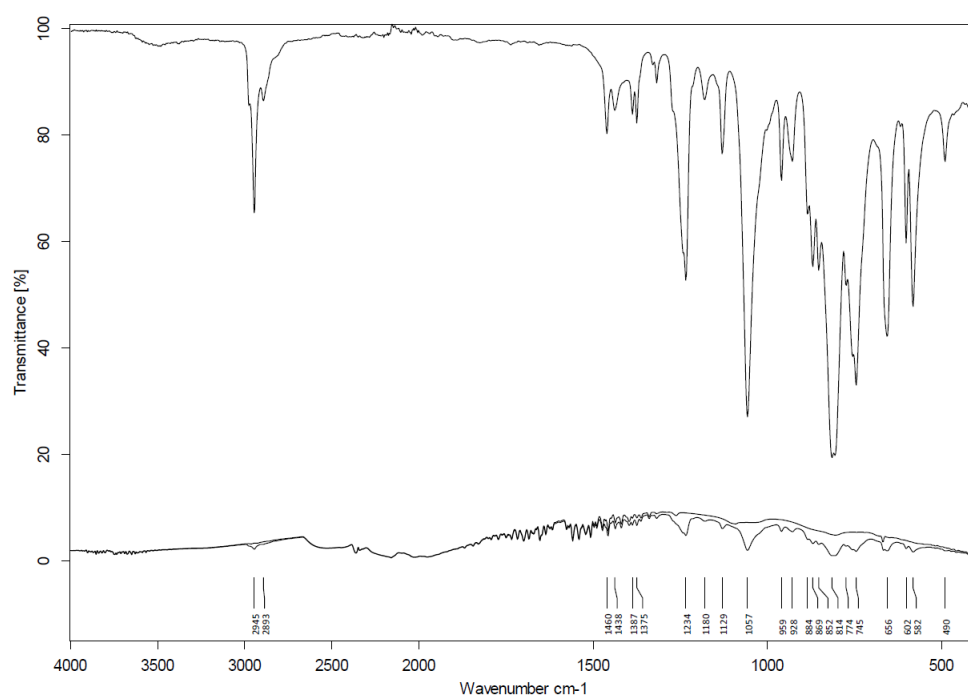


Figure S15: IR (ATR) of $[(\text{Me}_4\text{thf})\text{Sr}(\text{hmds})_2] \mathbf{3}$.

$[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ (**4**)

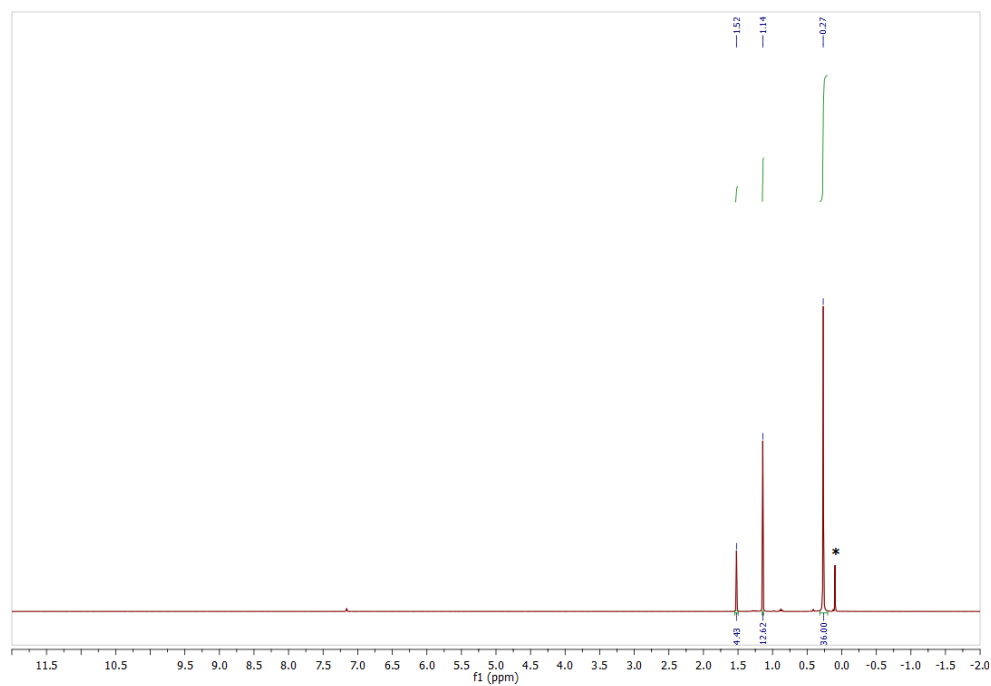


Figure S16: ^1H -NMR (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ **4** (*-H(hmds)).

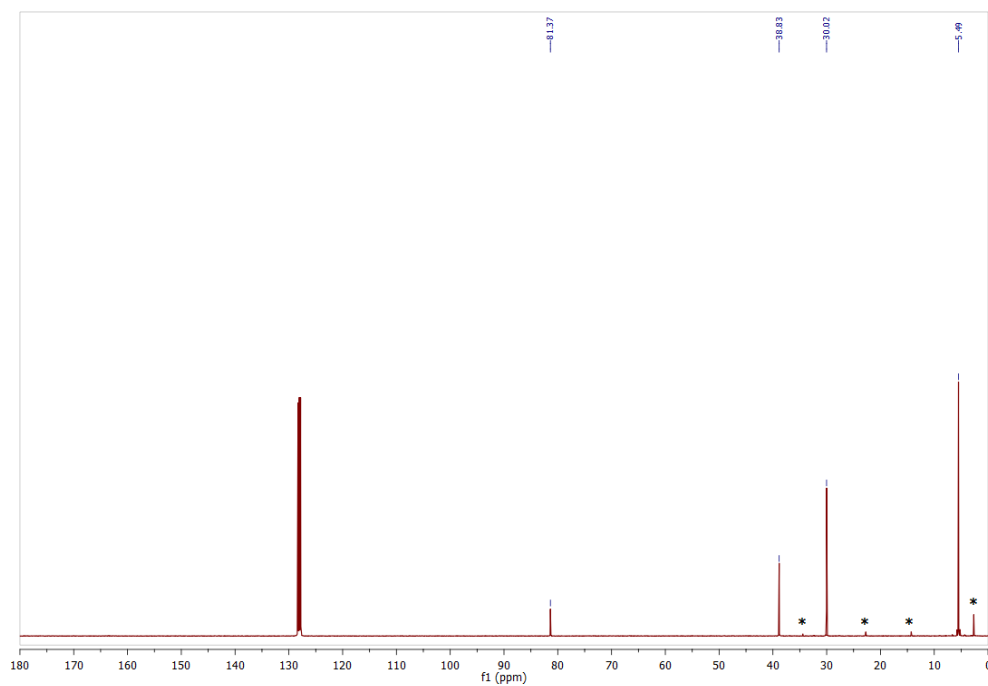


Figure S17: $^{13}\text{C}\{^1\text{H}\}$ -NMR (101 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ **4** (*-*n*-hexane, H(hmds)).

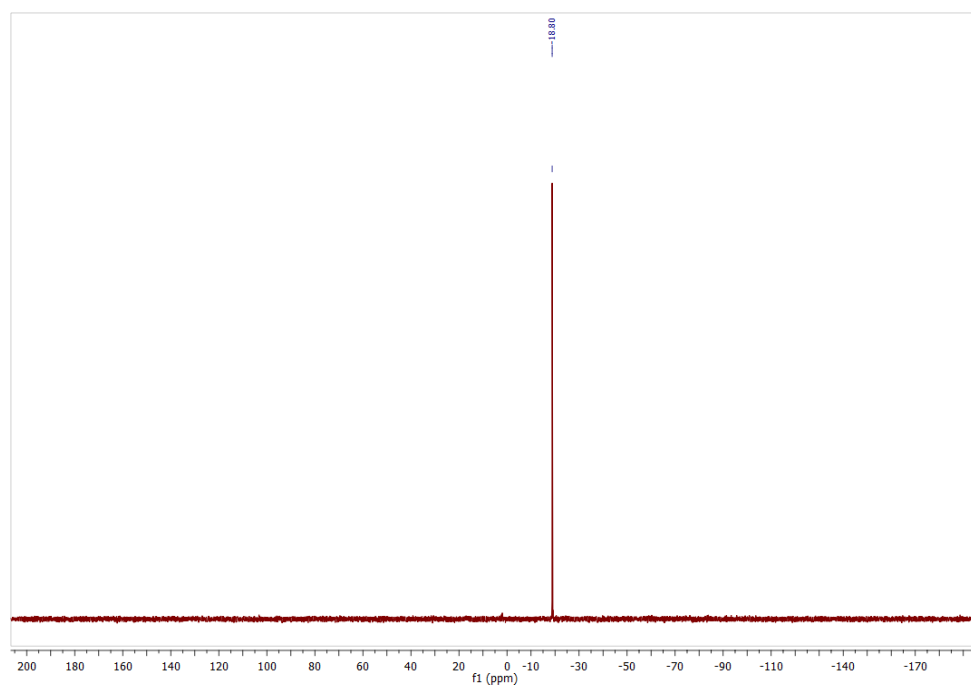


Figure S18: ^{29}Si - ^1H -DEPT (79,5 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ **4**.

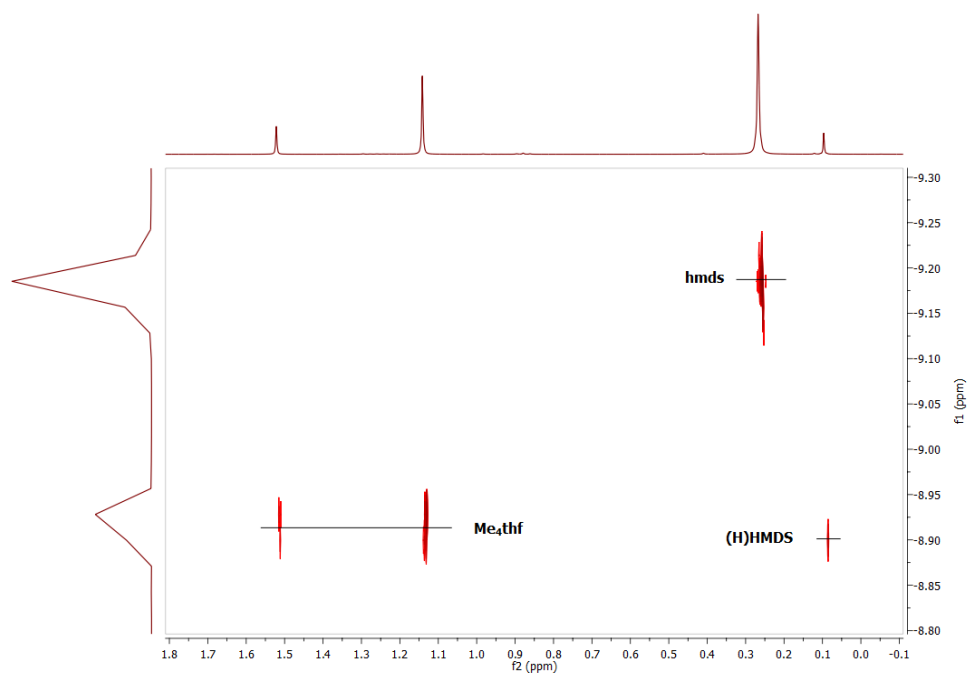


Figure S20: ^1H -DOSY (400 MHz, C_6D_6 , 297 K) of dissolved crystals of $[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ **4**.

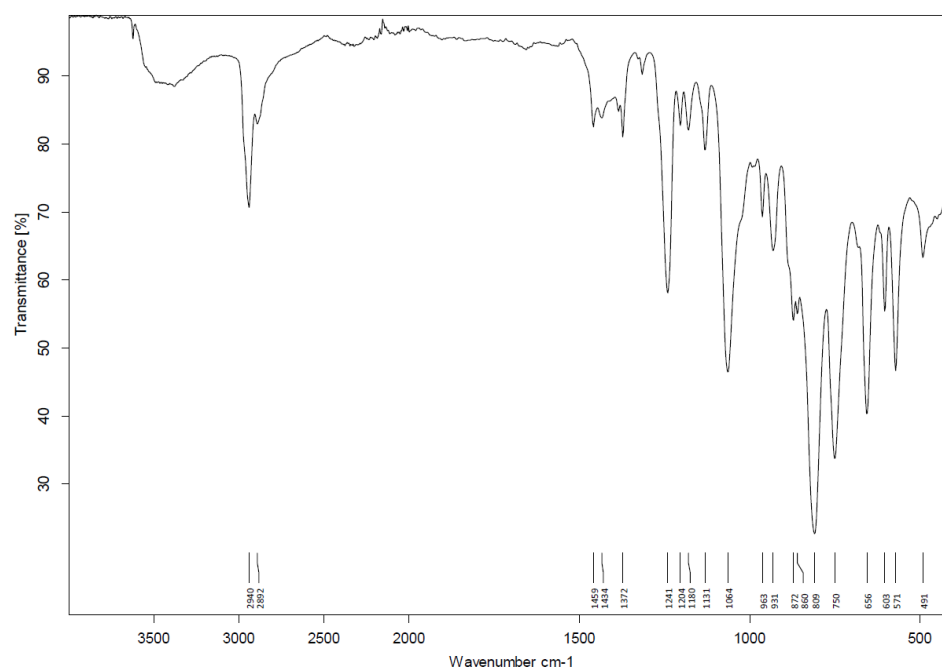


Figure S19: IR (ATR) of $[(\text{Me}_4\text{thf})\text{Ba}(\text{hmds})_2]$ **4**.

Table S1: Crystal data and refinement details for the X-ray structure determinations of the compounds **1** - **4**.

Compound	1	2	3	4
formula	C ₂₀ H ₅₂ MgN ₂ OSi ₄	C ₂₀ H ₅₂ CaN ₂ OSi ₄	C ₂₀ H ₅₂ N ₂ OSi ₄ Sr	C ₂₀ H ₅₂ BaN ₂ OSi ₄
fw (g·mol ⁻¹)	473.31	489.08	536.62	586.34
°C	-140(2)	-140(2)	-140(2)	-140(2)
crystal system	monoclinic	orthorhombic	orthorhombic	triclinic
space group	C 2	C 2 2 2 ₁	C 2 2 2 ₁	P $\bar{1}$
a/ Å	18.8311(3)	8.2638(1)	8.2667(17)	10.9264(2)
b/ Å	8.4579(1)	19.4277(4)	19.904(4)	11.4118(2)
c/ Å	18.8139(4)	18.9845(4)	18.906(4)	14.4972(2)
α /°	90	90	90	92.695(1)
β /°	92.4940(10)	90	90	93.785(1)
γ /°	90	90	90	115.177(1)
V/Å ³	2993.68(9)	3047.90(10)	3110.8(11)	1626.70(5)
Z	2	4	4	2
ρ (g·cm ⁻³)	1.050	1.066	1.146	1.197
μ (cm ⁻¹)	2.32	3.76	19	13.8
measured data	11820	9510	10782	10842
data with $I > 2\sigma(I)$	6573	3378	3390	7210
unique data (R_{int})	6661/0.0248	3504/0.0243	3492/0.0245	7421/0.0129
wR_2 (all data, on F^2) ^{a)}	0.1205	0.0595	0.0466	0.0450
R_1 ($I > 2\sigma(I)$) ^{a)}	0.0475	0.0251	0.0214	0.0192
s ^{b)}	1.220	1.047	1.053	1.086
Res. dens./e·Å ⁻³	0.583/-0.308	0.226/-0.149	0.195/-0.331	0.356/-0.302
Flack-parameter	0.08(14)	0.03(3)	0.000(4)	-
absorpt method	multi-scan	multi-scan	multi-scan	multi-scan
absorpt corr $T_{\text{min}}/T_{\text{max}}$	0.6624/0.7456	0.7123/0.7456	0.6456/0.7456	0.6913/0.7456
CCDC No.	1864053	1864054	1864055	1864056

^{a)} Definition of the R indices: $R_1 = (\sum ||F_o| - |F_c||) / \sum |F_o|$;

$wR_2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)]\}^{1/2}$ with $w^{-1} = \sigma^2(F_o^2) + (aP)^2 + bP$; $P = [2F_c^2 + \text{Max}(F_o^2)]/3$;

^{b)} $s = \{\sum [w(F_o^2 - F_c^2)^2] / (N_o - N_p)\}^{1/2}$.

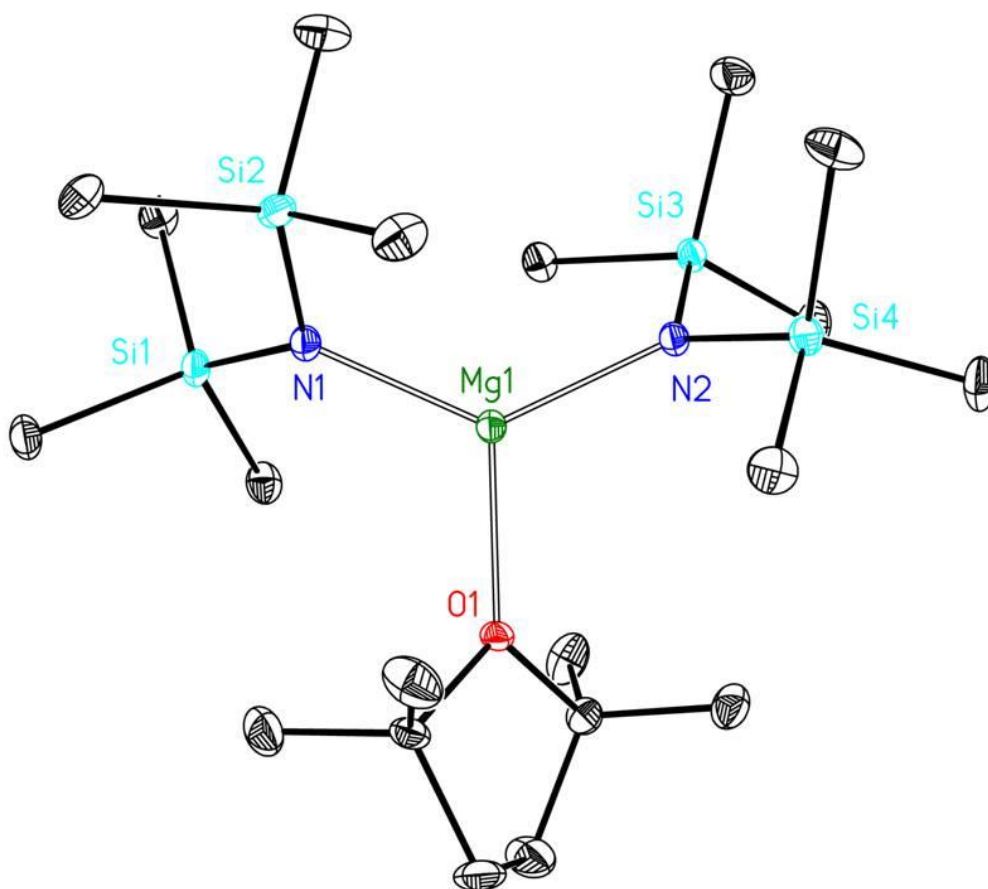


Figure S21: Molecular structure and numbering scheme of $[(\text{Me}_4\text{thf})\text{Mg}\{\text{N}(\text{SiMe}_3)_2\}_2]$ (**1**). The ellipsoids represent a probability of 30 %, hydrogen atoms are neglected for the sake of clarity. Selected bond lengths (pm): Mg1-N1 198.6(3), Mg1-N2 199.6(3), Mg1-O1 204.8(2), N1-Si1 171.4(3), N1-Si2 172.1(3), N2-Si3 170.6(3), N2-Si4 171.1(3), Mg1 $\cdots$ Si1 325.87(13), Mg1 $\cdots$ Si2 312.26(13), Mg1 $\cdots$ Si3 312.50(13); bond angles (deg.): N1-Mg1-N2 130.47(11), N1-Mg1-O1 115.54(10), N2-Mg1-O1 113.98(10), Mg1-N1-Si1 123.34(15), Mg1-N1-Si2 114.60(14), Si1-N1-Si2 120.97(15), Mg1-N2-Si3 114.97(14), Mg1-N2-Si4 123.95(15), Si3-N2-Si4 120.80(14).

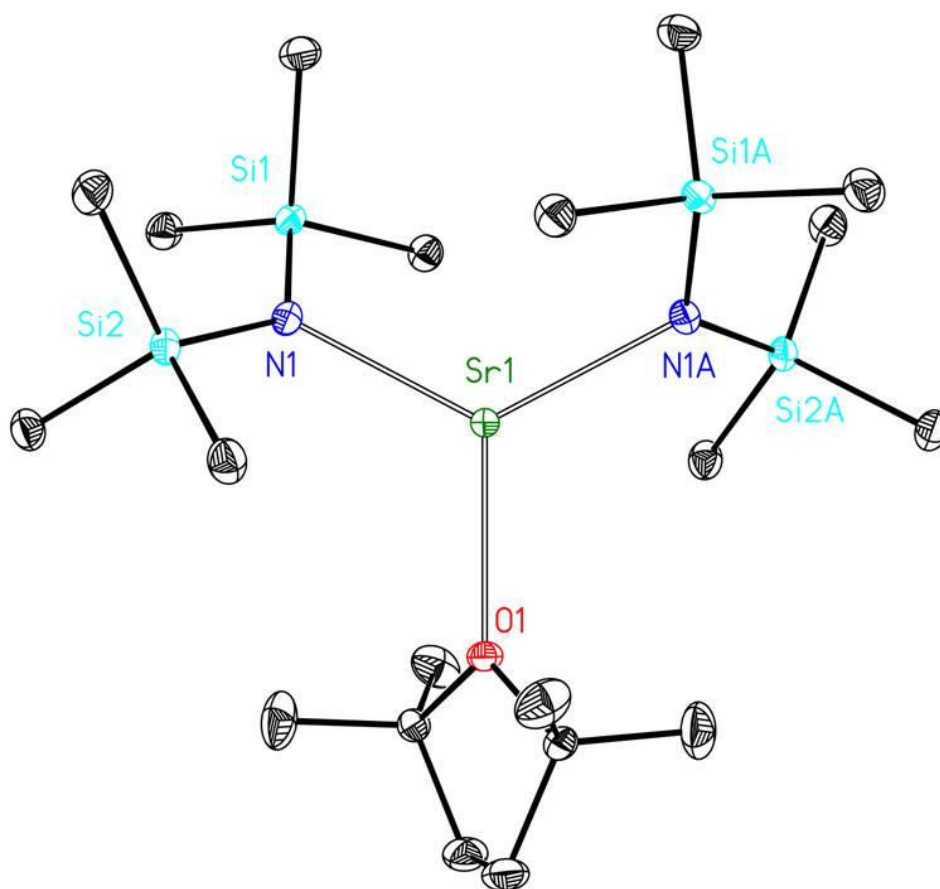


Figure S22: Molecular structure and numbering scheme of $[(\text{Me}_4\text{thf})\text{Sr}\{\text{N}(\text{SiMe}_3)_2\}_2]$ (**3**). The ellipsoids represent a probability of 30 %, hydrogen atoms are omitted for clarity reasons. Symmetry-related atoms $(-x+1, y, -z+1.5)$ are marked with the letter "A". Selected bond lengths (pm): Sr1-N1 240.90(13), Sr1-O1 249.61(14), N1-Si1 168.43(14), N1-Si2 168.97(14), Sr1 $\cdots$ Si1 344.33(6), Sr1 $\cdots$ Si2 353.23(7); bond angles (deg.): N1-Sr1-N1A 124.43(6), N1-Sr1-O1 117.79(3), Si1-N1-Si2 128.40(8), Sr1-N1-Si1 113.35(7), Sr1-N1-Si2 117.98(7).